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N-Benzyl-2-(2-chloro-5-methylphenoxy)-acetamide

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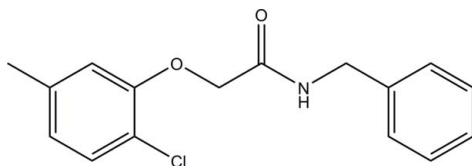
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Key indicators: single-crystal X-ray study; $T = 298$ K; mean $\sigma(\text{C}-\text{C}) = 0.006$ Å; R factor = 0.054; wR factor = 0.153; data-to-parameter ratio = 14.3.

The asymmetric unit of the title compound, $\text{C}_{16}\text{H}_{16}\text{ClNO}_2$, contains two crystallographically independent molecules, which differ mainly in the orientation of the benzyl group with respect to the rest of the molecule. In the crystal packing, centrosymmetrically related molecules are linked into dimers *via* intermolecular $\text{C}-\text{H}\cdots\text{O}$ hydrogen-bond interactions.

Related literature

For a related structure, see: Li *et al.* (2008). For hydrogen-bond motifs, see: Bernstein *et al.* (1995).



Experimental

Crystal data

$\text{C}_{16}\text{H}_{16}\text{ClNO}_2$
 $M_r = 289.75$

Triclinic, $P\bar{1}$
 $a = 9.5549$ (18) Å

$b = 11.086$ (2) Å
 $c = 14.725$ (3) Å
 $\alpha = 71.747$ (4)°
 $\beta = 89.062$ (4)°
 $\gamma = 85.655$ (4)°
 $V = 1477.0$ (5) Å³

$Z = 4$
Mo $K\alpha$ radiation
 $\mu = 0.26$ mm⁻¹
 $T = 298$ (2) K
 $0.12 \times 0.10 \times 0.06$ mm

Data collection

Bruker SMART CCD area-detector diffractometer
Absorption correction: multi-scan (SADABS; Bruker, 2005)
 $T_{\min} = 0.970$, $T_{\max} = 0.985$

7851 measured reflections
5194 independent reflections
2351 reflections with $I > 2\sigma(I)$
 $R_{\text{int}} = 0.033$

Refinement

$R[F^2 > 2\sigma(F^2)] = 0.054$
 $wR(F^2) = 0.153$
 $S = 0.98$
5194 reflections

362 parameters
H-atom parameters constrained
 $\Delta\rho_{\text{max}} = 0.18$ e Å⁻³
 $\Delta\rho_{\text{min}} = -0.23$ e Å⁻³

Table 1

Hydrogen-bond geometry (Å, °).

$D-H\cdots A$	$D-H$	$H\cdots A$	$D\cdots A$	$D-H\cdots A$
$\text{C3}-\text{H3}\cdots\text{O4}^i$	0.93	2.51	3.423 (4)	169
$\text{C19}-\text{H19}\cdots\text{O2}^i$	0.93	2.37	3.287 (4)	169

Symmetry code: (i) $-x + 1, -y, -z + 1$.

Data collection: SMART (Bruker, 2005); cell refinement: SAINT (Bruker, 2005); data reduction: SAINT; program(s) used to solve structure: SHELXS97 (Sheldrick, 2008); program(s) used to refine structure: SHELXL97 (Sheldrick, 2008); molecular graphics: XP in SHELXTL (Sheldrick, 2008); software used to prepare material for publication: SHELXL97.

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Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: RZ2267).

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Li, Z.-B., Luo, Y.-H., Dong, W.-L., Li, J. & Zuo, H. (2008). *Acta Cryst.* **E64**, o1610.
Sheldrick, G. M. (2008). *Acta Cryst.* **A64**, 112–122.

supporting information

Acta Cryst. (2008). E64, o2426 [doi:10.1107/S1600536808038555]

N*-Benzyl-2-(2-chloro-5-methylphenoxy)acetamide*Zhu-Bo Li, Xiao-Yan He, Yong-Sheng Xie and Guang-Cheng Qin****S1. Comment**

As part of our continuing project on the study of the interactions occurring between small molecules and proteins (Li *et al.*; 2008), we report here the synthesis and crystal structure of the title compound.

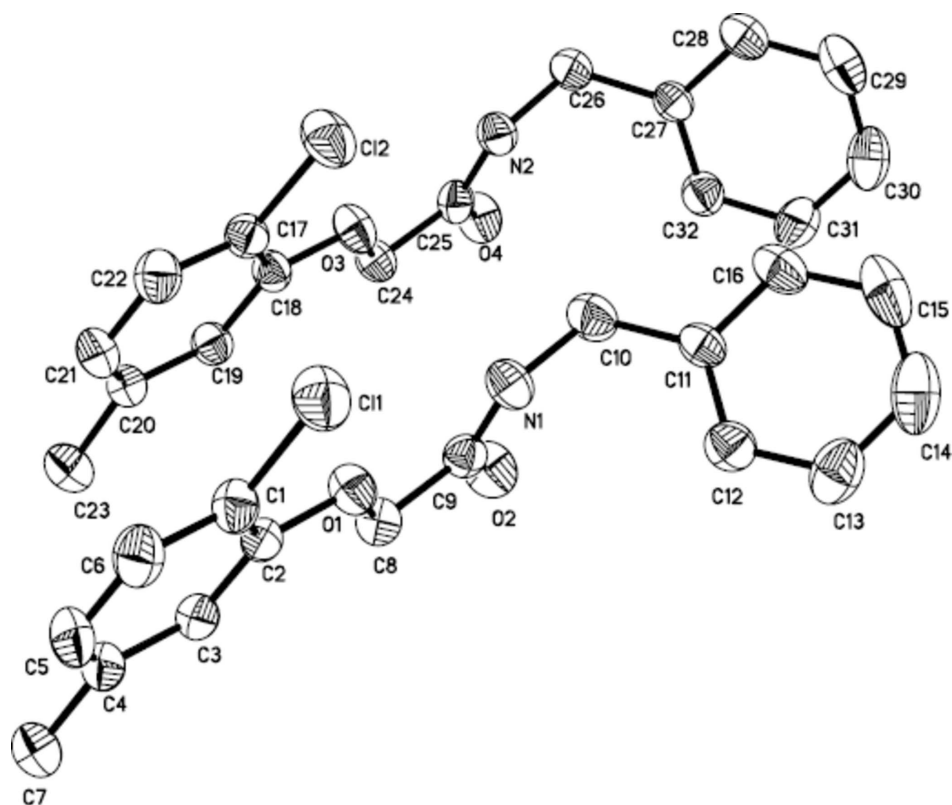
The asymmetric unit of the title compound (Fig. 1) contains two crystallographically independent molecules which differ mainly in the orientation of the benzyl groups (C10—C15 and C26—C32) bound to the amidic N atoms with respect of the mean plane through the rest of the molecule (dihedral angles of 87.03 (8) and 68.74 (6)° respectively). The O1—C8—C9—O2 and O3—C24—C25—O4 torsion angles are -176.5 (3) and -179.0 (3)°, respectively. These values can be compared with that of 10.5 (3)° found in the similar dichloro compound (Li *et al.*; 2008). In the crystal structure, centrosymmetrically related molecules are linked into dimers by intermolecular C—H···O hydrogen bonds (Table 1), forming fourteen-membered rings of graph set motif $R^2_2(14)$ (Bernstein *et al.*, 1995).

S2. Experimental

A solution of 2-chloro-5-methylphenol (1.0 mmol), *N*-benzyl-2-chloroacetamide (1.1 mmol), K₂CO₃ (1.1 mmol) and CH₃CN (20 ml) was refluxed for 3 h. After completion of the reaction (by TLC monitoring), the solution was cooled and the solvent was evaporated under reduced pressure. The residue was poured into water and adjusted to pH 6–7 with dilute hydrochloric acid (10%) and extracted with ethyl acetate, washed with brine and dried over anhydrous MgSO₄. The mixture was then filtered and the filtrate obtained was concentrated under reduced pressure to obtain the corresponding crude product. The product was purified by column chromatography on silica gel using ethyl acetate as eluent (yield 86%). Crystals suitable for X-ray diffraction were obtained by slow evaporation of an ethyl acetate/hexane (1:1 v/v) solution at room temperature for 10 days.

S3. Refinement

All H atoms were placed in calculated positions and refined as riding, with C—H = 0.93–0.97 Å, N—H = 0.86 Å and with $U_{\text{iso}}(\text{H}) = 1.2U_{\text{eq}}(\text{C}, \text{N})$ or $1.5U_{\text{eq}}(\text{C})$ for methyl H atoms.

**Figure 1**

The molecular structure of the title compound, showing displacement ellipsoids drawn at the 50% probability level for non-H atoms.

N-Benzyl-2-(2-chloro-5-methylphenoxy)acetamide

Crystal data

$C_{16}H_{16}ClNO_2$

$M_r = 289.75$

Triclinic, $P\bar{1}$

Hall symbol: -P 1

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$V = 1477.0$ (5) Å³

$Z = 4$

$F(000) = 608$

$D_x = 1.303$ Mg m⁻³

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Cell parameters from 818 reflections

$\theta = 5.5$ – 38.5 °

$\mu = 0.26$ mm⁻¹

$T = 298$ K

Block, colourless

$0.12 \times 0.10 \times 0.06$ mm

Data collection

Bruker SMART CCD area-detector
diffractometer

Radiation source: fine-focus sealed tube

Graphite monochromator

φ and ω scans

Absorption correction: multi-scan

(*SADABS*; Bruker, 2005)

$T_{\min} = 0.970$, $T_{\max} = 0.985$

7851 measured reflections

5194 independent reflections

2351 reflections with $I > 2\sigma(I)$

$R_{\text{int}} = 0.033$

$\theta_{\text{max}} = 25.1$ °, $\theta_{\text{min}} = 1.5$ °

$h = -11 \rightarrow 11$

$k = -12 \rightarrow 13$

$l = -14 \rightarrow 17$

*Refinement*Refinement on F^2

Least-squares matrix: full

 $R[F^2 > 2\sigma(F^2)] = 0.054$ $wR(F^2) = 0.153$ $S = 0.98$

5194 reflections

362 parameters

0 restraints

Primary atom site location: structure-invariant
direct methodsSecondary atom site location: difference Fourier
mapHydrogen site location: inferred from
neighbouring sites

H-atom parameters constrained

 $w = 1/[\sigma^2(F_o^2) + (0.0578P)^2]$ where $P = (F_o^2 + 2F_c^2)/3$ $(\Delta/\sigma)_{\max} < 0.001$ $\Delta\rho_{\max} = 0.18 \text{ e } \text{\AA}^{-3}$ $\Delta\rho_{\min} = -0.23 \text{ e } \text{\AA}^{-3}$ Extinction correction: *SHELXL97* (Sheldrick,
2008), $F_c^* = kF_c[1 + 0.001x F_c^2 \lambda^3 / \sin(2\theta)]^{-1/4}$

Extinction coefficient: 0.0039 (10)

Special details

Geometry. All e.s.d.'s (except the e.s.d. in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell e.s.d.'s are taken into account individually in the estimation of e.s.d.'s in distances, angles and torsion angles; correlations between e.s.d.'s in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell e.s.d.'s is used for estimating e.s.d.'s involving l.s. planes.

Refinement. Refinement of F^2 against ALL reflections. The weighted R -factor wR and goodness of fit S are based on F^2 , conventional R -factors R are based on F , with F set to zero for negative F^2 . The threshold expression of $F^2 > \sigma(F^2)$ is used only for calculating R -factors(gt) *etc.* and is not relevant to the choice of reflections for refinement. R -factors based on F^2 are statistically about twice as large as those based on F , and R -factors based on ALL data will be even larger.

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (\AA^2)

	x	y	z	$U_{\text{iso}}^*/U_{\text{eq}}$
Cl1	0.96319 (12)	0.34345 (9)	0.13691 (6)	0.0838 (4)
Cl2	0.50957 (12)	0.36619 (10)	0.12091 (6)	0.0843 (4)
O1	0.8583 (2)	0.1676 (2)	0.30620 (15)	0.0613 (7)
O2	0.6330 (3)	-0.0799 (2)	0.39861 (18)	0.0816 (8)
O3	0.3863 (2)	0.1968 (2)	0.28614 (14)	0.0612 (7)
O4	0.1458 (3)	-0.0409 (2)	0.37055 (17)	0.0717 (8)
N1	0.6903 (3)	0.0326 (3)	0.2484 (2)	0.0632 (8)
H1	0.7406	0.0909	0.2139	0.076*
N2	0.2154 (3)	0.0694 (2)	0.22282 (19)	0.0562 (8)
H2	0.2722	0.1231	0.1905	0.067*
C1	1.0023 (4)	0.3375 (3)	0.2529 (2)	0.0595 (10)
C2	0.9435 (3)	0.2484 (3)	0.3297 (2)	0.0513 (9)
C3	0.9725 (3)	0.2452 (3)	0.4215 (2)	0.0558 (9)
H3	0.9333	0.1851	0.4726	0.067*
C4	1.0594 (4)	0.3303 (4)	0.4391 (3)	0.0622 (10)
C5	1.1172 (4)	0.4175 (4)	0.3616 (3)	0.0755 (12)
H5	1.1763	0.4749	0.3720	0.091*
C6	1.0892 (4)	0.4211 (3)	0.2698 (3)	0.0739 (11)
H6	1.1293	0.4805	0.2187	0.089*
C7	1.0907 (4)	0.3245 (4)	0.5403 (3)	0.0881 (13)
H7A	1.0174	0.2841	0.5814	0.132*
H7B	1.0959	0.4093	0.5433	0.132*
H7C	1.1787	0.2763	0.5605	0.132*

C8	0.7986 (3)	0.0726 (3)	0.3822 (2)	0.0582 (10)
H8A	0.7484	0.1121	0.4246	0.070*
H8B	0.8726	0.0135	0.4189	0.070*
C9	0.6998 (4)	0.0018 (3)	0.3432 (3)	0.0553 (9)
C10	0.5997 (4)	-0.0276 (4)	0.2010 (3)	0.0732 (11)
H10A	0.5485	0.0379	0.1504	0.088*
H10B	0.5316	-0.0715	0.2466	0.088*
C11	0.6762 (4)	-0.1216 (3)	0.1586 (2)	0.0555 (9)
C12	0.8057 (4)	-0.1810 (4)	0.1908 (3)	0.0680 (11)
H12	0.8494	-0.1633	0.2407	0.082*
C13	0.8722 (5)	-0.2667 (4)	0.1503 (4)	0.0905 (14)
H13	0.9598	-0.3059	0.1731	0.109*
C14	0.8103 (7)	-0.2939 (5)	0.0774 (4)	0.1025 (17)
H14	0.8551	-0.3511	0.0498	0.123*
C15	0.6804 (7)	-0.2356 (5)	0.0448 (3)	0.0996 (17)
H15	0.6371	-0.2539	-0.0049	0.119*
C16	0.6137 (4)	-0.1502 (4)	0.0853 (3)	0.0738 (12)
H16	0.5257	-0.1118	0.0627	0.089*
C17	0.5350 (4)	0.3631 (3)	0.2384 (2)	0.0540 (9)
C18	0.4702 (3)	0.2756 (3)	0.3119 (2)	0.0496 (9)
C19	0.4939 (3)	0.2724 (3)	0.4050 (2)	0.0540 (9)
H19	0.4503	0.2141	0.4549	0.065*
C20	0.5820 (4)	0.3550 (3)	0.4249 (2)	0.0556 (9)
C21	0.6439 (4)	0.4419 (3)	0.3503 (3)	0.0657 (10)
H21	0.7020	0.4986	0.3627	0.079*
C22	0.6209 (4)	0.4461 (3)	0.2575 (3)	0.0663 (10)
H22	0.6636	0.5052	0.2076	0.080*
C23	0.6128 (4)	0.3443 (4)	0.5272 (2)	0.0848 (13)
H23A	0.7020	0.2976	0.5462	0.127*
H23B	0.5407	0.3005	0.5677	0.127*
H23C	0.6155	0.4280	0.5330	0.127*
C24	0.3147 (3)	0.1067 (3)	0.3610 (2)	0.0556 (9)
H24A	0.2613	0.1507	0.3993	0.067*
H24B	0.3826	0.0454	0.4024	0.067*
C25	0.2180 (4)	0.0391 (3)	0.3175 (3)	0.0551 (9)
C26	0.1215 (4)	0.0163 (3)	0.1709 (2)	0.0634 (10)
H26A	0.0899	0.0824	0.1128	0.076*
H26B	0.0395	-0.0100	0.2098	0.076*
C27	0.1878 (3)	-0.0956 (3)	0.1448 (2)	0.0514 (9)
C28	0.1796 (4)	-0.1007 (4)	0.0525 (3)	0.0666 (11)
H28	0.1372	-0.0315	0.0052	0.080*
C29	0.2332 (5)	-0.2062 (5)	0.0292 (3)	0.0821 (13)
H29	0.2261	-0.2079	-0.0333	0.099*
C30	0.2966 (5)	-0.3083 (4)	0.0978 (4)	0.0830 (13)
H30	0.3314	-0.3800	0.0824	0.100*
C31	0.3087 (4)	-0.3043 (4)	0.1899 (3)	0.0763 (12)
H31	0.3548	-0.3721	0.2363	0.092*
C32	0.2528 (4)	-0.2002 (4)	0.2128 (3)	0.0654 (10)

H32 0.2586 -0.1998 0.2757 0.078*

Atomic displacement parameters (Å²)

	U^{11}	U^{22}	U^{33}	U^{12}	U^{13}	U^{23}
Cl1	0.1148 (9)	0.0760 (7)	0.0597 (6)	-0.0167 (6)	0.0051 (6)	-0.0180 (5)
Cl2	0.1155 (9)	0.0892 (8)	0.0509 (6)	-0.0252 (7)	0.0039 (6)	-0.0218 (5)
O1	0.0719 (17)	0.0588 (15)	0.0535 (14)	-0.0184 (14)	0.0021 (13)	-0.0150 (12)
O2	0.082 (2)	0.0840 (19)	0.0773 (18)	-0.0361 (17)	0.0169 (15)	-0.0170 (15)
O3	0.0689 (16)	0.0691 (16)	0.0527 (14)	-0.0265 (14)	0.0068 (12)	-0.0246 (12)
O4	0.0743 (18)	0.0733 (17)	0.0716 (17)	-0.0294 (15)	0.0172 (14)	-0.0235 (14)
N1	0.062 (2)	0.072 (2)	0.060 (2)	-0.0189 (17)	0.0046 (16)	-0.0251 (16)
N2	0.0544 (19)	0.0574 (19)	0.0608 (19)	-0.0111 (15)	0.0006 (15)	-0.0224 (15)
C1	0.070 (3)	0.055 (2)	0.056 (2)	-0.006 (2)	0.007 (2)	-0.0213 (19)
C2	0.048 (2)	0.045 (2)	0.062 (2)	0.0001 (18)	-0.0048 (19)	-0.0204 (18)
C3	0.058 (2)	0.053 (2)	0.059 (2)	0.0026 (19)	-0.0026 (19)	-0.0226 (18)
C4	0.062 (3)	0.059 (2)	0.074 (3)	0.004 (2)	-0.008 (2)	-0.033 (2)
C5	0.082 (3)	0.061 (3)	0.095 (3)	-0.016 (2)	-0.008 (3)	-0.039 (2)
C6	0.082 (3)	0.056 (3)	0.086 (3)	-0.016 (2)	0.005 (2)	-0.023 (2)
C7	0.100 (3)	0.086 (3)	0.091 (3)	0.003 (3)	-0.027 (3)	-0.047 (2)
C8	0.059 (2)	0.058 (2)	0.057 (2)	-0.008 (2)	0.0072 (19)	-0.0167 (18)
C9	0.048 (2)	0.057 (2)	0.062 (2)	-0.0025 (19)	0.002 (2)	-0.0193 (19)
C10	0.061 (3)	0.084 (3)	0.081 (3)	-0.009 (2)	-0.010 (2)	-0.034 (2)
C11	0.053 (2)	0.063 (2)	0.051 (2)	-0.017 (2)	-0.0016 (19)	-0.0167 (18)
C12	0.058 (3)	0.073 (3)	0.074 (3)	-0.012 (2)	0.000 (2)	-0.024 (2)
C13	0.082 (3)	0.068 (3)	0.125 (4)	-0.012 (3)	0.017 (3)	-0.035 (3)
C14	0.130 (5)	0.082 (4)	0.113 (4)	-0.041 (4)	0.055 (4)	-0.051 (3)
C15	0.150 (5)	0.094 (4)	0.074 (3)	-0.054 (4)	0.016 (4)	-0.044 (3)
C16	0.082 (3)	0.078 (3)	0.062 (2)	-0.027 (2)	-0.011 (2)	-0.017 (2)
C17	0.059 (2)	0.055 (2)	0.049 (2)	-0.0040 (19)	0.0014 (18)	-0.0158 (18)
C18	0.046 (2)	0.052 (2)	0.056 (2)	-0.0069 (18)	-0.0008 (18)	-0.0223 (18)
C19	0.057 (2)	0.056 (2)	0.050 (2)	-0.0069 (19)	0.0021 (18)	-0.0176 (17)
C20	0.061 (2)	0.054 (2)	0.056 (2)	-0.001 (2)	-0.0069 (19)	-0.0225 (18)
C21	0.069 (3)	0.065 (3)	0.072 (3)	-0.015 (2)	-0.007 (2)	-0.031 (2)
C22	0.072 (3)	0.058 (2)	0.069 (3)	-0.013 (2)	0.009 (2)	-0.019 (2)
C23	0.098 (3)	0.089 (3)	0.074 (3)	-0.005 (3)	-0.020 (2)	-0.034 (2)
C24	0.056 (2)	0.063 (2)	0.051 (2)	-0.0128 (19)	0.0041 (18)	-0.0208 (18)
C25	0.056 (2)	0.055 (2)	0.060 (2)	-0.003 (2)	0.008 (2)	-0.0263 (19)
C26	0.057 (2)	0.067 (3)	0.070 (2)	-0.004 (2)	-0.011 (2)	-0.027 (2)
C27	0.048 (2)	0.058 (2)	0.051 (2)	-0.0118 (19)	0.0000 (18)	-0.0193 (19)
C28	0.066 (3)	0.080 (3)	0.056 (2)	-0.021 (2)	0.000 (2)	-0.021 (2)
C29	0.092 (3)	0.105 (4)	0.066 (3)	-0.039 (3)	0.020 (3)	-0.044 (3)
C30	0.089 (3)	0.070 (3)	0.104 (4)	-0.028 (3)	0.034 (3)	-0.045 (3)
C31	0.085 (3)	0.055 (3)	0.086 (3)	-0.003 (2)	0.009 (3)	-0.018 (2)
C32	0.075 (3)	0.063 (3)	0.059 (2)	-0.007 (2)	0.000 (2)	-0.020 (2)

Geometric parameters (Å, °)

C11—C1	1.735 (3)	C13—C14	1.357 (6)
C12—C17	1.741 (3)	C13—H13	0.9300
O1—C2	1.378 (3)	C14—C15	1.376 (6)
O1—C8	1.421 (3)	C14—H14	0.9300
O2—C9	1.225 (4)	C15—C16	1.380 (6)
O3—C18	1.369 (3)	C15—H15	0.9300
O3—C24	1.439 (3)	C16—H16	0.9300
O4—C25	1.231 (4)	C17—C22	1.372 (4)
N1—C9	1.332 (4)	C17—C18	1.381 (4)
N1—C10	1.442 (4)	C18—C19	1.381 (4)
N1—H1	0.8600	C19—C20	1.389 (4)
N2—C25	1.328 (4)	C19—H19	0.9300
N2—C26	1.456 (4)	C20—C21	1.373 (4)
N2—H2	0.8600	C20—C23	1.506 (4)
C1—C6	1.370 (5)	C21—C22	1.373 (4)
C1—C2	1.391 (4)	C21—H21	0.9300
C2—C3	1.373 (4)	C22—H22	0.9300
C3—C4	1.388 (4)	C23—H23A	0.9600
C3—H3	0.9300	C23—H23B	0.9600
C4—C5	1.383 (5)	C23—H23C	0.9600
C4—C7	1.504 (5)	C24—C25	1.500 (4)
C5—C6	1.371 (5)	C24—H24A	0.9700
C5—H5	0.9300	C24—H24B	0.9700
C6—H6	0.9300	C26—C27	1.504 (4)
C7—H7A	0.9600	C26—H26A	0.9700
C7—H7B	0.9600	C26—H26B	0.9700
C7—H7C	0.9600	C27—C28	1.382 (4)
C8—C9	1.500 (4)	C27—C32	1.384 (4)
C8—H8A	0.9700	C28—C29	1.380 (5)
C8—H8B	0.9700	C28—H28	0.9300
C10—C11	1.513 (5)	C29—C30	1.367 (6)
C10—H10A	0.9700	C29—H29	0.9300
C10—H10B	0.9700	C30—C31	1.377 (5)
C11—C16	1.374 (4)	C30—H30	0.9300
C11—C12	1.375 (5)	C31—C32	1.370 (5)
C12—C13	1.384 (5)	C31—H31	0.9300
C12—H12	0.9300	C32—H32	0.9300
C2—O1—C8	117.8 (2)	C16—C15—H15	119.7
C18—O3—C24	117.7 (2)	C11—C16—C15	120.5 (4)
C9—N1—C10	122.9 (3)	C11—C16—H16	119.7
C9—N1—H1	118.5	C15—C16—H16	119.7
C10—N1—H1	118.5	C22—C17—C18	120.5 (3)
C25—N2—C26	123.5 (3)	C22—C17—C12	119.8 (3)
C25—N2—H2	118.3	C18—C17—C12	119.7 (3)
C26—N2—H2	118.3	O3—C18—C19	124.6 (3)

C6—C1—C2	119.5 (3)	O3—C18—C17	116.4 (3)
C6—C1—C11	120.6 (3)	C19—C18—C17	119.0 (3)
C2—C1—C11	119.9 (3)	C18—C19—C20	120.8 (3)
C3—C2—O1	124.6 (3)	C18—C19—H19	119.6
C3—C2—C1	119.9 (3)	C20—C19—H19	119.6
O1—C2—C1	115.6 (3)	C21—C20—C19	118.9 (3)
C2—C3—C4	121.0 (3)	C21—C20—C23	121.3 (3)
C2—C3—H3	119.5	C19—C20—C23	119.7 (3)
C4—C3—H3	119.5	C22—C21—C20	120.8 (3)
C5—C4—C3	118.1 (3)	C22—C21—H21	119.6
C5—C4—C7	121.8 (3)	C20—C21—H21	119.6
C3—C4—C7	120.1 (3)	C17—C22—C21	120.0 (3)
C6—C5—C4	121.3 (3)	C17—C22—H22	120.0
C6—C5—H5	119.3	C21—C22—H22	120.0
C4—C5—H5	119.3	C20—C23—H23A	109.5
C1—C6—C5	120.2 (3)	C20—C23—H23B	109.5
C1—C6—H6	119.9	H23A—C23—H23B	109.5
C5—C6—H6	119.9	C20—C23—H23C	109.5
C4—C7—H7A	109.5	H23A—C23—H23C	109.5
C4—C7—H7B	109.5	H23B—C23—H23C	109.5
H7A—C7—H7B	109.5	O3—C24—C25	109.4 (3)
C4—C7—H7C	109.5	O3—C24—H24A	109.8
H7A—C7—H7C	109.5	C25—C24—H24A	109.8
H7B—C7—H7C	109.5	O3—C24—H24B	109.8
O1—C8—C9	110.1 (3)	C25—C24—H24B	109.8
O1—C8—H8A	109.6	H24A—C24—H24B	108.3
C9—C8—H8A	109.6	O4—C25—N2	123.4 (3)
O1—C8—H8B	109.6	O4—C25—C24	118.9 (3)
C9—C8—H8B	109.6	N2—C25—C24	117.7 (3)
H8A—C8—H8B	108.1	N2—C26—C27	113.6 (3)
O2—C9—N1	123.7 (3)	N2—C26—H26A	108.9
O2—C9—C8	119.4 (3)	C27—C26—H26A	108.9
N1—C9—C8	116.9 (3)	N2—C26—H26B	108.9
N1—C10—C11	114.0 (3)	C27—C26—H26B	108.9
N1—C10—H10A	108.7	H26A—C26—H26B	107.7
C11—C10—H10A	108.7	C28—C27—C32	117.4 (3)
N1—C10—H10B	108.7	C28—C27—C26	120.9 (3)
C11—C10—H10B	108.7	C32—C27—C26	121.7 (3)
H10A—C10—H10B	107.6	C29—C28—C27	121.3 (4)
C16—C11—C12	118.2 (4)	C29—C28—H28	119.4
C16—C11—C10	119.0 (4)	C27—C28—H28	119.4
C12—C11—C10	122.8 (3)	C30—C29—C28	120.1 (4)
C11—C12—C13	121.1 (4)	C30—C29—H29	120.0
C11—C12—H12	119.5	C28—C29—H29	120.0
C13—C12—H12	119.5	C29—C30—C31	119.6 (4)
C14—C13—C12	120.4 (5)	C29—C30—H30	120.2
C14—C13—H13	119.8	C31—C30—H30	120.2
C12—C13—H13	119.8	C32—C31—C30	119.8 (4)

C13—C14—C15	119.1 (5)	C32—C31—H31	120.1
C13—C14—H14	120.4	C30—C31—H31	120.1
C15—C14—H14	120.4	C31—C32—C27	121.7 (4)
C14—C15—C16	120.7 (5)	C31—C32—H32	119.1
C14—C15—H15	119.7	C27—C32—H32	119.1
C8—O1—C2—C3	-1.4 (5)	C24—O3—C18—C19	1.8 (5)
C8—O1—C2—C1	178.4 (3)	C24—O3—C18—C17	-178.5 (3)
C6—C1—C2—C3	0.2 (5)	C22—C17—C18—O3	179.9 (3)
C11—C1—C2—C3	-178.8 (3)	C12—C17—C18—O3	-1.1 (4)
C6—C1—C2—O1	-179.6 (3)	C22—C17—C18—C19	-0.4 (5)
C11—C1—C2—O1	1.4 (4)	C12—C17—C18—C19	178.5 (3)
O1—C2—C3—C4	-179.9 (3)	O3—C18—C19—C20	179.2 (3)
C1—C2—C3—C4	0.4 (5)	C17—C18—C19—C20	-0.4 (5)
C2—C3—C4—C5	-0.7 (5)	C18—C19—C20—C21	1.1 (5)
C2—C3—C4—C7	-179.5 (3)	C18—C19—C20—C23	-176.4 (3)
C3—C4—C5—C6	0.5 (6)	C19—C20—C21—C22	-1.0 (5)
C7—C4—C5—C6	179.3 (4)	C23—C20—C21—C22	176.4 (4)
C2—C1—C6—C5	-0.4 (6)	C18—C17—C22—C21	0.5 (6)
C11—C1—C6—C5	178.6 (3)	C12—C17—C22—C21	-178.5 (3)
C4—C5—C6—C1	0.0 (6)	C20—C21—C22—C17	0.2 (6)
C2—O1—C8—C9	175.0 (3)	C18—O3—C24—C25	174.4 (3)
C10—N1—C9—O2	0.1 (6)	C26—N2—C25—O4	3.4 (5)
C10—N1—C9—C8	180.0 (3)	C26—N2—C25—C24	-176.7 (3)
O1—C8—C9—O2	-176.5 (3)	O3—C24—C25—O4	-179.0 (3)
O1—C8—C9—N1	3.6 (4)	O3—C24—C25—N2	1.1 (4)
C9—N1—C10—C11	-105.2 (4)	C25—N2—C26—C27	-96.3 (4)
N1—C10—C11—C16	-157.2 (3)	N2—C26—C27—C28	-130.2 (3)
N1—C10—C11—C12	23.4 (5)	N2—C26—C27—C32	53.2 (4)
C16—C11—C12—C13	0.4 (5)	C32—C27—C28—C29	0.5 (5)
C10—C11—C12—C13	179.9 (3)	C26—C27—C28—C29	-176.2 (3)
C11—C12—C13—C14	0.0 (6)	C27—C28—C29—C30	-0.4 (6)
C12—C13—C14—C15	-0.4 (7)	C28—C29—C30—C31	-1.0 (6)
C13—C14—C15—C16	0.2 (7)	C29—C30—C31—C32	2.3 (6)
C12—C11—C16—C15	-0.6 (5)	C30—C31—C32—C27	-2.3 (6)
C10—C11—C16—C15	180.0 (3)	C28—C27—C32—C31	0.9 (5)
C14—C15—C16—C11	0.2 (6)	C26—C27—C32—C31	177.5 (3)

Hydrogen-bond geometry (\AA , $^\circ$)

$D-H\cdots A$	$D-H$	$H\cdots A$	$D\cdots A$	$D-H\cdots A$
C3—H3 \cdots O4 ⁱ	0.93	2.51	3.423 (4)	169
C19—H19 \cdots O2 ⁱ	0.93	2.37	3.287 (4)	169

Symmetry code: (i) $-x+1, -y, -z+1$.