

# Methyl 3-hydroxy-4-oxo-3,4-dihydro-2H-1,2-benzothiazine-3-carboxylate 1,1-dioxide monohydrate

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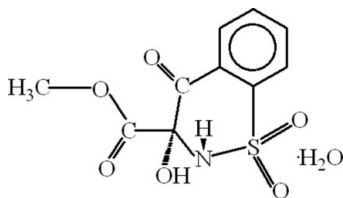
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Key indicators: single-crystal X-ray study;  $T = 296$  K; mean  $\sigma(\text{C}-\text{C}) = 0.004$  Å;  $R$  factor = 0.047;  $wR$  factor = 0.122; data-to-parameter ratio = 16.3.

In the molecule of the title compound,  $\text{C}_{10}\text{H}_9\text{NO}_6\text{S}\cdot\text{H}_2\text{O}$ , the benzothiazine ring adopts an envelope conformation. An intramolecular  $\text{N}-\text{H}\cdots\text{O}$  hydrogen bond results in the formation of a nonplanar five-membered ring which has a twisted conformation. In the crystal structure, intermolecular  $\text{N}-\text{H}\cdots\text{O}$ ,  $\text{O}-\text{H}\cdots\text{O}$  and  $\text{C}-\text{H}\cdots\text{O}$  hydrogen bonds link the molecules to form a three-dimensional network. There is a  $\pi-\pi$  contact between the benzene rings [centroid-centroid distance = 3.972 (2) Å].

## Related literature

For general background, see: Shafiq, Khan *et al.* (2008); Shafiq, Tahir *et al.* (2008); Tahir *et al.* (2008). For related literature, see: Antsyshkina *et al.* (2003); Allen (2002). For bond-length data, see: Allen *et al.* (1987). For ring puckering parameters, see: Cremer & Pople (1975).



## Experimental

### Crystal data

$\text{C}_{10}\text{H}_9\text{NO}_6\text{S}\cdot\text{H}_2\text{O}$

$M_r = 289.26$

Orthorhombic,  $Pbca$

$a = 7.7504$  (5) Å

$b = 14.5638$  (9) Å

$c = 21.0615$  (14) Å

$V = 2377.3$  (3) Å<sup>3</sup>

$Z = 8$

Mo  $K\alpha$  radiation

$\mu = 0.30$  mm<sup>-1</sup>

$T = 296$  (2) K

$0.24 \times 0.18 \times 0.15$  mm

### Data collection

Bruker Kappa APEXII CCD diffractometer

Absorption correction: multi-scan (SADABS; Bruker, 2005)

$T_{\min} = 0.934$ ,  $T_{\max} = 0.958$

14889 measured reflections

2998 independent reflections

1895 reflections with  $I > 2\sigma(I)$

$R_{\text{int}} = 0.065$

### Refinement

$R[F^2 > 2\sigma(F^2)] = 0.047$

$wR(F^2) = 0.122$

$S = 1.01$

2998 reflections

184 parameters

H atoms treated by a mixture of independent and constrained refinement

$\Delta\rho_{\text{max}} = 0.40$  e Å<sup>-3</sup>

$\Delta\rho_{\text{min}} = -0.32$  e Å<sup>-3</sup>

**Table 1**

Hydrogen-bond geometry (Å, °).

$D-H\cdots A$	$D-H$	$H\cdots A$	$D\cdots A$	$D-H\cdots A$
$\text{N1}-\text{H1}\cdots\text{O6}$	0.78 (3)	2.43 (3)	2.744 (3)	106 (2)
$\text{N1}-\text{H1}\cdots\text{O7}^i$	0.78 (3)	2.29 (3)	3.032 (3)	162 (3)
$\text{O4}-\text{H4O}\cdots\text{O7}^{ii}$	0.84 (3)	1.94 (3)	2.773 (3)	175 (2)
$\text{O7}-\text{H71}\cdots\text{O3}^{iii}$	0.83 (3)	2.45 (3)	3.107 (3)	137 (3)
$\text{O7}-\text{H71}\cdots\text{O5}^{iv}$	0.83 (3)	2.45 (3)	3.028 (3)	128 (3)
$\text{O7}-\text{H72}\cdots\text{O2}$	0.83 (4)	2.21 (4)	3.027 (3)	167 (3)
$\text{C5}-\text{H5}\cdots\text{O4}^{iv}$	0.9300	2.4800	3.374 (3)	162.00
$\text{C10}-\text{H10A}\cdots\text{O4}^i$	0.9600	2.3100	2.994 (3)	128.00

Symmetry codes: (i)  $x-1, y, z$ ; (ii)  $x-\frac{1}{2}, -y+\frac{1}{2}, -z$ ; (iii)  $-x+1, -y, -z$ ; (iv)  $-x+\frac{1}{2}, y-\frac{1}{2}, z$ .

Data collection: APEX2 (Bruker, 2007); cell refinement: APEX2; data reduction: SAINT (Bruker, 2007); program(s) used to solve structure: SHELXS97 (Sheldrick, 2008); program(s) used to refine structure: SHELXL97 (Sheldrick, 2008); molecular graphics: ORTEP-3 for Windows (Farrugia, 1997) and PLATON (Spek, 2003); software used to prepare material for publication: WinGX (Farrugia, 1999) and PLATON.

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Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: HK2537).

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## supporting information

*Acta Cryst.* (2008). E64, o2045 [doi:10.1107/S1600536808030948]

## Methyl 3-hydroxy-4-oxo-3,4-dihydro-2H-1,2-benzothiazine-3-carboxylate 1,1-dioxide monohydrate

Muhammad Nadeem Arshad, M. Nawaz Tahir, Islam Ullah Khan, Muhammad Shafiq and Waseeq Ahmad Siddiqui

### S1. Comment

The title compound has been prepared in continuation of research on benzo-thiazine derivatives (Shafiq, Khan *et al.*, 2008; Shafiq, Tahir *et al.*, 2008; Tahir *et al.*, 2008) by our research group. The CCDC search (Allen, 2002) shows that a single crystal structure has been reported, in which the same benzothiazine ring exists (Antsyshkina *et al.*, 2003). The title compound differs from the reported structure, due to the hydroxy and methylformate groups. Due to the hydroxy group, the S-configuration in the title compound has been confirmed.

In the molecule of the title compound (Fig. 1), the bond lengths (Allen *et al.*, 1987) and angles are within normal ranges. Ring A (C1-C6) is, of course, planar. Ring B (S1/N1/C1/C6-C8) is not planar, having total puckering amplitude,  $Q_T$ , of 0.733 (3) Å and envelope conformation [ $\varphi = 21.21$  (3)° and  $\theta = 76.70$  (3)°] (Cremer & Pople, 1975) with N1 atom displaced by 0.575 (3) Å from the plane of the other ring atoms. The intramolecular N-H...O hydrogen bond (Table 1) results in the formation of a nonplanar five-membered ring C (N1/O6/C8/C9/H1), having twisted conformation.

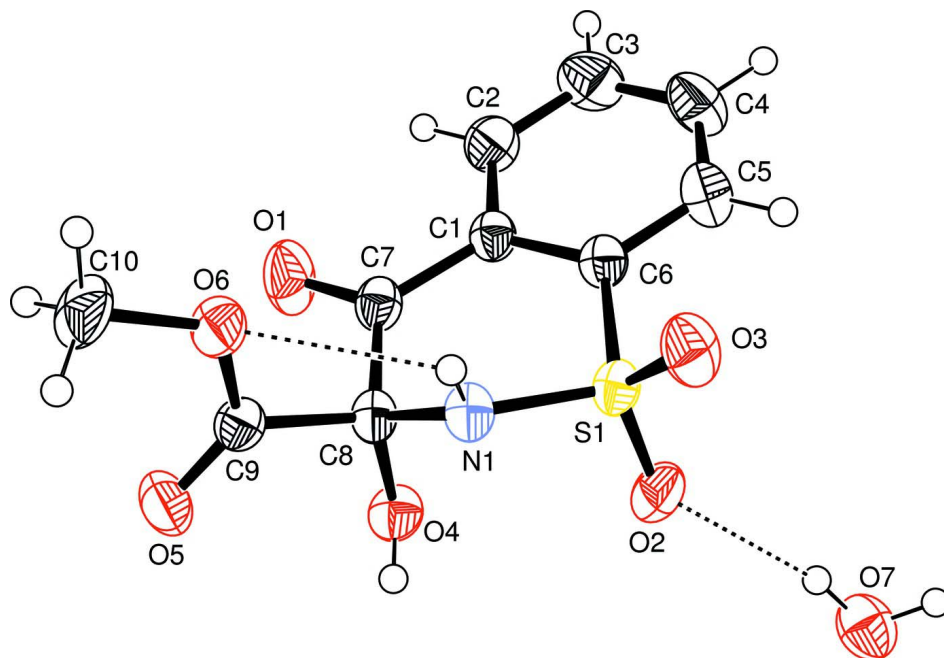
In the crystal structure, intermolecular N-H...O, O-H...O and C-H...O hydrogen bonds (Table 1) link the molecules to form a three dimensional network (Fig. 2), in which they may be effective in the stabilization of the structure. The  $\pi$ — $\pi$  contact between the benzene rings, Cg2...Cg2<sup>i</sup> [symmetry code: (i)  $-1/2 + x, y, 1/2 - z$ , where Cg2 is the centroid of the ring A (C1-C6)] may further stabilize the structure, with centroid-centroid distance of 3.972 (2) Å.

### S2. Experimental

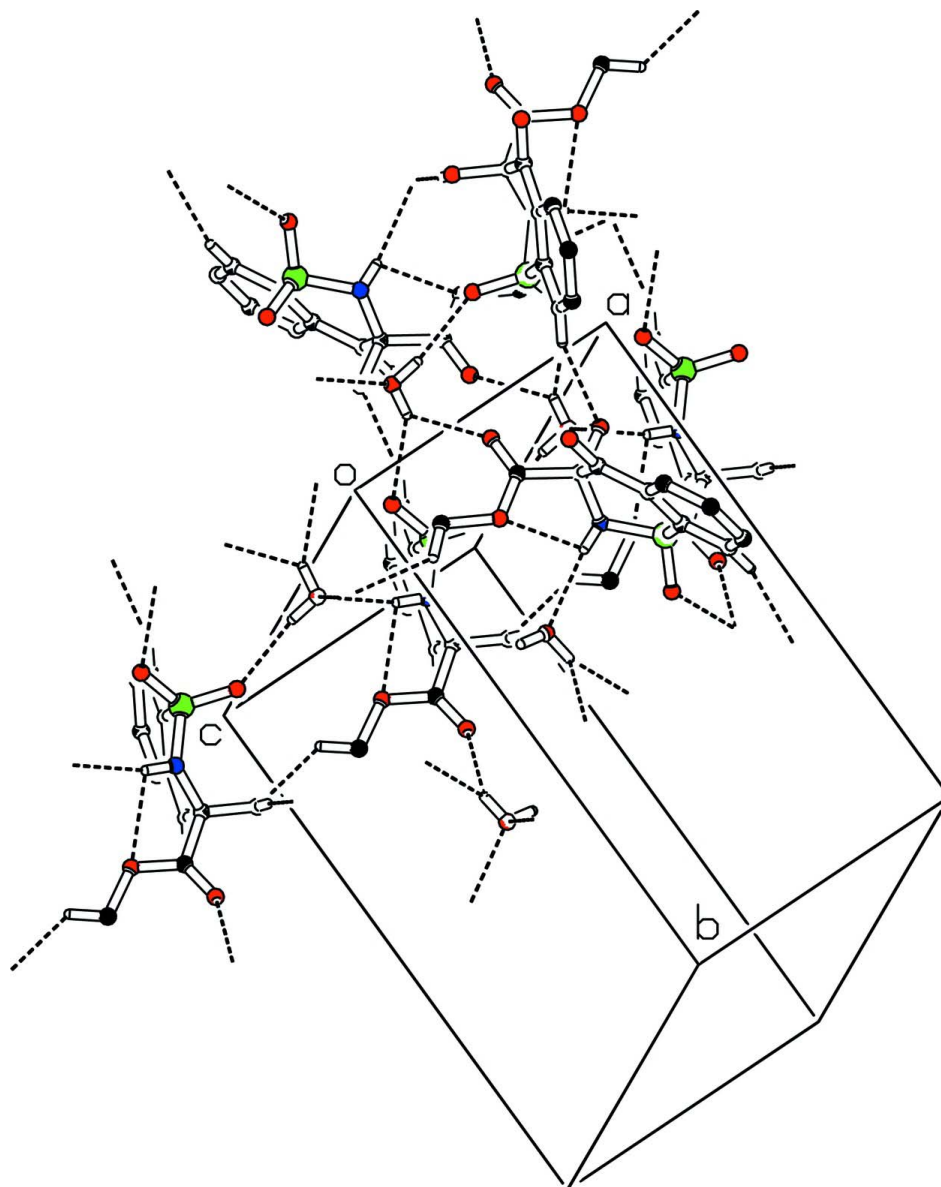
For the preparation of the title compound, methyl 4-hydroxy-2H-1,2-benzothiazine-3-carboxylate 1,1-dioxide (0.5 g, 1.95 mmol), *N*-bromo-succinamide (0.38 g, 2.145 mmol) and dibenzoyl peroxide (0.035 g, 0.15 mmol) were added in CCl<sub>4</sub> (10 ml). The mixture was refluxed for 2 h. After the completion of reaction, CCl<sub>4</sub> was distilled off under vacuum. The obtained residue was washed with hot water to remove other impurities. The solid product was recrystallized in water and methanol to obtain the suitable crystals for x-ray analysis.

### S3. Refinement

H atoms were located in difference syntheses and refined as [O-H = 0.84 (3) Å (for OH); O-H = 0.83 (3) and 0.83 (4) Å (for H<sub>2</sub>O); N-H = 0.78 (3) Å (for NH)]. The remaining H atoms were positioned geometrically, with C-H = 0.93 and 0.96 Å for aromatic and methyl H, respectively, and constrained to ride on their parent atoms with  $U_{\text{iso}}(\text{H}) = xU_{\text{eq}}(\text{C}, \text{N}, \text{O})$ , where  $x = 1.5$  for methyl H and  $x = 1.2$  for all other H atoms.

**Figure 1**

The molecular structure of the title molecule, with the atom-numbering scheme. Displacement ellipsoids are drawn at the 50% probability level. Hydrogen bonds are shown as dotted lines.



**Figure 2**

A packing diagram of the title compound. Hydrogen bonds are shown as dashed lines.

**Methyl 3-hydroxy-4-oxo-3,4-dihydro-2H-1,2-benzothiazine-3-carboxylate 1,1-dioxide monohydrate**

*Crystal data*

$C_{10}H_9NO_6S \cdot H_2O$

$M_r = 289.26$

Orthorhombic, *Pbca*

Hall symbol:  $-P\ 2ac\ 2ab$

$a = 7.7504\ (5)\ \text{\AA}$

$b = 14.5638\ (9)\ \text{\AA}$

$c = 21.0615\ (14)\ \text{\AA}$

$V = 2377.3\ (3)\ \text{\AA}^3$

$Z = 8$

$F(000) = 1200$

$D_x = 1.616\ \text{Mg m}^{-3}$

Mo  $K\alpha$  radiation,  $\lambda = 0.71073\ \text{\AA}$

Cell parameters from 2998 reflections

$\theta = 3.0\text{--}28.5^\circ$

$\mu = 0.30\ \text{mm}^{-1}$

$T = 296\ \text{K}$

Prismatic, colourless

$0.24 \times 0.18 \times 0.15\ \text{mm}$

*Data collection*

Bruker Kappa APEXII CCD  
diffractometer  
Radiation source: fine-focus sealed tube  
Graphite monochromator  
Detector resolution: 7.40 pixels mm<sup>-1</sup>  
 $\omega$  scans  
Absorption correction: multi-scan  
(*SADABS*; Bruker, 2005)  
 $T_{\min} = 0.934$ ,  $T_{\max} = 0.958$

14889 measured reflections  
2998 independent reflections  
1895 reflections with  $I > 2\sigma(I)$   
 $R_{\text{int}} = 0.065$   
 $\theta_{\max} = 28.5^\circ$ ,  $\theta_{\min} = 3.0^\circ$   
 $h = -10 \rightarrow 10$   
 $k = -19 \rightarrow 10$   
 $l = -28 \rightarrow 27$

*Refinement*

Refinement on  $F^2$   
Least-squares matrix: full  
 $R[F^2 > 2\sigma(F^2)] = 0.047$   
 $wR(F^2) = 0.122$   
 $S = 1.01$   
2998 reflections  
184 parameters  
0 restraints  
Primary atom site location: structure-invariant  
direct methods

Secondary atom site location: difference Fourier  
map  
Hydrogen site location: inferred from  
neighbouring sites  
H atoms treated by a mixture of independent  
and constrained refinement  
 $w = 1/[\sigma^2(F_o^2) + (0.0491P)^2 + 1.1022P]$   
where  $P = (F_o^2 + 2F_c^2)/3$   
 $(\Delta/\sigma)_{\max} < 0.001$   
 $\Delta\rho_{\max} = 0.40 \text{ e } \text{\AA}^{-3}$   
 $\Delta\rho_{\min} = -0.32 \text{ e } \text{\AA}^{-3}$

*Special details*

**Geometry.** Bond distances, angles *etc.* have been calculated using the rounded fractional coordinates. All su's are estimated from the variances of the (full) variance-covariance matrix. The cell e.s.d.'s are taken into account in the estimation of distances, angles and torsion angles

**Refinement.** Refinement of  $F^2$  against ALL reflections. The weighted  $R$ -factor  $wR$  and goodness of fit  $S$  are based on  $F^2$ , conventional  $R$ -factors  $R$  are based on  $F$ , with  $F$  set to zero for negative  $F^2$ . The threshold expression of  $F^2 > \sigma(F^2)$  is used only for calculating  $R$ -factors(gt) *etc.* and is not relevant to the choice of reflections for refinement.  $R$ -factors based on  $F^2$  are statistically about twice as large as those based on  $F$ , and  $R$ -factors based on ALL data will be even larger.

*Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters ( $\text{\AA}^2$ )*

	<i>x</i>	<i>y</i>	<i>z</i>	$U_{\text{iso}}^*/U_{\text{eq}}$
S1	0.18084 (8)	0.12546 (4)	0.08206 (3)	0.0314 (2)
O1	0.0037 (3)	0.32676 (13)	0.21865 (9)	0.0568 (8)
O2	0.3475 (2)	0.15746 (13)	0.06338 (9)	0.0450 (7)
O3	0.1137 (3)	0.04539 (12)	0.05197 (9)	0.0450 (6)
O4	0.2024 (2)	0.33701 (12)	0.09947 (10)	0.0360 (6)
H4O	0.204 (4)	0.3501 (19)	0.0608 (14)	0.0432*
O5	-0.0985 (3)	0.42862 (12)	0.08273 (10)	0.0460 (7)
O6	-0.2498 (2)	0.30222 (13)	0.10613 (9)	0.0413 (6)
O7	0.7145 (3)	0.10998 (15)	0.02689 (11)	0.0481 (7)
H71	0.731 (4)	0.054 (2)	0.0226 (15)	0.0577*
H72	0.615 (5)	0.116 (2)	0.0412 (16)	0.0577*
N1	0.0398 (3)	0.20645 (14)	0.07099 (10)	0.0308 (6)
H1	-0.052 (4)	0.1857 (18)	0.0675 (13)	0.0370*
C1	0.1243 (3)	0.17959 (16)	0.20533 (12)	0.0298 (7)
C2	0.1294 (4)	0.16328 (19)	0.27020 (13)	0.0414 (9)
H2	0.09091	0.20830	0.29812	0.0497*

C3	0.1908 (4)	0.0814 (2)	0.29384 (14)	0.0518 (10)
H3	0.19348	0.07164	0.33746	0.0623*
C4	0.2479 (5)	0.0141 (2)	0.25324 (15)	0.0525 (10)
H4	0.28955	-0.04093	0.26959	0.0629*
C5	0.2440 (4)	0.02761 (18)	0.18834 (14)	0.0444 (9)
H5	0.28239	-0.01804	0.16088	0.0533*
C6	0.1824 (3)	0.10985 (16)	0.16477 (12)	0.0306 (7)
C7	0.0552 (3)	0.26875 (16)	0.18234 (12)	0.0319 (8)
C8	0.0514 (3)	0.28972 (15)	0.11076 (11)	0.0284 (7)
C9	-0.1075 (3)	0.35000 (17)	0.09736 (12)	0.0312 (8)
C10	-0.4122 (3)	0.3507 (2)	0.10026 (15)	0.0500 (10)
H10A	-0.50577	0.30900	0.10785	0.0750*
H10B	-0.42207	0.37576	0.05825	0.0750*
H10C	-0.41623	0.39963	0.13080	0.0750*

*Atomic displacement parameters (Å<sup>2</sup>)*

	$U^{11}$	$U^{22}$	$U^{33}$	$U^{12}$	$U^{13}$	$U^{23}$
S1	0.0315 (3)	0.0292 (3)	0.0334 (3)	0.0025 (3)	0.0002 (3)	-0.0061 (3)
O1	0.0826 (16)	0.0444 (11)	0.0433 (12)	0.0237 (12)	-0.0002 (11)	-0.0134 (9)
O2	0.0328 (11)	0.0479 (11)	0.0544 (12)	0.0054 (9)	0.0128 (9)	-0.0020 (9)
O3	0.0591 (13)	0.0327 (9)	0.0433 (11)	0.0012 (9)	-0.0078 (10)	-0.0121 (8)
O4	0.0240 (9)	0.0376 (10)	0.0463 (11)	-0.0060 (8)	-0.0014 (8)	-0.0029 (9)
O5	0.0406 (12)	0.0312 (10)	0.0662 (14)	0.0041 (8)	-0.0023 (10)	0.0090 (9)
O6	0.0220 (9)	0.0402 (10)	0.0618 (13)	-0.0003 (8)	0.0007 (9)	0.0062 (9)
O7	0.0453 (13)	0.0400 (10)	0.0590 (14)	-0.0025 (10)	-0.0070 (10)	0.0005 (10)
N1	0.0258 (11)	0.0296 (10)	0.0370 (12)	-0.0008 (9)	-0.0049 (10)	-0.0052 (9)
C1	0.0261 (12)	0.0291 (12)	0.0342 (13)	0.0004 (10)	-0.0018 (10)	-0.0021 (10)
C2	0.0470 (16)	0.0411 (15)	0.0362 (14)	0.0006 (13)	0.0007 (13)	-0.0049 (12)
C3	0.063 (2)	0.0559 (18)	0.0364 (15)	0.0002 (17)	-0.0064 (15)	0.0089 (14)
C4	0.067 (2)	0.0393 (15)	0.0513 (18)	0.0054 (15)	-0.0086 (16)	0.0110 (14)
C5	0.0503 (17)	0.0342 (14)	0.0487 (17)	0.0069 (13)	-0.0052 (14)	-0.0021 (13)
C6	0.0269 (12)	0.0317 (12)	0.0332 (13)	0.0008 (11)	-0.0037 (11)	-0.0016 (10)
C7	0.0284 (13)	0.0296 (12)	0.0377 (14)	0.0005 (10)	-0.0013 (11)	-0.0055 (11)
C8	0.0240 (12)	0.0257 (11)	0.0356 (14)	-0.0001 (10)	-0.0027 (10)	-0.0041 (10)
C9	0.0272 (13)	0.0331 (13)	0.0332 (14)	-0.0001 (11)	-0.0003 (10)	-0.0017 (10)
C10	0.0234 (14)	0.0674 (19)	0.0591 (19)	0.0118 (14)	-0.0015 (13)	0.0030 (16)

*Geometric parameters (Å, °)*

S1—O2	1.4284 (17)	C1—C7	1.486 (3)
S1—O3	1.426 (2)	C1—C6	1.402 (3)
S1—N1	1.625 (2)	C2—C3	1.377 (4)
S1—C6	1.757 (3)	C3—C4	1.374 (4)
O1—C7	1.207 (3)	C4—C5	1.381 (4)
O4—C8	1.379 (3)	C5—C6	1.382 (4)
O5—C9	1.188 (3)	C7—C8	1.539 (3)
O6—C9	1.317 (3)	C8—C9	1.539 (3)

O6—C10	1.449 (3)	C2—H2	0.9300
O4—H4O	0.84 (3)	C3—H3	0.9300
O7—H71	0.83 (3)	C4—H4	0.9300
O7—H72	0.83 (4)	C5—H5	0.9300
N1—C8	1.477 (3)	C10—H10C	0.9600
N1—H1	0.78 (3)	C10—H10A	0.9600
C1—C2	1.387 (4)	C10—H10B	0.9600
S1…H10B <sup>i</sup>	3.0600	N1…O6	2.744 (3)
O1…O4	2.949 (3)	N1…O7 <sup>xii</sup>	3.032 (3)
O1…O6	3.099 (3)	C3…C5 <sup>x</sup>	3.570 (4)
O1…C4 <sup>ii</sup>	3.406 (4)	C4…O1 <sup>ix</sup>	3.418 (4)
O1…C4 <sup>iii</sup>	3.418 (4)	C4…O1 <sup>xiii</sup>	3.405 (4)
O2…O4	2.946 (3)	C5…C3 <sup>xi</sup>	3.570 (4)
O2…O7	3.027 (3)	C5…O4 <sup>ix</sup>	3.374 (3)
O2…C9 <sup>i</sup>	3.405 (3)	C9…O2 <sup>vii</sup>	3.405 (3)
O3…C10 <sup>iv</sup>	3.393 (3)	C10…O4 <sup>xii</sup>	2.994 (3)
O3…O7 <sup>v</sup>	3.107 (3)	C10…O3 <sup>xiv</sup>	3.393 (3)
O3…O3 <sup>vi</sup>	3.106 (3)	C10…H4O <sup>xii</sup>	3.09 (3)
O4…O5	2.710 (3)	C10…H2 <sup>x</sup>	2.9800
O4…C5 <sup>iii</sup>	3.374 (3)	H1…O6	2.43 (3)
O4…O7 <sup>vii</sup>	2.773 (3)	H1…O7 <sup>xii</sup>	2.29 (3)
O4…O2	2.946 (3)	H2…H10A <sup>xi</sup>	2.5800
O4…O1	2.949 (3)	H2…O1	2.5000
O4…C10 <sup>viii</sup>	2.994 (3)	H2…O6 <sup>xi</sup>	2.7300
O5…O7 <sup>iii</sup>	3.028 (3)	H2…C10 <sup>xi</sup>	2.9800
O5…O4	2.710 (3)	H3…O7 <sup>x</sup>	2.9200
O6…N1	2.744 (3)	H3…O5 <sup>xiii</sup>	2.7800
O6…O1	3.099 (3)	H4…H10C <sup>xiii</sup>	2.4700
O7…N1 <sup>viii</sup>	3.032 (3)	H4…O1 <sup>ix</sup>	2.7300
O7…O5 <sup>ix</sup>	3.028 (3)	H4O…C10 <sup>viii</sup>	3.09 (3)
O7…O4 <sup>i</sup>	2.773 (3)	H4O…H10A <sup>viii</sup>	2.5300
O7…O3 <sup>v</sup>	3.107 (3)	H4O…O5	2.65 (3)
O7…O2	3.027 (3)	H4O…O7 <sup>vii</sup>	1.94 (3)
O1…H4 <sup>iii</sup>	2.7300	H4O…H71 <sup>vii</sup>	2.25 (4)
O1…H2	2.5000	H4O…H72 <sup>vii</sup>	2.31 (4)
O2…H72	2.21 (4)	H5…O3	2.8000
O2…H10A <sup>viii</sup>	2.6500	H5…O4 <sup>ix</sup>	2.4800
O3…H10B <sup>i</sup>	2.6000	H10A…O2 <sup>xii</sup>	2.6500
O3…H5	2.8000	H10A…O4 <sup>xii</sup>	2.3100
O3…H71 <sup>v</sup>	2.45 (3)	H10A…H2 <sup>x</sup>	2.5800
O3…H10B <sup>iv</sup>	2.8900	H10A…H4O <sup>xii</sup>	2.5300
O4…H5 <sup>iii</sup>	2.4800	H10B…S1 <sup>vii</sup>	3.0600
O4…H10A <sup>viii</sup>	2.3100	H10B…O3 <sup>vii</sup>	2.6000
O5…H4O	2.65 (3)	H10B…O5	2.6700
O5…H10B	2.6700	H10B…O3 <sup>xiv</sup>	2.8900
O5…H3 <sup>ii</sup>	2.7800	H10C…O5	2.7000
O5…H72 <sup>iii</sup>	2.87 (3)	H10C…H4 <sup>ii</sup>	2.4700

O5...H10C	2.7000	H71...H4O <sup>i</sup>	2.25 (4)
O5...H71 <sup>iii</sup>	2.45 (3)	H71...O3 <sup>v</sup>	2.45 (3)
O6...H2 <sup>x</sup>	2.7300	H71...O5 <sup>ix</sup>	2.45 (3)
O6...H1	2.43 (3)	H72...O2	2.21 (4)
O7...H1 <sup>viii</sup>	2.29 (3)	H72...H4O <sup>i</sup>	2.31 (5)
O7...H4O <sup>i</sup>	1.94 (3)	H72...O5 <sup>ix</sup>	2.87 (3)
O7...H3 <sup>xi</sup>	2.9200		
O2—S1—O3	118.33 (12)	O1—C7—C8	118.4 (2)
O2—S1—N1	109.39 (11)	O4—C8—N1	111.36 (19)
O2—S1—C6	108.01 (11)	O4—C8—C7	104.59 (19)
O3—S1—N1	106.53 (12)	O4—C8—C9	111.27 (18)
O3—S1—C6	109.72 (11)	N1—C8—C7	113.20 (18)
N1—S1—C6	103.94 (11)	N1—C8—C9	108.41 (19)
C9—O6—C10	117.3 (2)	C7—C8—C9	107.96 (19)
C8—O4—H4O	107 (2)	O5—C9—C8	123.4 (2)
H71—O7—H72	107 (3)	O6—C9—C8	110.1 (2)
S1—N1—C8	118.28 (17)	O5—C9—O6	126.5 (2)
C8—N1—H1	115 (2)	C3—C2—H2	120.00
S1—N1—H1	110 (2)	C1—C2—H2	120.00
C2—C1—C6	117.8 (2)	C2—C3—H3	120.00
C2—C1—C7	118.7 (2)	C4—C3—H3	120.00
C6—C1—C7	123.4 (2)	C5—C4—H4	120.00
C1—C2—C3	120.9 (3)	C3—C4—H4	120.00
C2—C3—C4	120.3 (3)	C4—C5—H5	120.00
C3—C4—C5	120.5 (3)	C6—C5—H5	120.00
C4—C5—C6	119.1 (3)	O6—C10—H10B	109.00
S1—C6—C5	118.1 (2)	O6—C10—H10C	109.00
S1—C6—C1	120.56 (18)	O6—C10—H10A	110.00
C1—C6—C5	121.4 (2)	H10A—C10—H10C	109.00
O1—C7—C1	121.6 (2)	H10B—C10—H10C	109.00
C1—C7—C8	120.0 (2)	H10A—C10—H10B	110.00
O2—S1—N1—C8	-67.0 (2)	C2—C1—C7—C8	178.7 (2)
O3—S1—N1—C8	163.98 (18)	C6—C1—C7—O1	178.8 (2)
C6—S1—N1—C8	48.1 (2)	C6—C1—C7—C8	-2.4 (3)
O2—S1—C6—C1	94.5 (2)	C1—C2—C3—C4	-0.2 (5)
O2—S1—C6—C5	-84.8 (2)	C2—C3—C4—C5	-0.2 (5)
O3—S1—C6—C1	-135.2 (2)	C3—C4—C5—C6	0.2 (5)
O3—S1—C6—C5	45.5 (2)	C4—C5—C6—S1	179.4 (2)
N1—S1—C6—C1	-21.6 (2)	C4—C5—C6—C1	0.1 (4)
N1—S1—C6—C5	159.1 (2)	O1—C7—C8—O4	84.6 (3)
C10—O6—C9—O5	-3.0 (4)	O1—C7—C8—N1	-154.0 (2)
C10—O6—C9—C8	175.4 (2)	O1—C7—C8—C9	-33.9 (3)
S1—N1—C8—O4	64.3 (2)	C1—C7—C8—O4	-94.2 (2)
S1—N1—C8—C7	-53.3 (3)	C1—C7—C8—N1	27.2 (3)
S1—N1—C8—C9	-173.01 (16)	C1—C7—C8—C9	147.2 (2)
C6—C1—C2—C3	0.5 (4)	O4—C8—C9—O5	-4.0 (3)



C7—C1—C2—C3	179.4 (3)	O4—C8—C9—O6	177.6 (2)
C2—C1—C6—S1	-179.7 (2)	N1—C8—C9—O5	-126.8 (3)
C2—C1—C6—C5	-0.4 (4)	N1—C8—C9—O6	54.8 (3)
C7—C1—C6—S1	1.4 (3)	C7—C8—C9—O5	110.2 (3)
C7—C1—C6—C5	-179.3 (2)	C7—C8—C9—O6	-68.2 (2)
C2—C1—C7—O1	-0.1 (4)		

Symmetry codes: (i)  $x+1/2, -y+1/2, -z$ ; (ii)  $-x, y+1/2, -z+1/2$ ; (iii)  $-x+1/2, y+1/2, z$ ; (iv)  $-x-1/2, y-1/2, z$ ; (v)  $-x+1, -y, -z$ ; (vi)  $-x, -y, -z$ ; (vii)  $x-1/2, -y+1/2, -z$ ; (viii)  $x+1, y, z$ ; (ix)  $-x+1/2, y-1/2, z$ ; (x)  $x-1/2, y, -z+1/2$ ; (xi)  $x+1/2, y, -z+1/2$ ; (xii)  $x-1, y, z$ ; (xiii)  $-x, y-1/2, -z+1/2$ ; (xiv)  $-x-1/2, y+1/2, z$ .

Hydrogen-bond geometry ( $\text{\AA}, ^\circ$ )

$D-H\cdots A$	$D-H$	$H\cdots A$	$D\cdots A$	$D-H\cdots A$
N1—H1 $\cdots$ O6	0.78 (3)	2.43 (3)	2.744 (3)	106 (2)
N1—H1 $\cdots$ O7 <sup>xii</sup>	0.78 (3)	2.29 (3)	3.032 (3)	162 (3)
O4—H4O $\cdots$ O7 <sup>vii</sup>	0.84 (3)	1.94 (3)	2.773 (3)	175 (2)
O7—H71 $\cdots$ O3 <sup>v</sup>	0.83 (3)	2.45 (3)	3.107 (3)	137 (3)
O7—H71 $\cdots$ O5 <sup>ix</sup>	0.83 (3)	2.45 (3)	3.028 (3)	128 (3)
O7—H72 $\cdots$ O2	0.83 (4)	2.21 (4)	3.027 (3)	167 (3)
C5—H5 $\cdots$ O4 <sup>ix</sup>	0.9300	2.4800	3.374 (3)	162.00
C10—H10A $\cdots$ O4 <sup>xii</sup>	0.9600	2.3100	2.994 (3)	128.00

Symmetry codes: (v)  $-x+1, -y, -z$ ; (vii)  $x-1/2, -y+1/2, -z$ ; (ix)  $-x+1/2, y-1/2, z$ ; (xii)  $x-1, y, z$ .