

03.4-4 CARACURINE-II DIMETHOCHLORIDE OCTA-HYDRATE, A POTENT NEUROMUSCULAR BLOCKING AGENT. P. Bourne, S. Ginell, B.W. Low, & L. Lessinger, Columbia University, New York, N.Y., U.S.A.

$[C_{40}H_{44}N_4O_2]^{2+} \cdot 2Cl^- \cdot 8H_2O$, $P2_1$, $a=12.695(4)$, $b=7.424(2)$, $c=21.762(6)$ Å, $\beta=98.03(5)^\circ$, $Z = 2$. The structure could not be solved by the heavy atom Patterson method; it was solved by direct methods, and refined by least squares to $R=.10$.

The alkaloid cation, with two-fold molecular symmetry, has a highly fused ring system and is structurally rigid. This determination gives accurate stereochemical parameters for those atoms and groups (N^+ centers, aromatic rings, and hydrogen bond acceptors) postulated by various theories as involved in binding to the acetylcholine receptor.

In the crystal, layers of alkaloid cations parallel to the bc plane alternate with layers containing two chloride ions and eight water molecules distributed almost randomly over ten sites. Electrostatic attractions between N^+ and Cl^- bind the alternating layers together. Each of the ten sites is, on average, 4.60 Å from one or two N^+ , allowing the two Cl^- ions to be disordered. Binding interactions within the alkaloid layers are solely van der Waals attractions. Within each H_2O/Cl^- layer there is a complex hydrogen bond system, including four infinite spirals parallel to the b axis, with an average bonding distance of 2.94 Å. There are no hydrogen bonds between layers. The possible relevance to the activity of the alkaloid of its ability to organize large amounts of water is noted and discussed.

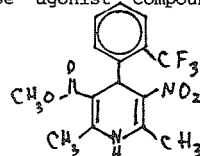
03.4-5 CONFORMATIONAL FEATURES OF CALCIUM CHANNEL AGONIST AND ANTAGONIST ANALOGS OF NIFEDIPINE. By D.A. Lings, Medical Foundation of Buffalo, Inc., Buffalo, NY 14203, and D. J. Trigg, Department of Biochemical Pharmacology, State University of New York at Buffalo, Buffalo, NY 14260, U.S.A.

Nifedipine analogs are 2,6-dimethyl-3,5-dicarboalkoxy-4(aryl-substituted)-1,4-dihydropyridine compounds which frequently exhibit important cardiovascular activity in that these drugs inhibit cardiac and smooth muscle contraction by blocking the flow of calcium ions through plasma membrane channels into the muscle cell. All such active nifedipine antagonists appear to act at a common dihydropyridine (DHP) plasma membrane binding site and correlations between pharmacologic and membrane binding activities establish that these binding sites are pharmacologically relevant. Quantitative structure-activity relationships have been derived for nifedipine analogs which correlate antagonist potency with large values of the minimum width steric Verloop parameter of the ortho- or meta- aryl substituent and lipophilic and steric factors of the ester groups (R. Rodenkirchen et al., Naunyn-Schied. Arch. Pharmacol. 310, 69-78 (1979)). Subsequent structural studies appeared to indicate a correlation between DHP ring flatness and activity for these ortho- and meta- substituted nifedipine antagonists (A.M. Trigg et al., J. Med. Chem. 23, 1442-1445 (1980); R. Fossheim et al., J. Med. Chem. 25, 126-131 (1982)).

Our present understanding of these physicochemical and conformational prerequisites for DHP receptor binding and calcium channel inhibition have been further complicated by the marked tissue selectivity shown by certain of the nifedipine analog antagonists which have dissymmetric ester groups. Such tissue selectivities are not shown by any of the symmetric ester analogs. In addition these dissymmetric ester analogs often show a chiral preference for receptor binding and calcium

channel inhibition which underscores the probable chiral nature of the putative endogenous hormone, which has yet to be discovered.

More recently several dissymmetric nifedipine analogs have been developed which surprisingly exhibit calcium channel agonism and stimulate cardiac and smooth muscle contraction (M. Schramm et al., Nature 303, 535-537 (1983); A. G. Truog, oral presentation at FASEB meeting, Chicago, April 1983). A diffraction study on the first of these agonist compounds, BAY K 8644, has



revealed that this compound has the flattest DHP ring of all the nifedipine analogs examined to date. Thus it appears that this conformational feature is not a characteristic of calcium channel antagonism, but rather a common feature which allows both agonists and antagonists to bind to the same DHP calcium channel receptor. Agonist or antagonist response must be encoded in other stereochemical and electronic characteristics which may be differentiated by the receptor. The crystal and molecular structure of BAY K 8644 suggests that the agonist behavior of this compound may in part be associated with a strong positive charge on the amine group brought about by a delocalization of electrons in the DHP ring as a consequence of the electron withdrawing effect of the 3-nitro substituent. Crystal data : BAY K 8644, $C_{16}H_{15}O_4N_2F_3$, $M_r = 356.3$,

monoclinic, $P2_1/c$, $a = 10.769(2)$, $b = 12.762(2)$, $c = 12.603(2)$ Å, $\beta = 108.61(2)^\circ$, $V = 1641$ Å³, $Z = 4$, $D_x = 1.44$ gm cm⁻³, $R = .064$ for 4059 data with $F > 2\sigma_F$. Research supported in part by Grant No. HL32303 from the National Heart, Lung, and Blood Institute.

03.4-6 SELECTIVITY AT THE μ OPIATE RECEPTOR: THE STRUCTURES OF α - AND β -FUNALTREXAMINE. Jane F. Griffin, Medical Foundation of Buffalo, Buffalo, NY 14203 and P. S. Portoghese, University of Minnesota, Minneapolis, Minn. 55455.

α - and β -Funaltrexamine (α - and β -FNA) are naltrexone derivatives differing only in chirality at C-6. Both α - and β -FNA bind to the μ opiate receptor in guinea pig ileum and mouse vas deferens preparations, but only the β -epimer selectively alkylates this receptor in both preparations. For this reason, β FNA has been used to "knock-out" μ receptors and study the remaining δ sites in these preparations and δ and κ sites in brain homogenate preparations. Sayre et al. (J. Med. Chem., 26, 1229-1235 (1983)) proposed a two step recognition process at the μ site with only the β -epimer in the proper orientation for the second recognition step which results in alkylation. Recent ¹H NMR studies of 6 α - and 6 β -oxymorphamine showed that the conformation of ring C was dramatically influenced by the stereochemistry of the 6-amino group: the 6 β -epimer existed in a chair conformation and the 6 α -epimer in the twist-boat conformation.

We have now determined by X-ray diffraction studies the molecular structures of both α - and β -FNA. The two epimers have almost identical conformations in the fused ring moiety except for ring C: in α -FNA ring C is observed in a twist-boat conformation, and in β -FNA ring C is in a chair conformation. The ring conformations result in the fumarate chain on C-6 being equatorial to ring C in both compounds. The fumarate moieties are approximately orthogonal to one another in the two structures.