

Acta Crystallographica Section E

## Structure Reports

Online

ISSN 1600-5368

# Methyl 2-diphenylphosphoryloxy-2-aza-bicyclo[2.2.1]hept-5-ene-3-exo-carboxylate

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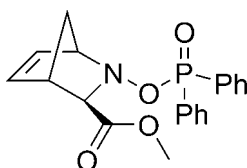
Received 2 December 2008; accepted 16 December 2008

Key indicators: single-crystal X-ray study;  $T = 100$  K; mean  $\sigma(\text{C}-\text{C}) = 0.002$  Å;  $R$  factor = 0.035;  $wR$  factor = 0.091; data-to-parameter ratio = 15.8.

In the title compound,  $\text{C}_{20}\text{H}_{20}\text{NO}_4\text{P}$ , the dihedral angle between the phenyl rings is  $68.52$  ( $7$ )°. In the crystal structure, the molecules are linked by a weak  $\text{C}-\text{H}\cdots\pi$ (arene) interaction along  $[010]$  involving the phenyl CH group and the phenyl rings. There are no further significant intermolecular interactions.

## Related literature

For the preparation of the precursor of the title compound, see: Sousa *et al.* (2008). For related literature about this type of bicyclic compound and their relevance see: Vale *et al.* (2006), Alves *et al.* (2006), Yoda *et al.* (1995).



## Experimental

### Crystal data

$\text{C}_{20}\text{H}_{20}\text{NO}_4\text{P}$   
 $M_r = 369.34$   
Monoclinic,  $P2_1/c$   
 $a = 18.4223$  (6) Å  
 $b = 8.5522$  (3) Å  
 $c = 11.6022$  (4) Å  
 $\beta = 97.1810$  (10)°

$V = 1813.60$  (11) Å<sup>3</sup>  
 $Z = 4$   
Mo  $K\alpha$  radiation  
 $\mu = 0.18$  mm<sup>-1</sup>  
 $T = 100$  (2) K  
 $0.37 \times 0.34 \times 0.34$  mm

### Data collection

Bruker APEXII CCD area-detector diffractometer  
Absorption correction: multi-scan (SADABS; Bruker, 2006)  
 $T_{\min} = 0.871$ ,  $T_{\max} = 0.940$   
14828 measured reflections  
3171 independent reflections  
3172 reflections with  $I > 2\sigma(I)$   
 $R_{\text{int}} = 0.033$

### Refinement

$R[F^2 > 2\sigma(F^2)] = 0.035$   
 $wR(F^2) = 0.091$   
 $S = 1.05$   
3717 reflections  
236 parameters  
H-atom parameters constrained  
 $\Delta\rho_{\max} = 0.29$  e Å<sup>-3</sup>  
 $\Delta\rho_{\min} = -0.41$  e Å<sup>-3</sup>

Table 1

Hydrogen-bond geometry (Å, °).

$D-H\cdots A$	$D-H$	$H\cdots A$	$D\cdots A$	$D-H\cdots A$
$\text{C12}-\text{H12}\cdots\text{Cg1}^i$	0.95	2.77	3.566 (2)	142

Symmetry code: (i)  $-x + 1, y + \frac{1}{2}, -z + \frac{1}{2}$ . Cg1 is the centroid of the C15–C20 ring.

Data collection: APEX2 (Bruker, 2006); cell refinement: SAINT (Bruker, 2006); data reduction: SAINT; program(s) used to solve structure: SIR97 (Altomare *et al.*, 1997); program(s) used to refine structure: SHELXL97 (Sheldrick, 2008); molecular graphics: ORTEP-3 for Windows (Farrugia, 1997) and PLATON (Spek, 2003); software used to prepare material for publication: WinGX publication routines (Farrugia, 1999).

This work was supported by the Centro de Investigação em Química of the University of Porto. The X-ray data were collected at the Unidade de Raios X, RIAIDT, University of Santiago de Compostela. The authors thank Antonio L. Llamas-Saiz for his help and the Fundação para a Ciência e Tecnologia (FCT) and Xunta de Galicia for financial support (grants POCTI/QUI/44471/2002 and 07CSA008203-PR, respectively). CADS thanks the FCT for grant No. SFRH/BD/31526/2006.

Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: BX2188).

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## supporting information

*Acta Cryst.* (2009). E65, o188 [doi:10.1107/S160053680804292X]

## Methyl 2-diphenylphosphoryloxy-2-azabicyclo[2.2.1]hept-5-ene-3-exo-carboxylate

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### S1. Comment

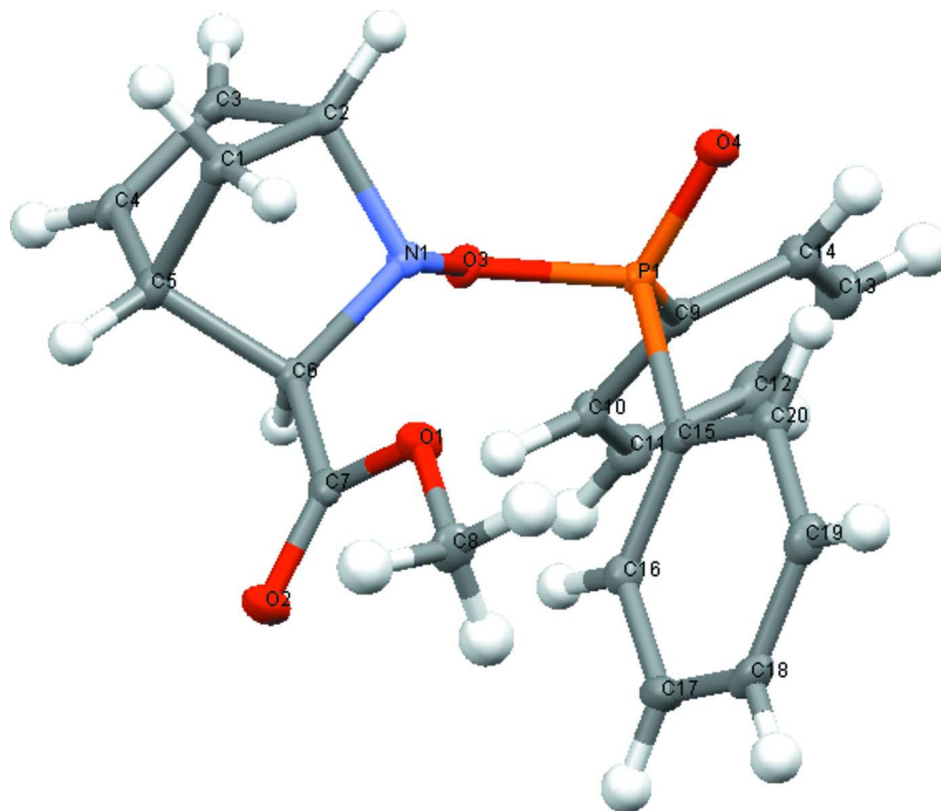
The structure of the title compound, (I), is shown in Fig. 1. It can be seen the existence of three chiral centers at C2 (*R*), C5 (*S*) and C6 (*R*). In the crystalline structure, the molecules are linked by a weak C—H $\cdots$  $\pi$  interaction, Fig. 2 [H12- $\pi^i$  2.77 Å, C12-H12- $\pi$  142°, C12- $\pi$  3.566 (2) Å, symmetry code: (i) 1-x, 1/2+y, 1/2-z] along [010] directions. There are no further significant intermolecular interactions.

### S2. Experimental

The title compound was synthesized from the previously prepared (3*exo*)-2-hydroxy-2-azabicyclo[2.2.1]hept-5-ene-3-carboxylate (Sousa *et al.* 2008). Equimolar amounts of (3*exo*)-2-hydroxy-2-azabicyclo[2.2.1]hept-5-ene-3-carboxylate (0.56 g, 3.3 mmol) and diphenylphosphinic chloride (0.63 ml, 3.3 mmol), in the presence of 1 eq. of anhydrous triethylamine and a catalytic quantity of DMAP, were let to react overnight in dichloromethane, at room temperature under argon atmosphere. Water was added and the product was extracted with dichloromethane (3  $\times$  15 ml). The organic layers were dried over sodium sulfate and the solvent was evaporated. The obtained product was purified by flash chromatography (eluent: dichloromethane/diethyl ether 1:1), leading to a light clear yellow oil in 80% yield. Crystals of (I) were made from a slow evaporation of a dichloromethane/hexane solution.

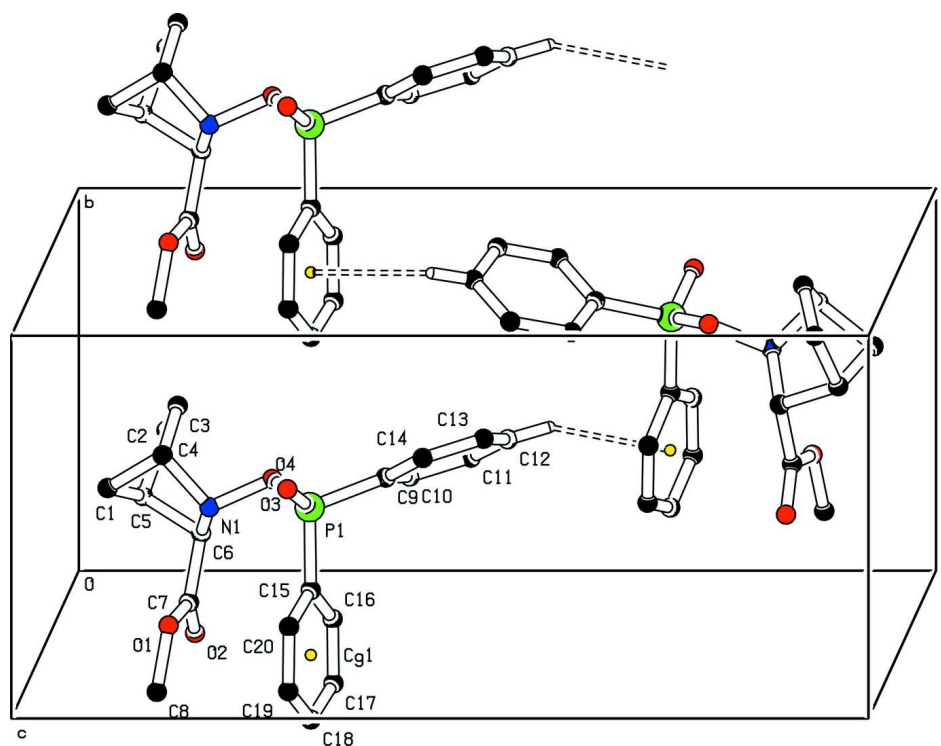
### S3. Refinement

All H atoms were found in a difference Fourier map and placed in geometrically idealized and constrained to ride on their parent atoms [C—H = 0.95–1.00 Å and  $U_{\text{iso}}(\text{H}) = 1.2$  or  $1.5U_{\text{eq}}(\text{C})$ ].



**Figure 1**

A view of (I), showing the three chiral carbons C2, C5 and C6 and the atom-labelling scheme. Displacement ellipsoids are drawn at the 50% probability level. H atoms are shown as spheres of arbitrary radius.



**Figure 2**

Part of the crystal structure of (I) viewed along the *c* axis. Dashed lines show C—H $\cdots$  $\pi$  (arene) interactions. Only H atoms participating in hydrogen bonding are shown.  $\pi$  is the centroid of the ring defined by atoms C15–C20.

### Methyl 2-diphenylphosphoryloxy-2-azabicyclo[2.2.1]hept-5-ene-3-exo- carboxylate

#### Crystal data

$C_{20}H_{20}NO_4P$   
 $M_r = 369.34$   
 Monoclinic,  $P2_1/c$   
 Hall symbol: -P 2ybc  
 $a = 18.4223$  (6) Å  
 $b = 8.5522$  (3) Å  
 $c = 11.6022$  (4) Å  
 $\beta = 97.181$  (1)°  
 $V = 1813.60$  (11) Å<sup>3</sup>  
 $Z = 4$

$F(000) = 776$   
 $D_x = 1.353$  Mg m<sup>-3</sup>  
 Mo  $K\alpha$  radiation,  $\lambda = 0.71073$  Å  
 Cell parameters from 1953 reflections  
 $\theta = 3.1$ – $25.9$ °  
 $\mu = 0.18$  mm<sup>-1</sup>  
 $T = 100$  K  
 Prism, colourless  
 $0.37 \times 0.34 \times 0.34$  mm

#### Data collection

Bruker ApexII CCD area-detector  
 diffractometer  
 Radiation source: sealed tube  
 Graphite monochromator  
 phi and  $\omega$  scans  
 Absorption correction: multi-scan  
 (SADABS; Bruker, 2006)  
 $T_{\min} = 0.871$ ,  $T_{\max} = 0.940$

14828 measured reflections  
 3717 independent reflections  
 3172 reflections with  $I > 2\sigma(I)$   
 $R_{\text{int}} = 0.033$   
 $\theta_{\text{max}} = 26.4$ °,  $\theta_{\text{min}} = 2.2$ °  
 $h = -23$ → $22$   
 $k = 0$ → $10$   
 $l = 0$ → $14$

Refinement

Refinement on  $F^2$   
 Least-squares matrix: full  
 $R[F^2 > 2\sigma(F^2)] = 0.035$   
 $wR(F^2) = 0.091$   
 $S = 1.05$   
 3717 reflections  
 236 parameters  
 0 restraints  
 Primary atom site location: structure-invariant  
 direct methods

Secondary atom site location: difference Fourier  
 map  
 Hydrogen site location: inferred from  
 neighbouring sites  
 H-atom parameters constrained  
 $w = 1/[\sigma^2(F_o^2) + (0.0429P)^2 + 0.8843P]$   
 where  $P = (F_o^2 + 2F_c^2)/3$   
 $(\Delta/\sigma)_{\max} = 0.001$   
 $\Delta\rho_{\max} = 0.29 \text{ e } \text{\AA}^{-3}$   
 $\Delta\rho_{\min} = -0.41 \text{ e } \text{\AA}^{-3}$

Special details

**Geometry.** All e.s.d.'s (except the e.s.d. in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell e.s.d.'s are taken into account individually in the estimation of e.s.d.'s in distances, angles and torsion angles; correlations between e.s.d.'s in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell e.s.d.'s is used for estimating e.s.d.'s involving l.s. planes.

**Refinement.** Refinement of  $F^2$  against ALL reflections. The weighted  $R$ -factor  $wR$  and goodness of fit  $S$  are based on  $F^2$ , conventional  $R$ -factors  $R$  are based on  $F$ , with  $F$  set to zero for negative  $F^2$ . The threshold expression of  $F^2 > \sigma(F^2)$  is used only for calculating  $R$ -factors(gt) *etc.* and is not relevant to the choice of reflections for refinement.  $R$ -factors based on  $F^2$  are statistically about twice as large as those based on  $F$ , and  $R$ -factors based on ALL data will be even larger.

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters ( $\text{\AA}^2$ )

	$x$	$y$	$z$	$U_{\text{iso}}^*/U_{\text{eq}}$
C1	0.04005 (8)	0.24766 (17)	0.08174 (13)	0.0156 (3)
H1A	-0.005	0.3099	0.0614	0.019*
H1B	0.0318	0.1646	0.1381	0.019*
C2	0.10739 (8)	0.34831 (17)	0.12086 (13)	0.0140 (3)
H2	0.1038	0.4159	0.1902	0.017*
C3	0.11615 (8)	0.43459 (18)	0.00937 (13)	0.0157 (3)
H3	0.1328	0.539	0.003	0.019*
C4	0.09614 (8)	0.33679 (18)	-0.07788 (13)	0.0170 (3)
H4	0.0965	0.358	-0.1582	0.02*
C5	0.07271 (8)	0.18529 (18)	-0.02579 (13)	0.0152 (3)
H5	0.0408	0.1144	-0.0788	0.018*
C6	0.14566 (8)	0.11298 (17)	0.03659 (12)	0.0124 (3)
H6	0.1849	0.1189	-0.0155	0.015*
C7	0.13407 (7)	-0.05517 (17)	0.07081 (12)	0.0126 (3)
C8	0.10791 (9)	-0.23359 (18)	0.21479 (14)	0.0188 (3)
H8A	0.0638	-0.2759	0.1694	0.028*
H8B	0.1021	-0.2364	0.2976	0.028*
H8C	0.1503	-0.2968	0.201	0.028*
C9	0.37374 (8)	0.32042 (17)	0.19955 (13)	0.0135 (3)
C10	0.39313 (8)	0.28107 (18)	0.09030 (13)	0.0156 (3)
H10	0.3592	0.2281	0.0353	0.019*
C11	0.46227 (8)	0.31986 (19)	0.06275 (14)	0.0196 (3)
H11	0.4758	0.2923	-0.011	0.024*
C12	0.51155 (8)	0.3985 (2)	0.14240 (15)	0.0220 (4)
H12	0.559	0.4232	0.1236	0.026*

C13	0.49180 (9)	0.4412 (2)	0.24924 (15)	0.0228 (4)
H13	0.5253	0.4974	0.3029	0.027*
C14	0.42326 (8)	0.40209 (18)	0.27826 (14)	0.0180 (3)
H14	0.41	0.431	0.3519	0.022*
C15	0.29184 (7)	0.05353 (17)	0.27096 (13)	0.0125 (3)
C16	0.31090 (8)	-0.05303 (18)	0.18853 (13)	0.0156 (3)
H16	0.3218	-0.017	0.1151	0.019*
C17	0.31393 (9)	-0.21192 (18)	0.21417 (14)	0.0190 (3)
H17	0.3284	-0.2841	0.1591	0.023*
C18	0.29590 (8)	-0.26536 (18)	0.31971 (15)	0.0200 (3)
H18	0.2972	-0.3742	0.3364	0.024*
C19	0.27594 (8)	-0.16006 (19)	0.40103 (14)	0.0185 (3)
H19	0.2629	-0.197	0.4729	0.022*
C20	0.27503 (8)	-0.00078 (18)	0.37771 (13)	0.0149 (3)
H20	0.2629	0.0713	0.4346	0.018*
N1	0.16332 (6)	0.21932 (14)	0.13812 (11)	0.0116 (3)
O1	0.11889 (6)	-0.07336 (12)	0.17981 (9)	0.0166 (2)
O2	0.13582 (6)	-0.16137 (12)	0.00278 (9)	0.0190 (2)
O3	0.23563 (5)	0.28855 (12)	0.12671 (9)	0.0131 (2)
O4	0.27071 (6)	0.34776 (12)	0.34900 (9)	0.0158 (2)
P1	0.288697 (19)	0.26111 (4)	0.24650 (3)	0.01112 (11)

*Atomic displacement parameters (Å<sup>2</sup>)*

	$U^{11}$	$U^{22}$	$U^{33}$	$U^{12}$	$U^{13}$	$U^{23}$
C1	0.0105 (7)	0.0152 (8)	0.0210 (8)	0.0011 (6)	0.0016 (6)	0.0022 (6)
C2	0.0121 (7)	0.0126 (7)	0.0175 (7)	0.0026 (6)	0.0025 (6)	-0.0004 (6)
C3	0.0118 (7)	0.0132 (7)	0.0218 (8)	0.0026 (6)	0.0016 (6)	0.0041 (6)
C4	0.0154 (7)	0.0179 (8)	0.0172 (7)	0.0029 (6)	0.0000 (6)	0.0060 (6)
C5	0.0130 (7)	0.0149 (8)	0.0167 (7)	-0.0011 (6)	-0.0020 (6)	0.0021 (6)
C6	0.0112 (7)	0.0123 (7)	0.0135 (7)	-0.0005 (6)	0.0008 (5)	0.0000 (6)
C7	0.0081 (6)	0.0149 (7)	0.0145 (7)	0.0003 (6)	-0.0001 (5)	-0.0004 (6)
C8	0.0206 (8)	0.0145 (8)	0.0209 (8)	-0.0045 (6)	0.0010 (6)	0.0058 (6)
C9	0.0106 (7)	0.0116 (7)	0.0180 (7)	0.0001 (6)	0.0005 (6)	0.0035 (6)
C10	0.0133 (7)	0.0161 (8)	0.0169 (7)	-0.0018 (6)	-0.0003 (6)	0.0041 (6)
C11	0.0170 (8)	0.0232 (8)	0.0192 (8)	0.0001 (6)	0.0047 (6)	0.0046 (6)
C12	0.0131 (7)	0.0242 (9)	0.0289 (9)	-0.0026 (6)	0.0031 (6)	0.0075 (7)
C13	0.0152 (8)	0.0235 (9)	0.0283 (9)	-0.0056 (7)	-0.0031 (7)	-0.0009 (7)
C14	0.0157 (7)	0.0174 (8)	0.0205 (8)	-0.0009 (6)	0.0002 (6)	-0.0009 (6)
C15	0.0086 (7)	0.0114 (7)	0.0165 (7)	-0.0001 (5)	-0.0021 (5)	0.0011 (6)
C16	0.0139 (7)	0.0163 (8)	0.0162 (7)	0.0004 (6)	0.0008 (6)	0.0004 (6)
C17	0.0188 (8)	0.0145 (8)	0.0226 (8)	0.0031 (6)	-0.0022 (6)	-0.0060 (6)
C18	0.0166 (8)	0.0120 (8)	0.0296 (9)	-0.0001 (6)	-0.0043 (7)	0.0042 (7)
C19	0.0151 (7)	0.0198 (8)	0.0202 (8)	-0.0022 (6)	0.0002 (6)	0.0068 (6)
C20	0.0121 (7)	0.0154 (8)	0.0169 (7)	-0.0008 (6)	0.0008 (6)	-0.0007 (6)
N1	0.0071 (6)	0.0117 (6)	0.0161 (6)	-0.0022 (5)	0.0017 (5)	-0.0004 (5)
O1	0.0224 (6)	0.0119 (5)	0.0163 (5)	-0.0032 (4)	0.0049 (4)	0.0011 (4)
O2	0.0245 (6)	0.0139 (6)	0.0189 (6)	-0.0012 (4)	0.0038 (5)	-0.0031 (5)

O3	0.0079 (5)	0.0142 (5)	0.0168 (5)	-0.0033 (4)	0.0003 (4)	0.0029 (4)
O4	0.0165 (5)	0.0135 (5)	0.0177 (5)	-0.0003 (4)	0.0030 (4)	-0.0014 (4)
P1	0.00981 (19)	0.01024 (19)	0.0131 (2)	-0.00066 (14)	0.00083 (14)	0.00053 (14)

*Geometric parameters (Å, °)*

C1—C2	1.531 (2)	C10—C11	1.392 (2)
C1—C5	1.546 (2)	C10—H10	0.95
C1—H1A	0.99	C11—C12	1.386 (2)
C1—H1B	0.99	C11—H11	0.95
C2—N1	1.5057 (18)	C12—C13	1.384 (2)
C2—C3	1.515 (2)	C12—H12	0.95
C2—H2	1	C13—C14	1.388 (2)
C3—C4	1.329 (2)	C13—H13	0.95
C3—H3	0.95	C14—H14	0.95
C4—C5	1.515 (2)	C15—C20	1.393 (2)
C4—H4	0.95	C15—C16	1.397 (2)
C5—C6	1.571 (2)	C15—P1	1.7976 (15)
C5—H5	1	C16—C17	1.391 (2)
C6—N1	1.4915 (18)	C16—H16	0.95
C6—C7	1.514 (2)	C17—C18	1.386 (2)
C6—H6	1	C17—H17	0.95
C7—O2	1.2064 (18)	C18—C19	1.387 (2)
C7—O1	1.3379 (17)	C18—H18	0.95
C8—O1	1.4505 (18)	C19—C20	1.389 (2)
C8—H8A	0.98	C19—H19	0.95
C8—H8B	0.98	C20—H20	0.95
C8—H8C	0.98	N1—O3	1.4786 (15)
C9—C14	1.395 (2)	O3—P1	1.6133 (10)
C9—C10	1.400 (2)	O4—P1	1.4737 (11)
C9—P1	1.7950 (15)		
C2—C1—C5	92.89 (11)	C11—C10—H10	120.2
C2—C1—H1A	113.1	C9—C10—H10	120.2
C5—C1—H1A	113.1	C12—C11—C10	120.30 (15)
C2—C1—H1B	113.1	C12—C11—H11	119.8
C5—C1—H1B	113.1	C10—C11—H11	119.8
H1A—C1—H1B	110.5	C13—C12—C11	120.11 (14)
N1—C2—C3	109.04 (11)	C13—C12—H12	119.9
N1—C2—C1	98.23 (11)	C11—C12—H12	119.9
C3—C2—C1	100.92 (12)	C12—C13—C14	120.20 (15)
N1—C2—H2	115.5	C12—C13—H13	119.9
C3—C2—H2	115.5	C14—C13—H13	119.9
C1—C2—H2	115.5	C13—C14—C9	120.14 (15)
C4—C3—C2	107.14 (13)	C13—C14—H14	119.9
C4—C3—H3	126.4	C9—C14—H14	119.9
C2—C3—H3	126.4	C20—C15—C16	119.59 (14)
C3—C4—C5	107.46 (13)	C20—C15—P1	117.59 (11)

C3—C4—H4	126.3	C16—C15—P1	122.82 (11)
C5—C4—H4	126.3	C17—C16—C15	119.85 (14)
C4—C5—C1	100.71 (12)	C17—C16—H16	120.1
C4—C5—C6	104.50 (12)	C15—C16—H16	120.1
C1—C5—C6	99.25 (11)	C18—C17—C16	120.24 (15)
C4—C5—H5	116.6	C18—C17—H17	119.9
C1—C5—H5	116.6	C16—C17—H17	119.9
C6—C5—H5	116.6	C17—C18—C19	120.02 (14)
N1—C6—C7	113.33 (11)	C17—C18—H18	120
N1—C6—C5	102.32 (11)	C19—C18—H18	120
C7—C6—C5	110.74 (12)	C18—C19—C20	120.14 (15)
N1—C6—H6	110.1	C18—C19—H19	119.9
C7—C6—H6	110.1	C20—C19—H19	119.9
C5—C6—H6	110.1	C19—C20—C15	120.11 (14)
O2—C7—O1	123.83 (14)	C19—C20—H20	119.9
O2—C7—C6	121.82 (13)	C15—C20—H20	119.9
O1—C7—C6	114.29 (12)	O3—N1—C6	106.47 (10)
O1—C8—H8A	109.5	O3—N1—C2	107.69 (10)
O1—C8—H8B	109.5	C6—N1—C2	105.28 (11)
H8A—C8—H8B	109.5	C7—O1—C8	115.30 (12)
O1—C8—H8C	109.5	N1—O3—P1	108.68 (8)
H8A—C8—H8C	109.5	O4—P1—O3	116.67 (6)
H8B—C8—H8C	109.5	O4—P1—C9	113.30 (7)
C14—C9—C10	119.54 (13)	O3—P1—C9	98.94 (6)
C14—C9—P1	117.84 (12)	O4—P1—C15	112.04 (7)
C10—C9—P1	122.54 (11)	O3—P1—C15	106.49 (6)
C11—C10—C9	119.67 (14)	C9—P1—C15	108.35 (7)
C5—C1—C2—N1	-60.85 (12)	C18—C19—C20—C15	2.1 (2)
C5—C1—C2—C3	50.47 (12)	C16—C15—C20—C19	-1.2 (2)
N1—C2—C3—C4	68.17 (15)	P1—C15—C20—C19	179.43 (11)
C1—C2—C3—C4	-34.58 (15)	C7—C6—N1—O3	120.70 (12)
C2—C3—C4—C5	0.88 (16)	C5—C6—N1—O3	-120.03 (11)
C3—C4—C5—C1	32.73 (15)	C7—C6—N1—C2	-125.14 (12)
C3—C4—C5—C6	-69.86 (15)	C5—C6—N1—C2	-5.87 (13)
C2—C1—C5—C4	-49.87 (12)	C3—C2—N1—O3	51.25 (14)
C2—C1—C5—C6	56.93 (12)	C1—C2—N1—O3	155.87 (10)
C4—C5—C6—N1	71.20 (13)	C3—C2—N1—C6	-62.06 (14)
C1—C5—C6—N1	-32.48 (13)	C1—C2—N1—C6	42.56 (13)
C4—C5—C6—C7	-167.72 (12)	O2—C7—O1—C8	3.0 (2)
C1—C5—C6—C7	88.59 (13)	C6—C7—O1—C8	-179.72 (12)
N1—C6—C7—O2	-162.95 (13)	C6—N1—O3—P1	-127.24 (9)
C5—C6—C7—O2	82.74 (17)	C2—N1—O3—P1	120.25 (10)
N1—C6—C7—O1	19.76 (17)	N1—O3—P1—O4	-68.47 (10)
C5—C6—C7—O1	-94.56 (14)	N1—O3—P1—C9	169.71 (9)
C14—C9—C10—C11	1.8 (2)	N1—O3—P1—C15	57.44 (9)
P1—C9—C10—C11	-174.85 (12)	C14—C9—P1—O4	18.47 (14)
C9—C10—C11—C12	-0.7 (2)	C10—C9—P1—O4	-164.79 (12)



C10—C11—C12—C13	-1.0 (2)	C14—C9—P1—O3	142.70 (12)
C11—C12—C13—C14	1.6 (3)	C10—C9—P1—O3	-40.56 (14)
C12—C13—C14—C9	-0.4 (2)	C14—C9—P1—C15	-106.51 (13)
C10—C9—C14—C13	-1.3 (2)	C10—C9—P1—C15	70.23 (14)
P1—C9—C14—C13	175.54 (12)	C20—C15—P1—O4	2.24 (13)
C20—C15—C16—C17	-0.9 (2)	C16—C15—P1—O4	-177.14 (11)
P1—C15—C16—C17	178.48 (11)	C20—C15—P1—O3	-126.43 (11)
C15—C16—C17—C18	2.0 (2)	C16—C15—P1—O3	54.19 (13)
C16—C17—C18—C19	-1.1 (2)	C20—C15—P1—C9	127.97 (11)
C17—C18—C19—C20	-1.0 (2)	C16—C15—P1—C9	-51.41 (14)

*Hydrogen-bond geometry (Å, °)*

<i>D—H...A</i>	<i>D—H</i>	<i>H...A</i>	<i>D...A</i>	<i>D—H...A</i>
C12—H12...Cg1 <sup>i</sup>	0.95	2.77	3.566 (2)	142

Symmetry code: (i)  $-x+1, y+1/2, -z+1/2$ .