

## Self-assembled molecular squares as supramolecular tectons

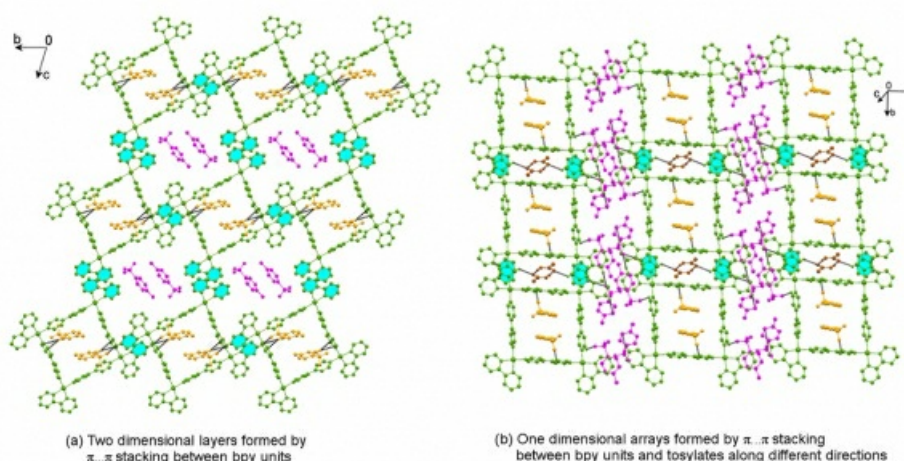
Shobhana Krishnaswamy<sup>1</sup>, Soumyakanta Prusty<sup>1</sup>, Daniel Chartrand<sup>2</sup>, Garry Hanan<sup>2</sup>, Dillip Kumar Chand<sup>1</sup><sup>1</sup>Dept. Of Chemistry, Indian Institute Of Technology Madras, Chennai, India, <sup>2</sup>Dept. of Chemistry, University of Montreal, Montreal, Canada

E-mail: cy14ipf04@smail.iitm.ac.in

A concentration dependent equilibrating mixture of molecular squares  $[\text{Pd}4(\text{L}')4(\text{L})4](\text{NO}_3)_8$  and triangles  $[\text{Pd}3(\text{L}')3(\text{L})3](\text{NO}_3)_6$  was obtained when cis-protected Pd(II) units  $[\text{Pd}(\text{L}')(\text{NO}_3)_2]$  ( $\text{L}' = \text{tmeda}$ , [1] 2,2'-bpy, [2] phen) were combined in turn with 4,4'-bipyridine (L) in water. The addition of AgOTs to the mixture led to a shift in the equilibrium, resulting in the disappearance of the triangles and exclusive formation of the squares in case of all the complexes. The crystal structures of the molecular squares  $[\text{Pd}4(\text{L}')4(\text{L})4](\text{OTs})_8$  revealed a pair of tosylate anions encapsulated in the hydrophobic cavity of the square and the presence of several water molecules outside the cavity. The complexes  $[\text{Pd}4(\text{bpy})4(\text{L})4](\text{OTs})_8$  and  $[\text{Pd}4(\text{phen})4(\text{L})4](\text{OTs})_8$  exhibited solvatomorphism and yielded two crystalline forms each, respectively. The cationic units in these crystals associate through  $\pi \dots \pi$  stacking interactions between the aromatic rings of the four bpy/phen units and form either one-dimensional arrays or two-dimensional layers. The formation of a one-dimensional array occurs when one pair of bpy/phen units of the square engage in  $\pi \dots \pi$  stacking interactions with the tosylate anions instead of other adjacent bpy/phen units. Therefore, the cations in the bpy and phen squares may be considered as 'tectons' which contain four supramolecular 'synthons' apiece, i.e. the cis-protecting units bpy/phen. The knowledge of common patterns of molecular association and identification of supramolecular synthons in these structures can help in the crystal engineering of coordination compounds with desired solid-state properties/functions.

[1] Uehara, K. et al. (2007). Inorg. Chem., 46, 2563-2570.

[2] Fujita, M. et al. (1996). Chem. Commun., 1535-1536.



**Keywords:** [crystal engineering](#), [Pd\(II\) molecular square](#), [solvatomorphism](#)