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2-[1-(4-Ethoxyphenyl)-2-oxo-4-styryl-azetid-3-yl]isoindoline-1,3-dione

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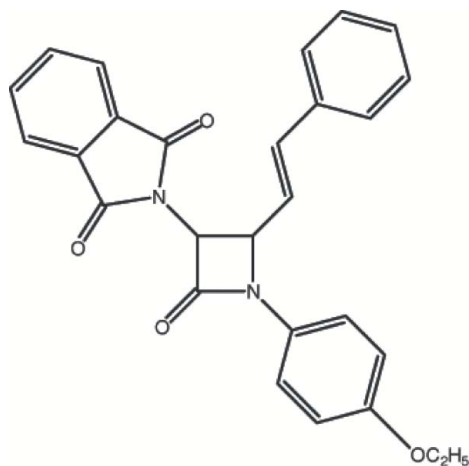
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Key indicators: single-crystal X-ray study; $T = 293$ K; mean $\sigma(\text{C}-\text{C}) = 0.003$ Å; R factor = 0.042; wR factor = 0.100; data-to-parameter ratio = 16.5.

The title compound, $\text{C}_{27}\text{H}_{22}\text{N}_2\text{O}_4$, contains a nearly planar four-membered β -lactam ring, which makes dihedral angles of 74.64 (12), 1.70 (11) and 73.67 (12)° with the nine-membered ring system, the benzene ring and the phenyl ring, respectively. The crystal structure is stabilized by $\text{C}-\text{H}\cdots\text{O}$ and $\text{C}-\text{H}\cdots\pi$ interactions and a $\pi-\pi$ interaction [centroid-centroid distance = 3.4505 (19) Å] is also present.

Related literature

For related structures, see: Pinar *et al.* (2006); Akkurt *et al.* (2007). For background, see: Halve *et al.* (2007); Aoyama *et al.* (2001). For related literature, see: Jarrahpour & Zarei (2007).



Experimental

Crystal data

 $\text{C}_{27}\text{H}_{22}\text{N}_2\text{O}_4$ $M_r = 438.47$

Monoclinic, $C2/c$
 $a = 33.7560$ (17) Å
 $b = 7.0403$ (2) Å
 $c = 31.0482$ (17) Å
 $\beta = 140.454$ (3)°
 $V = 4698.0$ (5) Å³

$Z = 8$
Mo $K\alpha$ radiation
 $\mu = 0.08$ mm⁻¹
 $T = 293$ (2) K
 $0.53 \times 0.45 \times 0.14$ mm

Data collection

Stoe IPDS-2 diffractometer
Absorption correction: none
22562 measured reflections

4934 independent reflections
3071 reflections with $I > 2\sigma(I)$
 $R_{\text{int}} = 0.052$

Refinement

$R[F^2 > 2\sigma(F^2)] = 0.042$
 $wR(F^2) = 0.099$
 $S = 0.96$
4934 reflections

299 parameters
H-atom parameters constrained
 $\Delta\rho_{\text{max}} = 0.17$ e Å⁻³
 $\Delta\rho_{\text{min}} = -0.11$ e Å⁻³

Table 1

Hydrogen-bond geometry (Å, °).

$D-H\cdots A$	$D-H$	$H\cdots A$	$D\cdots A$	$D-H\cdots A$
$\text{C3}-\text{H3}\cdots\text{O1}^{\text{i}}$	0.93	2.53	3.283 (3)	138
$\text{C5}-\text{H5}\cdots\text{O3}^{\text{ii}}$	0.93	2.47	3.257 (3)	142
$\text{C13}-\text{H13}\cdots\text{O3}$	0.93	2.53	3.139 (2)	123
$\text{C20}-\text{H20}\cdots\text{O2}^{\text{iii}}$	0.93	2.51	3.374 (2)	156
$\text{C9}-\text{H9}\cdots\text{Cg1}^{\text{iv}}$	0.98	2.84	3.7938 (16)	166
$\text{C19}-\text{H19C}\cdots\text{Cg2}^{\text{v}}$	0.96	2.82	3.633 (4)	143

Symmetry codes: (i) $-x+1, -y+1, -z+1$; (ii) $-x+1, -y+2, -z+1$; (iii) $x, y-1, z$; (iv) $x, y+1, z$; (v) $x+\frac{1}{2}, -y+\frac{1}{2}, z+\frac{1}{2}$. Cg1 is the centroid of atoms C12-C17 and Cg2 is the centroid of atoms C2-C7.

Data collection: *X-AREA* (Stoe & Cie, 2002); cell refinement: *X-AREA*; data reduction: *X-RED32* (Stoe & Cie, 2002); program(s) used to solve structure: *SIR97* (Altomare *et al.*, 1999); program(s) used to refine structure: *SHELXL97* (Sheldrick, 2008); molecular graphics: *ORTEP-3* (Farrugia, 1997); software used to prepare material for publication: *WinGX* (Farrugia, 1999).

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Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: HB2715).

References

- Akkurt, M., Yalçın, Ş. P., Jarrahpour, A. A., Nazari, M. & Büyükgüngör, O. (2007). *Acta Cryst.* **E63**, o3729–o3730.
Altomare, A., Burla, M. C., Camalli, M., Cascarano, G. L., Giacovazzo, C., Guagliardi, A., Moliterni, A. G. G., Polidori, G. & Spagna, R. (1999). *J. Appl. Cryst.* **32**, 115–119.
Aoyama, Y., Uenaka, M., Kii, M., Tanaka, M., Konoike, T., Hayasakikajiwaru, Y., Naya, N. & Nakajima, M. (2001). *Bioorg. Med. Chem.* **9**, 3065–3075.
Farrugia, L. J. (1997). *J. Appl. Cryst.* **30**, 565.
Farrugia, L. J. (1999). *J. Appl. Cryst.* **32**, 837–838.
Halve, A. K., Bhadauria, D. & Dubey, R. (2007). *Bioorg. Med. Chem. Lett.* **17**, 341–345.
Jarrahpour, A. A. & Zarei, M. (2007). *Molecules*, **12**, 2364–2379.
Pinar, S., Akkurt, M., Jarrahpour, A. A., Khalili, D. & Büyükgüngör, O. (2006). *Acta Cryst.* **E62**, o804–o806.
Sheldrick, G. M. (2008). *Acta Cryst.* **A64**, 112–122.
Stoe & Cie (2002). *X-AREA* and *X-RED32*. Stoe & Cie, Darmstadt, Germany.

supporting information

Acta Cryst. (2008). E64, o924 [doi:10.1107/S1600536808011586]

2-[1-(4-Ethoxyphenyl)-2-oxo-4-styrylazetid-3-yl]isoindoline-1,3-dione

Mehmet Akkurt, Selvi Karaca, Ali Asghar Jarrahpour, Maarof Zarei and Orhan Büyükgüngör

S1. Comment

The 2-azetidinone ring system is the common structural feature of a number of broad spectrum β -lactam antibiotics (Halve *et al.*, 2007) and also possesses other pharmacological properties (Aoyama *et al.*, 2001). As part of our ongoing studies of such materials (Pinar *et al.*, 2006; Akkurt *et al.*, 2007), we now report the synthesis and structure of the title compound, (I), (Fig. 1).

The four-membered β -lactam ring in (I) is nearly planar, with maximum deviations of 0.022 (1) Å for N2 and -0.021 (2) Å for C10. Within the lactam ring, the bond lengths are similar to those observed in our previous studies (Pinar *et al.*, 2006; Akkurt *et al.*, 2007).

The four-membered β -lactam ring (N2/C9–C11) in (I) makes dihedral angles of 74.64 (12), 1.70 (11) and 73.67 (12)° with the nine-membered ring system (A: N1/C1–C8) {max. deviations from planarity = 0.048 (2) for N1 and -0.029 (3) Å for C2 and C7}, the benzene ring (B: C12–C17) and the phenyl ring (C: C22–C27), respectively. The other dihedral angles are A/B = 74.74 (9), A/C = 55.97 (11) and B/C = 74.84 (10)°. The sum of the bond angles about atom N2 is 360.0°.

The packing and hydrogen bonding of the title compound is shown in Fig. 2. The crystal structure is stabilized by inter- and intramolecular C—H \cdots O interactions and C—H \cdots π contacts (Table 1). Finally, an aromatic π — π stacking interaction { $C_g3 \cdots C_g2(1-x, 2-y, 1-z) = 3.4505$ (19)Å, where C_g3 is the centroid of the N1/C1/C2/C7/C8 five-membered ring} is observed in the crystal structure.

S2. Experimental

A solution of Schiff base (4-cinnamylidene)-(4-ethoxyphenyl)amine (1.0 eq.) was stirred with Phthaloylglycine (1.5 eq.), p-toluenesulfonyl chloride (1.5 eq.) and triethylamine (5 eq.) in dry CH₂Cl₂ at room temperature. After 10 h, the mixture was washed with saturated sodium bicarbonate solution and brine, dried over sodium sulfate and the solvent was evaporated to give the crude product which was then purified by recrystallization from EtOAc (Jarrahpour & Zarei, 2007) [mp: 434–436 K]. IR (CHCl₃) cm⁻¹: 1724.2, 1758.5 (CO, phth), 1774.7 (CO, β -lactam); ¹H NMR (250 MHz, CDCl₃) δ 1.37 (Me, t, 3H), 2.33 (Me, s, 3H), 3.97 (OCH₂, q, 2H), 5.03 (H-4, dd, 1H, J=5.5, 8.5), 5.68 (H-3, d, 1H, J=5.5), 6.32 (H-5, dd, J=8.5, 16.0), 6.85 (H-6, d, 1H, J=9.0), 7.19–7.82 (ArH, m, 13H); ¹³C NMR (62.9 MHz, CDCl₃) δ 14.78 (Me), 57.69 (OCH₂), 61.04 (C-4), 63.67 (C-3), 114.99–155.82 (C=C, aromatic carbons), 160.56 (CO, phth), 167.28 (CO, β -lactam); GC-MS m/z = 438 [M⁺]. Analysis calculated for C₂₇H₂₂N₂O₄: C 73.96, H 5.06, N 6.39%. Found: C 74.02, H 5.09, N 6.33%.

S3. Refinement

All the H atoms were geometrically generated (C–H = 0.93–0.98 Å) and refined as riding with $U_{iso}(H) = 1.2U_{eq}(C)$ or $1.5U_{eq}(\text{methyl C})$.

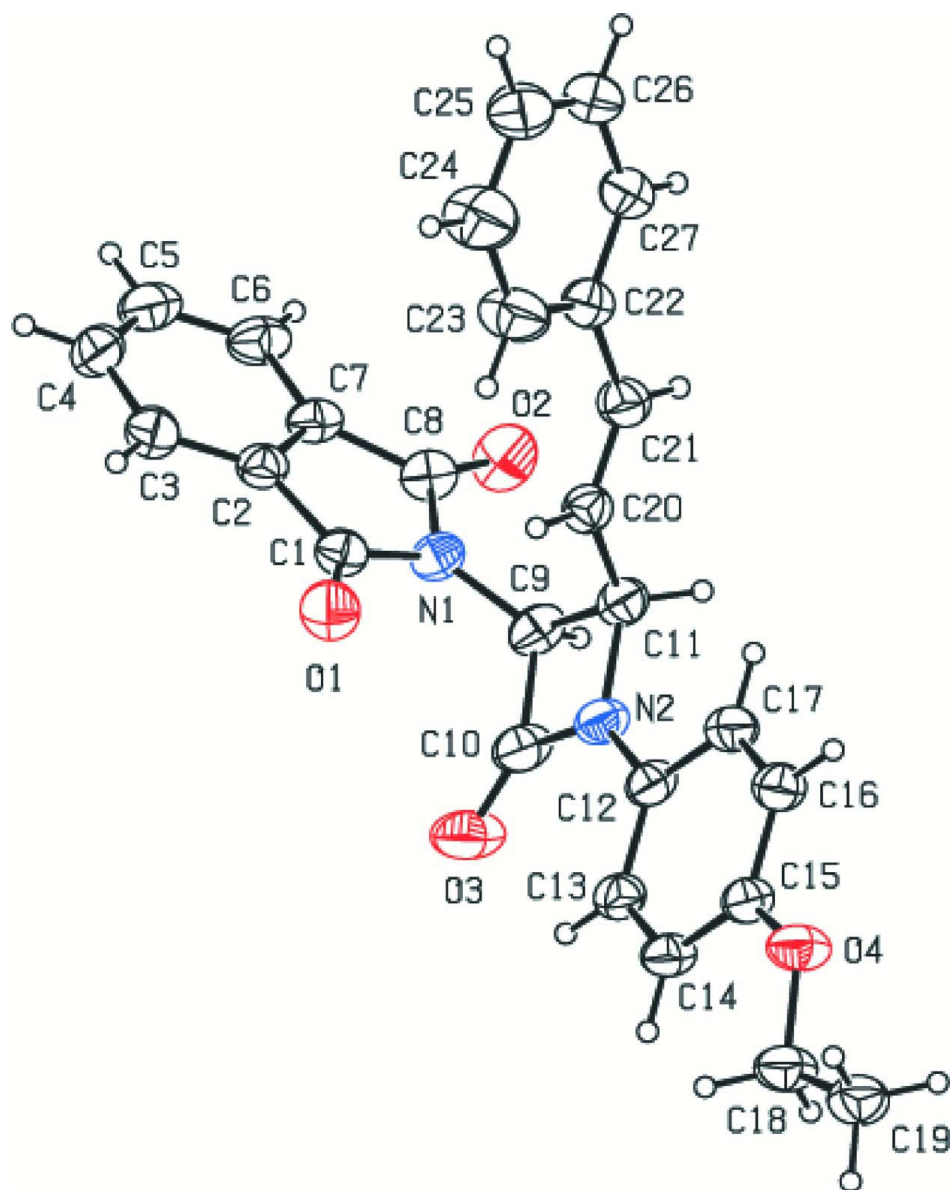
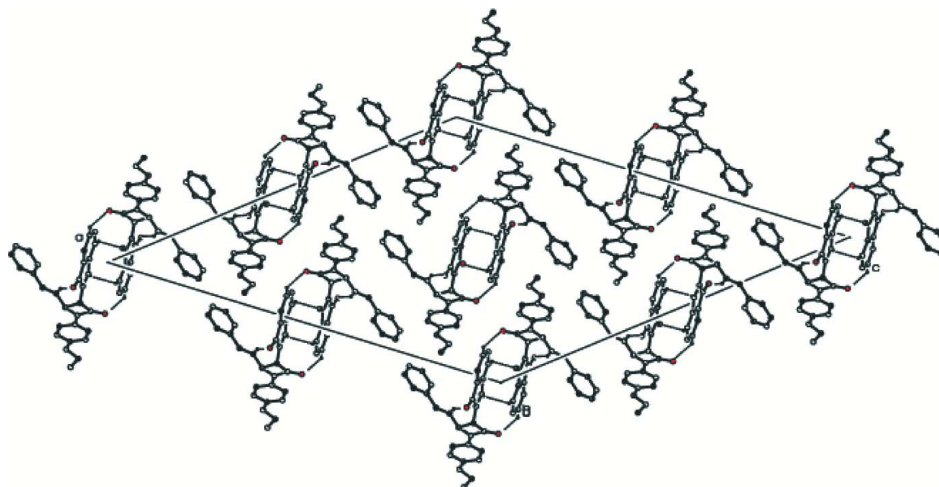


Figure 1

View of the molecular structure of (I), with 30% probability displacement ellipsoids for the non-hydrogen atoms.

**Figure 2**

View of the packing and hydrogen bonding interactions for (I). H atoms not involved in hydrogen bonding have been omitted for clarity.

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Crystal data

$C_{27}H_{22}N_2O_4$

$M_r = 438.47$

Monoclinic, $C2/c$

Hall symbol: $-C 2yc$

$a = 33.7560$ (17) Å

$b = 7.0403$ (2) Å

$c = 31.0482$ (17) Å

$\beta = 140.454$ (3)°

$V = 4698.0$ (5) Å³

$Z = 8$

$F(000) = 1840$

$D_x = 1.240$ Mg m⁻³

Mo $K\alpha$ radiation, $\lambda = 0.71073$ Å

Cell parameters from 19454 reflections

$\theta = 1.3$ – 27.3 °

$\mu = 0.08$ mm⁻¹

$T = 293$ K

Prism, colourless

$0.53 \times 0.45 \times 0.14$ mm

Data collection

Stoe IPDS-2
diffractometer

Radiation source: sealed X-ray tube, 12 x 0.4
mm long-fine focus

Plane graphite monochromator

Detector resolution: 6.67 pixels mm⁻¹

ω scans

22562 measured reflections

4934 independent reflections

3071 reflections with $I > 2\sigma(I)$

$R_{int} = 0.052$

$\theta_{max} = 26.8$ °, $\theta_{min} = 1.4$ °

$h = -42 \rightarrow 42$

$k = -8 \rightarrow 8$

$l = -39 \rightarrow 39$

Refinement

Refinement on F^2

Least-squares matrix: full

$R[F^2 > 2\sigma(F^2)] = 0.042$

$wR(F^2) = 0.099$

$S = 0.97$

4934 reflections

299 parameters

0 restraints

Primary atom site location: structure-invariant
direct methods

Secondary atom site location: difference Fourier
map

Hydrogen site location: inferred from
neighbouring sites

H-atom parameters constrained

$w = 1/[\sigma^2(F_o^2) + (0.05P)^2]$

where $P = (F_o^2 + 2F_c^2)/3$

$(\Delta/\sigma)_{max} < 0.001$

$\Delta\rho_{max} = 0.17$ e Å⁻³

$\Delta\rho_{min} = -0.11$ e Å⁻³

Extinction correction: *SHELXL97* (Sheldrick, 2008), $F_c^* = kFc[1 + 0.001xFc^2\lambda^3/\sin(2\theta)]^{-1/4}$

Extinction coefficient: 0.0032 (2)

Special details

Geometry. Bond distances, angles *etc.* have been calculated using the rounded fractional coordinates. All su's are estimated from the variances of the (full) variance-covariance matrix. The cell e.s.d.'s are taken into account in the estimation of distances, angles and torsion angles

Refinement. Refinement on F^2 for ALL reflections except those flagged by the user for potential systematic errors. Weighted R -factors wR and all goodnesses of fit S are based on F^2 , conventional R -factors R are based on F , with F set to zero for negative F^2 . The observed criterion of $F^2 > \sigma(F^2)$ is used only for calculating $-R$ -factor-obs *etc.* and is not relevant to the choice of reflections for refinement. R -factors based on F^2 are statistically about twice as large as those based on F , and R -factors based on ALL data will be even larger.

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (\AA^2)

	<i>x</i>	<i>y</i>	<i>z</i>	$U_{\text{iso}}^*/U_{\text{eq}}$
O1	0.52753 (6)	0.55052 (15)	0.57556 (7)	0.0765 (5)
O2	0.54065 (6)	1.14698 (16)	0.64241 (7)	0.0887 (5)
O3	0.67821 (7)	0.6142 (2)	0.70712 (8)	0.1034 (6)
O4	0.74347 (5)	-0.13874 (15)	0.89253 (6)	0.0729 (4)
N1	0.54930 (5)	0.83949 (16)	0.62525 (6)	0.0555 (4)
N2	0.64313 (6)	0.53412 (18)	0.74672 (7)	0.0617 (4)
C1	0.51510 (7)	0.7165 (2)	0.57070 (8)	0.0562 (5)
C2	0.46331 (7)	0.8305 (2)	0.50943 (8)	0.0554 (5)
C3	0.41750 (8)	0.7826 (3)	0.44180 (9)	0.0708 (6)
C4	0.37641 (8)	0.9267 (3)	0.39554 (10)	0.0814 (8)
C5	0.38151 (9)	1.1085 (3)	0.41602 (11)	0.0795 (8)
C6	0.42714 (8)	1.1549 (2)	0.48327 (10)	0.0714 (7)
C7	0.46791 (7)	1.0133 (2)	0.52986 (8)	0.0566 (5)
C8	0.52172 (7)	1.0197 (2)	0.60414 (9)	0.0598 (5)
C9	0.60433 (7)	0.7905 (2)	0.69508 (8)	0.0641 (5)
C10	0.64799 (8)	0.6362 (3)	0.71402 (9)	0.0719 (6)
C11	0.59771 (7)	0.6599 (2)	0.72999 (8)	0.0597 (5)
C12	0.67075 (7)	0.3647 (2)	0.78458 (8)	0.0567 (5)
C13	0.71296 (7)	0.2662 (2)	0.79469 (8)	0.0597 (5)
C14	0.73897 (7)	0.0980 (2)	0.83145 (8)	0.0601 (5)
C15	0.72250 (7)	0.0284 (2)	0.85749 (8)	0.0580 (5)
C16	0.68038 (7)	0.1281 (2)	0.84731 (9)	0.0639 (6)
C17	0.65484 (7)	0.2938 (2)	0.81140 (8)	0.0640 (6)
C18	0.79389 (8)	-0.2335 (3)	0.91444 (10)	0.0753 (7)
C19	0.81168 (9)	-0.3978 (3)	0.95687 (10)	0.0799 (7)
C20	0.53523 (7)	0.5771 (2)	0.68565 (8)	0.0564 (5)
C21	0.49520 (7)	0.6482 (2)	0.67920 (9)	0.0672 (6)
C22	0.43110 (8)	0.5844 (2)	0.63340 (9)	0.0655 (6)
C23	0.40642 (9)	0.4242 (3)	0.59366 (12)	0.0936 (8)
C24	0.34510 (11)	0.3722 (3)	0.54989 (13)	0.1084 (10)
C25	0.30869 (9)	0.4779 (3)	0.54607 (12)	0.0919 (8)
C26	0.33206 (9)	0.6339 (3)	0.58488 (10)	0.0815 (7)
C27	0.39250 (8)	0.6879 (3)	0.62798 (9)	0.0726 (6)

H3	0.41420	0.66050	0.42780	0.0850*
H4	0.34460	0.89950	0.34950	0.0980*
H5	0.35350	1.20120	0.38360	0.0950*
H6	0.43060	1.27740	0.49710	0.0860*
H9	0.62800	0.90510	0.72220	0.0770*
H11	0.61500	0.71940	0.77040	0.0720*
H13	0.72390	0.31270	0.77680	0.0720*
H14	0.76750	0.03250	0.83850	0.0720*
H16	0.66940	0.08180	0.86510	0.0770*
H17	0.62660	0.35940	0.80490	0.0770*
H18A	0.82890	-0.14760	0.94060	0.0900*
H18B	0.78150	-0.27720	0.87570	0.0900*
H19A	0.77670	-0.48160	0.93060	0.0960*
H19B	0.82430	-0.35290	0.99530	0.0960*
H19C	0.84560	-0.46500	0.97210	0.0960*
H20	0.52400	0.46840	0.66110	0.0680*
H21	0.50880	0.75120	0.70680	0.0810*
H23	0.43100	0.35050	0.59620	0.1120*
H24	0.32880	0.26450	0.52300	0.1300*
H25	0.26760	0.44230	0.51660	0.1100*
H26	0.30720	0.70510	0.58250	0.0980*
H27	0.40790	0.79670	0.65410	0.0870*

Atomic displacement parameters (Å²)

	U^{11}	U^{22}	U^{33}	U^{12}	U^{13}	U^{23}
O1	0.0929 (9)	0.0524 (6)	0.0847 (9)	0.0128 (6)	0.0686 (8)	0.0056 (6)
O2	0.0932 (9)	0.0554 (6)	0.0838 (9)	0.0000 (6)	0.0594 (8)	-0.0081 (7)
O3	0.0824 (9)	0.1325 (11)	0.1212 (12)	0.0437 (8)	0.0853 (10)	0.0642 (9)
O4	0.0679 (7)	0.0707 (7)	0.0925 (9)	0.0208 (6)	0.0651 (7)	0.0242 (6)
N1	0.0501 (7)	0.0471 (6)	0.0569 (8)	0.0026 (5)	0.0380 (7)	0.0046 (6)
N2	0.0499 (7)	0.0691 (8)	0.0602 (8)	0.0122 (6)	0.0409 (7)	0.0166 (7)
C1	0.0574 (9)	0.0528 (8)	0.0646 (10)	0.0019 (7)	0.0487 (9)	0.0029 (8)
C2	0.0488 (8)	0.0622 (9)	0.0601 (10)	0.0012 (7)	0.0433 (8)	0.0039 (8)
C3	0.0620 (10)	0.0890 (11)	0.0675 (12)	-0.0043 (9)	0.0515 (10)	-0.0026 (10)
C4	0.0527 (10)	0.1283 (17)	0.0578 (11)	0.0072 (11)	0.0412 (10)	0.0174 (11)
C5	0.0597 (11)	0.0968 (14)	0.0827 (15)	0.0211 (10)	0.0551 (12)	0.0319 (11)
C6	0.0641 (10)	0.0658 (10)	0.0866 (14)	0.0161 (8)	0.0587 (11)	0.0234 (9)
C7	0.0491 (8)	0.0547 (8)	0.0671 (10)	0.0053 (7)	0.0451 (9)	0.0103 (7)
C8	0.0593 (9)	0.0477 (8)	0.0698 (11)	-0.0010 (7)	0.0491 (10)	0.0023 (8)
C9	0.0488 (8)	0.0616 (9)	0.0589 (10)	-0.0025 (7)	0.0355 (8)	0.0039 (8)
C10	0.0504 (9)	0.0870 (12)	0.0687 (11)	0.0113 (8)	0.0434 (9)	0.0225 (9)
C11	0.0522 (8)	0.0608 (8)	0.0548 (9)	0.0044 (7)	0.0383 (8)	0.0025 (7)
C12	0.0421 (7)	0.0642 (9)	0.0496 (9)	0.0026 (7)	0.0316 (7)	0.0061 (7)
C13	0.0462 (8)	0.0729 (9)	0.0568 (9)	0.0006 (7)	0.0389 (8)	0.0050 (8)
C14	0.0456 (8)	0.0676 (9)	0.0649 (10)	0.0045 (7)	0.0420 (8)	0.0050 (8)
C15	0.0477 (8)	0.0627 (9)	0.0574 (10)	0.0050 (7)	0.0389 (8)	0.0057 (7)
C16	0.0607 (9)	0.0717 (10)	0.0670 (11)	0.0111 (8)	0.0513 (9)	0.0141 (8)

C17	0.0562 (9)	0.0744 (10)	0.0639 (10)	0.0152 (8)	0.0470 (9)	0.0130 (8)
C18	0.0659 (10)	0.0775 (11)	0.0890 (13)	0.0206 (9)	0.0614 (11)	0.0174 (10)
C19	0.0763 (12)	0.0824 (11)	0.0857 (13)	0.0266 (9)	0.0637 (12)	0.0206 (10)
C20	0.0555 (9)	0.0539 (8)	0.0560 (9)	0.0039 (7)	0.0420 (8)	0.0030 (7)
C21	0.0574 (9)	0.0721 (10)	0.0643 (11)	-0.0014 (8)	0.0449 (9)	-0.0129 (8)
C22	0.0579 (9)	0.0755 (10)	0.0638 (10)	-0.0002 (8)	0.0471 (9)	-0.0060 (8)
C23	0.0762 (13)	0.0912 (13)	0.1146 (17)	-0.0156 (10)	0.0739 (14)	-0.0352 (12)
C24	0.0876 (15)	0.1084 (16)	0.124 (2)	-0.0347 (13)	0.0802 (16)	-0.0474 (14)
C25	0.0647 (11)	0.1170 (16)	0.0931 (15)	-0.0159 (12)	0.0606 (12)	-0.0111 (13)
C26	0.0662 (11)	0.1067 (14)	0.0831 (13)	0.0048 (11)	0.0606 (11)	0.0017 (12)
C27	0.0659 (11)	0.0863 (11)	0.0715 (11)	0.0024 (9)	0.0545 (10)	-0.0063 (9)

Geometric parameters (Å, °)

O1—C1	1.211 (2)	C21—C22	1.470 (4)
O2—C8	1.205 (2)	C22—C23	1.380 (3)
O3—C10	1.203 (5)	C22—C27	1.385 (4)
O4—C15	1.3717 (18)	C23—C24	1.388 (5)
O4—C18	1.421 (4)	C24—C25	1.362 (6)
N1—C1	1.3913 (19)	C25—C26	1.347 (3)
N1—C8	1.402 (2)	C26—C27	1.374 (4)
N1—C9	1.435 (2)	C3—H3	0.9300
N2—C10	1.353 (3)	C4—H4	0.9300
N2—C11	1.475 (3)	C5—H5	0.9300
N2—C12	1.408 (2)	C6—H6	0.9300
C1—C2	1.480 (2)	C9—H9	0.9800
C2—C3	1.381 (2)	C11—H11	0.9800
C2—C7	1.390 (2)	C13—H13	0.9300
C3—C4	1.394 (3)	C14—H14	0.9300
C4—C5	1.380 (3)	C16—H16	0.9300
C5—C6	1.371 (3)	C17—H17	0.9300
C6—C7	1.382 (2)	C18—H18A	0.9700
C7—C8	1.470 (2)	C18—H18B	0.9700
C9—C10	1.537 (4)	C19—H19A	0.9600
C9—C11	1.563 (3)	C19—H19B	0.9600
C11—C20	1.485 (3)	C19—H19C	0.9600
C12—C13	1.386 (4)	C20—H20	0.9300
C12—C17	1.386 (4)	C21—H21	0.9300
C13—C14	1.389 (2)	C23—H23	0.9300
C14—C15	1.377 (4)	C24—H24	0.9300
C15—C16	1.386 (4)	C25—H25	0.9300
C16—C17	1.366 (2)	C26—H26	0.9300
C18—C19	1.488 (3)	C27—H27	0.9300
C20—C21	1.304 (4)		
C15—O4—C18	118.5 (2)	C24—C25—C26	120.1 (3)
C1—N1—C8	111.45 (13)	C25—C26—C27	120.2 (3)
C1—N1—C9	125.59 (13)	C22—C27—C26	121.6 (2)

C8—N1—C9	122.91 (13)	C2—C3—H3	122.00
C10—N2—C11	96.40 (17)	C4—C3—H3	122.00
C10—N2—C12	133.7 (3)	C3—C4—H4	119.00
C11—N2—C12	129.9 (2)	C5—C4—H4	119.00
O1—C1—N1	124.49 (15)	C4—C5—H5	119.00
O1—C1—C2	129.48 (15)	C6—C5—H5	119.00
N1—C1—C2	106.03 (13)	C5—C6—H6	121.00
C1—C2—C3	130.54 (16)	C7—C6—H6	121.00
C1—C2—C7	108.06 (14)	N1—C9—H9	110.00
C3—C2—C7	121.35 (16)	C10—C9—H9	110.00
C2—C3—C4	116.54 (19)	C11—C9—H9	110.00
C3—C4—C5	121.93 (19)	N2—C11—H11	112.00
C4—C5—C6	121.16 (19)	C9—C11—H11	112.00
C5—C6—C7	117.70 (16)	C20—C11—H11	112.00
C2—C7—C6	121.31 (15)	C12—C13—H13	120.00
C2—C7—C8	108.22 (14)	C14—C13—H13	120.00
C6—C7—C8	130.44 (15)	C13—C14—H14	120.00
O2—C8—N1	123.78 (16)	C15—C14—H14	120.00
O2—C8—C7	130.10 (16)	C15—C16—H16	120.00
N1—C8—C7	106.12 (13)	C17—C16—H16	120.00
N1—C9—C10	119.36 (17)	C12—C17—H17	120.00
N1—C9—C11	118.6 (2)	C16—C17—H17	120.00
C10—C9—C11	85.79 (16)	O4—C18—H18A	110.00
O3—C10—N2	132.5 (2)	O4—C18—H18B	110.00
O3—C10—C9	135.9 (2)	C19—C18—H18A	110.00
N2—C10—C9	91.6 (2)	C19—C18—H18B	110.00
N2—C11—C9	86.09 (19)	H18A—C18—H18B	109.00
N2—C11—C20	115.99 (14)	C18—C19—H19A	109.00
C9—C11—C20	116.63 (14)	C18—C19—H19B	110.00
N2—C12—C13	121.1 (2)	C18—C19—H19C	109.00
N2—C12—C17	119.5 (2)	H19A—C19—H19B	110.00
C13—C12—C17	119.40 (16)	H19A—C19—H19C	109.00
C12—C13—C14	120.1 (2)	H19B—C19—H19C	109.00
C13—C14—C15	120.0 (2)	C11—C20—H20	118.00
O4—C15—C14	125.2 (2)	C21—C20—H20	118.00
O4—C15—C16	115.3 (2)	C20—C21—H21	116.00
C14—C15—C16	119.57 (16)	C22—C21—H21	116.00
C15—C16—C17	120.6 (3)	C22—C23—H23	120.00
C12—C17—C16	120.3 (3)	C24—C23—H23	120.00
O4—C18—C19	107.8 (3)	C23—C24—H24	120.00
C11—C20—C21	123.68 (16)	C25—C24—H24	120.00
C20—C21—C22	127.56 (17)	C24—C25—H25	120.00
C21—C22—C23	122.6 (3)	C26—C25—H25	120.00
C21—C22—C27	120.12 (16)	C25—C26—H26	120.00
C23—C22—C27	117.3 (3)	C27—C26—H26	120.00
C22—C23—C24	120.5 (3)	C22—C27—H27	119.00
C23—C24—C25	120.4 (2)	C26—C27—H27	119.00

C18—O4—C15—C16	-170.02 (15)	C4—C5—C6—C7	0.1 (5)
C18—O4—C15—C14	11.6 (2)	C5—C6—C7—C8	178.3 (3)
C15—O4—C18—C19	174.18 (14)	C5—C6—C7—C2	0.6 (5)
C9—N1—C1—C2	-179.5 (2)	C6—C7—C8—N1	-175.1 (3)
C9—N1—C1—O1	0.5 (5)	C2—C7—C8—N1	2.8 (3)
C1—N1—C9—C10	30.8 (4)	C6—C7—C8—O2	5.8 (6)
C8—N1—C9—C10	-152.3 (2)	C2—C7—C8—O2	-176.3 (3)
C8—N1—C1—C2	3.2 (3)	C10—C9—C11—C20	-114.35 (19)
C1—N1—C8—O2	175.4 (3)	N1—C9—C10—O3	60.1 (3)
C9—N1—C8—O2	-1.9 (5)	N1—C9—C11—C20	7.0 (3)
C1—N1—C8—C7	-3.8 (3)	C11—C9—C10—O3	-179.3 (2)
C9—N1—C8—C7	178.9 (2)	N1—C9—C11—N2	124.16 (17)
C8—N1—C1—O1	-176.8 (3)	C10—C9—C11—N2	2.85 (12)
C8—N1—C9—C11	105.6 (2)	N1—C9—C10—N2	-123.7 (2)
C1—N1—C9—C11	-71.3 (3)	C11—C9—C10—N2	-3.10 (13)
C10—N2—C12—C17	-179.02 (18)	C9—C11—C20—C21	-98.1 (3)
C11—N2—C12—C17	0.9 (3)	N2—C11—C20—C21	162.74 (18)
C12—N2—C11—C9	176.79 (16)	C13—C12—C17—C16	-0.2 (2)
C10—N2—C12—C13	0.1 (3)	N2—C12—C13—C14	-179.13 (15)
C11—N2—C12—C13	-179.98 (15)	N2—C12—C17—C16	178.93 (15)
C12—N2—C10—O3	-0.4 (4)	C17—C12—C13—C14	0.0 (2)
C12—N2—C11—C20	-65.4 (2)	C12—C13—C14—C15	0.4 (2)
C10—N2—C11—C20	114.56 (19)	C13—C14—C15—C16	-0.6 (2)
C10—N2—C11—C9	-3.25 (13)	C13—C14—C15—O4	177.73 (15)
C11—N2—C10—O3	179.7 (2)	C14—C15—C16—C17	0.4 (3)
C11—N2—C10—C9	3.30 (14)	O4—C15—C16—C17	-178.09 (15)
C12—N2—C10—C9	-176.74 (17)	C15—C16—C17—C12	0.0 (3)
N1—C1—C2—C7	-1.4 (3)	C11—C20—C21—C22	175.49 (18)
N1—C1—C2—C3	176.2 (3)	C20—C21—C22—C23	5.4 (3)
O1—C1—C2—C7	178.6 (3)	C20—C21—C22—C27	-173.5 (2)
O1—C1—C2—C3	-3.8 (6)	C21—C22—C23—C24	-178.4 (2)
C3—C2—C7—C6	-0.6 (5)	C27—C22—C23—C24	0.6 (4)
C1—C2—C3—C4	-177.3 (3)	C21—C22—C27—C26	179.0 (2)
C7—C2—C3—C4	-0.1 (5)	C23—C22—C27—C26	0.0 (3)
C3—C2—C7—C8	-178.7 (3)	C22—C23—C24—C25	-0.6 (4)
C1—C2—C7—C8	-0.9 (3)	C23—C24—C25—C26	0.0 (4)
C1—C2—C7—C6	177.2 (3)	C24—C25—C26—C27	0.6 (4)
C2—C3—C4—C5	0.7 (5)	C25—C26—C27—C22	-0.6 (4)
C3—C4—C5—C6	-0.7 (6)		

Hydrogen-bond geometry (\AA , $^\circ$)

$D-H\cdots A$	$D-H$	$H\cdots A$	$D\cdots A$	$D-H\cdots A$
C3—H3 \cdots O1 ⁱ	0.93	2.53	3.283 (3)	138
C5—H5 \cdots O3 ⁱⁱ	0.93	2.47	3.257 (3)	142
C13—H13 \cdots O3	0.93	2.53	3.139 (2)	123
C20—H20 \cdots O2 ⁱⁱⁱ	0.93	2.51	3.374 (2)	156

C9—H9 \cdots C _g 1 ^{iv}	0.98	2.84	3.7938 (16)	166
C19—H19C \cdots C _g 2 ^v	0.96	2.82	3.633 (4)	143

Symmetry codes: (i) $-x+1, -y+1, -z+1$; (ii) $-x+1, -y+2, -z+1$; (iii) $x, y-1, z$; (iv) $x, y+1, z$; (v) $x+1/2, -y+1/2, z+1/2$.