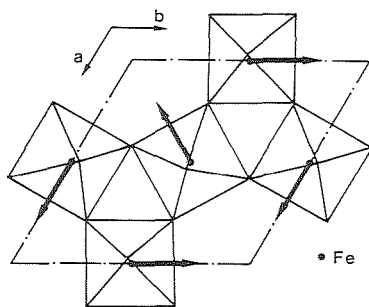


08.5-5 FRUSTRATED MAGNETIC STRUCTURE OF THE HEXAGONAL BRONZOID FeF_3 . By M. Leblanc, G. Ferey, J. Pannetier* and R. De Pape, ERA 609, Faculté des Sciences, 72017 Le Mans Cédex, France and *Institut Laue Langevin 38042 Grenoble Cédex, France.

A new variety of hydrated iron trifluoride $(\text{H}_2\text{O})_{0.33}\text{FeF}_3$ with HTB structure was recently grown by the hydrothermal method (Leblanc, Ferey, Chevallier, Calage, De Pape, J. Solid State Chem. (1983) 47, 53). It dehydrates at 122°C and leads to the HTB new form of FeF_3 . However, the small quantity of crystals obtained by this way lead us to imagine another synthesis yielding larger amounts of material for neutron powder diffraction experiments: the flash evaporation of a concentrated solution of FeF_3 in 49 % HF was followed by heating at 150°C under vacuum of the resulting powder. Then, chemical analysis, within the accuracy of the methods, are consistent with the FeF_3 formulation. Neutron diffraction patterns reveal the presence of $\alpha\text{-FeF}_3$ R3C (20 % molar) together with HTB- FeF_3 . So, their analysis was performed using a multipattern profile refinement program (Thomas, Bendall, Acta Cryst. (1978) A34, S351). At room-temperature the structure of HTB- FeF_3 ($a = 7.413(2)$ Å, $c = 3.795(1)$ Å) is related to that of ideal tungsten bronze with empty tunnels. At 4.2K, the magnetic order and the tilting of the octahedra imply the doubling of the c parameter ($a = 7.402(2)$ Å, $c = 7.569(1)$ Å).

The refined magnetic structure can be described with three antiferromagnetic sublattices. The spins of Fe^{3+} ($\mu = 4.07(8)\mu_B$) lie in the (a, b) plane at 120° from each other (figure). The magnetic interactions are antiferromagnetic between successive planes along c . This disposition of the spins is due to the 2D-frustration which occurs in the triangles of metallic ions.



08.5-6 MAGNETIC AND CRYSTAL STRUCTURES OF $\text{Nd}_4\text{MnFeF}_6$. By M. Leblanc, G. Ferey, J. Pannetier* and R. De Pape, ERA 609, Faculté des Sciences, 72017 Le Mans Cédex, France and *Institut Laue Langevin 38042 Grenoble Cédex, France.

Using hydrothermal conditions (380°C, 200 MPa), a new form of $\text{Nd}_4\text{MnFeF}_6$ is obtained (Leblanc, Ferey, Calage, De Pape, J. Solid State Chem. (1983) 47, 24). The 3D structure, built from MnFeF_{10} bioctahedra (Figure) is related to that of BaNb_2O_6 (Galasso, Layden, Ganung, Mater. Res. Bull. (1968) 3, 397). The compound is antiferromagnetic ($T_N = 117.7 \pm 0.5\text{K}$) with a parasitic superimposed ferromagnetism ($\sigma_r = 0.005 \mu_B \cdot \text{mole}^{-1}$ at 4.2K).

Neutron powder diffraction patterns were recorded at 130K and 4.2K (Pnc2, $Z = 4$, $a = 10.5280(4)$ Å, $b = 7.7973(3)$ Å, $c = 12.8158(5)$ Å, $\lambda = 1.909$ Å). The magnetic and nuclear cells are identical, so the Bertaut's theory was applied (Bertaut, Magnetism III, Rado and Shull Ed. (1963) 149). The metallic atoms form 4 magnetic independent sublattices.

The Rietveld-Hewatt profile refinement method (Rietveld, J. Appl. Cryst. (1969) 2, 65; Hewatt, Harwell Report AERE-R7350 (1973)) was used ($12^\circ < 2\theta < 115.5^\circ$, 452 hkl triplets) and deuterium atoms were localized from Fourier maps. As a consequence of the complexity of the structure (29 independent positions) simplifying hypothesis were applied: isotropic thermal motion identical for each type of atoms, pure antiferromagnetism with $\mu_{\text{Mn}^{2+}} = \mu_{\text{Fe}^{3+}}$. The best fit ($R_{\text{Nuc}} = 0.050$, $R_{\text{Mag}} = 0.040$, $R_{\text{Prof}} = 0.073$) between observed and calculated intensities was obtained when the magnetic moments ($\mu = 4.51(5)\mu_B$) lie along b with spins alternatively up and down (Figure). A slight canting apart this direction may occur but could not be refined ($C_x F_y G_z$ mode).

According to these results, all the magnetic interactions are antiferromagnetic, particularly inside the MnFeF_{10} unit where 90° superexchange occurs; the double correlation superexchange mechanism is excluded. The previously described cationic order between Mn^{2+} and Fe^{3+} inside the MnFeF_{10} bioctahedra is clearly confirmed.

