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Ethyl 4'-ethenyl-2'-oxo-4-phenyl-2-(3,4,5-trimethoxyphenyl)spiro[pyrrolidine-3,3'-indoline]-5-carboxylate monohydrate

M. Sathyanarayanan,^a P. Ramesh,^b Ramalingam Murugan,^c S. Sriman Narayanan^c and M. N. Ponnuswamy^{d*}

^aPG & Research Department of Physics, A. M. Jain College, Meenambakkam, Chennai 600 114, India, ^bDepartment of Physics, Presidency College (Autonomous), Chennai 600 005, India, ^cDepartment of Analytical Chemistry, University of Madras, Guindy Campus, Chennai 600 025, India, and ^dCentre of Advanced Study in Crystallography and Biophysics, University of Madras, Guindy Campus, Chennai 600 025, India

Correspondence e-mail: mnpsy2004@yahoo.com

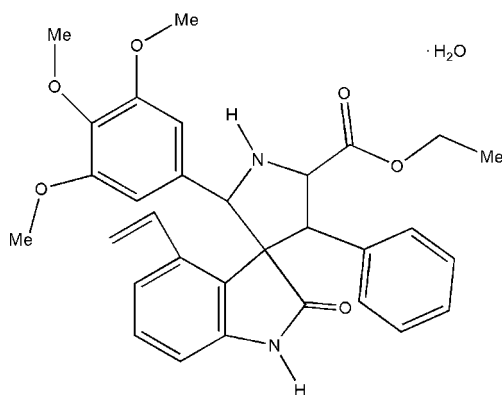
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Key indicators: single-crystal X-ray study; $T = 293$ K; mean $\sigma(\text{C}-\text{C}) = 0.003$ Å; disorder in main residue; R factor = 0.046; wR factor = 0.141; data-to-parameter ratio = 19.8.

In the title compound, $\text{C}_{31}\text{H}_{32}\text{N}_2\text{O}_6 \cdot \text{H}_2\text{O}$, the pyrrolidine ring adopts an envelope conformation. The ethyl C atoms of the ethoxycarbonyl group are disordered over two positions with occupancies of *ca* 0.80 and 0.20. Intramolecular $\text{N}-\text{H} \cdots \text{O}$ hydrogen bonds form $S(5)$ and $S(6)$ ring motifs. Molecules are linked into a three-dimensional framework by $\text{O}-\text{H} \cdots \text{O}$, $\text{N}-\text{H} \cdots \text{O}$ and $\text{C}-\text{H} \cdots \text{O}$ hydrogen bonds, and by $\text{C}-\text{H} \cdots \pi$ interactions.

Related literature

For related literature, see: Amalraj *et al.* (2003); Beddoes *et al.* (1986); Cordell (1981); Suzuki *et al.* (1994). For hydrogen-bond motifs, see: Bernstein *et al.* (1995). For ring conformational analysis, see: Cremer & Pople (1975); Nardelli (1983).



Experimental

Crystal data

$\text{C}_{31}\text{H}_{32}\text{N}_2\text{O}_6 \cdot \text{H}_2\text{O}$
 $M_r = 546.60$
 Hexagonal, $R\bar{3}$
 $a = 38.8029$ (10) Å
 $c = 11.0307$ (3) Å
 $V = 14383.4$ (7) Å³

$Z = 18$
 Mo $K\alpha$ radiation
 $\mu = 0.08$ mm⁻¹
 $T = 293$ (2) K
 $0.23 \times 0.21 \times 0.17$ mm

Data collection

Bruker Kappa APEXII area-detector diffractometer
 Absorption correction: multi-scan (SADABS, Sheldrick, 2001)
 $T_{\min} = 0.982$, $T_{\max} = 0.986$

111058 measured reflections
 7909 independent reflections
 5572 reflections with $I > 2\sigma(I)$
 $R_{\text{int}} = 0.035$

Refinement

$R[F^2 > 2\sigma(F^2)] = 0.046$
 $wR(F^2) = 0.141$
 $S = 1.05$
 7909 reflections
 399 parameters
 29 restraints

H atoms treated by a mixture of independent and constrained refinement
 $\Delta\rho_{\text{max}} = 0.26$ e Å⁻³
 $\Delta\rho_{\text{min}} = -0.21$ e Å⁻³

Table 1

Hydrogen-bond geometry (Å, °).

| $D-\text{H} \cdots A$ | $D-\text{H}$ | $\text{H} \cdots A$ | $D \cdots A$ | $D-\text{H} \cdots A$ |
|--|--------------|---------------------|--------------|-----------------------|
| $\text{N1}-\text{H1} \cdots \text{O1}$ | 0.86 (2) | 2.419 (18) | 2.8088 (18) | 108 (1) |
| $\text{N1}-\text{H1} \cdots \text{O3}$ | 0.86 (2) | 2.376 (18) | 2.9395 (17) | 123 (1) |
| $\text{O4}-\text{H4B} \cdots \text{O1}^i$ | 0.85 (3) | 2.142 (18) | 2.909 (2) | 150 (3) |
| $\text{N16}-\text{H16} \cdots \text{O3}^{ii}$ | 0.87 (2) | 1.99 (2) | 2.8449 (17) | 166 (2) |
| $\text{C5}-\text{H5} \cdots \text{O5}^{iii}$ | 0.98 | 2.45 | 3.3197 (18) | 147 |
| $\text{C18}-\text{H18} \cdots \text{O4}^{iv}$ | 0.93 | 2.46 | 3.357 (2) | 162 |
| $\text{C24}-\text{H24B} \cdots \text{Cg1}^{iii}$ | 0.93 | 2.93 | 3.776 (2) | 153 |

Symmetry codes: (i) $y + \frac{1}{3}, -x + y + \frac{2}{3}, -z + \frac{2}{3}$; (ii) $-x + 1, -y, -z + 1$; (iii) $-y + \frac{1}{3}, x - y - \frac{1}{3}, z - \frac{1}{3}$; (iv) $-x + y + \frac{2}{3}, -x + \frac{1}{3}, z + \frac{1}{3}$.

Data collection: APEX2 (Bruker, 2004); cell refinement: APEX2; data reduction: SAINT (Bruker, 2004); program(s) used to solve structure: SHELXS97 (Sheldrick, 2008); program(s) used to refine structure: SHELXL97 (Sheldrick, 2008); molecular graphics: ORTEP-3 (Farrugia, 1997); software used to prepare material for publication: SHELXL97 and PLATON (Spek, 2003).

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Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: CI2651).

References

- Amalraj, A., Raghunathan, R., Sridevi Kumari, M. R. & Raman, N. (2003). *Bioorg. Med. Chem.* **11**, 407–419.
 Beddoes, R. L., Dalton, L., Joule, T. A., Mills, O. S., Street, J. D. & Watt, C. I. F. (1986). *J. Chem. Soc. Perkin Trans. 2*, pp. 787–797.
 Bernstein, J., Davis, R. E., Shimoni, L. & Chang, N. L. (1995). *Angew. Chem. Int. Ed. Engl.* **34**, 1555–1573.
 Bruker (2004). APEX2 and SAINT. Bruker AXS Inc., Madison, Wisconsin, USA.
 Cordell, G. (1981). *Introduction to Alkaloids: A Biogenic Approach*. New York: Wiley International.
 Cremer, D. & Pople, J. A. (1975). *J. Am. Chem. Soc.* **97**, 1354–1358.
 Farrugia, L. J. (1997). *J. Appl. Cryst.* **30**, 565.
 Nardelli, M. (1983). *Acta Cryst.* **C39**, 1141–1142.

Sheldrick, G. M. (2001). *SADABS*. University of Göttingen, Germany.
Sheldrick, G. M. (2008). *Acta Cryst.* **A64**, 112–122.

Spek, A. L. (2003). *J. Appl. Cryst.* **36**, 7–13.
Suzuki, H., Aoyagi, S. & Kibayashi, C. (1994). *Tetrahedron Lett.* **35**, 6119–6122.

supporting information

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Ethyl 4'-ethenyl-2'-oxo-4-phenyl-2-(3,4,5-trimethoxyphenyl)spiro-[pyrrolidine-3,3'-indoline]-5-carboxylate monohydrate

M. Sathyanarayanan, P. Ramesh, Ramalingam Murugan, S. Sriman Narayanan and M. N. Ponnuswamy

S1. Comment

Substituted pyrrolidine compounds possess antimicrobial and antifungal activity against various pathogens (Amalraj *et al.*, 2003). Several optically active pyrrolidine compounds are used as intermediates in controlled asymmetric synthesis (Suzuki *et al.*, 1994). The spiro-indole-pyrrolidine ring system is a frequently encountered structural motif in many biologically important and pharmacologically relevant alkaloids, *e.g.* vincristine, vinblastine and spirotopostatins (Cordell, 1981). Against this background and to ascertain the detailed information on its molecular conformation, the structure determination of the title compound has been carried out.

The pyrrolidine ring (N1—C5) adopts an envelope conformation, with puckering (Cremer & Pople, 1975) and asymmetry (Nardelli, 1983) parameters $q_2 = 0.416$ (2) Å, $\varphi = 137.5$ (2)° and $\Delta_s(C5) = 2.8$ (2)°. The indoline ring system is planar and the keto atom O3 lies on the plane. The sum of angles at atom N1 of the pyrrolidine ring (323.3°) is in accordance with sp^3 hybridization (Beddoes *et al.*, 1986). The ethoxycarbonyl group is in an extended conformation as evidenced by torsion angles C2—C6—O2—C7 of -170.3 (3)° and C6—O2—C7—C8 of 170.3 (3)°.

Intramolecular N1—H1...O1 and N1—H1...O3 hydrogen bonds generate S(5) and S(6) ring motifs (Bernstein *et al.* 1995), respectively. The crystal packing is stabilized by O—H...O, O—H...N, N—H...O and C—H...O hydrogen bonds, and C—H... π intermolecular interactions (Table 1) which link the molecules into a three-dimensional framework.

S2. Experimental

3-Arylidene-4-vinylindoline-2-one (0.5 g, 1.0 mol) and (*E*)-ethyl-2-(3,4,5-trimethoxybenzylideneamino)acetate (0.15 g, 1.0 mol) in acetonitrile (10 ml) was stirred in the presence of catalytic amount of AgOAc and triethylamine. The obtained crude product was recrystallized in n-hexane-acetone (8:2 v/v).

S3. Refinement

The ethyl C atoms of the ethoxycarbonyl group are disordered over two positions (C7/C7A and C8/C8A) with refined occupancies of 0.797 (8) and 0.203 (8). The corresponding bond distances involving the disordered atoms were restrained to 1.54 (5) Å, and also the U^{ij} parameters of atoms C7, C7A, C8 and C8A were restrained to an approximate isotropic behaviour. The O- and N-bound H atoms were located in a difference map and refined with O—H and H...H distances restrained to 0.84 (1) and 1.37 (1) Å, respectively. The remaining H atoms were positioned geometrically (C—H = 0.93–0.98 Å) and allowed to ride on their parent atoms, with $U_{iso}(H) = 1.2$ –1.5 (methyl) $U_{eq}(C)$. A search for solvent-accessible voids in the crystal structure using *PLATON* showed a potential solvent volume of 2189.3 Å³ and subsequent application of *SQUEEZE* procedures showed three relevant voids each with a solvent-accessible volume of 730 Å³. However, this procedure showed no electrons in the voids. This indicates that the crystal lost nearly all of its solvent of crystallization

by the time it was used for data collection, without collapse of the structure.

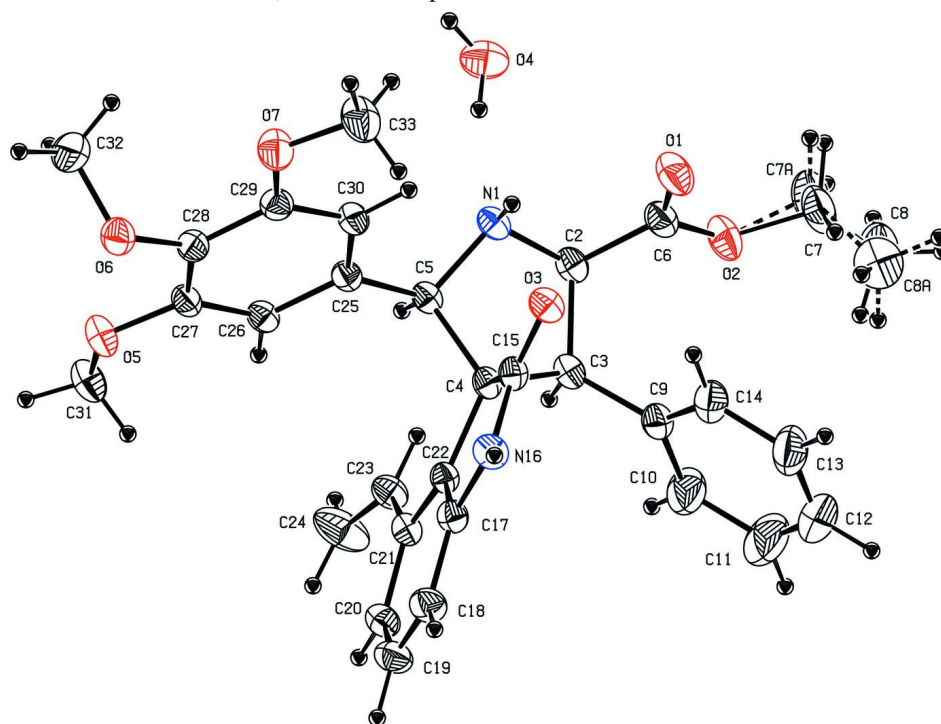


Figure 1

The molecular structure of the title compound, showing 30% probability displacement ellipsoids. Both disorder components are shown.

Ethyl 4'-ethenyl-2'-oxo-4-phenyl-2-(3,4,5-trimethoxyphenyl)spiro[pyrrolidine-3,3'-indoline]-5-carboxylate monohydrate

Crystal data

$C_{31}H_{32}N_2O_6 \cdot H_2O$

$M_r = 546.60$

Hexagonal, $R\bar{3}$

Hall symbol: $-R\ 3$

$a = 38.8029 (10) \text{ \AA}$

$c = 11.0307 (3) \text{ \AA}$

$V = 14383.4 (7) \text{ \AA}^3$

$Z = 18$

$F(000) = 5220$

$D_x = 1.136 \text{ Mg m}^{-3}$

Mo $K\alpha$ radiation, $\lambda = 0.71073 \text{ \AA}$

Cell parameters from 5683 reflections

$\theta = 1.1\text{--}28.2^\circ$

$\mu = 0.08 \text{ mm}^{-1}$

$T = 293 \text{ K}$

Block, colourless

$0.23 \times 0.21 \times 0.17 \text{ mm}$

Data collection

Bruker Kappa APEXII area-detector
diffractometer

Radiation source: fine-focus sealed tube

Graphite monochromator

ω and ϕ scans

Absorption correction: multi-scan

(*SADABS*, Sheldrick, 2001)

$T_{\min} = 0.982$, $T_{\max} = 0.986$

111058 measured reflections

7909 independent reflections

5572 reflections with $I > 2\sigma(I)$

$R_{\text{int}} = 0.035$

$\theta_{\max} = 28.2^\circ$, $\theta_{\min} = 1.9^\circ$

$h = -51 \rightarrow 51$

$k = -51 \rightarrow 51$

$l = -14 \rightarrow 14$

Refinement

Refinement on F^2
 Least-squares matrix: full
 $R[F^2 > 2\sigma(F^2)] = 0.046$
 $wR(F^2) = 0.142$
 $S = 1.05$
 7909 reflections
 399 parameters
 29 restraints
 Primary atom site location: structure-invariant
 direct methods

Secondary atom site location: difference Fourier
 map
 Hydrogen site location: inferred from
 neighbouring sites
 H atoms treated by a mixture of independent
 and constrained refinement
 $w = 1/[\sigma^2(F_o^2) + (0.0548P)^2 + 16.9913P]$
 where $P = (F_o^2 + 2F_c^2)/3$
 $(\Delta/\sigma)_{\max} = 0.001$
 $\Delta\rho_{\max} = 0.26 \text{ e } \text{Å}^{-3}$
 $\Delta\rho_{\min} = -0.21 \text{ e } \text{Å}^{-3}$

Special details

Geometry. All e.s.d.'s (except the e.s.d. in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell e.s.d.'s are taken into account individually in the estimation of e.s.d.'s in distances, angles and torsion angles; correlations between e.s.d.'s in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell e.s.d.'s is used for estimating e.s.d.'s involving l.s. planes.

Refinement. Refinement of F^2 against ALL reflections. The weighted R -factor wR and goodness of fit S are based on F^2 , conventional R -factors R are based on F , with F set to zero for negative F^2 . The threshold expression of $F^2 > \sigma(F^2)$ is used only for calculating R -factors(gt) etc. and is not relevant to the choice of reflections for refinement. R -factors based on F^2 are statistically about twice as large as those based on F , and R -factors based on ALL data will be even larger.

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (Å^2)

| | <i>x</i> | <i>y</i> | <i>z</i> | $U_{\text{iso}}^*/U_{\text{eq}}$ | Occ. (<1) |
|-----|--------------|--------------|--------------|----------------------------------|-----------|
| O1 | 0.56007 (4) | 0.06180 (4) | 0.10291 (14) | 0.0633 (4) | |
| O2 | 0.56908 (4) | 0.12184 (4) | 0.05914 (13) | 0.0619 (4) | |
| O3 | 0.50835 (3) | 0.01984 (3) | 0.34955 (10) | 0.0412 (3) | |
| O5 | 0.31168 (3) | -0.07849 (4) | 0.31585 (11) | 0.0499 (3) | |
| O6 | 0.33393 (4) | -0.13259 (3) | 0.34016 (11) | 0.0501 (3) | |
| O7 | 0.40791 (4) | -0.11581 (4) | 0.28214 (13) | 0.0556 (3) | |
| N1 | 0.47767 (4) | 0.03182 (4) | 0.11926 (11) | 0.0385 (3) | |
| H1 | 0.4898 (5) | 0.0196 (5) | 0.1441 (16) | 0.041 (5)* | |
| N16 | 0.47798 (4) | 0.03293 (4) | 0.50612 (12) | 0.0400 (3) | |
| H16 | 0.4835 (5) | 0.0200 (6) | 0.5595 (18) | 0.048 (5)* | |
| C2 | 0.50558 (5) | 0.07457 (5) | 0.12126 (13) | 0.0392 (3) | |
| H2 | 0.4975 | 0.0866 | 0.0572 | 0.047* | |
| C3 | 0.50071 (4) | 0.09151 (5) | 0.24589 (13) | 0.0376 (3) | |
| H3 | 0.4867 | 0.1059 | 0.2262 | 0.045* | |
| C4 | 0.46974 (4) | 0.05359 (4) | 0.31510 (12) | 0.0326 (3) | |
| C5 | 0.44655 (4) | 0.02552 (4) | 0.20623 (12) | 0.0337 (3) | |
| H5 | 0.4309 | 0.0362 | 0.1690 | 0.040* | |
| C6 | 0.54749 (5) | 0.08427 (6) | 0.09354 (15) | 0.0467 (4) | |
| C7 | 0.61222 (16) | 0.13632 (13) | 0.0495 (5) | 0.0722 (13) | 0.797 (8) |
| H7A | 0.6178 | 0.1242 | -0.0189 | 0.087* | 0.797 (8) |
| H7B | 0.6221 | 0.1305 | 0.1229 | 0.087* | 0.797 (8) |
| C8 | 0.63069 (12) | 0.18077 (11) | 0.0316 (6) | 0.121 (2) | 0.797 (8) |
| H8A | 0.6259 | 0.1860 | -0.0499 | 0.182* | 0.797 (8) |
| H8B | 0.6588 | 0.1935 | 0.0456 | 0.182* | 0.797 (8) |

| | | | | | |
|------|--------------|--------------|--------------|-------------|-----------|
| H8C | 0.6191 | 0.1910 | 0.0876 | 0.182* | 0.797 (8) |
| C7A | 0.6071 (7) | 0.1388 (7) | 0.0101 (16) | 0.068 (5) | 0.203 (8) |
| H7C | 0.6102 | 0.1568 | -0.0554 | 0.081* | 0.203 (8) |
| H7D | 0.6121 | 0.1184 | -0.0212 | 0.081* | 0.203 (8) |
| C8A | 0.6359 (5) | 0.1615 (6) | 0.1138 (16) | 0.109 (6) | 0.203 (8) |
| H8D | 0.6324 | 0.1434 | 0.1780 | 0.163* | 0.203 (8) |
| H8E | 0.6305 | 0.1816 | 0.1439 | 0.163* | 0.203 (8) |
| H8F | 0.6628 | 0.1739 | 0.0846 | 0.163* | 0.203 (8) |
| C9 | 0.53732 (5) | 0.12089 (5) | 0.31560 (16) | 0.0463 (4) | |
| C10 | 0.54234 (7) | 0.15785 (7) | 0.3395 (3) | 0.0837 (8) | |
| H10 | 0.5233 | 0.1640 | 0.3128 | 0.100* | |
| C11 | 0.57526 (10) | 0.18618 (9) | 0.4028 (4) | 0.1240 (13) | |
| H11 | 0.5782 | 0.2111 | 0.4177 | 0.149* | |
| C12 | 0.60336 (9) | 0.17751 (10) | 0.4432 (3) | 0.1093 (11) | |
| H12 | 0.6255 | 0.1965 | 0.4855 | 0.131* | |
| C13 | 0.59885 (7) | 0.14127 (8) | 0.4215 (2) | 0.0780 (7) | |
| H13 | 0.6178 | 0.1353 | 0.4496 | 0.094* | |
| C14 | 0.56632 (5) | 0.11301 (6) | 0.35802 (17) | 0.0554 (5) | |
| H14 | 0.5638 | 0.0882 | 0.3434 | 0.066* | |
| C15 | 0.48836 (4) | 0.03372 (4) | 0.38989 (13) | 0.0345 (3) | |
| C17 | 0.45453 (4) | 0.05069 (5) | 0.52273 (13) | 0.0370 (3) | |
| C18 | 0.44120 (5) | 0.05730 (5) | 0.63084 (14) | 0.0469 (4) | |
| H18 | 0.4468 | 0.0495 | 0.7044 | 0.056* | |
| C19 | 0.41904 (5) | 0.07617 (6) | 0.62473 (15) | 0.0507 (4) | |
| H19 | 0.4099 | 0.0816 | 0.6960 | 0.061* | |
| C20 | 0.41024 (5) | 0.08710 (5) | 0.51543 (15) | 0.0446 (4) | |
| H20 | 0.3951 | 0.0996 | 0.5148 | 0.054* | |
| C21 | 0.42347 (4) | 0.07994 (4) | 0.40474 (13) | 0.0366 (3) | |
| C22 | 0.44694 (4) | 0.06215 (4) | 0.41045 (12) | 0.0329 (3) | |
| C23 | 0.41162 (5) | 0.09042 (5) | 0.28964 (15) | 0.0430 (4) | |
| H23 | 0.4249 | 0.0898 | 0.2202 | 0.052* | |
| C24 | 0.38483 (8) | 0.10039 (9) | 0.2753 (2) | 0.0826 (8) | |
| H24A | 0.3707 | 0.1015 | 0.3419 | 0.099* | |
| H24B | 0.3796 | 0.1065 | 0.1984 | 0.099* | |
| C25 | 0.41775 (4) | -0.01738 (4) | 0.23731 (12) | 0.0344 (3) | |
| C26 | 0.37872 (4) | -0.02733 (5) | 0.26238 (13) | 0.0363 (3) | |
| H26 | 0.3714 | -0.0079 | 0.2588 | 0.044* | |
| C27 | 0.35070 (4) | -0.06595 (5) | 0.29254 (13) | 0.0377 (3) | |
| C28 | 0.36146 (5) | -0.09500 (5) | 0.30097 (14) | 0.0392 (3) | |
| C29 | 0.40046 (5) | -0.08511 (5) | 0.27477 (14) | 0.0401 (3) | |
| C30 | 0.42853 (5) | -0.04645 (5) | 0.24184 (14) | 0.0396 (3) | |
| H30 | 0.4544 | -0.0401 | 0.2230 | 0.048* | |
| C31 | 0.29816 (5) | -0.05142 (6) | 0.29070 (19) | 0.0546 (5) | |
| H31A | 0.2701 | -0.0640 | 0.3062 | 0.082* | |
| H31B | 0.3032 | -0.0435 | 0.2072 | 0.082* | |
| H31C | 0.3120 | -0.0284 | 0.3416 | 0.082* | |
| C32 | 0.31732 (7) | -0.16139 (7) | 0.2477 (2) | 0.0876 (9) | |
| H32A | 0.2983 | -0.1865 | 0.2822 | 0.131* | |

| | | | | |
|------|-------------|--------------|--------------|-------------|
| H32B | 0.3380 | -0.1639 | 0.2085 | 0.131* |
| H32C | 0.3044 | -0.1533 | 0.1894 | 0.131* |
| C33 | 0.44799 (6) | -0.10687 (6) | 0.2791 (2) | 0.0670 (6) |
| H33A | 0.4490 | -0.1309 | 0.2889 | 0.100* |
| H33B | 0.4625 | -0.0888 | 0.3437 | 0.100* |
| H33C | 0.4597 | -0.0948 | 0.2028 | 0.100* |
| O4 | 0.31014 (5) | 0.11716 (5) | 0.52433 (13) | 0.0667 (4) |
| H4A | 0.3088 (8) | 0.1132 (9) | 0.6004 (10) | 0.110 (11)* |
| H4B | 0.3348 (4) | 0.1306 (9) | 0.507 (2) | 0.122 (12)* |

Atomic displacement parameters (Å²)

| | U^{11} | U^{22} | U^{33} | U^{12} | U^{13} | U^{23} |
|-----|-------------|-------------|-------------|-------------|--------------|--------------|
| O1 | 0.0469 (7) | 0.0699 (9) | 0.0813 (10) | 0.0354 (7) | 0.0131 (7) | 0.0068 (7) |
| O2 | 0.0454 (7) | 0.0612 (8) | 0.0703 (9) | 0.0201 (6) | 0.0217 (6) | 0.0187 (7) |
| O3 | 0.0417 (6) | 0.0510 (7) | 0.0404 (6) | 0.0303 (5) | 0.0012 (5) | 0.0064 (5) |
| O5 | 0.0351 (6) | 0.0546 (7) | 0.0608 (7) | 0.0231 (6) | 0.0113 (5) | 0.0135 (6) |
| O6 | 0.0465 (7) | 0.0408 (6) | 0.0553 (7) | 0.0161 (5) | 0.0091 (5) | 0.0048 (5) |
| O7 | 0.0459 (7) | 0.0424 (7) | 0.0841 (9) | 0.0262 (6) | 0.0031 (6) | 0.0014 (6) |
| N1 | 0.0389 (7) | 0.0463 (8) | 0.0349 (6) | 0.0247 (6) | 0.0067 (5) | 0.0077 (5) |
| N16 | 0.0442 (7) | 0.0492 (8) | 0.0330 (6) | 0.0281 (7) | -0.0007 (5) | 0.0077 (6) |
| C2 | 0.0381 (8) | 0.0459 (9) | 0.0357 (7) | 0.0226 (7) | 0.0052 (6) | 0.0112 (6) |
| C3 | 0.0355 (8) | 0.0409 (8) | 0.0398 (8) | 0.0218 (7) | 0.0056 (6) | 0.0095 (6) |
| C4 | 0.0295 (7) | 0.0374 (8) | 0.0331 (7) | 0.0184 (6) | 0.0013 (5) | 0.0054 (6) |
| C5 | 0.0321 (7) | 0.0427 (8) | 0.0307 (7) | 0.0219 (7) | 0.0010 (5) | 0.0049 (6) |
| C6 | 0.0409 (9) | 0.0566 (11) | 0.0407 (8) | 0.0231 (8) | 0.0088 (7) | 0.0068 (7) |
| C7 | 0.043 (2) | 0.079 (2) | 0.078 (3) | 0.0183 (18) | 0.019 (2) | 0.013 (2) |
| C8 | 0.073 (2) | 0.080 (3) | 0.164 (5) | 0.0039 (19) | 0.033 (3) | 0.003 (3) |
| C7A | 0.050 (7) | 0.077 (8) | 0.069 (8) | 0.026 (5) | 0.003 (6) | 0.024 (6) |
| C8A | 0.089 (8) | 0.114 (10) | 0.111 (10) | 0.042 (7) | 0.006 (7) | -0.015 (7) |
| C9 | 0.0394 (9) | 0.0424 (9) | 0.0493 (9) | 0.0146 (7) | 0.0091 (7) | 0.0039 (7) |
| C10 | 0.0583 (13) | 0.0541 (13) | 0.134 (2) | 0.0244 (11) | -0.0018 (14) | -0.0212 (14) |
| C11 | 0.085 (2) | 0.0677 (18) | 0.199 (4) | 0.0227 (16) | -0.010 (2) | -0.057 (2) |
| C12 | 0.0620 (16) | 0.101 (2) | 0.129 (3) | 0.0137 (16) | -0.0146 (16) | -0.050 (2) |
| C13 | 0.0464 (12) | 0.0876 (18) | 0.0723 (14) | 0.0127 (11) | -0.0080 (10) | -0.0029 (12) |
| C14 | 0.0428 (10) | 0.0570 (11) | 0.0559 (10) | 0.0170 (9) | -0.0032 (8) | 0.0038 (8) |
| C15 | 0.0306 (7) | 0.0368 (8) | 0.0356 (7) | 0.0165 (6) | -0.0016 (5) | 0.0049 (6) |
| C17 | 0.0352 (8) | 0.0393 (8) | 0.0357 (7) | 0.0181 (7) | 0.0010 (6) | 0.0039 (6) |
| C18 | 0.0512 (10) | 0.0565 (10) | 0.0331 (7) | 0.0271 (8) | 0.0021 (7) | 0.0032 (7) |
| C19 | 0.0534 (10) | 0.0622 (11) | 0.0385 (8) | 0.0304 (9) | 0.0083 (7) | -0.0018 (7) |
| C20 | 0.0424 (9) | 0.0494 (9) | 0.0472 (9) | 0.0268 (8) | 0.0073 (7) | 0.0005 (7) |
| C21 | 0.0333 (7) | 0.0380 (8) | 0.0387 (7) | 0.0180 (6) | 0.0038 (6) | 0.0037 (6) |
| C22 | 0.0305 (7) | 0.0347 (7) | 0.0322 (7) | 0.0154 (6) | 0.0020 (5) | 0.0036 (5) |
| C23 | 0.0446 (9) | 0.0526 (10) | 0.0425 (8) | 0.0324 (8) | 0.0042 (7) | 0.0052 (7) |
| C24 | 0.0934 (17) | 0.149 (2) | 0.0544 (12) | 0.0973 (19) | 0.0075 (11) | 0.0120 (13) |
| C25 | 0.0329 (7) | 0.0418 (8) | 0.0292 (6) | 0.0193 (6) | -0.0002 (5) | 0.0013 (6) |
| C26 | 0.0367 (8) | 0.0432 (8) | 0.0340 (7) | 0.0236 (7) | 0.0018 (6) | 0.0034 (6) |
| C27 | 0.0317 (7) | 0.0483 (9) | 0.0321 (7) | 0.0194 (7) | 0.0031 (5) | 0.0020 (6) |

| | | | | | | |
|-----|-------------|-------------|-------------|-------------|-------------|--------------|
| C28 | 0.0381 (8) | 0.0400 (8) | 0.0355 (7) | 0.0164 (7) | 0.0023 (6) | 0.0005 (6) |
| C29 | 0.0406 (8) | 0.0410 (8) | 0.0418 (8) | 0.0226 (7) | -0.0008 (6) | -0.0028 (6) |
| C30 | 0.0334 (8) | 0.0447 (9) | 0.0432 (8) | 0.0213 (7) | 0.0017 (6) | -0.0012 (6) |
| C31 | 0.0411 (9) | 0.0628 (12) | 0.0665 (11) | 0.0310 (9) | 0.0115 (8) | 0.0136 (9) |
| C32 | 0.0689 (15) | 0.0558 (13) | 0.0984 (18) | 0.0013 (11) | 0.0245 (13) | -0.0236 (12) |
| C33 | 0.0528 (12) | 0.0571 (12) | 0.1029 (17) | 0.0365 (10) | 0.0078 (11) | 0.0063 (11) |
| O4 | 0.0785 (11) | 0.0625 (9) | 0.0496 (8) | 0.0280 (8) | 0.0013 (7) | 0.0025 (7) |

Geometric parameters (Å, °)

| | | | |
|---------|-------------|----------|-----------|
| O1—C6 | 1.198 (2) | C10—C11 | 1.387 (4) |
| O2—C6 | 1.323 (2) | C10—H10 | 0.93 |
| O2—C7A | 1.39 (2) | C11—C12 | 1.367 (5) |
| O2—C7 | 1.479 (6) | C11—H11 | 0.93 |
| O3—C15 | 1.2281 (18) | C12—C13 | 1.349 (4) |
| O5—C27 | 1.3632 (18) | C12—H12 | 0.93 |
| O5—C31 | 1.417 (2) | C13—C14 | 1.379 (3) |
| O6—C28 | 1.3774 (19) | C13—H13 | 0.93 |
| O6—C32 | 1.409 (3) | C14—H14 | 0.93 |
| O7—C29 | 1.3618 (19) | C17—C18 | 1.374 (2) |
| O7—C33 | 1.414 (2) | C17—C22 | 1.396 (2) |
| N1—C2 | 1.459 (2) | C18—C19 | 1.382 (3) |
| N1—C5 | 1.4639 (18) | C18—H18 | 0.93 |
| N1—H1 | 0.864 (19) | C19—C20 | 1.377 (2) |
| N16—C15 | 1.3397 (19) | C19—H19 | 0.93 |
| N16—C17 | 1.401 (2) | C20—C21 | 1.405 (2) |
| N16—H16 | 0.87 (2) | C20—H20 | 0.93 |
| C2—C6 | 1.506 (2) | C21—C22 | 1.392 (2) |
| C2—C3 | 1.575 (2) | C21—C23 | 1.475 (2) |
| C2—H2 | 0.98 | C23—C24 | 1.288 (3) |
| C3—C9 | 1.513 (2) | C23—H23 | 0.93 |
| C3—C4 | 1.557 (2) | C24—H24A | 0.93 |
| C3—H3 | 0.98 | C24—H24B | 0.93 |
| C4—C22 | 1.514 (2) | C25—C30 | 1.386 (2) |
| C4—C15 | 1.5344 (19) | C25—C26 | 1.391 (2) |
| C4—C5 | 1.568 (2) | C26—C27 | 1.382 (2) |
| C5—C25 | 1.509 (2) | C26—H26 | 0.93 |
| C5—H5 | 0.98 | C27—C28 | 1.387 (2) |
| C7—C8 | 1.514 (4) | C28—C29 | 1.393 (2) |
| C7—H7A | 0.97 | C29—C30 | 1.391 (2) |
| C7—H7B | 0.97 | C30—H30 | 0.93 |
| C8—H8A | 0.96 | C31—H31A | 0.96 |
| C8—H8B | 0.96 | C31—H31B | 0.96 |
| C8—H8C | 0.96 | C31—H31C | 0.96 |
| C7A—C8A | 1.532 (5) | C32—H32A | 0.96 |
| C7A—H7C | 0.97 | C32—H32B | 0.96 |
| C7A—H7D | 0.97 | C32—H32C | 0.96 |
| C8A—H8D | 0.96 | C33—H33A | 0.96 |

| | | | |
|-------------|-------------|---------------|-------------|
| C8A—H8E | 0.96 | C33—H33B | 0.96 |
| C8A—H8F | 0.96 | C33—H33C | 0.96 |
| C9—C10 | 1.373 (3) | O4—H4A | 0.850 (10) |
| C9—C14 | 1.387 (3) | O4—H4B | 0.85 (3) |
| C6—O2—C7A | 124.8 (11) | C13—C12—H12 | 120.2 |
| C6—O2—C7 | 114.3 (2) | C11—C12—H12 | 120.2 |
| C7A—O2—C7 | 20.0 (6) | C12—C13—C14 | 120.6 (3) |
| C27—O5—C31 | 117.10 (13) | C12—C13—H13 | 119.7 |
| C28—O6—C32 | 114.69 (15) | C14—C13—H13 | 119.7 |
| C29—O7—C33 | 118.19 (14) | C13—C14—C9 | 121.1 (2) |
| C2—N1—C5 | 105.16 (12) | C13—C14—H14 | 119.4 |
| C2—N1—H1 | 108.6 (12) | C9—C14—H14 | 119.4 |
| C5—N1—H1 | 109.4 (12) | O3—C15—N16 | 125.83 (13) |
| C15—N16—C17 | 112.06 (12) | O3—C15—C4 | 125.81 (13) |
| C15—N16—H16 | 120.3 (13) | N16—C15—C4 | 108.32 (12) |
| C17—N16—H16 | 127.2 (13) | C18—C17—C22 | 123.46 (15) |
| N1—C2—C6 | 112.10 (14) | C18—C17—N16 | 127.07 (14) |
| N1—C2—C3 | 108.17 (11) | C22—C17—N16 | 109.45 (13) |
| C6—C2—C3 | 114.37 (13) | C17—C18—C19 | 116.65 (15) |
| N1—C2—H2 | 107.3 | C17—C18—H18 | 121.7 |
| C6—C2—H2 | 107.3 | C19—C18—H18 | 121.7 |
| C3—C2—H2 | 107.3 | C20—C19—C18 | 121.46 (15) |
| C9—C3—C4 | 116.95 (12) | C20—C19—H19 | 119.3 |
| C9—C3—C2 | 119.65 (13) | C18—C19—H19 | 119.3 |
| C4—C3—C2 | 103.33 (12) | C19—C20—C21 | 121.92 (16) |
| C9—C3—H3 | 105.2 | C19—C20—H20 | 119.0 |
| C4—C3—H3 | 105.2 | C21—C20—H20 | 119.0 |
| C2—C3—H3 | 105.2 | C22—C21—C20 | 116.93 (14) |
| C22—C4—C15 | 102.14 (11) | C22—C21—C23 | 123.17 (13) |
| C22—C4—C3 | 113.14 (12) | C20—C21—C23 | 119.89 (14) |
| C15—C4—C3 | 113.80 (12) | C21—C22—C17 | 119.51 (13) |
| C22—C4—C5 | 119.27 (12) | C21—C22—C4 | 132.54 (13) |
| C15—C4—C5 | 108.28 (12) | C17—C22—C4 | 107.86 (12) |
| C3—C4—C5 | 100.64 (11) | C24—C23—C21 | 126.75 (17) |
| N1—C5—C25 | 115.29 (12) | C24—C23—H23 | 116.6 |
| N1—C5—C4 | 104.29 (11) | C21—C23—H23 | 116.6 |
| C25—C5—C4 | 116.34 (11) | C23—C24—H24A | 120.0 |
| N1—C5—H5 | 106.8 | C23—C24—H24B | 120.0 |
| C25—C5—H5 | 106.8 | H24A—C24—H24B | 120.0 |
| C4—C5—H5 | 106.8 | C30—C25—C26 | 119.79 (14) |
| O1—C6—O2 | 124.28 (16) | C30—C25—C5 | 123.19 (13) |
| O1—C6—C2 | 125.63 (16) | C26—C25—C5 | 117.02 (13) |
| O2—C6—C2 | 110.08 (15) | C27—C26—C25 | 120.38 (14) |
| O2—C7—C8 | 104.1 (4) | C27—C26—H26 | 119.8 |
| O2—C7—H7A | 110.9 | C25—C26—H26 | 119.8 |
| C8—C7—H7A | 110.9 | O5—C27—C26 | 124.30 (14) |
| O2—C7—H7B | 110.9 | O5—C27—C28 | 115.41 (14) |

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|--------------|--------------|-----------------|--------------|
| C8—C7—H7B | 110.9 | C26—C27—C28 | 120.29 (14) |
| H7A—C7—H7B | 109.0 | O6—C28—C27 | 119.52 (14) |
| C7—C8—H8A | 109.5 | O6—C28—C29 | 121.14 (15) |
| C7—C8—H8B | 109.5 | C27—C28—C29 | 119.28 (14) |
| H8A—C8—H8B | 109.5 | O7—C29—C30 | 124.38 (14) |
| C7—C8—H8C | 109.5 | O7—C29—C28 | 115.04 (14) |
| H8A—C8—H8C | 109.5 | C30—C29—C28 | 120.56 (14) |
| H8B—C8—H8C | 109.5 | C25—C30—C29 | 119.65 (14) |
| O2—C7A—C8A | 106.2 (15) | C25—C30—H30 | 120.2 |
| O2—C7A—H7C | 110.5 | C29—C30—H30 | 120.2 |
| C8A—C7A—H7C | 110.5 | O5—C31—H31A | 109.5 |
| O2—C7A—H7D | 110.5 | O5—C31—H31B | 109.5 |
| C8A—C7A—H7D | 110.5 | H31A—C31—H31B | 109.5 |
| H7C—C7A—H7D | 108.7 | O5—C31—H31C | 109.5 |
| C7A—C8A—H8D | 109.5 | H31A—C31—H31C | 109.5 |
| C7A—C8A—H8E | 109.5 | H31B—C31—H31C | 109.5 |
| H8D—C8A—H8E | 109.5 | O6—C32—H32A | 109.5 |
| C7A—C8A—H8F | 109.5 | O6—C32—H32B | 109.5 |
| H8D—C8A—H8F | 109.5 | H32A—C32—H32B | 109.5 |
| H8E—C8A—H8F | 109.5 | O6—C32—H32C | 109.5 |
| C10—C9—C14 | 117.30 (19) | H32A—C32—H32C | 109.5 |
| C10—C9—C3 | 118.31 (18) | H32B—C32—H32C | 109.5 |
| C14—C9—C3 | 124.39 (16) | O7—C33—H33A | 109.5 |
| C9—C10—C11 | 121.2 (3) | O7—C33—H33B | 109.5 |
| C9—C10—H10 | 119.4 | H33A—C33—H33B | 109.5 |
| C11—C10—H10 | 119.4 | O7—C33—H33C | 109.5 |
| C12—C11—C10 | 120.1 (3) | H33A—C33—H33C | 109.5 |
| C12—C11—H11 | 120.0 | H33B—C33—H33C | 109.5 |
| C10—C11—H11 | 120.0 | H4A—O4—H4B | 106.1 (15) |
| C13—C12—C11 | 119.7 (3) | | |
| C5—N1—C2—C6 | 149.93 (13) | C5—C4—C15—N16 | -128.16 (13) |
| C5—N1—C2—C3 | 22.92 (15) | C15—N16—C17—C18 | -175.19 (16) |
| N1—C2—C3—C9 | 135.98 (14) | C15—N16—C17—C22 | 3.36 (18) |
| C6—C2—C3—C9 | 10.3 (2) | C22—C17—C18—C19 | 0.3 (3) |
| N1—C2—C3—C4 | 3.84 (15) | N16—C17—C18—C19 | 178.67 (16) |
| C6—C2—C3—C4 | -121.84 (14) | C17—C18—C19—C20 | 1.1 (3) |
| C9—C3—C4—C22 | 71.14 (17) | C18—C19—C20—C21 | -0.4 (3) |
| C2—C3—C4—C22 | -155.16 (12) | C19—C20—C21—C22 | -1.7 (2) |
| C9—C3—C4—C15 | -44.87 (18) | C19—C20—C21—C23 | 177.00 (16) |
| C2—C3—C4—C15 | 88.82 (14) | C20—C21—C22—C17 | 3.0 (2) |
| C9—C3—C4—C5 | -160.45 (13) | C23—C21—C22—C17 | -175.60 (15) |
| C2—C3—C4—C5 | -26.75 (13) | C20—C21—C22—C4 | -172.97 (15) |
| C2—N1—C5—C25 | -169.62 (12) | C23—C21—C22—C4 | 8.4 (3) |
| C2—N1—C5—C4 | -40.81 (14) | C18—C17—C22—C21 | -2.5 (2) |
| C22—C4—C5—N1 | 166.34 (12) | N16—C17—C22—C21 | 178.91 (13) |
| C15—C4—C5—N1 | -77.61 (13) | C18—C17—C22—C4 | 174.44 (15) |
| C3—C4—C5—N1 | 42.03 (13) | N16—C17—C22—C4 | -4.17 (17) |

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|-----------------|--------------|-----------------|--------------|
| C22—C4—C5—C25 | -65.48 (17) | C15—C4—C22—C21 | 179.73 (16) |
| C15—C4—C5—C25 | 50.57 (16) | C3—C4—C22—C21 | 57.0 (2) |
| C3—C4—C5—C25 | 170.21 (12) | C5—C4—C22—C21 | -61.0 (2) |
| C7A—O2—C6—O1 | 11.2 (8) | C15—C4—C22—C17 | 3.37 (15) |
| C7—O2—C6—O1 | -8.5 (4) | C3—C4—C22—C17 | -119.37 (13) |
| C7A—O2—C6—C2 | -170.0 (7) | C5—C4—C22—C17 | 122.61 (14) |
| C7—O2—C6—C2 | 170.3 (3) | C22—C21—C23—C24 | 167.1 (2) |
| N1—C2—C6—O1 | -20.7 (2) | C20—C21—C23—C24 | -11.5 (3) |
| C3—C2—C6—O1 | 102.9 (2) | N1—C5—C25—C30 | 33.13 (19) |
| N1—C2—C6—O2 | 160.55 (14) | C4—C5—C25—C30 | -89.46 (17) |
| C3—C2—C6—O2 | -75.85 (17) | N1—C5—C25—C26 | -146.98 (13) |
| C6—O2—C7—C8 | -170.3 (4) | C4—C5—C25—C26 | 90.44 (16) |
| C7A—O2—C7—C8 | 64 (3) | C30—C25—C26—C27 | 0.4 (2) |
| C6—O2—C7A—C8A | -100.1 (19) | C5—C25—C26—C27 | -179.48 (13) |
| C7—O2—C7A—C8A | -36 (2) | C31—O5—C27—C26 | 9.3 (2) |
| C4—C3—C9—C10 | -113.8 (2) | C31—O5—C27—C28 | -170.80 (15) |
| C2—C3—C9—C10 | 120.2 (2) | C25—C26—C27—O5 | -178.51 (14) |
| C4—C3—C9—C14 | 65.8 (2) | C25—C26—C27—C28 | 1.6 (2) |
| C2—C3—C9—C14 | -60.2 (2) | C32—O6—C28—C27 | 103.6 (2) |
| C14—C9—C10—C11 | 0.5 (4) | C32—O6—C28—C29 | -79.3 (2) |
| C3—C9—C10—C11 | -179.9 (3) | O5—C27—C28—O6 | -4.9 (2) |
| C9—C10—C11—C12 | -0.4 (5) | C26—C27—C28—O6 | 175.03 (14) |
| C10—C11—C12—C13 | -0.2 (6) | O5—C27—C28—C29 | 177.94 (14) |
| C11—C12—C13—C14 | 0.6 (5) | C26—C27—C28—C29 | -2.1 (2) |
| C12—C13—C14—C9 | -0.4 (4) | C33—O7—C29—C30 | 12.7 (3) |
| C10—C9—C14—C13 | -0.1 (3) | C33—O7—C29—C28 | -168.46 (17) |
| C3—C9—C14—C13 | -179.70 (18) | O6—C28—C29—O7 | 4.7 (2) |
| C17—N16—C15—O3 | -179.00 (15) | C27—C28—C29—O7 | -178.19 (14) |
| C17—N16—C15—C4 | -1.03 (17) | O6—C28—C29—C30 | -176.37 (14) |
| C22—C4—C15—O3 | 176.54 (14) | C27—C28—C29—C30 | 0.7 (2) |
| C3—C4—C15—O3 | -61.2 (2) | C26—C25—C30—C29 | -1.8 (2) |
| C5—C4—C15—O3 | 49.82 (19) | C5—C25—C30—C29 | 178.09 (14) |
| C22—C4—C15—N16 | -1.44 (15) | O7—C29—C30—C25 | -179.95 (15) |
| C3—C4—C15—N16 | 120.85 (14) | C28—C29—C30—C25 | 1.2 (2) |

Hydrogen-bond geometry (\AA , $^\circ$)

| $D-H\cdots A$ | $D-H$ | $H\cdots A$ | $D\cdots A$ | $D-H\cdots A$ |
|--------------------------------------|----------|-------------|-------------|---------------|
| N1—H1 \cdots O1 | 0.86 (2) | 2.419 (18) | 2.8088 (18) | 108 (1) |
| N1—H1 \cdots O3 | 0.86 (2) | 2.376 (18) | 2.9395 (17) | 123 (1) |
| O4—H4B \cdots O1 ⁱ | 0.85 (3) | 2.14 (2) | 2.909 (2) | 150 (3) |
| N16—H16 \cdots O3 ⁱⁱ | 0.87 (2) | 1.99 (2) | 2.8449 (17) | 166 (2) |
| C5—H5 \cdots O5 ⁱⁱⁱ | 0.98 | 2.45 | 3.3197 (18) | 147 |
| C18—H18 \cdots O4 ^{iv} | 0.93 | 2.46 | 3.357 (2) | 162 |
| C24—H24B \cdots Cg1 ⁱⁱⁱ | 0.93 | 2.93 | 3.776 (2) | 153 |

Symmetry codes: (i) $y+1/3, -x+y+2/3, -z+2/3$; (ii) $-x+1, -y, -z+1$; (iii) $-y+1/3, x-y-1/3, z-1/3$; (iv) $-x+y+2/3, -x+1/3, z+1/3$.