

## MS40 Operando and in situ crystallographic studies

### MS40-01

Operando X-ray diffraction studies of NASICON-type positive electrodes for Na-ion batteries

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### Abstract

Polyanionic materials (phosphates in particular) are of special interest as positive electrodes for Li-Ion or Na-ion batteries since they offer competitive electro-chemical performances compared to sodiated or lithiated transition metal oxides [1,2]. They are based upon stable 3D frameworks, which provide long-term structural stability and demonstrate a unique variety of atomic arrangements in their crystal structures. Recent electrochemical and structural investigations of vanadium-based phosphate compounds ( $\text{LiVPO}_4\text{O-LiVPO}_4\text{F}$ ,  $\text{Na}_3\text{V}_2(\text{PO}_4)_2\text{F}_3$ ,  $\text{Na}_3\text{V}_2(\text{PO}_4)_3$  ....) revealed promising perspectives [3-5].

The NASICON structural family with its large panel of compositions,  $\text{Na}_x\text{MM}'(\text{PO}_4)_3$  ( $0 \leq x \leq 4$ ; M,M' = Ti, Fe, V, Cr, Mn) is among the most widely investigated due to its specific three-dimensional framework structure, stable long-term cycling ability and high  $\text{Na}^+$  mobility [1-2, 5-6]. Among them, the vanadium phosphate  $\text{Na}_3\text{V}_2(\text{PO}_4)_3$  [7] is of particular interest. We will present several new structures that we determined, from pristine powders or for intermediate compositions spotted by operando X-Ray diffraction.

Recently, we succeeded in synthesizing Fe-substituted  $\text{Na}_4\text{FeV}(\text{PO}_4)_3$  that allows the reversible extraction of close to 3  $\text{Na}^+$  (for two transition metals) and we will report on its crystal structure and on that of  $\text{Na}_3\text{FeV}(\text{PO}_4)_3$  for which new  $\text{Na}^+$  order-disorder phenomena have been spotted [8, 9]. Even more recently, we reported on the existence of an intriguing definite phase of composition  $\text{Na}_2\text{V}_2(\text{PO}_4)_3$  through computational methods [10] and operando X-ray diffraction and X-ray absorption spectroscopy during battery operation [11].

### References

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