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Retraction of articles

This article reports the retraction of articles published in *Acta Crystallographica Section E* between 2005 and 2009.

After further thorough investigation (see Harrison *et al.*, 2010), articles are retracted as a result of problems with the data sets or incorrect atom assignments. Full details of all the articles are given in Table 1.

Table 1
Details of articles to be retracted, in order of publication.

Title	Reference	DOI	Refcode
Poly[diaquadi-μ_3-malonato-μ-pyrazine-dinickel(II)]	Liu *et al.* (2005)	10.1107/S1600536805026358	GATWAA
catena-Poly[[[diaqua(6-carboxypyridine-2-carboxylato)samarium(II)]-μ-pyridine-2,6-dicarboxylato] tetrahydrate]	Liu *et al.* (2006)	10.1107/S1600536806038141	FONCUH03
Poly[[[μ_4-4,4′-carbonylbis(benzene-3,4-dicarboxylato)]tetrakis(1,10-phenanthroline)-dipalladium(II)] dihydrate]	Li, Wang, Zhang & Yu (2007*e*)	10.1107/S1600536807039050	AFELAZ
Poly[diaqua-μ_3-malonato-μ-pyrazine-diiron(II)]	Li, Liu *et al.* (2007)	10.1107/S1600536807038743	AFELON
Poly[diaqua-di-μ_3-malonato-μ-pyrazine-dimanganese(II)]	Li, Wang, Zhang & Yu (2007*f*)	10.1107/S1600536807039773	VIJZAQ
Poly[[aqua(2,2-bipyridine)(μ_3-pyridine-3,4-dicarboxylato)cobalt(II)] monohydrate]	Li, Wang, Zhang & Yu (2007*g*)	10.1107/S1600536807040275	VIKCIC
catena-Poly[[[diaqua(6-carboxypyridine-2-carboxylato)holmium(III)]-μ-pyridine-2,6-dicarboxylato] tetrahydrate]	Li, Wang, Zhang & Yu (2007*a*)	10.1107/S1600536807041657	DILGEL
catena-Poly[[(2,2′-bipyridine-κ^2N,N')iron(II)]-μ-5-carboxy-4-carboxylatoimidazol-1-ido-$\kappa^4N^3,O^4{:}N^1,O^5$]	Li, Wang, Zhang & Yu (2007*h*)	10.1107/S1600536807042122	XIKWAQ
Poly[[aqua(2,2′-bipyridine)(μ_3-pyridine-3,4-dicarboxylato)nickel(II)] monohydrate]	Li, Wang, Zhang & Yu (2007*b*)	10.1107/S1600536807046466	LEVZAO01
2-(Benzyliminomethyl)-6-methoxyphenol	Li, Wang, Zhang & Yu (2007*i*)	10.1107/S1600536807042134	SILDEX
Poly[aqua(2,2′-bipyridine)(μ_3-pyridine-2,4-dicarboxylato)palladium(II)]	Li, Wang, Zhang & Yu (2007*c*)	10.1107/S1600536807047575	SILXAN
μ-Oxido-bis{chlorido[tris(2-pyridylmethyl)amine]iron(III)} bis(hexafluoridophosphate)	Liu, Dou, Li & Zhang (2007)	10.1107/S1600536807049665	TINRIS
μ-Oxido-bis({4,4′-dibromo-2,2′-[ethane-1,2-diylbis(nitrilomethylidyne)]diphenolato}-manganese(III))	Liu, Dou, Niu & Zhang (2007*a*)	10.1107/S1600536807051008	GIMZAE
Bis[N-(8-quinolyl)pyridine-2-carboxamidato]iron(III) perchlorate monohydrate	Li, Wang, Zhang & Yu (2007*d*)	10.1107/S1600536807048556	WIMZIC
μ-Oxido-bis({4,4′-dibromo-2,2′-[ethane-1,2-diylbis(nitrilomethylidyne)]diphenolato}-chromium(III))	Liu, Dou, Niu & Zhang (2007*b*)	10.1107/S1600536807057996	HIQFIX
μ-Oxido-bis{chlorido[tris(2-pyridylmethyl)amine]chromium(III)} bis(hexafluorido-phosphate)	Li, Wang *et al.* (2008)	10.1107/S1600536807061296	MIRNAD
μ-Oxido-bis({4,4′-dibromo-2,2′-ethane-1,2-diylbis(nitrilomethylidyne)]diphenolato}-iron(III))	Meng *et al.* (2008*a*)	10.1107/S1600536807063143	MIRWUG
catena-Poly[[bis(1H-benzimidazole-κN^3)palladium(II)]-μ-benzene-1,4-dicarboxylato-$\kappa^2O^1{:}O^4$]	Meng *et al.* (2008*b*)	10.1107/S1600536807065051	XISCAE
Oxalatobis(propane-1,3-diamine)manganese(II) chloride monohydrate	Meng *et al.* (2008*e*)	10.1107/S1600536807065361	SISWIB
μ-Oxido-bis{chlorido[tris(2-pyridylmethyl)amine]manganese(III)} bis(hexafluorido-phosphate)	Meng *et al.* (2008*c*)	10.1107/S1600536807066512	RISRIV
Bis[N-(8-quinolyl)pyridine-2-carboxamidato-κ^3N,N',N'']manganese(III) perchlorate monohydrate	Meng *et al.* (2008*d*)	10.1107/S1600536808000287	GISLEA
Diaquabis(pyridine-2-carboxylato-κ^2N,O)cobalt(II)	Huang (2008)	10.1107/S1600536808010507	WIZPOL
Tetra-μ-2,5-difluorobenzoato-bis[(2,2′-bipyridine)(2,5-difluorobenzoato)gadolinium(III)]	Li, Zhang *et al.* (2008)	10.1107/S1600536808023507	BOFQIX
catena-Poly[[(2,2′-bipyridine-κ^2N,N')nickel(II)]-μ-oxalato-$\kappa^4O^1,O^2{:}O^{1'},O^{2'}$]	Li, Yan *et al.* (2008)	10.1107/S1600536808028389	NOHYUF
catena-Poly[[aqua(2,2′-bipyridyl)cobalt(II)]-μ-5-nitroisophthlalato]	Liu *et al.* (2008)	10.1107/S1600536808038178	AFIREN
Diaquabis(pyridine-2-carboxylato-κ^2N,O)iron(II)	Xia & Sun (2009)	10.1107/S1600536809005765	RONFEG
catena-Poly[[[diaquathulium(III)]-μ-6-carboxynicotinato-μ-pyridine-2,5-dicarboxylato] dihydrate]	Li *et al.* (2009)	10.1107/S1600536809008836	NOQNIR
1-Phenyl-3-(2,4,6-trimethoxyphenyl)prop-2-en-1-one	Liu *et al.* (2009)	10.1107/S1600536809040227	PUGLOT

  doi:10.1107/S1600536811037603

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catena-Poly[[bis(1*H*-benzimidazole-κN^3)palladium(II)]-μ-benzene-1,4-dicarboxylato-κ^2O^1:O^4]

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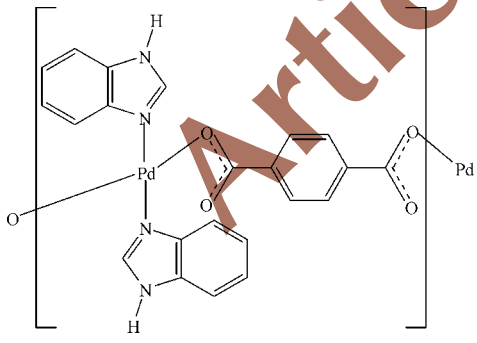
Received 28 November 2007; accepted 2 December 2007

Key indicators: single-crystal X-ray study; T = 273 K; mean σ(C–C) = 0.003 Å; R factor = 0.021; wR factor = 0.044; data-to-parameter ratio = 13.4.

In the title compound, $[Pd(C_8H_4O_4)(C_7H_6N_2)_2]_n$, the Pd atom is tetracoordinated by two carboxylate O atoms from two benzene-1,4-dicarboxylate (bdc) dianions and two N atoms from two benzimidazole ligands, resulting in a slightly distorted tetrahedral PdO_2N_2 geometry. The bdc ligand acts as a bridge, linking the Pd atoms into a chain. Inter-chain N—H···O hydrogen bonds help to stabilize the crystal structure.

Related literature

For background, see: Okabe & Oya (2000).



Experimental

Crystal data

$[Pd(C_8H_4O_4)(C_7H_6N_2)_2]$
M_r = 506.79
Monoclinic, $P2_1/n$
a = 17.0627 (5) Å
b = 7.3612 (10) Å
c = 18.0210 (5) Å
β = 114.362 (3)°
V = 2061.9 (3) Å^3
Z = 4
Mo $K\alpha$ radiation
μ = 0.94 mm^{-1}
T = 273 (2) K
0.43 × 0.28 × 0.22 mm

Data collection

Bruker APEXII CCD diffractometer
Absorption correction: multi-scan (*SADABS*; Bruker, 2001)
T_{min} = 0.689, T_{max} = 0.820
10244 measured reflections
3763 independent reflections
3136 reflections with $I > 2\sigma(I)$
R_{int} = 0.021

Refinement

$R[F^2 > 2\sigma(F^2)]$ = 0.021
$wR(F^2)$ = 0.044
S = 1.00
3763 reflections
280 parameters
H-atom parameters constrained
$\Delta\rho_{max}$ = 0.25 e Å^{-3}
$\Delta\rho_{min}$ = −0.30 e Å^{-3}

Table 1
Selected bond lengths (Å).

Pd1—O4	2.1945 (13)	Pd1—N4	2.2355 (17)
Pd1—N2	2.1987 (17)	Pd1—O3^{i}	2.2382 (13)

Symmetry code: (i) $x-\frac{1}{2}, -y+\frac{3}{2}, z-\frac{1}{2}$.

Table 2
Hydrogen-bond geometry (Å, °).

D—H···A	D—H	H···A	D···A	D—H···A
N3—H3A···O3^{ii}	0.86	2.04	2.762 (2)	142
N1—H1A···O2^{iii}	0.86	1.91	2.699 (2)	152

Symmetry codes: (ii) $x-\frac{1}{2}, -y+\frac{3}{2}, z-\frac{1}{2}$; (iii) $-x+\frac{1}{2}, y-\frac{1}{2}, -z+\frac{1}{2}$.

Data collection: *APEX2* (Bruker, 2004); cell refinement: *SAINT-Plus* (Bruker, 2001); data reduction: *SAINT-Plus*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 1997); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *SHELXTL* (Bruker, 2001); software used to prepare material for publication: *SHELXTL*.

The authors thank the Education Department of Shandong Province for research and development projects (J06A55).

Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: HB2670).

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supporting information

Acta Cryst. (2008). E**64**, m133 [https://doi.org/10.1107/S1600536807065051]

catena-Poly[[bis(1*H*-benzimidazole-κN^3)palladium(II)]-μ-benzene-1,4-dicarboxylato-κ^2O^1:O^4]

Qingguo Meng, Lintong Wang, Yanzhen Liu and Yan Pang

S1. Comment

Dicarboxylic acids are key components in the synthesis of coordination polymers (*e.g.* Okabe & Oya, 2000). In this paper, we report the structure of the title compound, (I).

In compound (I) the Pd atom is tetra-coordinated by two oxygen atoms from two benzene-1,4-dicarboxylate (bdc) dianions and two nitrogen atoms from two benzimidazole ligands, resulting in a slightly distorterd tetrahedral PdO_4N_2 geometry for the metal (Fig. 1, Table 1). Two short Pd···O contacts arise from this arrangement [Pd1···O2 = 2.7015 (15)Å and Pd1···O1^{i} = 2.5324 (14) Å; i = *x* - 1/2,-*y* + 5/2,*z* - 1/2] The bdc acts as a bridge to link Pd atoms into a chain (Fig. 2).

Two N—H···O hydrogen bonds (Table 2) help to link the chains into a three-dimensional structure.

S2. Experimental

A mixture of palladium acetate (1 mmol), benzene-1,4-dicarboxylic acid (1 mmol), benzimidazole (2 mmol), and 8 ml H_2O was sealed in a 25 ml autoclave and heated to 413 K for 2 days. On cooling to room temperature, colourless blocks of (I) were obtained with a yield of 12%. Anal. Calc. for $C_{22}H_{16}N_4O_4Pd$: C 60.77, H 3.16, N 11.05%; Found: C 60.71, H 3.22, N 47.01%.

S3. Refinement

All H atoms were placed in calculated positions with C—H = 0.93Å and N—H = 0.86Å and refined as riding with U_{iso}(H) = 1.2U_{eq}(carrier).

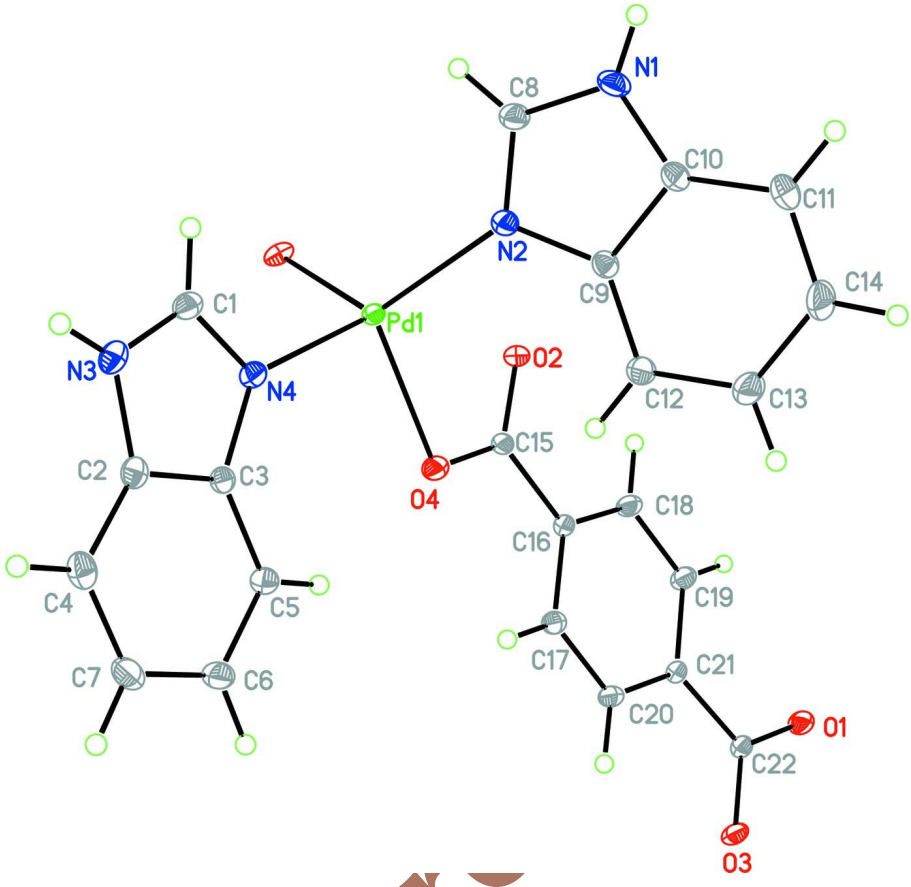

Figure 1

The molecular structure of (I), drawn with 30% probability displacement ellipsoids for the non-hydrogen atoms. The unlabelled O3 atom is at the symmetry position (x - 1/2, -y + 5/2, z - 1/2).

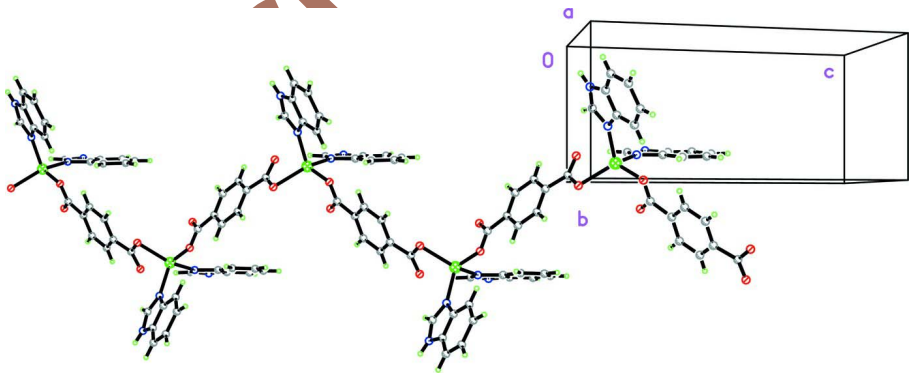

Figure 2

Fragment of a one dimensional chain in (I).

***catena*-Poly[[bis(1*H*-benzimidazole-κN^3)palladium(II)]- μ-benzene-1,4-dicarboxylato-κ^2O^1:O^4]**

Crystal data

$[Pd(C_8H_4O_4)(C_7H_6N_2)_2]$
M_r = 506.79

Monoclinic, $P2_1/n$
Hall symbol: -P 2yn

a = 17.0627 (5) Å
b = 7.3612 (10) Å
c = 18.0210 (5) Å
β = 114.362 (3)°
V = 2061.9 (3) Å^3
Z = 4
F(000) = 1016
D_x = 1.633 Mg m^{-3}

Mo $K\alpha$ radiation, λ = 0.71073 Å
Cell parameters from 3823 reflections
θ = 2.5–25.5°
μ = 0.94 mm^{-1}
T = 273 K
Block, colourless
0.43 × 0.28 × 0.22 mm

Data collection

Bruker APEXII CCD diffractometer
Radiation source: fine-focus sealed tube
Graphite monochromator
φ and ω scans
Absorption correction: multi-scan (*SADABS*; Bruker, 2001)
T_{min} = 0.689, T_{max} = 0.820

10244 measured reflections
3763 independent reflections
3136 reflections with $I > 2\sigma(I)$
R_{int} = 0.021
θ_{max} = 25.5°, θ_{min} = 2.5°
h = −20→18
k = −8→8
l = −16→21

Refinement

Refinement on F^2
Least-squares matrix: full
$R[F^2 > 2\sigma(F^2)]$ = 0.021
$wR(F^2)$ = 0.044
S = 1.00
3763 reflections
280 parameters
0 restraints
Primary atom site location: structure-invariant direct methods

Secondary atom site location: difference Fourier map
Hydrogen site location: inferred from neighbouring sites
H-atom parameters constrained
$w = 1/[\sigma^2(F_o^2) + (0.0204P)^2 + 0.1114P]$ where $P = (F_o^2 + 2F_c^2)/3$
$(\Delta/\sigma)_{max}$ = 0.007
$\Delta\rho_{max}$ = 0.25 e Å^{-3}
$\Delta\rho_{min}$ = −0.30 e Å^{-3}

Special details

Geometry. All e.s.d.'s (except the e.s.d. in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell e.s.d.'s are taken into account individually in the estimation of e.s.d.'s in distances, angles and torsion angles; correlations between e.s.d.'s in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell e.s.d.'s is used for estimating e.s.d.'s involving l.s. planes.

Refinement. Refinement of F^2 against ALL reflections. The weighted R-factor wR and goodness of fit S are based on F^2, conventional R-factors R are based on F, with F set to zero for negative F^2. The threshold expression of $F^2 > \sigma(F^2)$ is used only for calculating R-factors(gt) *etc*. and is not relevant to the choice of reflections for refinement. R-factors based on F^2 are statistically about twice as large as those based on F, and R- factors based on ALL data will be even larger.

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (Å^2)

	x	y	z	U_{iso}*/U_{eq}
Pd1	0.386453 (9)	0.87505 (2)	0.131047 (8)	0.03602 (6)
C1	0.38692 (14)	0.4922 (3)	0.05522 (12)	0.0501 (5)
H1	0.3273	0.5028	0.0320	0.060*
C2	0.51274 (14)	0.3771 (3)	0.08708 (12)	0.0465 (5)
C3	0.51865 (12)	0.5488 (3)	0.12239 (11)	0.0414 (5)
C4	0.58302 (16)	0.2724 (3)	0.09742 (14)	0.0600 (6)
H4	0.5781	0.1595	0.0727	0.072*
C5	0.59695 (13)	0.6213 (3)	0.17173 (13)	0.0514 (5)
H5	0.6021	0.7348	0.1959	0.062*

C6	0.66649 (15)	0.5167 (4)	0.18295 (14)	0.0642 (6)
H6	0.7208	0.5601	0.2164	0.077*
C7	0.65997 (17)	0.3453 (4)	0.14617 (15)	0.0677 (7)
H7	0.7100	0.2803	0.1556	0.081*
C8	0.23606 (14)	0.7875 (3)	0.18459 (14)	0.0558 (6)
H8	0.1970	0.7811	0.1303	0.067*
C9	0.35186 (13)	0.8107 (3)	0.29148 (12)	0.0442 (5)
C10	0.28633 (14)	0.7875 (3)	0.31634 (13)	0.0498 (5)
C11	0.30073 (17)	0.7824 (4)	0.39652 (15)	0.0698 (7)
H11	0.2559	0.7668	0.4126	0.084*
C12	0.43443 (14)	0.8292 (3)	0.34715 (13)	0.0549 (6)
H12	0.4793	0.8457	0.3312	0.066*
C13	0.44869 (17)	0.8227 (4)	0.42700 (15)	0.0749 (8)
H13	0.5047	0.8334	0.4665	0.090*
C14	0.3826 (2)	0.8007 (4)	0.45121 (16)	0.0843 (8)
H14	0.3948	0.7985	0.5065	0.101*
C15	0.49046 (12)	1.1428 (3)	0.23451 (11)	0.0402 (5)
C16	0.56002 (11)	1.2527 (3)	0.29649 (11)	0.0381 (4)
C17	0.64130 (13)	1.1842 (3)	0.32852 (12)	0.0508 (5)
H17	0.6528	1.0730	0.3107	0.061*
C18	0.54456 (13)	1.4201 (3)	0.32342 (12)	0.0500 (5)
H18	0.4893	1.4683	0.3010	0.060*
C19	0.60938 (13)	1.5157 (3)	0.38254 (12)	0.0517 (6)
H19	0.5980	1.6274	0.4001	0.062*
C20	0.70605 (13)	1.2803 (3)	0.38725 (12)	0.0512 (5)
H20	0.7617	1.2338	0.4088	0.061*
C21	0.68998 (12)	1.4463 (3)	0.41515 (11)	0.0392 (4)
C22	0.75824 (12)	1.5416 (3)	0.48470 (11)	0.0423 (5)
N1	0.21398 (12)	0.7742 (3)	0.24689 (12)	0.0596 (5)
H1A	0.1627	0.7596	0.2436	0.072*
N2	0.31782 (11)	0.8103 (2)	0.20759 (10)	0.0477 (4)
N3	0.42774 (12)	0.3463 (2)	0.04448 (10)	0.0527 (5)
H3A	0.4045	0.2514	0.0160	0.063*
N4	0.43772 (11)	0.6190 (2)	0.10114 (10)	0.0456 (4)
O1	0.74180 (9)	1.6814 (2)	0.51415 (9)	0.0616 (4)
O2	0.41604 (8)	1.1972 (2)	0.20972 (8)	0.0517 (4)
O3	0.83114 (8)	1.46812 (19)	0.51434 (8)	0.0510 (4)
O4	0.51057 (8)	0.99635 (19)	0.20976 (8)	0.0516 (4)

Atomic displacement parameters (Å²)

	U^{11}	U^{22}	U^{33}	U^{12}	U^{13}	U^{23}
Pd1	0.02818 (8)	0.04723 (9)	0.02930 (9)	−0.00258 (7)	0.00849 (6)	0.00106 (6)
C1	0.0435 (12)	0.0563 (13)	0.0459 (12)	−0.0098 (10)	0.0138 (10)	−0.0010 (10)
C2	0.0561 (13)	0.0490 (12)	0.0371 (11)	−0.0015 (11)	0.0221 (10)	0.0066 (10)
C3	0.0410 (12)	0.0494 (12)	0.0360 (11)	−0.0018 (9)	0.0180 (9)	0.0026 (9)
C4	0.0762 (18)	0.0561 (14)	0.0552 (14)	0.0118 (13)	0.0348 (13)	0.0061 (11)
C5	0.0415 (12)	0.0627 (13)	0.0506 (13)	−0.0062 (11)	0.0196 (10)	−0.0052 (11)

C6	0.0425 (14)	0.0929 (19)	0.0581 (15)	−0.0005 (13)	0.0217 (11)	0.0025 (14)
C7	0.0602 (17)	0.090 (2)	0.0626 (16)	0.0250 (14)	0.0351 (14)	0.0149 (14)
C8	0.0415 (13)	0.0714 (15)	0.0541 (14)	−0.0101 (11)	0.0194 (11)	0.0052 (12)
C9	0.0449 (12)	0.0463 (11)	0.0448 (12)	0.0012 (9)	0.0219 (10)	0.0057 (9)
C10	0.0491 (13)	0.0530 (12)	0.0538 (14)	−0.0008 (10)	0.0277 (11)	0.0044 (11)
C11	0.0730 (18)	0.0872 (18)	0.0654 (17)	−0.0066 (15)	0.0448 (15)	0.0048 (14)
C12	0.0468 (13)	0.0694 (15)	0.0496 (13)	0.0032 (11)	0.0210 (11)	0.0093 (11)
C13	0.0617 (16)	0.108 (2)	0.0487 (15)	−0.0052 (14)	0.0164 (13)	0.0122 (14)
C14	0.094 (2)	0.115 (2)	0.0515 (16)	−0.0136 (18)	0.0375 (16)	0.0058 (15)
C15	0.0316 (11)	0.0570 (13)	0.0334 (10)	−0.0034 (9)	0.0148 (8)	0.0020 (9)
C16	0.0337 (10)	0.0493 (12)	0.0310 (10)	−0.0023 (9)	0.0128 (8)	−0.0015 (8)
C17	0.0377 (12)	0.0596 (13)	0.0492 (13)	0.0047 (10)	0.0120 (10)	−0.0170 (10)
C18	0.0344 (11)	0.0571 (13)	0.0470 (12)	0.0087 (9)	0.0053 (9)	−0.0043 (10)
C19	0.0425 (12)	0.0527 (13)	0.0475 (13)	0.0081 (10)	0.0060 (10)	−0.0107 (10)
C20	0.0320 (11)	0.0648 (14)	0.0487 (13)	0.0076 (10)	0.0084 (9)	−0.0102 (11)
C21	0.0361 (11)	0.0482 (11)	0.0311 (10)	0.0000 (9)	0.0117 (8)	−0.0008 (9)
C22	0.0386 (12)	0.0510 (12)	0.0335 (11)	−0.0001 (10)	0.0110 (9)	0.0021 (9)
N1	0.0430 (11)	0.0751 (13)	0.0693 (13)	−0.0076 (10)	0.0317 (10)	0.0075 (11)
N2	0.0404 (10)	0.0595 (11)	0.0452 (10)	−0.0043 (8)	0.0196 (8)	0.0014 (8)
N3	0.0598 (12)	0.0479 (11)	0.0458 (10)	−0.0114 (9)	0.0171 (9)	−0.0077 (8)
N4	0.0412 (10)	0.0491 (10)	0.0432 (10)	−0.0059 (8)	0.0141 (8)	−0.0053 (8)
O1	0.0449 (9)	0.0736 (10)	0.0552 (10)	0.0035 (7)	0.0097 (7)	−0.0251 (8)
O2	0.0290 (8)	0.0759 (10)	0.0461 (8)	−0.0005 (7)	0.0114 (6)	−0.0057 (7)
O3	0.0378 (8)	0.0546 (9)	0.0441 (8)	0.0077 (7)	0.0002 (6)	−0.0024 (7)
O4	0.0390 (8)	0.0575 (9)	0.0528 (9)	−0.0046 (7)	0.0134 (7)	−0.0161 (7)

Geometric parameters (Å, º)

Pd1—O4	2.1945 (13)	C11—C14	1.343 (3)
Pd1—N2	2.1987 (17)	C11—H11	0.9300
Pd1—N4	2.2355 (17)	C12—C13	1.358 (3)
Pd1—O3^{i}	2.2382 (13)	C12—H12	0.9300
Pd1—O1^{i}	2.5324 (14)	C13—C14	1.377 (4)
C1—N4	1.310 (2)	C13—H13	0.9300
C1—N3	1.337 (3)	C14—H14	0.9300
C1—H1	0.9300	C15—O2	1.227 (2)
C2—N3	1.351 (3)	C15—O4	1.267 (2)
C2—C4	1.372 (3)	C15—C16	1.489 (3)
C2—C3	1.400 (3)	C16—C17	1.360 (3)
C3—C5	1.371 (3)	C16—C18	1.389 (3)
C3—N4	1.373 (2)	C17—C20	1.370 (3)
C4—C7	1.354 (3)	C17—H17	0.9300
C4—H4	0.9300	C18—C19	1.372 (3)
C5—C6	1.358 (3)	C18—H18	0.9300
C5—H5	0.9300	C19—C21	1.353 (3)
C6—C7	1.408 (3)	C19—H19	0.9300
C6—H6	0.9300	C20—C21	1.391 (3)
C7—H7	0.9300	C20—H20	0.9300

C8—N2	1.292 (2)	C21—C22	1.488 (3)
C8—N1	1.327 (3)	C22—O1	1.242 (2)
C8—H8	0.9300	C22—O3	1.256 (2)
C9—C12	1.358 (3)	N1—H1A	0.8600
C9—C10	1.376 (3)	N3—H3A	0.8600
C9—N2	1.378 (2)	O1—Pd1^{ii}	2.5324 (14)
C10—N1	1.351 (3)	O3—Pd1^{ii}	2.2382 (13)
C10—C11	1.363 (3)		

O4—Pd1—N2	107.72 (6)	C14—C13—H13	119.1
O4—Pd1—N4	97.45 (6)	C11—C14—C13	121.3 (2)
N2—Pd1—N4	109.34 (6)	C11—C14—H14	119.4
O4—Pd1—O3^{i}	108.54 (5)	C13—C14—H14	119.4
N2—Pd1—O3^{i}	125.04 (6)	O2—C15—O4	122.67 (18)
N4—Pd1—O3^{i}	105.28 (6)	O2—C15—C16	118.73 (19)
N4—C1—N3	114.5 (2)	O4—C15—C16	118.60 (17)
N4—C1—H1	122.7	C17—C16—C18	119.19 (18)
N3—C1—H1	122.7	C17—C16—C15	118.35 (18)
N3—C2—C4	131.0 (2)	C18—C16—C15	122.46 (17)
N3—C2—C3	105.61 (18)	C16—C17—C20	119.53 (19)
C4—C2—C3	123.4 (2)	C16—C17—H17	120.2
C5—C3—N4	129.27 (19)	C20—C17—H17	120.2
C5—C3—C2	121.0 (2)	C19—C18—C16	121.20 (18)
N4—C3—C2	109.73 (17)	C19—C18—H18	119.4
C7—C4—C2	115.1 (2)	C16—C18—H18	119.4
C7—C4—H4	122.4	C21—C19—C18	119.55 (19)
C2—C4—H4	122.4	C21—C19—H19	120.2
C6—C5—C3	115.7 (2)	C18—C19—H19	120.2
C6—C5—H5	122.2	C17—C20—C21	121.10 (18)
C3—C5—H5	122.2	C17—C20—H20	119.4
C5—C6—C7	122.9 (2)	C21—C20—H20	119.4
C5—C6—H6	118.5	C19—C21—C20	119.41 (18)
C7—C6—H6	118.5	C19—C21—C22	119.34 (18)
C4—C7—C6	121.9 (2)	C20—C21—C22	121.06 (17)
C4—C7—H7	119.1	O1—C22—O3	122.26 (18)
C6—C7—H7	119.1	O1—C22—C21	120.94 (18)
N2—C8—N1	112.6 (2)	O3—C22—C21	116.68 (18)
N2—C8—H8	123.7	C8—N1—C10	107.97 (19)
N1—C8—H8	123.7	C8—N1—H1A	126.0
C12—C9—C10	120.5 (2)	C10—N1—H1A	126.0
C12—C9—N2	130.5 (2)	C8—N2—C9	105.25 (18)
C10—C9—N2	109.00 (18)	C8—N2—Pd1	127.66 (15)
N1—C10—C11	132.6 (2)	C9—N2—Pd1	126.54 (13)
N1—C10—C9	105.20 (19)	C1—N3—C2	106.50 (17)
C11—C10—C9	122.2 (2)	C1—N3—H3A	126.7
C14—C11—C10	117.0 (2)	C2—N3—H3A	126.7
C14—C11—H11	121.5	C1—N4—C3	103.61 (17)
C10—C11—H11	121.5	C1—N4—Pd1	122.03 (15)

C9—C12—C13	117.2 (2)	C3—N4—Pd1	134.34 (13)
C9—C12—H12	121.4	C22—O1—Pd1^{ii}	84.55 (11)
C13—C12—H12	121.4	C22—O3—Pd1^{ii}	97.89 (12)
C12—C13—C14	121.9 (2)	C15—O4—Pd1	104.15 (12)
C12—C13—H13	119.1		

Symmetry codes: (i) $x-1/2, -y+5/2, z-1/2$; (ii) $x+1/2, -y+5/2, z+1/2$.

Hydrogen-bond geometry (Å, º)

D—H···*A*	*D*—H	H···*A*	*D*···*A*	*D*—H···*A*
N3—H3*A*···O3^{iii}	0.86	2.04	2.762 (2)	142
N1—H1*A*···O2^{iv}	0.86	1.91	2.699 (2)	152

Symmetry codes: (iii) $x-1/2, -y+3/2, z-1/2$; (iv) $-x+1/2, y-1/2, -z+1/2$.