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## Structure Reports

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## N-Benzyl-2-hydroxybenzamide

Qiu-Xia Zhang\* and Bi-Song Zhang

College of Materials Science and Chemical Engineering, Jinhua College of Profession and Technology, Jinhua, Zhejiang 321017, People's Republic of China

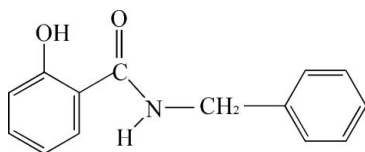
Correspondence e-mail: zbs\_jy@163.com

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Key indicators: single-crystal X-ray study;  $T = 290$  K; mean  $\sigma(\text{C}-\text{C}) = 0.003$  Å;  $R$  factor = 0.043;  $wR$  factor = 0.117; data-to-parameter ratio = 12.9.

In the title compound,  $\text{C}_{14}\text{H}_{13}\text{NO}_2$ , the mean planes through the benzyl and 2-hydroxybenzamide units make a dihedral angle of  $68.81(7)^\circ$ . There is an intramolecular  $\text{O}-\text{H}\cdots\text{O}$  hydrogen bond involving the carbonyl O atom and the 2-hydroxy substituent. In the crystal structure,  $\text{N}-\text{H}\cdots\text{O}$  hydrogen bonds link symmetry-related molecules into one-dimensional chains extending along the  $a$ -axis direction. These chains are further connected via  $\text{C}-\text{H}\cdots\text{O}$  hydrogen bonds, forming a sheet-like structure

## Related literature

For related literature, see: Agwade (1982); Allen *et al.* (1987).

## Experimental

## Crystal data

 $\text{C}_{14}\text{H}_{13}\text{NO}_2$  $M_r = 227.25$ Monoclinic,  $P2_1/n$  $a = 12.478(3)$  Å $b = 8.3503(17)$  Å $c = 12.664(3)$  Å $\beta = 118.02(3)^\circ$  $V = 1164.9(6)$  Å<sup>3</sup> $Z = 4$ Mo  $K\alpha$  radiation $\mu = 0.09$  mm<sup>-1</sup> $T = 290(2)$  K $0.33 \times 0.22 \times 0.20$  mm

## Data collection

Rigaku R-Axis RAPID

diffractometer

Absorption correction: multi-scan

(ABSCOR; Higashi, 1995)

 $T_{\min} = 0.977$ ,  $T_{\max} = 0.983$ 

11112 measured reflections

2665 independent reflections

1648 reflections with  $I > 2\sigma(I)$  $R_{\text{int}} = 0.037$ 

## Refinement

 $R[F^2 > 2\sigma(F^2)] = 0.043$  $wR(F^2) = 0.116$  $S = 1.03$ 

2665 reflections

207 parameters

All H-atom parameters refined

 $\Delta\rho_{\text{max}} = 0.17$  e Å<sup>-3</sup> $\Delta\rho_{\text{min}} = -0.14$  e Å<sup>-3</sup>

Table 1

Hydrogen-bond geometry (Å, °).

$D-\text{H}\cdots A$	$D-\text{H}$	$\text{H}\cdots A$	$D\cdots A$	$D-\text{H}\cdots A$
$\text{N1}-\text{HN1}\cdots\text{O1}^i$	0.882 (18)	2.083 (18)	2.9191 (18)	158.1 (18)
$\text{O1}-\text{HO1}\cdots\text{O2}$	0.98 (3)	1.56 (3)	2.4886 (19)	157 (2)
$\text{C2}-\text{H2}\cdots\text{O1}^i$	0.956 (19)	2.58 (2)	3.507 (2)	162.5 (15)

Symmetry code: (i)  $x + \frac{1}{2}, -y + \frac{3}{2}, z + \frac{1}{2}$ .

Data collection: *RAPID-AUTO* (Rigaku, 1998); cell refinement: *RAPID-AUTO*; data reduction: *CrystalStructure* (Rigaku/MS, 2002); program(s) used to solve structure: *SHELXS97* (Sheldrick, 2008); program(s) used to refine structure: *SHELXL97* (Sheldrick, 2008); molecular graphics: *SHELXTL* (Sheldrick, 2008); software used to prepare material for publication: *SHELXL97*.

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Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: SU2048).

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## supporting information

*Acta Cryst.* (2008). E64, o884 [doi:10.1107/S1600536808010453]

## N-Benzyl-2-hydroxybenzamide

Qiu-Xia Zhang and Bi-Song Zhang

### S1. Comment

Over a quarter of a century ago (Agwade, 1982) published a thesis on Potential Central Nervous System Active Agents, which included the synthesis of aromatic *N*-benzyl amides. One of the compounds synthesized was *N*-benzyl-2-hydroxybenzamide, (I), whose crystal structure has not been described until now.

The molecular structure of compound I is illustrated in Fig. 1. The bond lengths and angles are close to normal values (Allen *et al.*, 1987). There is an intramolecular O-H $\cdots$ O hydrogen bond in the molecule involving the carbonyl O-atom and the 2-hydroxyl substituent (Table 1). The best planes through the benzyl (atoms C8,C9-C14) and the 2-hydroxybenzamide (atoms C1-C6,C7,N1,O1,O2) moieties are inclined to one another by 68.81 (7)°.

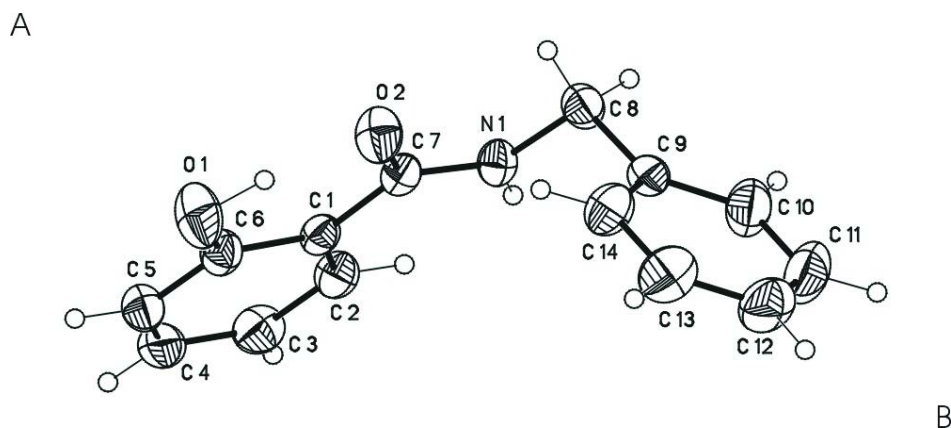
In the crystal structure of I symmetry related molecules are connected via an N-H $\cdots$ O hydrogen bond to form chains running along the *a* direction. These chains are further connected via C-H $\cdots$ O hydrogen bonds (Table 1) to form a sheet-like structure (Fig 2).

### S2. Experimental

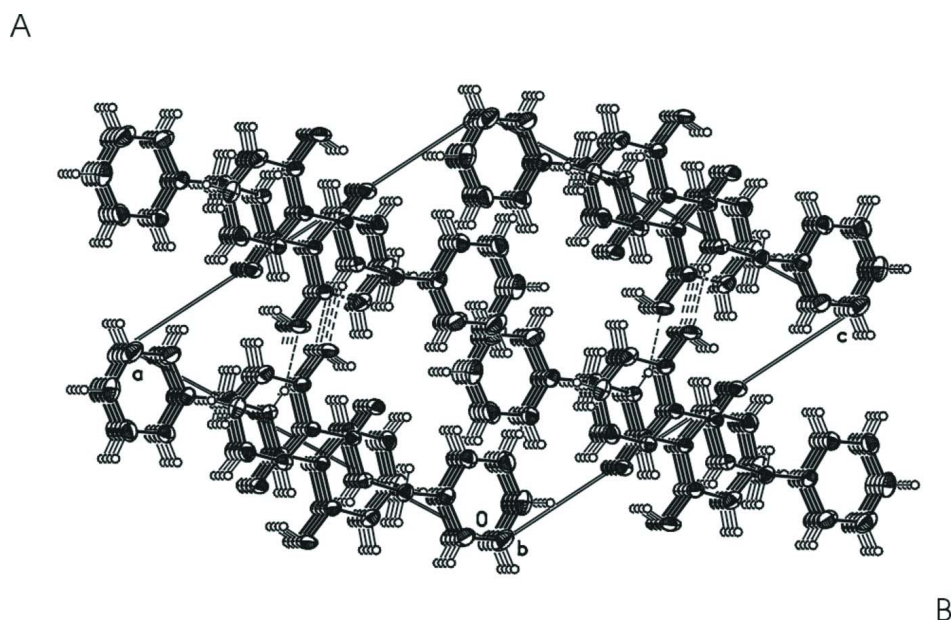
Freshly prepared CuCO<sub>3</sub> (0.310 g 2.50 mmol), [C<sub>6</sub>H<sub>4</sub>(COOC<sub>6</sub>H<sub>4</sub>CONHCH<sub>2</sub>ph)<sub>2</sub>](0.350 g 0.601 mmol), 2-chloro-benzoic acid (0.160 g 1.022 mmol), and 15 ml CH<sub>3</sub>OH/H<sub>2</sub>O (1:2,*v/v*) were mixed and stirred for *ca.* 1.5 h. The resulting suspension was heated in a 23 ml Teflon-lined stainless steel autoclave at 373 K for 7 days. After the autoclave was cooled to room temperature, and colorless block-like crystals, suitable for X-ray analysis, were obtained.

### S3. Refinement

All of the H atoms were located in difference Fourier syntheses and were freely refined: O-H = 0.98 (3), N-H = 0.882 (18), and C-H = 0.94 (2) - 1.010 (16) Å.

**Figure 1**

A view of the molecular structure of I, showing the atom-labelling scheme. Displacement ellipsoids are drawn at the 40% probability level.

**Figure 2**

The crystal packing diagram for compound I, viewed down the *b* axis. Dashed lines indicate N-H...O hydrogen bonds.

### N-Benzyl-2-hydroxybenzamide

#### Crystal data

$C_{14}H_{13}NO_2$

$M_r = 227.25$

Monoclinic,  $P2_1/n$

Hall symbol:  $-P 2_1n$

$a = 12.478 (3) \text{ \AA}$

$b = 8.3503 (17) \text{ \AA}$

$c = 12.664 (3) \text{ \AA}$

$\beta = 118.02 (3)^\circ$

$V = 1164.9 (6) \text{ \AA}^3$

$Z = 4$

$F(000) = 480$

$D_x = 1.296 \text{ Mg m}^{-3}$

Mo  $K\alpha$  radiation,  $\lambda = 0.71073 \text{ \AA}$

Cell parameters from 6913 reflections

$\theta = 3.0\text{--}27.5^\circ$

$\mu = 0.09 \text{ mm}^{-1}$

$T = 290 \text{ K}$

Block, colorless

$0.33 \times 0.22 \times 0.20 \text{ mm}$

*Data collection*

Rigaku R-AXIS RAPID  
diffractometer

Radiation source: fine-focus sealed tube

Graphite monochromator

Detector resolution: 10 pixels mm<sup>-1</sup>

$\omega$  scans

Absorption correction: multi-scan  
(*ABSCOR*; Higashi, 1995)

$T_{\min} = 0.977$ ,  $T_{\max} = 0.983$

11112 measured reflections

2665 independent reflections

1648 reflections with  $I > 2\sigma(I)$

$R_{\text{int}} = 0.037$

$\theta_{\max} = 27.5^\circ$ ,  $\theta_{\min} = 3.0^\circ$

$h = -15 \rightarrow 16$

$k = -10 \rightarrow 10$

$l = -16 \rightarrow 16$

*Refinement*

Refinement on  $F^2$

Least-squares matrix: full

$R[F^2 > 2\sigma(F^2)] = 0.043$

$wR(F^2) = 0.116$

$S = 1.03$

2665 reflections

207 parameters

0 restraints

Primary atom site location: structure-invariant  
direct methods

Secondary atom site location: difference Fourier  
map

Hydrogen site location: inferred from  
neighbouring sites

All H-atom parameters refined

$w = 1/[\sigma^2(F_o^2) + (0.0564P)^2 + 0.0491P]$

where  $P = (F_o^2 + 2F_c^2)/3$

$(\Delta/\sigma)_{\max} < 0.001$

$\Delta\rho_{\max} = 0.17 \text{ e } \text{\AA}^{-3}$

$\Delta\rho_{\min} = -0.14 \text{ e } \text{\AA}^{-3}$

Extinction correction: *SHELXL97* (Sheldrick,  
2008),  $F_c^* = kFc[1 + 0.001x\lambda^3/\sin(2\theta)]^{-1/4}$

Extinction coefficient: 0.011 (2)

*Special details*

**Geometry.** Bond distances, angles etc. have been calculated using the rounded fractional coordinates. All su's are estimated from the variances of the (full) variance-covariance matrix. The cell esds are taken into account in the estimation of distances, angles and torsion angles

**Refinement.** Refinement of  $F^2$  against ALL reflections. The weighted  $R$ -factor  $wR$  and goodness of fit  $S$  are based on  $F^2$ , conventional  $R$ -factors  $R$  are based on  $F$ , with  $F$  set to zero for negative  $F^2$ . The threshold expression of  $F^2 > \sigma(F^2)$  is used only for calculating  $R$ -factors(gt) etc. and is not relevant to the choice of reflections for refinement.  $R$ -factors based on  $F^2$  are statistically about twice as large as those based on  $F$ , and  $R$ -factors based on ALL data will be even larger.

*Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters ( $\text{\AA}^2$ )*

	<i>x</i>	<i>y</i>	<i>z</i>	$U_{\text{iso}}^*/U_{\text{eq}}$
O1	0.29704 (10)	0.74227 (17)	-0.23736 (9)	0.0731 (4)
O2	0.49480 (9)	0.60089 (14)	-0.16419 (9)	0.0600 (4)
N1	0.64585 (9)	0.63326 (15)	0.02206 (11)	0.0449 (4)
C1	0.45255 (11)	0.76116 (16)	-0.03318 (11)	0.0396 (4)
C2	0.48553 (14)	0.82179 (19)	0.07995 (14)	0.0493 (5)
C3	0.40715 (15)	0.9140 (2)	0.10230 (16)	0.0592 (6)
C4	0.29300 (15)	0.9477 (2)	0.01061 (17)	0.0604 (6)
C5	0.25762 (14)	0.8909 (2)	-0.10172 (16)	0.0576 (6)
C6	0.33637 (12)	0.79771 (18)	-0.12499 (13)	0.0478 (5)
C7	0.53270 (11)	0.66022 (17)	-0.06264 (12)	0.0412 (4)
C8	0.73259 (14)	0.5375 (2)	0.00128 (16)	0.0508 (5)
C9	0.80422 (12)	0.63678 (17)	-0.04347 (14)	0.0460 (5)
C10	0.92320 (15)	0.6768 (3)	0.03099 (19)	0.0722 (7)
C11	0.98716 (18)	0.7699 (3)	-0.0108 (2)	0.0875 (9)
C12	0.9333 (2)	0.8228 (3)	-0.1254 (2)	0.0785 (9)

C13	0.81540 (19)	0.7850 (3)	-0.20000 (19)	0.0721 (8)
C14	0.75139 (16)	0.6917 (2)	-0.15893 (16)	0.0596 (6)
HN1	0.6728 (14)	0.6819 (19)	0.0917 (16)	0.059 (5)*
H2	0.5651 (15)	0.7995 (18)	0.1431 (15)	0.060 (5)*
HO1	0.367 (2)	0.681 (3)	-0.230 (2)	0.106 (7)*
H3	0.4331 (15)	0.952 (2)	0.1817 (17)	0.078 (5)*
H4	0.2386 (15)	1.009 (2)	0.0252 (15)	0.071 (5)*
H5	0.1770 (17)	0.913 (2)	-0.1697 (16)	0.081 (5)*
H8A	0.6833 (14)	0.455 (2)	-0.0610 (14)	0.062 (5)*
H8B	0.7896 (14)	0.4913 (19)	0.0805 (15)	0.062 (5)*
H10	0.9607 (17)	0.633 (2)	0.1088 (19)	0.087 (6)*
H11	1.070 (2)	0.793 (3)	0.042 (2)	0.107 (7)*
H12	0.9743 (19)	0.887 (3)	-0.1597 (18)	0.099 (7)*
H13	0.7716 (18)	0.825 (3)	-0.283 (2)	0.098 (7)*
H14	0.6696 (16)	0.668 (2)	-0.2122 (15)	0.070 (5)*

*Atomic displacement parameters (Å<sup>2</sup>)*

	$U^{11}$	$U^{22}$	$U^{33}$	$U^{12}$	$U^{13}$	$U^{23}$
O1	0.0512 (6)	0.1059 (10)	0.0395 (6)	0.0150 (7)	0.0024 (5)	-0.0056 (6)
O2	0.0503 (6)	0.0798 (8)	0.0421 (6)	0.0005 (5)	0.0152 (5)	-0.0147 (5)
N1	0.0376 (6)	0.0549 (8)	0.0399 (7)	0.0001 (5)	0.0162 (5)	0.0025 (6)
C1	0.0369 (6)	0.0420 (8)	0.0378 (7)	-0.0052 (6)	0.0159 (5)	0.0030 (6)
C2	0.0465 (8)	0.0557 (10)	0.0433 (8)	-0.0040 (7)	0.0190 (7)	-0.0014 (7)
C3	0.0664 (10)	0.0594 (10)	0.0604 (11)	-0.0070 (8)	0.0370 (9)	-0.0106 (8)
C4	0.0586 (10)	0.0493 (10)	0.0853 (13)	0.0013 (8)	0.0438 (9)	0.0000 (9)
C5	0.0440 (8)	0.0553 (10)	0.0696 (11)	0.0052 (7)	0.0235 (8)	0.0109 (8)
C6	0.0408 (7)	0.0533 (9)	0.0436 (8)	-0.0018 (7)	0.0150 (6)	0.0048 (7)
C7	0.0379 (7)	0.0458 (8)	0.0374 (8)	-0.0052 (6)	0.0156 (5)	0.0021 (6)
C8	0.0454 (8)	0.0501 (9)	0.0583 (10)	0.0065 (7)	0.0256 (7)	0.0099 (8)
C9	0.0447 (7)	0.0410 (8)	0.0558 (9)	0.0018 (6)	0.0266 (6)	-0.0020 (7)
C10	0.0477 (9)	0.0840 (14)	0.0753 (13)	-0.0025 (9)	0.0209 (8)	0.0173 (11)
C11	0.0510 (11)	0.0993 (17)	0.1074 (18)	-0.0151 (11)	0.0333 (11)	0.0117 (13)
C12	0.0808 (13)	0.0764 (14)	0.1031 (17)	-0.0141 (11)	0.0639 (13)	-0.0013 (12)
C13	0.0835 (13)	0.0814 (14)	0.0652 (12)	-0.0113 (10)	0.0464 (10)	0.0005 (10)
C14	0.0579 (10)	0.0695 (11)	0.0538 (10)	-0.0107 (8)	0.0283 (8)	-0.0039 (8)

*Geometric parameters (Å, °)*

O1—C6	1.3506 (18)	C10—C11	1.385 (3)
O2—C7	1.2456 (17)	C11—C12	1.355 (3)
O1—HO1	0.98 (3)	C12—C13	1.360 (4)
N1—C7	1.3311 (19)	C13—C14	1.381 (3)
N1—C8	1.466 (2)	C2—H2	0.956 (19)
N1—HN1	0.882 (18)	C3—H3	0.954 (19)
C1—C7	1.483 (2)	C4—H4	0.94 (2)
C1—C6	1.401 (2)	C5—H5	0.99 (2)
C1—C2	1.387 (2)	C8—H8A	1.010 (16)

C2—C3	1.375 (3)	C8—H8B	0.995 (17)
C3—C4	1.379 (3)	C10—H10	0.94 (2)
C4—C5	1.361 (3)	C11—H11	0.95 (3)
C5—C6	1.389 (3)	C12—H12	0.97 (3)
C8—C9	1.511 (3)	C13—H13	0.99 (2)
C9—C14	1.370 (2)	C14—H14	0.94 (2)
C9—C10	1.375 (3)		
O1…O2	2.4886 (19)	C9…H4 <sup>iv</sup>	3.034 (17)
O1…N1 <sup>i</sup>	2.9191 (18)	C14…H4 <sup>iv</sup>	2.989 (17)
O2…C14	3.260 (3)	HN1…C2	2.554 (19)
O2…O1	2.4886 (19)	HN1…H2	2.00 (3)
O1…H2 <sup>i</sup>	2.58 (2)	HN1…O1 <sup>iii</sup>	2.083 (18)
O1…HN1 <sup>i</sup>	2.083 (18)	H2…N1	2.597 (18)
O2…HO1	1.56 (3)	H2…HN1	2.00 (3)
O2…H8A	2.416 (18)	H2…O1 <sup>iii</sup>	2.58 (2)
O2…H14	2.59 (2)	HO1…O2	1.56 (3)
O2…H5 <sup>ii</sup>	2.692 (19)	HO1…C7	2.16 (2)
N1…O1 <sup>iii</sup>	2.9191 (18)	HO1…H5 <sup>ii</sup>	2.50 (3)
N1…H2	2.597 (18)	H4…C9 <sup>iv</sup>	3.034 (17)
C1…C3 <sup>iv</sup>	3.554 (2)	H4…C14 <sup>iv</sup>	2.989 (17)
C1…C8 <sup>v</sup>	3.549 (2)	H5…O2 <sup>vi</sup>	2.692 (19)
C3…C1 <sup>iv</sup>	3.554 (2)	H5…HO1 <sup>vi</sup>	2.50 (3)
C6…C8 <sup>v</sup>	3.507 (2)	H8A…O2	2.416 (18)
C7…C7 <sup>v</sup>	3.400 (2)	H8A…H14	2.56 (2)
C7…C14	3.483 (3)	H8A…H13 <sup>vii</sup>	2.54 (3)
C8…C1 <sup>v</sup>	3.549 (2)	H8A…C1 <sup>v</sup>	3.075 (18)
C8…C6 <sup>v</sup>	3.507 (2)	H8A…C2 <sup>v</sup>	3.060 (18)
C14…C7	3.483 (3)	H8B…H10	2.32 (3)
C14…O2	3.260 (3)	H8B…C6 <sup>v</sup>	3.073 (17)
C1…H8A <sup>v</sup>	3.075 (18)	H10…H8B	2.32 (3)
C2…H8A <sup>v</sup>	3.060 (18)	H13…H8A <sup>viii</sup>	2.54 (3)
C2…HN1	2.554 (19)	H14…O2	2.59 (2)
C6…H8B <sup>v</sup>	3.073 (17)	H14…C7	3.09 (2)
C7…H14	3.09 (2)	H14…H8A	2.56 (2)
C7…HO1	2.16 (2)		
C6—O1—HO1	102.3 (13)	C9—C14—C13	121.17 (19)
C7—N1—C8	122.74 (13)	C1—C2—H2	118.7 (11)
C7—N1—HN1	119.3 (12)	C3—C2—H2	119.8 (11)
C8—N1—HN1	117.7 (13)	C2—C3—H3	118.8 (12)
C6—C1—C7	118.17 (12)	C4—C3—H3	121.6 (12)
C2—C1—C6	117.83 (15)	C3—C4—H4	120.2 (11)
C2—C1—C7	124.00 (14)	C5—C4—H4	119.2 (11)
C1—C2—C3	121.50 (16)	C4—C5—H5	122.9 (12)
C2—C3—C4	119.62 (17)	C6—C5—H5	116.9 (12)
C3—C4—C5	120.55 (19)	N1—C8—H8A	106.4 (11)
C4—C5—C6	120.15 (17)	N1—C8—H8B	105.6 (11)

C1—C6—C5	120.35 (14)	C9—C8—H8A	109.4 (10)
O1—C6—C1	121.46 (15)	C9—C8—H8B	108.8 (11)
O1—C6—C5	118.19 (15)	H8A—C8—H8B	114.1 (13)
O2—C7—C1	120.79 (13)	C9—C10—H10	117.9 (14)
O2—C7—N1	120.61 (14)	C11—C10—H10	121.6 (14)
N1—C7—C1	118.60 (12)	C10—C11—H11	118.5 (15)
N1—C8—C9	112.64 (13)	C12—C11—H11	121.0 (15)
C8—C9—C10	120.97 (16)	C11—C12—H12	123.9 (13)
C8—C9—C14	120.74 (16)	C13—C12—H12	116.2 (13)
C10—C9—C14	118.28 (18)	C12—C13—H13	122.5 (15)
C9—C10—C11	120.3 (2)	C14—C13—H13	117.6 (15)
C10—C11—C12	120.5 (2)	C9—C14—H14	120.7 (11)
C11—C12—C13	119.9 (2)	C13—C14—H14	118.2 (11)
C12—C13—C14	119.8 (2)		
C8—N1—C7—O2	1.6 (2)	C2—C3—C4—C5	-0.2 (3)
C8—N1—C7—C1	-178.95 (14)	C3—C4—C5—C6	0.2 (3)
C7—N1—C8—C9	89.06 (18)	C4—C5—C6—O1	179.22 (16)
C6—C1—C2—C3	0.4 (2)	C4—C5—C6—C1	0.1 (2)
C7—C1—C2—C3	-179.59 (16)	N1—C8—C9—C10	104.4 (2)
C2—C1—C6—O1	-179.48 (15)	N1—C8—C9—C14	-74.4 (2)
C2—C1—C6—C5	-0.4 (2)	C8—C9—C10—C11	-179.0 (2)
C7—C1—C6—O1	0.5 (2)	C14—C9—C10—C11	-0.2 (3)
C7—C1—C6—C5	179.56 (15)	C8—C9—C14—C13	178.77 (18)
C2—C1—C7—O2	175.27 (15)	C10—C9—C14—C13	-0.1 (3)
C2—C1—C7—N1	-4.2 (2)	C9—C10—C11—C12	0.1 (4)
C6—C1—C7—O2	-4.7 (2)	C10—C11—C12—C13	0.3 (4)
C6—C1—C7—N1	175.80 (14)	C11—C12—C13—C14	-0.6 (4)
C1—C2—C3—C4	-0.1 (3)	C12—C13—C14—C9	0.5 (3)

Symmetry codes: (i)  $x-1/2, -y+3/2, z-1/2$ ; (ii)  $-x+1/2, y-1/2, -z-1/2$ ; (iii)  $x+1/2, -y+3/2, z+1/2$ ; (iv)  $-x+1, -y+2, -z$ ; (v)  $-x+1, -y+1, -z$ ; (vi)  $-x+1/2, y+1/2, -z-1/2$ ; (vii)  $-x+3/2, y-1/2, -z-1/2$ ; (viii)  $-x+3/2, y+1/2, -z-1/2$ .

#### Hydrogen-bond geometry ( $\text{\AA}, ^\circ$ )

$D-H\cdots A$	$D-H$	$H\cdots A$	$D\cdots A$	$D-H\cdots A$
N1—HN1 $\cdots$ O1 <sup>iii</sup>	0.882 (18)	2.083 (18)	2.9191 (18)	158.1 (18)
O1—HO1 $\cdots$ O2	0.98 (3)	1.56 (3)	2.4886 (19)	157 (2)
C2—H2 $\cdots$ O1 <sup>iii</sup>	0.956 (19)	2.58 (2)	3.507 (2)	162.5 (15)
C14—H14 $\cdots$ O2	0.94 (2)	2.59 (2)	3.260 (3)	128.8 (14)

Symmetry code: (iii)  $x+1/2, -y+3/2, z+1/2$ .