

## Poster Presentation

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### Synthesis and crystal structure of $\text{Sr}_2\text{MnGaO}_{5-d}\text{F}_x$ , layered oxyfluorides

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$\text{Sr}_2\text{MnGaO}_5$  brownmillerite has an interesting structure. In this compound,  $\text{MnO}_6$  octahedra and  $\text{GaO}_4$  tetrahedra are ordered along c-axis. The  $\text{MnO}_6$  octahedra are deformed due to Jahn-Teller effect with four short equatorial Mn-O distances and two long apical one. Oxygen vacancies in the GaO layer provide distorted tetrahedral coordination for the Ga cation. The oxide and/or fluoride ions are introduced into the vacancies in GaO layer. The synthesis of  $\text{Sr}_2\text{MnGaO}_{4.78}\text{F}_{1.22}$  using  $\text{XeF}_2$  as a fluorinating agent is reported, while the small amount  $\text{SrF}_2$  impurity is contained[1]. On the other hand, low temperature fluorination using polyvinylidene fluoride (PVDF) is powerful method for the synthesis of oxyfluorides from brownmillerite[2]. In this study, we synthesize the  $\text{Sr}_2\text{MnGaO}_{5-d}\text{F}_x$  ( $x=0.5, 1.0, 1.5, 2.0$ ) using PVDF and investigate the variation of crystal structure with  $x$ .  $\text{Sr}_2\text{MnGaO}_5$  brownmillerites were prepared solid state reaction method. Starting materials were  $\text{SrCO}_3$ ,  $\text{Mn}_2\text{O}_3$  and  $\text{Ga}_2\text{O}_3$ . The mixture was heated twice in Ar at  $1300^\circ\text{C}$  for 72h. For the preparation of  $\text{Sr}_2\text{MnGaO}_{5-d}\text{F}_x$  ( $x=0.5, 1.0, 1.5, 2.0$ ), stoichiometric amounts of  $\text{Sr}_2\text{MnGaO}_5$  brownmillerite and PVDF were ground and the mixture were heated in  $\text{N}_2$  at  $400^\circ\text{C}$  for 8h. And then, the as prepared materials were subsequently heated in  $\text{O}_2$  at  $400^\circ\text{C}$  for 8h. The  $\text{Sr}_2\text{MnGaO}_{5-d}\text{F}_x$  were obtained without impurities. As shown in Fig., the orthorhombic distortion reduced with the increase in  $x$ . The X-ray diffraction data of  $\text{Sr}_2\text{MnGaO}_{5-d}\text{F}_{1.0}$  could be refined on the structure model belonging the space group  $\text{Icmm}$ . the bond valence sum of  $\text{Ga}^{3+}$  calculated from Ga-O bond length indicates that the fluoride ions are introduced into GaO layer preferentially. The measurements of magnetic properties for  $\text{Sr}_2\text{MnGaO}_{5-d}\text{F}_x$  are in progress.

[1] A. M. Alekseeva et al., *J. Solid State Chem.*, 177 (2004) 731., [2] F. J. Berry et al., *J. Solid State Chem.*, 184 (2011) 1361.

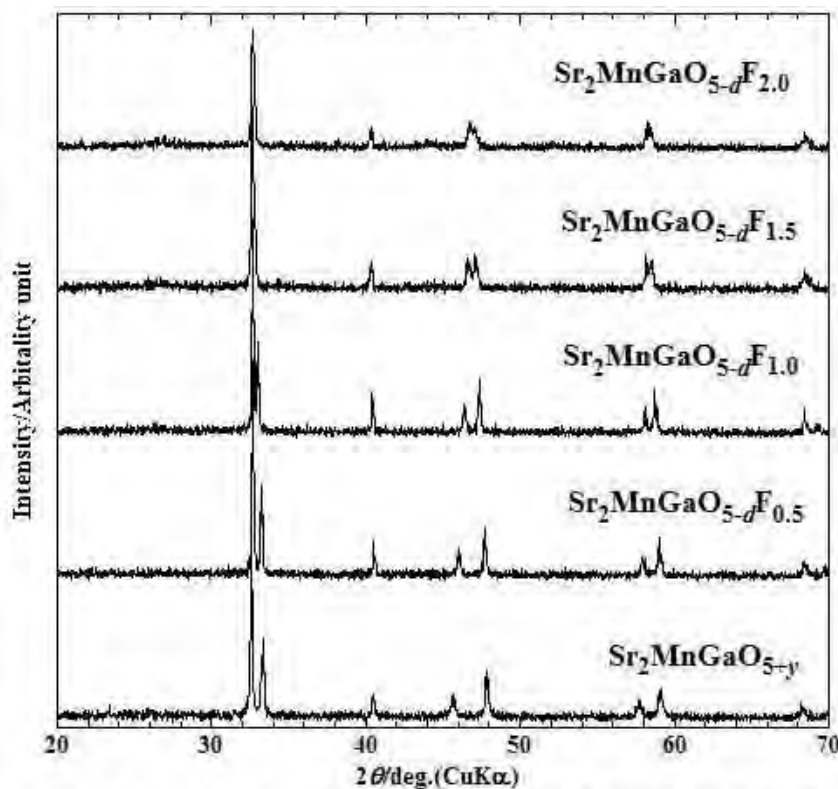


Fig.1 X-ray diffraction patterns for  $\text{Sr}_2\text{MnGaO}_{5-d}\text{F}_x$  ( $x=0.5, 1.0, 1.5, 2.0$ )

**Keywords:** crystal structure, oxyfluorides, bond valence sum