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Pyrrolidinium chloride

Helene Giglmeier, Tobias Kerscher, Peter Klüfers* and Peter Mayer

Ludwig-Maximilians Universität, Department Chemie und Biochemie, Butenandtstrasse 5–13 (Haus D), 81377 München, Germany

Correspondence e-mail: kluef@cup.uni-muenchen.de

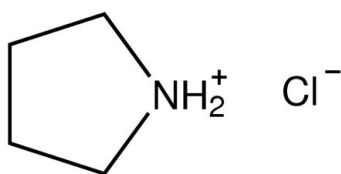
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Key indicators: single-crystal X-ray study; $T = 200$ K; mean $\sigma(\text{C}-\text{C}) = 0.004$ Å; R factor = 0.040; wR factor = 0.120; data-to-parameter ratio = 24.4.

The title compound, $\text{C}_4\text{H}_{10}\text{N}^+\cdot\text{Cl}^-$, was obtained as a decomposition product from 2,6-bis(pyrrolidinyl)pyridine. The anion lies on the same crystallographic mirror plane as the N atom of the cation, the complete cation being generated by mirror symmetry. The anions and cations are connected by $\text{N}^+-\text{H}\cdots\text{Cl}^-$ hydrogen bonds into chains along [100]. The pyrrolidinium cation is puckered in an envelope conformation E_{N1} .

Related literature

For details of the synthesis of 2,6-bis(pyrrolidinyl)pyridine, see: Folmer-Anderson *et al.* (2005). For related structures containing the pyrrolidinium cation, see: Kashino *et al.* (1978); Moritani *et al.* (1987); Jakubas *et al.* (2005). For a description of the E_{N1} conformation of the five-membered ring, see: Cremer & Pople (1975).



Experimental

Crystal data

 $\text{C}_4\text{H}_{10}\text{N}^+\cdot\text{Cl}^-$
 $M_r = 107.58$ Orthorhombic, $Pnma$
 $a = 7.4429$ (4) Å $b = 9.4104$ (5) Å
 $c = 8.9021$ (4) Å
 $V = 623.51$ (5) Å³
 $Z = 4$ Mo $K\alpha$ radiation
 $\mu = 0.48$ mm⁻¹
 $T = 200$ K
 $0.22 \times 0.13 \times 0.12$ mm

Data collection

Nonius KappaCCD diffractometer
Absorption correction: none
4239 measured reflections756 independent reflections
608 reflections with $I > 2\sigma(I)$
 $R_{\text{int}} = 0.037$

Refinement

 $R[F^2 > 2\sigma(F^2)] = 0.040$
 $wR(F^2) = 0.120$
 $S = 1.07$
756 reflections31 parameters
H-atom parameters constrained
 $\Delta\rho_{\text{max}} = 0.21$ e Å⁻³
 $\Delta\rho_{\text{min}} = -0.24$ e Å⁻³

Table 1

Hydrogen-bond geometry (Å, °).

$D-H\cdots A$	$D-H$	$H\cdots A$	$D\cdots A$	$D-H\cdots A$
$\text{N1}-\text{H101}\cdots\text{Cl1}$	0.92	2.17	3.091 (3)	180
$\text{N1}-\text{H102}\cdots\text{Cl1}^i$	0.92	2.18	3.097 (2)	177

Symmetry code: (i) $x - \frac{1}{2}, y, -z + \frac{1}{2}$

Data collection: *COLLECT* (Nonius, 2004); cell refinement: *SCALEPACK* (Otwinowski & Minor, 1997); data reduction: *DENZO* (Otwinowski & Minor, 1997) and *SCALEPACK*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 2008); program(s) used to refine structure: *SHELXL97* (Sheldrick, 2008); molecular graphics: *ORTEP-3* (Farrugia, 1997); software used to prepare material for publication: *SHELXL97*.

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Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: BI2348).

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supporting information

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Pyrrolidinium chloride

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S1. Comment

The title compound was obtained as a decomposition product. The organic salt is composed of the pyrrolidinium cation and a chloride anion (Fig. 1). The crystal packing is shown in Fig. 2. In the crystal, both H atoms bonded to N1 of the pyrrolidinium cation are involved in hydrogen bonds with chloride as acceptor. Both can be described according to graph set analysis with a $D^1_1(2)$ descriptor on the unitary level. This bonding pattern leads to chains along [1 0 0] which, starting from chloride, can be described according to graph set analysis with a $C^2_1(4)$ descriptor on the binary level. The hydrogen bonding pattern is shown in Fig. 3.

The C_s symmetric five-membered pyrrolidinium ring can be described according to Cremer & Pople (1975) by the puckering parameters $q_2 = 0.3061 \text{ \AA}$ and $\Phi_2 = 180.0000$. The closest pucker descriptor is an envelope E_{N1} .

S2. Experimental

The title compound was obtained as decomposition product of 2,6-bis(pyrrolidinyl)pyridine, which was synthesized according to Folmer-Anderson *et al.* (2005), after 4 months at room temperature.

S3. Refinement

H atoms were placed in calculated positions (C—H = 0.99 Å, N—H = 0.92 Å) and were included in the refinement in the riding model approximation with $U_{\text{iso}}(\text{H}) = 1.2 U_{\text{eq}}(\text{C/N})$.

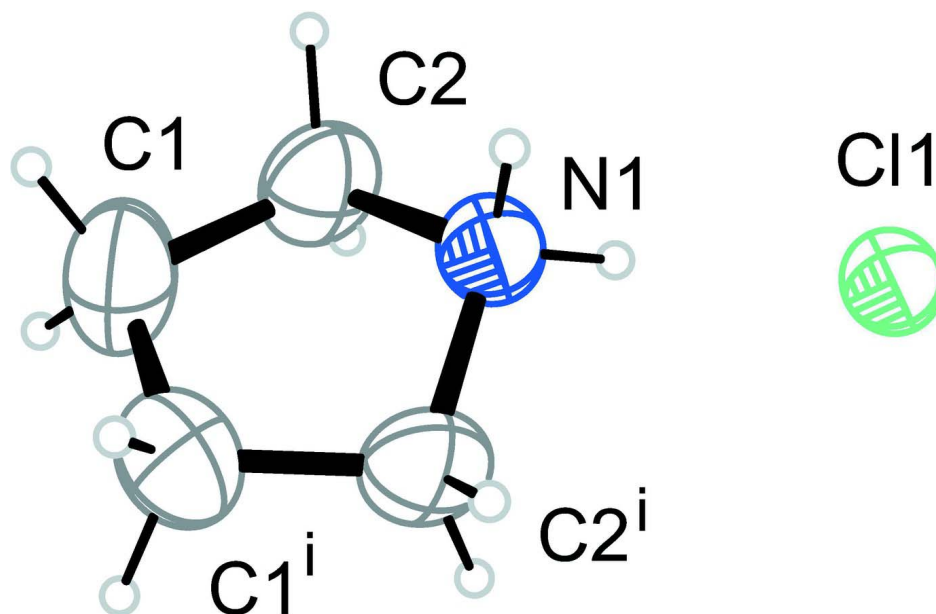


Figure 1

The molecular structure of the *C*₁ symmetric title compound with anisotropic displacement ellipsoids drawn at 50% probability for non-H atoms. Symmetry code: (i) *x*, -*y* + 1/2, *z*.

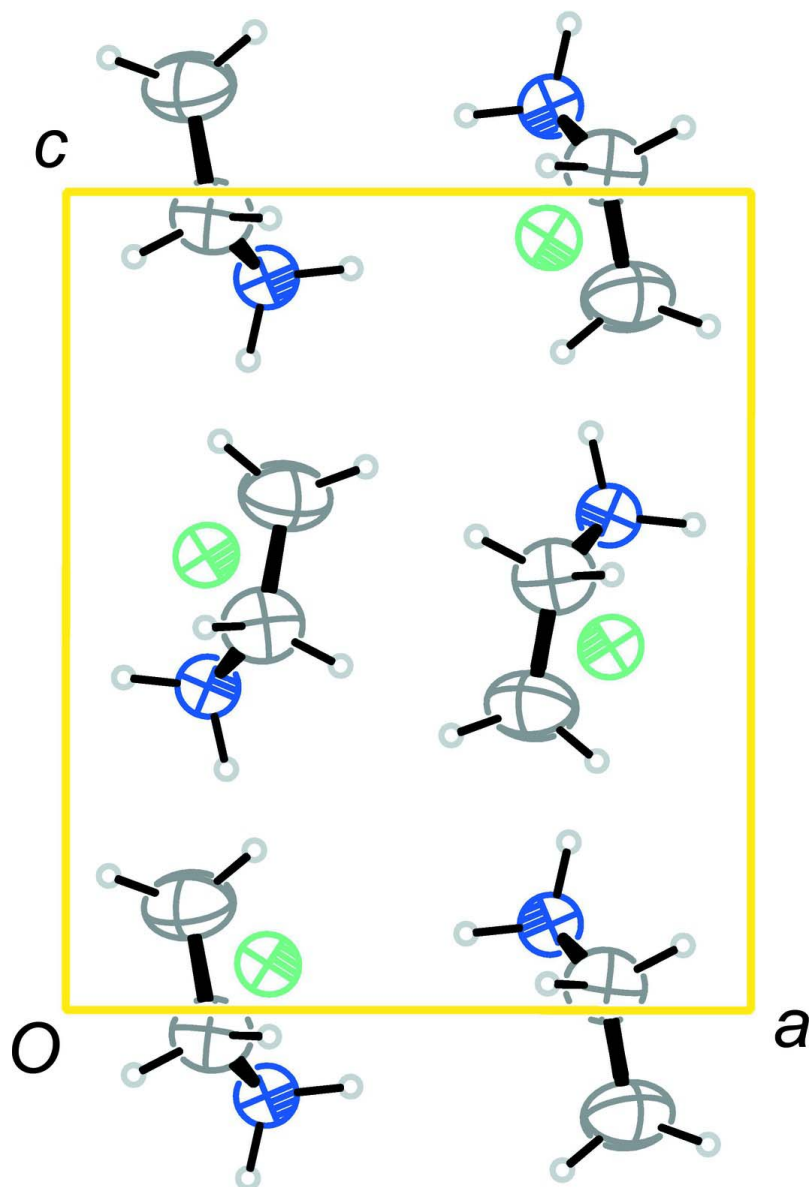


Figure 2
Packing of the title compound, viewed along $[0\ 1\ 0]$.

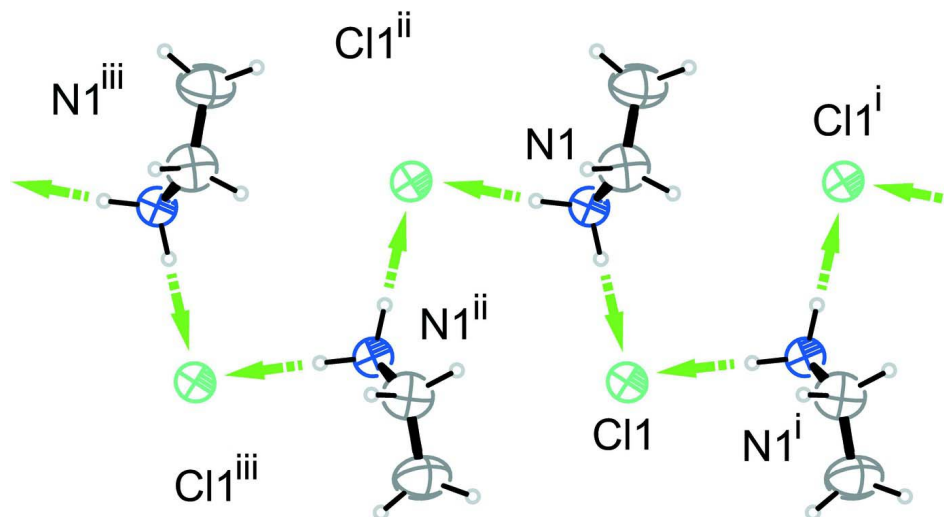


Figure 3

N—H...Cl hydrogen bonds lead to chain-like structures in the crystal structure along [1 0 0], shown here normal to [0 1 0]. Symmetry codes: (i) $x + 1/2, -y + 1/2, -z + 1/2$; (ii) $x - 1/2, -y + 1/2, -z + 1/2$; (iii) $x - 1, y, z$.

Pyrrolidinium chloride

Crystal data

$C_4H_{10}N^+ \cdot Cl^-$

$M_r = 107.58$

Orthorhombic, $Pnma$

Hall symbol: $-P\ 2ac\ 2n$

$a = 7.4429$ (4) Å

$b = 9.4104$ (5) Å

$c = 8.9021$ (4) Å

$V = 623.51$ (5) Å³

$Z = 4$

$F(000) = 232$

$D_x = 1.146$ Mg m⁻³

Mo $K\alpha$ radiation, $\lambda = 0.71073$ Å

Cell parameters from 2321 reflections

$\theta = 3.1\text{--}27.5^\circ$

$\mu = 0.48$ mm⁻¹

$T = 200$ K

Block, colourless

$0.22 \times 0.13 \times 0.12$ mm

Data collection

Nonius KappaCCD

diffractometer

Radiation source: rotating anode

MONTEL, graded multilayered X-ray optics

monochromator

φ and ω scans

4239 measured reflections

756 independent reflections

608 reflections with $I > 2\sigma(I)$

$R_{int} = 0.037$

$\theta_{max} = 27.5^\circ, \theta_{min} = 3.2^\circ$

$h = -8 \rightarrow 9$

$k = -12 \rightarrow 12$

$l = -11 \rightarrow 10$

Refinement

Refinement on F^2

Least-squares matrix: full

$R[F^2 > 2\sigma(F^2)] = 0.040$

$wR(F^2) = 0.120$

$S = 1.07$

756 reflections

31 parameters

0 restraints

Primary atom site location: structure-invariant

direct methods

Secondary atom site location: difference Fourier map

Hydrogen site location: inferred from neighbouring sites

H-atom parameters constrained

$w = 1/[\sigma^2(F_o^2) + (0.0625P)^2 + 0.1854P]$

where $P = (F_o^2 + 2F_c^2)/3$

$(\Delta/\sigma)_{max} < 0.001$

$\Delta\rho_{max} = 0.21$ e Å⁻³

$\Delta\rho_{min} = -0.24$ e Å⁻³

Special details

Refinement. Hydrogen atoms were placed in calculated positions (C–H 0.99 Å for methylene C atoms and N–H 0.92 Å for N atoms) and were included in the refinement in the riding model approximation with $U(\text{H})$ set to 1.2 $U_{\text{eq}}(\text{C})$ for C atoms and 1.2 $U_{\text{eq}}(\text{N})$ for N atoms.

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (\AA^2)

	<i>x</i>	<i>y</i>	<i>z</i>	$U_{\text{iso}}^*/U_{\text{eq}}$
N1	0.2072 (3)	0.2500	0.3955 (3)	0.0473 (6)
H101	0.2341	0.2500	0.2946	0.057*
H102	0.0843	0.2500	0.4065	0.057*
C1	0.3204 (4)	0.1712 (3)	0.6283 (3)	0.0781 (8)
H11	0.2249	0.1348	0.6954	0.094*
H12	0.4376	0.1348	0.6641	0.094*
C2	0.2871 (3)	0.1242 (2)	0.4702 (3)	0.0623 (6)
H21	0.4009	0.0963	0.4208	0.075*
H22	0.2032	0.0427	0.4677	0.075*
Cl1	0.29515 (8)	0.2500	0.05610 (7)	0.0488 (3)

Atomic displacement parameters (\AA^2)

	U^{11}	U^{22}	U^{33}	U^{12}	U^{13}	U^{23}
N1	0.0466 (12)	0.0500 (13)	0.0452 (12)	0.000	−0.0007 (10)	0.000
C1	0.102 (2)	0.0775 (16)	0.0553 (14)	0.0109 (14)	−0.0060 (13)	0.0106 (12)
C2	0.0803 (16)	0.0411 (11)	0.0657 (14)	0.0045 (10)	0.0013 (11)	0.0050 (9)
Cl1	0.0477 (4)	0.0528 (4)	0.0459 (4)	0.000	−0.0020 (3)	0.000

Geometric parameters (\AA , $^\circ$)

N1—C2 ⁱ	1.482 (3)	C1—C2	1.495 (4)
N1—C2	1.482 (2)	C1—H11	0.9900
N1—H101	0.9200	C1—H12	0.9900
N1—H102	0.9200	C2—H21	0.9900
C1—C1 ⁱ	1.482 (5)	C2—H22	0.9900
C2 ⁱ —N1—C2	105.9 (2)	C1 ⁱ —C1—H12	110.3
C2 ⁱ —N1—H101	110.5	C2—C1—H12	110.3
C2—N1—H101	110.5	H11—C1—H12	108.5
C2 ⁱ —N1—H102	110.5	N1—C2—C1	104.64 (19)
C2—N1—H102	110.5	N1—C2—H21	110.8
H101—N1—H102	108.7	C1—C2—H21	110.8
C1 ⁱ —C1—C2	107.20 (13)	N1—C2—H22	110.8
C1 ⁱ —C1—H11	110.3	C1—C2—H22	110.8
C2—C1—H11	110.3	H21—C2—H22	108.9
C2 ⁱ —N1—C2—C1	31.4 (3)	C1 ⁱ —C1—C2—N1	−19.16 (18)

Symmetry code: (i) $x, -y+1/2, z$.

Hydrogen-bond geometry (Å, °)

<i>D</i> —H \cdots <i>A</i>	<i>D</i> —H	H \cdots <i>A</i>	<i>D</i> \cdots <i>A</i>	<i>D</i> —H \cdots <i>A</i>
N1—H101 \cdots C11	0.92	2.17	3.091 (3)	180
N1—H102 \cdots C11 ⁱⁱ	0.92	2.18	3.097 (2)	177

Symmetry code: (ii) $x-1/2, y, -z+1/2$.