

Exploring liquid-liquid' transitions in ZnSe at extreme conditions

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Here, we report on the fingerprints of 4-fold to 6-fold transition in liquid ZnSe at extreme pressure and temperature conditions up to 42 GPa and more than 3000 K using X-ray absorption spectroscopy (XAS) combined with laser-heated diamond anvil cell techniques and complementary ab initio molecular dynamics simulations. Among pressure induced structural modifications, liquid-liquid' (L-L') transitions have attracted much interest both experimentally and theoretically due to their peculiarity. Unlike most L-L' transitions that only show slight modifications of the bond distances between atoms (i.e., P, Na, supercooled Si) [1, 2, 3], strong L-L' transitions have been theoretically predicted for a few II-VI semiconductor compounds such as ZnSe, CdSe and CdTe, given the following two conditions are satisfied: Firstly, the melting of the sp³ bonded phase (4-fold coordination) at ambient pressure results in a liquid structure that remains approximately 4-fold coordinated; and secondly, the presence of a semiconductor to metal phase transition under pressure [4, 5, 6, 7]. Our results show that solid and liquid ZnSe undergoes a series of structural modifications at various P, T values that satisfy the conditions above. The red shift in Zn K edge energy observed at around 7 GPa upon increasing temperature suggests a metallization event before melting to a possible 6-fold coordinated liquid structure. These findings are supported by our simulation results showing a pronounced difference in the first diffraction peak of the calculated structure factor at high pressures indicating a 4-fold to 6-fold coordination change in liquid ZnSe. Our results may provide additional insight for such transitions that may be observed for similar tetrahedrally coordinated II-VI systems.

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