

## MS22-01 | 'PANCAKE' BONDING - A CHARGE DENSITY PERSPECTIVE

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The 1,2,3,5-dithiadiazolyls (R-CNSSN<sup>\*</sup>, hereafter DTDA), a family of thiazyl radicals, have been the focus of much investigation due to their potential as building blocks for magnetic and conducting materials. [1] However, these molecules tend to dimerise in the solid state *via* a spin-pairing interaction known as 'pancake bonding', a two-electron multi-centre interaction between  $\pi$ -radicals [2]. This dimerization renders the resulting materials diamagnetic. Much effort has been devoted to overcoming this interaction to produce magnetic materials.

In order to gain a deeper understanding of pancake bonding, experimental charge density analysis has been carried out on a number of DTDA homodimers, heterodimers and monomers. [3] These data, as well as various computational results, are assessed to probe the nature of the pancake bonds in DTDA.

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[3] S. Domagała, K. Kość, S. W. Robinson, D. A. Haynes and K. Wozniak, *Cryst. Growth Des.*, 2014, **14**, 4834-4848; S. Domagała and D. A. Haynes, *CrystEngComm*, 2016, **18**, 7116-7125; A. B. Voufack, N. Claiser, A. B. Dippenaar, C. Esterhuysen, D. A. Haynes, C. Lecomte and M. Souhassou' *manuscript in preparation*.