

Acta Crystallographica Section E

Structure Reports

Online

ISSN 1600-5368

8,9-Isopropylidenedioxy-3-*p*-tolyl-1,6-dioxa-3-azaspiro[4.5]decane-2,10-dione

 Chun-Sheng Ling,^a Qiang Wu^{b*} and Shan-Shan Li^c

^aPharmaceutical College of Henan University, Kaifeng 475004, People's Republic of China, ^bInstitute of Pharmacy, Henan University, Kaifeng 475004, People's Republic of China, and ^cCollege of Chemistry and Environmental Engineering, Beijing Technology and Business University, Beijing 100037, People's Republic of China
Correspondence e-mail: ysywu@126.com

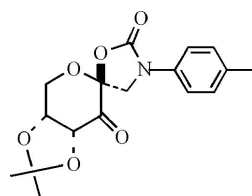
Received 13 October 2008; accepted 20 October 2008

Key indicators: single-crystal X-ray study; $T = 296$ K; mean $\sigma(\text{C}-\text{C}) = 0.004$ Å; R factor = 0.034; wR factor = 0.075; data-to-parameter ratio = 8.2.

In the title compound, $\text{C}_{17}\text{H}_{19}\text{NO}_6$, which may serve as a ketone catalyst for the asymmetric epoxidation of olefins, the crystal packing is consolidated by $\text{C}-\text{H}\cdots\text{O}$ interactions.

Related literature

For general background, see: Denmark & Wu (1999); Shi (2004); Yang (2004). For the synthesis, see: Zhao *et al.* (2006).



Experimental

Crystal data

$\text{C}_{17}\text{H}_{19}\text{NO}_6$
 $M_r = 333.33$
 Monoclinic, $P2_1$
 $a = 11.1268$ (8) Å
 $b = 6.3163$ (5) Å
 $c = 11.8697$ (8) Å
 $\beta = 94.084$ (1)°

$V = 832.09$ (11) Å³
 $Z = 2$
 Mo $K\alpha$ radiation
 $\mu = 0.10$ mm⁻¹
 $T = 296$ (2) K
 $0.18 \times 0.15 \times 0.13$ mm

Data collection

Bruker SMART CCD diffractometer
 Absorption correction: multi-scan (SADABS; Sheldrick, 2001)
 $T_{\min} = 0.982$, $T_{\max} = 0.987$

8821 measured reflections
 1795 independent reflections
 1389 reflections with $I > 2\sigma(I)$
 $R_{\text{int}} = 0.028$

Refinement

$R[F^2 > 2\sigma(F^2)] = 0.034$
 $wR(F^2) = 0.075$
 $S = 1.06$
 1795 reflections
 220 parameters

1 restraint
 H-atom parameters constrained
 $\Delta\rho_{\max} = 0.10$ e Å⁻³
 $\Delta\rho_{\min} = -0.13$ e Å⁻³

Table 1
 Hydrogen-bond geometry (Å, °).

$D-H\cdots A$	$D-H$	$H\cdots A$	$D\cdots A$	$D-H\cdots A$
$\text{C9}-\text{H9B}\cdots\text{O1}^{\text{i}}$	0.97	2.54	3.093 (3)	116
$\text{C14}-\text{H14B}\cdots\text{O4}^{\text{ii}}$	0.97	2.55	3.426 (3)	151

Symmetry codes: (i) $-x + 2, y - \frac{1}{2}, -z$; (ii) $-x + 1, y + \frac{1}{2}, -z$.

Data collection: SMART (Bruker, 2001); cell refinement: SAINT-Plus (Bruker, 2001); data reduction: SAINT-Plus; program(s) used to solve structure: SHELXS97 (Sheldrick, 2008); program(s) used to refine structure: SHELXL97 (Sheldrick, 2008); molecular graphics: PLATON (Spek, 2003); software used to prepare material for publication: PLATON.

This work was supported by the Basic Research Foundation for Natural Science of Henan University.

Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: HB2819).

References

- Bruker (2001). SAINT-Plus and SMART. Bruker AXS Inc., Madison, Wisconsin, USA.
 Denmark, S. E. & Wu, Z. (1999). *Synlett*, pp. 847–859.
 Sheldrick, G. M. (2001). SADABS. Bruker AXS Inc., Madison, Wisconsin, USA.
 Sheldrick, G. M. (2008). *Acta Cryst.* **A64**, 112–122.
 Shi, Y. (2004). *Acc. Chem. Res.* **37**, 488–496.
 Spek, A. L. (2003). *J. Appl. Cryst.* **36**, 7–13.
 Yang, D. (2004). *Acc. Chem. Res.* **37**, 497–505.
 Zhao, M.-X., Goeddel, D., Li, K. & Shi, Y. (2006). *Tetrahedron*, **62**, 8064–8068.

supporting information

Acta Cryst. (2008). E64, o2181 [doi:10.1107/S1600536808034259]

8,9-Isopropylidenedioxy-3-*p*-tolyl-1,6-dioxaspiro[4.5]decane-2,10-dione

Chun-Sheng Ling, Qiang Wu and Shan-Shan Li

S1. Comment

Dioxiranes generated *in situ* from chiral ketones are effective for the asymmetric epoxidation of olefins (Denmark & Wu, 1999; Shi, 2004; Yang, 2004). As part of our own studies in this area, we now report the synthesis and structure of the title compound, (I).

The compound (I) consists of a four-ring system, including a phenyl ring, a pyran ring, a dioxolane ring and an oxazolidine ring, and which displays a chair molecular framework (Fig. 1). In the structure of (I), the S(6) ring of O4/C10/C11/C12/C13/C14 is nonplanar, characterized by a O4–C10–C11–C12 torsion angle of 61.1 (2)°. The stereogenic centres C10, C12 and C13 were assigned R, S, and S configurations, respectively.

In the crystal, some short C—H···O interactions (Table 1) may help to establish the packing (Fig. 2).

S2. Experimental

The title compound was made by the method of Zhao *et al.* (2006), starting from D-glucose and 4-methyl-benzenamine to yield colorless blocks of (I). The molecular formula, C₁₇H₁₉NO₆, was established by ESI-MS, *m/z*: 356(*M*+Na), 334(*M*+H), 232, 204, 108. Spectroscopic analysis, ¹H NMR (400 MHz, CDCl₃): δ 7.40 (d, *J*=5.4 Hz, 2H, ArH), 7.19 (d, *J*=5.4 Hz, 2H, ArH), 4.87 (d, *J*=5.7 Hz, 1H), 4.74 (d, *J*=10.2 Hz, 1H), 4.66–4.61 (m, 2H), 4.27 (d, *J*=13.8 Hz, 1H), 3.74 (d, *J*=10.2 Hz, 1H), 2.34 (s, 3H, ArCH₃), 1.49 (s, 3H, –CH₃), 1.44 (s, 3H, –CH₃).

S3. Refinement

Anomalous dispersion was negligible and Friedel pairs were merged before refinement. The H atoms were geometrically placed (C—H = 0.93–0.98 Å) and refined as riding with $U_{\text{iso}}(\text{H}) = 1.2U_{\text{eq}}(\text{C})$ 1.5 $U_{\text{eq}}(\text{methyl C})$.

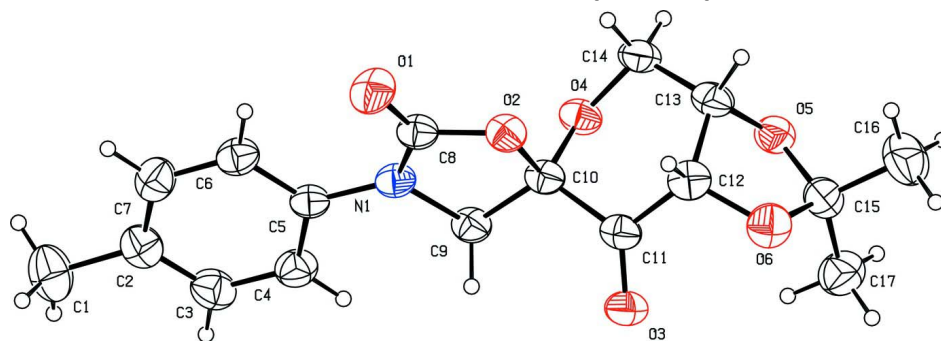


Figure 1

The molecular structure of (I) with displacement ellipsoids for the non-hydrogen atoms drawn at the 50% probability level.

8,9-Isopropylidenedioxy-3-p-tolyl-1,6-dioxo-3-azaspiro[4.5]decane-2,10-dione*Crystal data*C₁₇H₁₉NO₆ $M_r = 333.33$ Monoclinic, $P2_1$ Hall symbol: $P\ 2y\ b$ $a = 11.1268\ (8)\ \text{\AA}$ $b = 6.3163\ (5)\ \text{\AA}$ $c = 11.8697\ (8)\ \text{\AA}$ $\beta = 94.084\ (1)^\circ$ $V = 832.09\ (11)\ \text{\AA}^3$ $Z = 2$ $F(000) = 352$ $D_x = 1.330\ \text{Mg m}^{-3}$ Mo $K\alpha$ radiation, $\lambda = 0.71073\ \text{\AA}$

Cell parameters from 2427 reflections

 $\theta = 2.4\text{--}21.1^\circ$ $\mu = 0.10\ \text{mm}^{-1}$ $T = 296\ \text{K}$

Block, colorless

 $0.18 \times 0.15 \times 0.13\ \text{mm}$ *Data collection*

Bruker SMART CCD

diffractometer

Radiation source: fine-focus sealed tube

Graphite monochromator

 ω scans

Absorption correction: multi-scan

(SADABS; Sheldrick, 2001)

 $T_{\min} = 0.982$, $T_{\max} = 0.987$

8821 measured reflections

1795 independent reflections

1389 reflections with $I > 2\sigma(I)$ $R_{\text{int}} = 0.028$ $\theta_{\max} = 26.0^\circ$, $\theta_{\min} = 1.7^\circ$ $h = -13 \rightarrow 13$ $k = -7 \rightarrow 7$ $l = -14 \rightarrow 14$ *Refinement*Refinement on F^2

Least-squares matrix: full

 $R[F^2 > 2\sigma(F^2)] = 0.034$ $wR(F^2) = 0.075$ $S = 1.06$

1795 reflections

220 parameters

1 restraint

Primary atom site location: structure-invariant

direct methods

Secondary atom site location: difference Fourier map

Hydrogen site location: inferred from neighbouring sites

H-atom parameters constrained

 $w = 1/[\sigma^2(F_o^2) + (0.0303P)^2 + 0.0894P]$ where $P = (F_o^2 + 2F_c^2)/3$ $(\Delta/\sigma)_{\max} < 0.001$ $\Delta\rho_{\max} = 0.10\ \text{e \AA}^{-3}$ $\Delta\rho_{\min} = -0.13\ \text{e \AA}^{-3}$ *Special details*

Geometry. All e.s.d.'s (except the e.s.d. in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell e.s.d.'s are taken into account individually in the estimation of e.s.d.'s in distances, angles and torsion angles; correlations between e.s.d.'s in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell e.s.d.'s is used for estimating e.s.d.'s involving l.s. planes.

Refinement. Refinement of F^2 against ALL reflections. The weighted R -factor wR and goodness of fit S are based on F^2 , conventional R -factors R are based on F , with F set to zero for negative F^2 . The threshold expression of $F^2 > \sigma(F^2)$ is used only for calculating R -factors(gt) *etc.* and is not relevant to the choice of reflections for refinement. R -factors based on F^2 are statistically about twice as large as those based on F , and R -factors based on ALL data will be even larger.

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (\AA^2)

	x	y	z	$U_{\text{iso}}^*/U_{\text{eq}}$
N1	0.89223 (17)	0.9579 (3)	0.12879 (17)	0.0543 (5)
O1	1.00555 (17)	1.2504 (3)	0.08456 (16)	0.0792 (6)
O2	0.84070 (14)	1.1560 (3)	-0.02227 (14)	0.0588 (5)
O3	0.70022 (18)	0.7396 (3)	-0.15295 (19)	0.0810 (6)

O4	0.64278 (14)	1.1097 (3)	0.02397 (14)	0.0600 (5)
O5	0.47051 (15)	1.1373 (3)	-0.18543 (15)	0.0665 (5)
O6	0.59983 (16)	1.0190 (4)	-0.31240 (15)	0.0766 (6)
C1	1.1700 (3)	0.5776 (7)	0.5018 (3)	0.0997 (12)
H1A	1.1778	0.6722	0.5652	0.150*
H1B	1.2484	0.5455	0.4775	0.150*
H1C	1.1316	0.4492	0.5235	0.150*
C2	1.0944 (3)	0.6815 (5)	0.4059 (2)	0.0705 (8)
C3	1.0017 (3)	0.5773 (5)	0.3486 (2)	0.0760 (8)
H3	0.9824	0.4413	0.3714	0.091*
C4	0.9355 (3)	0.6666 (5)	0.2580 (2)	0.0687 (8)
H4	0.8727	0.5907	0.2212	0.082*
C5	0.9620 (2)	0.8692 (4)	0.2214 (2)	0.0534 (6)
C6	1.0555 (2)	0.9785 (5)	0.2790 (2)	0.0648 (7)
H6	1.0751	1.1145	0.2567	0.078*
C7	1.1195 (2)	0.8837 (6)	0.3702 (2)	0.0727 (8)
H7	1.1815	0.9590	0.4086	0.087*
C8	0.9231 (2)	1.1294 (4)	0.0685 (2)	0.0563 (6)
C9	0.7901 (2)	0.8463 (4)	0.0715 (2)	0.0568 (6)
H9A	0.7294	0.8121	0.1234	0.068*
H9B	0.8156	0.7175	0.0357	0.068*
C10	0.74453 (19)	1.0086 (4)	-0.0148 (2)	0.0524 (6)
C11	0.7064 (2)	0.9266 (4)	-0.1330 (2)	0.0566 (6)
C12	0.6676 (2)	1.0984 (5)	-0.2160 (2)	0.0628 (7)
H12	0.7381	1.1759	-0.2389	0.075*
C13	0.5811 (2)	1.2495 (4)	-0.1640 (2)	0.0647 (7)
H13	0.5778	1.3830	-0.2062	0.078*
C14	0.6057 (2)	1.2941 (4)	-0.0408 (2)	0.0677 (7)
H14A	0.6684	1.4007	-0.0314	0.081*
H14B	0.5335	1.3517	-0.0114	0.081*
C15	0.4747 (2)	1.0412 (5)	-0.2935 (2)	0.0696 (8)
C16	0.4154 (3)	1.1789 (7)	-0.3854 (3)	0.1116 (13)
H16A	0.3316	1.1956	-0.3732	0.167*
H16B	0.4234	1.1136	-0.4576	0.167*
H16C	0.4537	1.3152	-0.3840	0.167*
C17	0.4167 (3)	0.8262 (6)	-0.2888 (3)	0.0914 (10)
H17A	0.4574	0.7441	-0.2296	0.137*
H17B	0.4222	0.7553	-0.3597	0.137*
H17C	0.3335	0.8422	-0.2739	0.137*

Atomic displacement parameters (\AA^2)

	U^{11}	U^{22}	U^{33}	U^{12}	U^{13}	U^{23}
N1	0.0565 (12)	0.0435 (12)	0.0639 (12)	-0.0099 (10)	0.0112 (10)	-0.0006 (10)
O1	0.0775 (12)	0.0680 (12)	0.0918 (13)	-0.0344 (12)	0.0032 (10)	0.0086 (11)
O2	0.0532 (9)	0.0492 (10)	0.0746 (11)	-0.0102 (8)	0.0097 (8)	0.0076 (9)
O3	0.0798 (13)	0.0514 (12)	0.1092 (15)	0.0067 (11)	-0.0109 (11)	-0.0152 (12)
O4	0.0554 (9)	0.0535 (11)	0.0729 (11)	0.0052 (9)	0.0179 (8)	0.0002 (9)

O5	0.0617 (10)	0.0617 (11)	0.0763 (12)	0.0064 (10)	0.0058 (8)	-0.0065 (11)
O6	0.0769 (12)	0.0874 (14)	0.0670 (11)	0.0026 (12)	0.0163 (10)	-0.0039 (11)
C1	0.087 (2)	0.131 (3)	0.082 (2)	0.014 (2)	0.0079 (17)	0.023 (2)
C2	0.0640 (17)	0.081 (2)	0.0681 (18)	0.0086 (16)	0.0169 (14)	0.0026 (16)
C3	0.089 (2)	0.0619 (19)	0.0775 (19)	0.0032 (17)	0.0113 (16)	0.0108 (16)
C4	0.0782 (18)	0.0552 (17)	0.0727 (18)	-0.0080 (15)	0.0053 (14)	-0.0008 (16)
C5	0.0565 (14)	0.0485 (15)	0.0563 (15)	0.0007 (12)	0.0131 (12)	-0.0047 (12)
C6	0.0637 (16)	0.0599 (17)	0.0719 (17)	-0.0069 (14)	0.0128 (14)	-0.0061 (15)
C7	0.0608 (17)	0.088 (2)	0.0695 (19)	-0.0023 (17)	0.0066 (14)	-0.0095 (17)
C8	0.0560 (14)	0.0470 (15)	0.0672 (15)	-0.0056 (14)	0.0142 (12)	-0.0002 (14)
C9	0.0501 (13)	0.0438 (13)	0.0771 (16)	-0.0094 (12)	0.0082 (12)	0.0038 (13)
C10	0.0439 (12)	0.0439 (13)	0.0708 (16)	-0.0061 (12)	0.0147 (11)	0.0010 (12)
C11	0.0418 (13)	0.0497 (15)	0.0799 (18)	-0.0014 (11)	0.0168 (12)	-0.0023 (14)
C12	0.0629 (15)	0.0604 (16)	0.0671 (16)	-0.0096 (14)	0.0182 (13)	0.0037 (14)
C13	0.0712 (17)	0.0429 (14)	0.0803 (19)	0.0024 (14)	0.0080 (14)	0.0057 (15)
C14	0.0699 (17)	0.0472 (16)	0.086 (2)	0.0098 (14)	0.0052 (14)	-0.0095 (15)
C15	0.0694 (18)	0.0691 (19)	0.0700 (18)	0.0084 (15)	0.0032 (14)	0.0037 (16)
C16	0.131 (3)	0.108 (3)	0.093 (2)	0.025 (3)	-0.012 (2)	0.023 (2)
C17	0.086 (2)	0.084 (3)	0.104 (2)	-0.009 (2)	0.0007 (18)	-0.012 (2)

Geometric parameters (Å, °)

N1—C8	1.356 (3)	C5—C6	1.387 (3)
N1—C5	1.415 (3)	C6—C7	1.389 (4)
N1—C9	1.463 (3)	C6—H6	0.9300
O1—C8	1.199 (3)	C7—H7	0.9300
O2—C8	1.374 (3)	C9—C10	1.511 (3)
O2—C10	1.426 (3)	C9—H9A	0.9700
O3—C11	1.205 (3)	C9—H9B	0.9700
O4—C10	1.405 (3)	C10—C11	1.528 (4)
O4—C14	1.440 (3)	C11—C12	1.508 (4)
O5—C15	1.423 (3)	C12—C13	1.517 (4)
O5—C13	1.427 (3)	C12—H12	0.9800
O6—C12	1.417 (3)	C13—C14	1.495 (4)
O6—C15	1.432 (3)	C13—H13	0.9800
C1—C2	1.516 (4)	C14—H14A	0.9700
C1—H1A	0.9600	C14—H14B	0.9700
C1—H1B	0.9600	C15—C17	1.506 (4)
C1—H1C	0.9600	C15—C16	1.510 (4)
C2—C3	1.364 (4)	C16—H16A	0.9600
C2—C7	1.380 (5)	C16—H16B	0.9600
C3—C4	1.380 (4)	C16—H16C	0.9600
C3—H3	0.9300	C17—H17A	0.9600
C4—C5	1.390 (4)	C17—H17B	0.9600
C4—H4	0.9300	C17—H17C	0.9600
C8—N1—C5	125.5 (2)	O4—C10—C11	106.10 (18)
C8—N1—C9	110.9 (2)	O2—C10—C11	108.94 (19)

C5—N1—C9	122.3 (2)	C9—C10—C11	116.8 (2)
C8—O2—C10	109.46 (18)	O3—C11—C12	124.6 (3)
C10—O4—C14	113.54 (19)	O3—C11—C10	121.4 (3)
C15—O5—C13	106.81 (19)	C12—C11—C10	113.8 (2)
C12—O6—C15	107.8 (2)	O6—C12—C11	112.6 (2)
C2—C1—H1A	109.5	O6—C12—C13	103.6 (2)
C2—C1—H1B	109.5	C11—C12—C13	110.3 (2)
H1A—C1—H1B	109.5	O6—C12—H12	110.1
C2—C1—H1C	109.5	C11—C12—H12	110.1
H1A—C1—H1C	109.5	C13—C12—H12	110.1
H1B—C1—H1C	109.5	O5—C13—C14	111.3 (2)
C3—C2—C7	117.1 (3)	O5—C13—C12	100.3 (2)
C3—C2—C1	121.7 (3)	C14—C13—C12	116.0 (2)
C7—C2—C1	121.2 (3)	O5—C13—H13	109.6
C2—C3—C4	122.3 (3)	C14—C13—H13	109.6
C2—C3—H3	118.9	C12—C13—H13	109.6
C4—C3—H3	118.9	O4—C14—C13	113.3 (2)
C3—C4—C5	120.4 (3)	O4—C14—H14A	108.9
C3—C4—H4	119.8	C13—C14—H14A	108.9
C5—C4—H4	119.8	O4—C14—H14B	108.9
C6—C5—C4	118.2 (3)	C13—C14—H14B	108.9
C6—C5—N1	122.4 (2)	H14A—C14—H14B	107.7
C4—C5—N1	119.4 (2)	O5—C15—O6	106.1 (2)
C5—C6—C7	119.7 (3)	O5—C15—C17	108.0 (3)
C5—C6—H6	120.2	O6—C15—C17	110.0 (2)
C7—C6—H6	120.2	O5—C15—C16	111.4 (3)
C2—C7—C6	122.3 (3)	O6—C15—C16	108.8 (3)
C2—C7—H7	118.9	C17—C15—C16	112.3 (3)
C6—C7—H7	118.9	C15—C16—H16A	109.5
O1—C8—N1	130.1 (2)	C15—C16—H16B	109.5
O1—C8—O2	120.5 (2)	H16A—C16—H16B	109.5
N1—C8—O2	109.4 (2)	C15—C16—H16C	109.5
N1—C9—C10	101.55 (19)	H16A—C16—H16C	109.5
N1—C9—H9A	111.5	H16B—C16—H16C	109.5
C10—C9—H9A	111.5	C15—C17—H17A	109.5
N1—C9—H9B	111.5	C15—C17—H17B	109.5
C10—C9—H9B	111.5	H17A—C17—H17B	109.5
H9A—C9—H9B	109.3	C15—C17—H17C	109.5
O4—C10—O2	110.5 (2)	H17A—C17—H17C	109.5
O4—C10—C9	109.00 (19)	H17B—C17—H17C	109.5
O2—C10—C9	105.54 (17)		
C7—C2—C3—C4	0.6 (4)	N1—C9—C10—O2	17.4 (2)
C1—C2—C3—C4	-177.1 (3)	N1—C9—C10—C11	138.56 (19)
C2—C3—C4—C5	0.2 (4)	O4—C10—C11—O3	-113.6 (3)
C3—C4—C5—C6	-0.6 (4)	O2—C10—C11—O3	127.4 (3)
C3—C4—C5—N1	-179.6 (2)	C9—C10—C11—O3	8.0 (3)
C8—N1—C5—C6	16.7 (4)	O4—C10—C11—C12	61.1 (2)

C9—N1—C5—C6	-177.0 (2)	O2—C10—C11—C12	-57.9 (2)
C8—N1—C5—C4	-164.3 (2)	C9—C10—C11—C12	-177.25 (19)
C9—N1—C5—C4	1.9 (3)	C15—O6—C12—C11	95.7 (3)
C4—C5—C6—C7	0.2 (4)	C15—O6—C12—C13	-23.5 (3)
N1—C5—C6—C7	179.2 (2)	O3—C11—C12—O6	12.3 (4)
C3—C2—C7—C6	-1.0 (4)	C10—C11—C12—O6	-162.21 (18)
C1—C2—C7—C6	176.7 (3)	O3—C11—C12—C13	127.5 (3)
C5—C6—C7—C2	0.6 (4)	C10—C11—C12—C13	-47.0 (3)
C5—N1—C8—O1	-7.5 (4)	C15—O5—C13—C14	-160.6 (2)
C9—N1—C8—O1	-175.0 (3)	C15—O5—C13—C12	-37.3 (2)
C5—N1—C8—O2	173.46 (19)	O6—C12—C13—O5	36.9 (2)
C9—N1—C8—O2	5.9 (3)	C11—C12—C13—O5	-83.8 (2)
C10—O2—C8—O1	-172.9 (2)	O6—C12—C13—C14	157.0 (2)
C10—O2—C8—N1	6.2 (3)	C11—C12—C13—C14	36.3 (3)
C8—N1—C9—C10	-14.5 (2)	C10—O4—C14—C13	56.3 (3)
C5—N1—C9—C10	177.48 (19)	O5—C13—C14—O4	73.7 (3)
C14—O4—C10—O2	53.3 (3)	C12—C13—C14—O4	-40.2 (3)
C14—O4—C10—C9	168.83 (19)	C13—O5—C15—O6	24.2 (3)
C14—O4—C10—C11	-64.6 (2)	C13—O5—C15—C17	142.0 (2)
C8—O2—C10—O4	102.5 (2)	C13—O5—C15—C16	-94.1 (3)
C8—O2—C10—C9	-15.2 (3)	C12—O6—C15—O5	0.8 (3)
C8—O2—C10—C11	-141.4 (2)	C12—O6—C15—C17	-115.8 (3)
N1—C9—C10—O4	-101.3 (2)	C12—O6—C15—C16	120.8 (3)

Hydrogen-bond geometry (\AA , $^\circ$)

$D-H\cdots A$	$D-H$	$H\cdots A$	$D\cdots A$	$D-H\cdots A$
C9—H9B \cdots O1 ⁱ	0.97	2.54	3.093 (3)	116
C14—H14B \cdots O4 ⁱⁱ	0.97	2.55	3.426 (3)	151

Symmetry codes: (i) $-x+2, y-1/2, -z$; (ii) $-x+1, y+1/2, -z$.