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3-Methyl-5-(4-methylphenyl)cyclohex-2-enone

R. T. Sabapathy Mohan,^a S. Kamatchi,^a M. Subramanyam,^b
A. Thiruvalluvar^{b*} and A. Linden^c^aDepartment of Chemistry, Annamalai University, Annamalai Nagar 608 002, Tamil Nadu, India, ^bPG Research Department of Physics, Rajah Serfoji Government College (Autonomous), Thanjavur 613 005, Tamil Nadu, India, and ^cInstitute of Organic Chemistry, University of Zürich, Winterthurerstrasse 190, CH-8057 Zürich, Switzerland.

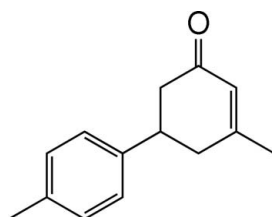
Correspondence e-mail: athiru@vsnl.net

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Key indicators: single-crystal X-ray study; $T = 160$ K; mean $\sigma(\text{C}-\text{C}) = 0.004$ Å; R factor = 0.073; wR factor = 0.231; data-to-parameter ratio = 14.6.

In the title molecule, $\text{C}_{14}\text{H}_{16}\text{O}$, the cyclohexene ring adopts an envelope conformation, with all substituents equatorial. Molecules are linked by $\text{C}-\text{H}\cdots\text{O}$ hydrogen bonds. A $\text{C}-\text{H}\cdots\pi$ interaction involving the benzene ring is also found in the crystal structure. The H atoms of both methyl groups are disordered equally over two positions.

Related literature

For related literature, see: Padmavathi *et al.* (2000).

Experimental

Crystal data

 $\text{C}_{14}\text{H}_{16}\text{O}$
 $M_r = 200.27$
Monoclinic, $P2_1/n$ $a = 5.2623$ (3) Å
 $b = 11.1583$ (7) Å
 $c = 19.3341$ (11) Å $\beta = 94.994$ (4)°
 $V = 1130.96$ (12) Å³
 $Z = 4$
Mo $K\alpha$ radiation $\mu = 0.07$ mm⁻¹
 $T = 160$ (1) K
 $0.25 \times 0.18 \times 0.18$ mm

Data collection

Nonius KappaCCD area-detector
diffractometer
Absorption correction: none
16716 measured reflections2002 independent reflections
1316 reflections with $I > 2\sigma(I)$
 $R_{\text{int}} = 0.085$

Refinement

 $R[F^2 > 2\sigma(F^2)] = 0.072$
 $wR(F^2) = 0.231$
 $S = 1.09$
2002 reflections137 parameters
H-atom parameters constrained
 $\Delta\rho_{\text{max}} = 0.48$ e Å⁻³
 $\Delta\rho_{\text{min}} = -0.30$ e Å⁻³

Table 1

Hydrogen-bond geometry (Å, °).

| $D-\text{H}\cdots A$ | $D-\text{H}$ | $\text{H}\cdots A$ | $D\cdots A$ | $D-\text{H}\cdots A$ |
|--|--------------|--------------------|-------------|----------------------|
| $\text{C}2-\text{H}2\cdots\text{O}1^{\text{i}}$ | 0.95 | 2.48 | 3.425 (3) | 173 |
| $\text{C}5-\text{H}5\cdots\text{C}g^{\text{ii}}$ | 1.00 | 2.94 | 3.818 (3) | 147 |

Symmetry codes: (i) $-x + 2, -y + 1, -z + 1$; (ii) $x + 1, y, z$. $\text{C}g$ is the centroid of the benzene ring.

Data collection: *COLLECT* (Nonius, 2000); cell refinement: *DENZO-SMN* (Otwinowski & Minor, 1997); data reduction: *DENZO-SMN* and *SCALEPACK* (Otwinowski & Minor, 1997); program(s) used to solve structure: *SHELXS97* (Sheldrick, 2008); program(s) used to refine structure: *SHELXL97* (Sheldrick, 2008); molecular graphics: *ORTEP-3* (Farrugia, 1997); software used to prepare material for publication: *PLATON* (Spek, 2003).

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Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: WN2260).

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supporting information

Acta Cryst. (2008). E64, o1066 [doi:10.1107/S1600536808013822]

3-Methyl-5-(4-methylphenyl)cyclohex-2-enone

R. T. Sabapathy Mohan, S. Kamatchi, M. Subramanyam, A. Thiruvalluvar and A. Linden

S1. Comment

The title compound, has been analysed as part of our crystallographic studies on substituted cyclohexenes. The molecular structure of the title compound, with atomic numbering scheme, is shown in Fig. 1. The cyclohexene ring adopts an envelope conformation, with all substituents equatorial. Molecules are linked by C2—H2 \cdots O1(2 - x, 1 - y, 1 - z) hydrogen bonds (Fig. 2). A C—H \cdots π interaction involving the benzene ring is also found in the crystal structure.

S2. Experimental

The title compound was prepared according to the general procedure reported by Padmavathi *et al.* (2000). A mixture of 2,4-bis(ethoxycarbonyl)-5-hydroxy-5-methyl-3,4'-methylphenylcyclohexanone (3.62 g, 0.01 mol) in glacial acetic acid (25 ml) and concentrated hydrochloric acid (50 ml) was refluxed for 12 h. After completion of the reaction, the reaction mixture was neutralized with aqueous ammonia and separated using chloroform. The product was purified by column chromatography (benzene-EtOAc, 9.5:0.5 v/v). The yield of the isolated product was 1.07 g (87%).

S3. Refinement

H atoms were positioned geometrically and allowed to ride on their parent atoms, with C—H = 0.95–1.00 Å and $U_{\text{iso}}(\text{H}) = xU_{\text{eq}}(\text{carrier atom})$, where $x = 1.5$ for methyl and 1.2 for all other C atoms. The H atoms of both methyl groups were found to be disordered equally over two positions rotated from each other by 60°. They were refined as idealized.

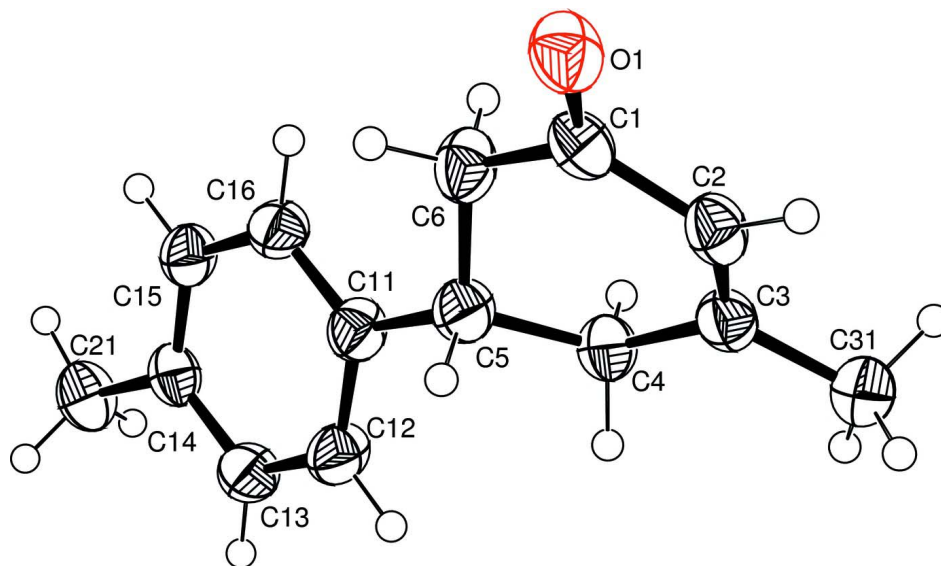


Figure 1

The molecular structure of the title compound, showing the atom-numbering scheme and displacement ellipsoids drawn at the 50% probability level. Hydrogen atoms are represented by spheres of arbitrary radius.

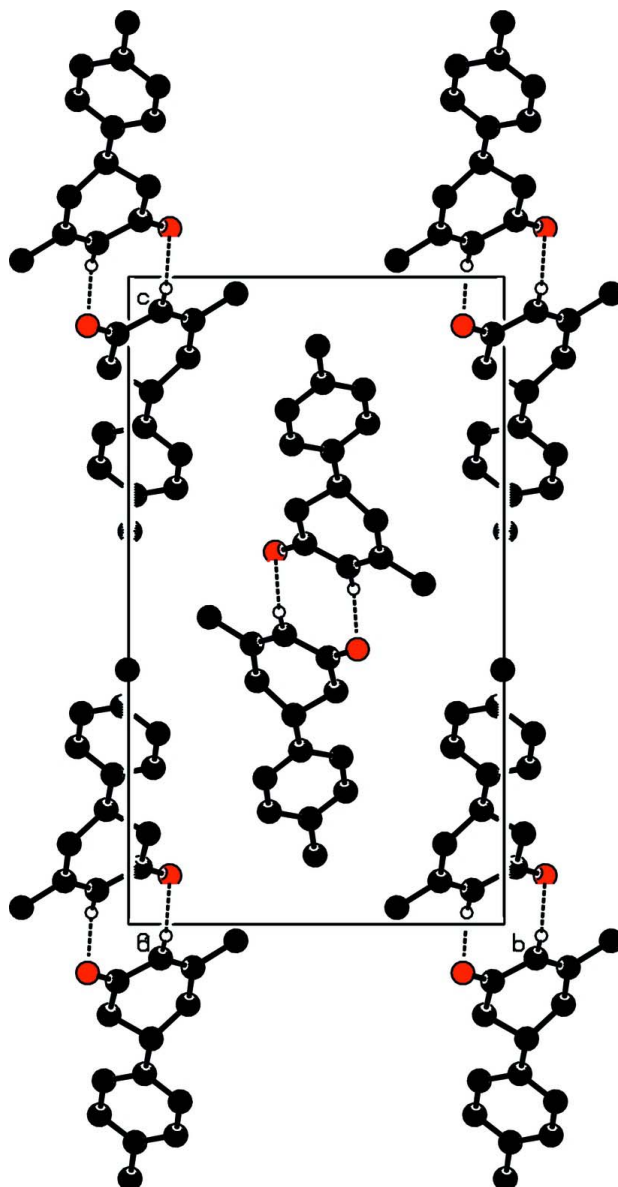


Figure 2

The molecular packing of the title compound, viewed down the *a* axis showing the C—H...O (dashed lines) interactions.

3-Methyl-5-(4-methylphenyl)cyclohex-2-enone

Crystal data

$C_{14}H_{16}O$

$M_r = 200.27$

Monoclinic, $P2_1/n$

Hall symbol: $-P 2_1n$

$a = 5.2623 (3) \text{ \AA}$

$b = 11.1583 (7) \text{ \AA}$

$c = 19.3341 (11) \text{ \AA}$

$\beta = 94.994 (4)^\circ$

$V = 1130.96 (12) \text{ \AA}^3$

$Z = 4$

$F(000) = 432$

$D_x = 1.176 \text{ Mg m}^{-3}$

Melting point: 315 K

Mo $K\alpha$ radiation, $\lambda = 0.71073 \text{ \AA}$

Cell parameters from 2105 reflections

$\theta = 2.0\text{--}25.0^\circ$

$\mu = 0.07 \text{ mm}^{-1}$

$T = 160 \text{ K}$

Prism, colourless

$0.25 \times 0.18 \times 0.18 \text{ mm}$

Data collection

| | |
|--|---|
| Nonius KappaCCD area-detector diffractometer | 16716 measured reflections 2002 independent reflections |
| Radiation source: Nonius FR590 sealed tube generator | 1316 reflections with $I > 2\sigma(I)$ $R_{\text{int}} = 0.085$ |
| Horizontally mounted graphite crystal monochromator | $\theta_{\text{max}} = 25.0^\circ$, $\theta_{\text{min}} = 3.7^\circ$ $h = 0 \rightarrow 6$ |
| Detector resolution: 9 pixels mm^{-1} ω scans with κ offsets | $k = 0 \rightarrow 13$ $l = -22 \rightarrow 22$ |

Refinement

| | |
|--|---|
| Refinement on F^2 | Hydrogen site location: inferred from neighbouring sites |
| Least-squares matrix: full | H-atom parameters constrained |
| $R[F^2 > 2\sigma(F^2)] = 0.072$ | $w = 1/[\sigma^2(F_o^2) + (0.1423P)^2]$ |
| $wR(F^2) = 0.231$ | where $P = (F_o^2 + 2F_c^2)/3$ |
| $S = 1.09$ | $(\Delta/\sigma)_{\text{max}} < 0.001$ |
| 2002 reflections | $\Delta\rho_{\text{max}} = 0.48 \text{ e } \text{\AA}^{-3}$ |
| 137 parameters | $\Delta\rho_{\text{min}} = -0.30 \text{ e } \text{\AA}^{-3}$ |
| 0 restraints | Extinction correction: <i>SHELXL97</i> (Sheldrick, 2008), $F_c^* = kFc[1 + 0.001xFc^2\lambda^3/\sin(2\theta)]^{-1/4}$ |
| Primary atom site location: structure-invariant direct methods | Extinction coefficient: 0.16 (2) |
| Secondary atom site location: difference Fourier map | |

Special details

Experimental. Solvent used: ? Cooling Device: Oxford Cryosystems Cryostream 700 Crystal mount: glued on a glass fibre Mosaicity ($^\circ$): 0.728 (3) Frames collected: 237 Seconds exposure per frame: 18 Degrees rotation per frame: 1.8 Crystal-Detector distance (mm): 30.0

Geometry. Bond distances, angles *etc.* have been calculated using the rounded fractional coordinates. All su's are estimated from the variances of the (full) variance-covariance matrix. The cell e.s.d.'s are taken into account in the estimation of distances, angles and torsion angles

Refinement. Refinement of F^2 against ALL reflections. The weighted R -factor wR and goodness of fit S are based on F^2 , conventional R -factors R are based on F , with F set to zero for negative F^2 . The threshold expression of $F^2 > \sigma(F^2)$ is used only for calculating R -factors(gt) *etc.* and is not relevant to the choice of reflections for refinement. R -factors based on F^2 are statistically about twice as large as those based on F , and R -factors based on ALL data will be even larger.

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (\AA^2)

| | x | y | z | $U_{\text{iso}}^*/U_{\text{eq}}$ | Occ. (<1) |
|-----|-------------|--------------|--------------|----------------------------------|-----------|
| O1 | 0.8289 (3) | 0.60964 (17) | 0.42477 (10) | 0.0480 (7) | |
| C1 | 0.6687 (5) | 0.5310 (2) | 0.41187 (13) | 0.0380 (9) | |
| C2 | 0.6807 (5) | 0.4150 (3) | 0.44740 (14) | 0.0392 (9) | |
| C3 | 0.5102 (5) | 0.3288 (2) | 0.43313 (13) | 0.0373 (9) | |
| C4 | 0.2974 (5) | 0.3417 (2) | 0.37608 (13) | 0.0397 (9) | |
| C5 | 0.3519 (5) | 0.4399 (2) | 0.32304 (14) | 0.0410 (9) | |
| C6 | 0.4432 (5) | 0.5523 (3) | 0.36013 (15) | 0.0446 (10) | |
| C11 | 0.1274 (5) | 0.4579 (2) | 0.26887 (14) | 0.0399 (9) | |
| C12 | 0.0467 (5) | 0.3630 (3) | 0.22587 (15) | 0.0449 (10) | |
| C13 | -0.1517 (5) | 0.3746 (3) | 0.17411 (15) | 0.0437 (10) | |
| C14 | -0.2758 (5) | 0.4829 (2) | 0.16303 (13) | 0.0386 (9) | |
| C15 | -0.1973 (5) | 0.5781 (2) | 0.20557 (14) | 0.0412 (9) | |
| C16 | -0.0006 (5) | 0.5656 (2) | 0.25810 (14) | 0.0411 (9) | |

| | | | | | |
|------|-------------|------------|--------------|-------------|-------|
| C21 | -0.4928 (5) | 0.4965 (3) | 0.10693 (15) | 0.0505 (10) | |
| C31 | 0.5112 (6) | 0.2145 (2) | 0.47430 (15) | 0.0496 (10) | |
| H2 | 0.81624 | 0.40075 | 0.48216 | 0.0471* | |
| H4A | 0.27208 | 0.26420 | 0.35156 | 0.0476* | |
| H4B | 0.13730 | 0.36148 | 0.39696 | 0.0476* | |
| H5 | 0.49701 | 0.41021 | 0.29761 | 0.0492* | |
| H6A | 0.49029 | 0.61187 | 0.32552 | 0.0536* | |
| H6B | 0.30195 | 0.58644 | 0.38449 | 0.0536* | |
| H12 | 0.13007 | 0.28779 | 0.23210 | 0.0539* | |
| H13 | -0.20263 | 0.30744 | 0.14604 | 0.0525* | |
| H15 | -0.27932 | 0.65353 | 0.19880 | 0.0494* | |
| H16 | 0.04671 | 0.63215 | 0.28705 | 0.0493* | |
| H21A | -0.55656 | 0.57899 | 0.10675 | 0.0758* | 0.500 |
| H21B | -0.43159 | 0.47801 | 0.06168 | 0.0758* | 0.500 |
| H21C | -0.63066 | 0.44117 | 0.11600 | 0.0758* | 0.500 |
| H21D | -0.52265 | 0.41979 | 0.08287 | 0.0758* | 0.500 |
| H21E | -0.64762 | 0.52077 | 0.12794 | 0.0758* | 0.500 |
| H21F | -0.44855 | 0.55761 | 0.07362 | 0.0758* | 0.500 |
| H31A | 0.37072 | 0.16309 | 0.45556 | 0.0744* | 0.500 |
| H31B | 0.67370 | 0.17279 | 0.47117 | 0.0744* | 0.500 |
| H31C | 0.48990 | 0.23312 | 0.52299 | 0.0744* | 0.500 |
| H31D | 0.65216 | 0.21624 | 0.51092 | 0.0744* | 0.500 |
| H31E | 0.34918 | 0.20654 | 0.49531 | 0.0744* | 0.500 |
| H31F | 0.53298 | 0.14621 | 0.44349 | 0.0744* | 0.500 |

Atomic displacement parameters (Å²)

| | U^{11} | U^{22} | U^{33} | U^{12} | U^{13} | U^{23} |
|-----|-------------|-------------|-------------|--------------|--------------|--------------|
| O1 | 0.0439 (12) | 0.0548 (13) | 0.0434 (13) | -0.0079 (10) | -0.0074 (9) | 0.0016 (9) |
| C1 | 0.0331 (15) | 0.0480 (17) | 0.0324 (15) | -0.0016 (13) | 0.0006 (12) | -0.0033 (12) |
| C2 | 0.0340 (14) | 0.0505 (17) | 0.0324 (15) | 0.0048 (13) | -0.0014 (12) | 0.0012 (12) |
| C3 | 0.0368 (15) | 0.0427 (16) | 0.0323 (15) | 0.0072 (12) | 0.0025 (12) | -0.0009 (12) |
| C4 | 0.0379 (16) | 0.0433 (17) | 0.0370 (16) | 0.0007 (12) | -0.0012 (12) | 0.0020 (12) |
| C5 | 0.0421 (16) | 0.0400 (16) | 0.0393 (16) | 0.0006 (12) | -0.0052 (13) | 0.0000 (12) |
| C6 | 0.0429 (16) | 0.0458 (17) | 0.0432 (17) | -0.0035 (13) | -0.0076 (13) | 0.0081 (13) |
| C11 | 0.0393 (16) | 0.0390 (16) | 0.0399 (16) | -0.0028 (12) | -0.0043 (12) | 0.0024 (12) |
| C12 | 0.0485 (17) | 0.0373 (16) | 0.0463 (18) | 0.0028 (12) | -0.0114 (14) | -0.0005 (12) |
| C13 | 0.0471 (17) | 0.0418 (17) | 0.0406 (17) | -0.0031 (13) | -0.0058 (13) | -0.0043 (12) |
| C14 | 0.0366 (15) | 0.0468 (17) | 0.0315 (15) | -0.0018 (12) | -0.0024 (11) | 0.0029 (12) |
| C15 | 0.0403 (16) | 0.0424 (16) | 0.0395 (17) | 0.0052 (12) | -0.0049 (13) | 0.0046 (12) |
| C16 | 0.0449 (16) | 0.0360 (16) | 0.0409 (17) | -0.0035 (12) | -0.0042 (13) | -0.0015 (12) |
| C21 | 0.0443 (17) | 0.065 (2) | 0.0398 (17) | 0.0002 (15) | -0.0103 (13) | 0.0021 (14) |
| C31 | 0.0583 (19) | 0.0449 (17) | 0.0441 (17) | 0.0061 (14) | -0.0038 (14) | 0.0024 (13) |

Geometric parameters (Å, °)

| | | | |
|-------|-----------|--------|--------|
| O1—C1 | 1.227 (3) | C5—H5 | 1.0000 |
| C1—C2 | 1.464 (4) | C6—H6A | 0.9900 |

| | | | |
|-------------------------|-----------|-------------------------|--------|
| C1—C6 | 1.503 (4) | C6—H6B | 0.9900 |
| C2—C3 | 1.328 (4) | C12—H12 | 0.9500 |
| C3—C4 | 1.509 (4) | C13—H13 | 0.9500 |
| C3—C31 | 1.503 (3) | C15—H15 | 0.9500 |
| C4—C5 | 1.545 (3) | C16—H16 | 0.9500 |
| C5—C6 | 1.503 (4) | C21—H21A | 0.9800 |
| C5—C11 | 1.522 (4) | C21—H21B | 0.9800 |
| C11—C12 | 1.390 (4) | C21—H21C | 0.9800 |
| C11—C16 | 1.385 (3) | C21—H21D | 0.9800 |
| C12—C13 | 1.388 (4) | C21—H21E | 0.9800 |
| C13—C14 | 1.381 (4) | C21—H21F | 0.9800 |
| C14—C15 | 1.384 (3) | C31—H31A | 0.9800 |
| C14—C21 | 1.513 (4) | C31—H31B | 0.9800 |
| C15—C16 | 1.393 (4) | C31—H31C | 0.9800 |
| C2—H2 | 0.9500 | C31—H31D | 0.9800 |
| C4—H4A | 0.9900 | C31—H31E | 0.9800 |
| C4—H4B | 0.9900 | C31—H31F | 0.9800 |
| O1…C31 ⁱ | 3.387 (3) | H4B…H2 ^v | 2.5000 |
| O1…H6B ⁱⁱ | 2.6800 | H5…C2 | 2.9700 |
| O1…H13 ⁱⁱⁱ | 2.6500 | H5…C14 ⁱⁱ | 3.0700 |
| O1…H2 ^{iv} | 2.4800 | H6A…C15 ⁱⁱ | 2.9800 |
| O1…H31C ⁱ | 2.6800 | H6A…C16 | 2.8400 |
| O1…H31E ⁱ | 2.7800 | H6A…H16 | 2.4000 |
| C1…C16 ⁱⁱ | 3.594 (4) | H6A…C12 ⁱⁱⁱ | 2.9700 |
| C1…C2 ⁱ | 3.467 (4) | H6A…C13 ⁱⁱⁱ | 3.0500 |
| C1…C3 ⁱ | 3.579 (4) | H6A…H12 ⁱⁱⁱ | 2.3200 |
| C2…C1 ⁱ | 3.467 (4) | H6A…H13 ⁱⁱⁱ | 2.4900 |
| C2…C2 ⁱ | 3.468 (4) | H6B…O1 ^v | 2.6800 |
| C3…C1 ⁱ | 3.579 (4) | H6B…C16 | 2.8100 |
| C16…C1 ^v | 3.594 (4) | H6B…H16 | 2.2700 |
| C31…O1 ⁱ | 3.387 (3) | H12…C4 | 2.9100 |
| C1…H31C ⁱ | 3.0600 | H12…H4A | 2.3800 |
| C2…H5 | 2.9700 | H12…H6A ^{vi} | 2.3200 |
| C2…H4B ⁱⁱ | 2.7400 | H12…H16 ^{vi} | 2.4800 |
| C4…H12 | 2.9100 | H13…H21D | 2.3500 |
| C6…H16 | 2.5700 | H13…O1 ^{vi} | 2.6500 |
| C12…H21C ⁱⁱ | 2.9600 | H13…H6A ^{vi} | 2.4900 |
| C12…H4A | 2.8300 | H15…H21A | 2.3500 |
| C12…H6A ^{vi} | 2.9700 | H16…C6 | 2.5700 |
| C13…H6A ^{vi} | 3.0500 | H16…H6A | 2.4000 |
| C14…H5 ^v | 3.0700 | H16…H6B | 2.2700 |
| C15…H6A ^v | 2.9800 | H16…H12 ⁱⁱⁱ | 2.4800 |
| C16…H6A | 2.8400 | H21A…H15 | 2.3500 |
| C16…H6B | 2.8100 | H21A…H4A ^{vii} | 2.5200 |
| C21…H31B ⁱⁱⁱ | 3.0700 | H21C…C12 ^v | 2.9600 |
| C21…H31A ^{vii} | 2.9100 | H21D…H13 | 2.3500 |
| C31…H21F ^{vi} | 3.1000 | H21F…C31 ⁱⁱⁱ | 3.1000 |

| | | | |
|----------------------------|-----------|----------------------------|--------|
| H2...H4B ⁱⁱ | 2.5000 | H31A...H4A | 2.3200 |
| H2...H31D | 2.3200 | H31A...C21 ^{viii} | 2.9100 |
| H2...O1 ^{iv} | 2.4800 | H31B...C21 ^{vi} | 3.0700 |
| H4A...C12 | 2.8300 | H31C...O1 ⁱ | 2.6800 |
| H4A...H12 | 2.3800 | H31C...C1 ⁱ | 3.0600 |
| H4A...H31A | 2.3200 | H31D...H2 | 2.3200 |
| H4A...H31F | 2.5200 | H31E...O1 ⁱ | 2.7800 |
| H4A...H21A ^{viii} | 2.5200 | H31F...H4A | 2.5200 |
| H4B...C2 ^v | 2.7400 | | |
| O1—C1—C2 | 122.4 (2) | C15—C16—H16 | 119.00 |
| O1—C1—C6 | 120.7 (2) | C14—C21—H21A | 109.00 |
| C2—C1—C6 | 116.8 (2) | C14—C21—H21B | 109.00 |
| C1—C2—C3 | 122.8 (2) | C14—C21—H21C | 109.00 |
| C2—C3—C4 | 121.9 (2) | C14—C21—H21D | 109.00 |
| C2—C3—C31 | 122.2 (2) | C14—C21—H21E | 109.00 |
| C4—C3—C31 | 115.9 (2) | C14—C21—H21F | 109.00 |
| C3—C4—C5 | 112.6 (2) | H21A—C21—H21B | 109.00 |
| C4—C5—C6 | 110.2 (2) | H21A—C21—H21C | 110.00 |
| C4—C5—C11 | 111.9 (2) | H21A—C21—H21D | 141.00 |
| C6—C5—C11 | 114.5 (2) | H21A—C21—H21E | 56.00 |
| C1—C6—C5 | 112.8 (2) | H21A—C21—H21F | 56.00 |
| C5—C11—C12 | 119.3 (2) | H21B—C21—H21C | 109.00 |
| C5—C11—C16 | 123.8 (2) | H21B—C21—H21D | 56.00 |
| C12—C11—C16 | 116.9 (2) | H21B—C21—H21E | 141.00 |
| C11—C12—C13 | 122.1 (3) | H21B—C21—H21F | 56.00 |
| C12—C13—C14 | 120.6 (3) | H21C—C21—H21D | 56.00 |
| C13—C14—C15 | 117.8 (2) | H21C—C21—H21E | 56.00 |
| C13—C14—C21 | 121.1 (2) | H21C—C21—H21F | 141.00 |
| C15—C14—C21 | 121.1 (2) | H21D—C21—H21E | 109.00 |
| C14—C15—C16 | 121.4 (2) | H21D—C21—H21F | 109.00 |
| C11—C16—C15 | 121.2 (2) | H21E—C21—H21F | 110.00 |
| C1—C2—H2 | 119.00 | C3—C31—H31A | 109.00 |
| C3—C2—H2 | 119.00 | C3—C31—H31B | 109.00 |
| C3—C4—H4A | 109.00 | C3—C31—H31C | 109.00 |
| C3—C4—H4B | 109.00 | C3—C31—H31D | 109.00 |
| C5—C4—H4A | 109.00 | C3—C31—H31E | 109.00 |
| C5—C4—H4B | 109.00 | C3—C31—H31F | 109.00 |
| H4A—C4—H4B | 108.00 | H31A—C31—H31B | 109.00 |
| C4—C5—H5 | 107.00 | H31A—C31—H31C | 109.00 |
| C6—C5—H5 | 107.00 | H31A—C31—H31D | 141.00 |
| C11—C5—H5 | 107.00 | H31A—C31—H31E | 56.00 |
| C1—C6—H6A | 109.00 | H31A—C31—H31F | 56.00 |
| C1—C6—H6B | 109.00 | H31B—C31—H31C | 109.00 |
| C5—C6—H6A | 109.00 | H31B—C31—H31D | 56.00 |
| C5—C6—H6B | 109.00 | H31B—C31—H31E | 141.00 |
| H6A—C6—H6B | 108.00 | H31B—C31—H31F | 56.00 |
| C11—C12—H12 | 119.00 | H31C—C31—H31D | 56.00 |

| | | | |
|---------------|------------|-----------------|------------|
| C13—C12—H12 | 119.00 | H31C—C31—H31E | 56.00 |
| C12—C13—H13 | 120.00 | H31C—C31—H31F | 141.00 |
| C14—C13—H13 | 120.00 | H31D—C31—H31E | 109.00 |
| C14—C15—H15 | 119.00 | H31D—C31—H31F | 109.00 |
| C16—C15—H15 | 119.00 | H31E—C31—H31F | 109.00 |
| C11—C16—H16 | 119.00 | | |
| O1—C1—C2—C3 | -179.1 (3) | C4—C5—C11—C16 | 120.6 (3) |
| C6—C1—C2—C3 | 3.3 (4) | C6—C5—C11—C12 | 172.5 (2) |
| O1—C1—C6—C5 | 149.3 (2) | C6—C5—C11—C16 | -5.7 (4) |
| C2—C1—C6—C5 | -33.1 (3) | C5—C11—C12—C13 | -178.0 (3) |
| C1—C2—C3—C4 | 3.3 (4) | C16—C11—C12—C13 | 0.4 (4) |
| C1—C2—C3—C31 | -174.4 (2) | C5—C11—C16—C15 | 177.0 (2) |
| C2—C3—C4—C5 | 19.3 (3) | C12—C11—C16—C15 | -1.3 (4) |
| C31—C3—C4—C5 | -162.8 (2) | C11—C12—C13—C14 | 0.6 (4) |
| C3—C4—C5—C6 | -47.3 (3) | C12—C13—C14—C15 | -0.7 (4) |
| C3—C4—C5—C11 | -175.9 (2) | C12—C13—C14—C21 | -179.8 (3) |
| C4—C5—C6—C1 | 54.2 (3) | C13—C14—C15—C16 | -0.3 (4) |
| C11—C5—C6—C1 | -178.6 (2) | C21—C14—C15—C16 | 178.9 (2) |
| C4—C5—C11—C12 | -61.2 (3) | C14—C15—C16—C11 | 1.3 (4) |

Symmetry codes: (i) $-x+1, -y+1, -z+1$; (ii) $x+1, y, z$; (iii) $-x+1/2, y+1/2, -z+1/2$; (iv) $-x+2, -y+1, -z+1$; (v) $x-1, y, z$; (vi) $-x+1/2, y-1/2, -z+1/2$; (vii) $-x-1/2, y+1/2, -z+1/2$; (viii) $-x-1/2, y-1/2, -z+1/2$.

Hydrogen-bond geometry (Å, °)

| <i>D</i> —H \cdots <i>A</i> | <i>D</i> —H | H \cdots <i>A</i> | <i>D</i> \cdots <i>A</i> | <i>D</i> —H \cdots <i>A</i> |
|---------------------------------|-------------|---------------------|----------------------------|-------------------------------|
| C2—H2 \cdots O1 ^{iv} | 0.95 | 2.48 | 3.425 (3) | 173 |
| C5—H5 \cdots Cg ⁱⁱ | 1.00 | 2.94 | 3.818 (3) | 147 |

Symmetry codes: (ii) $x+1, y, z$; (iv) $-x+2, -y+1, -z+1$.