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(E)-1-(4-Chlorophenyl)-3-[4-(dimethylamino)phenyl]prop-2-en-1-one

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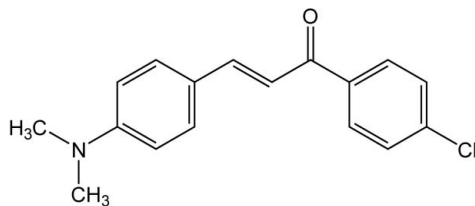
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 Key indicators: single-crystal X-ray study; $T = 298$ K; mean $\sigma(\text{C}-\text{C}) = 0.005$ Å; disorder in main residue; R factor = 0.059; wR factor = 0.170; data-to-parameter ratio = 11.0.

The title compound, $\text{C}_{17}\text{H}_{16}\text{ClNO}$, was synthesized using a solvent-free method by reaction of 4-(dimethylamino)-benzaldehyde with 4-chloroacetophenone and NaOH. The chlorophenyl ring makes a dihedral angle of $18.1(3)^\circ$ with the central propenone unit, while the (dimethylamino)phenyl group is disordered over two orientations of equal occupancies, which make dihedral angles with the central propenone unit of $32.9(3)^\circ$ and $57.4(3)^\circ$, respectively.

Related literature

 For a related structure, see: Li *et al.* (1992).


Experimental

Crystal data

$\text{C}_{17}\text{H}_{16}\text{ClNO}$	$V = 1479.5(3) \text{ \AA}^3$
$M_r = 285.76$	$Z = 4$
Monoclinic, $P2_1/c$	Mo $K\alpha$ radiation
$a = 16.792(2) \text{ \AA}$	$\mu = 0.25 \text{ mm}^{-1}$
$b = 14.5602(16) \text{ \AA}$	$T = 298 \text{ K}$
$c = 6.1160(8) \text{ \AA}$	$0.42 \times 0.20 \times 0.13 \text{ mm}$
$\beta = 98.333(2)^\circ$	

Data collection

Bruker SMART CCD diffractometer	7357 measured reflections
Absorption correction: multi-scan (SADABS; Sheldrick, 2003)	2605 independent reflections
$T_{\min} = 0.901$, $T_{\max} = 0.968$	1066 reflections with $I > 2\sigma(I)$
	$R_{\text{int}} = 0.068$

Refinement

$R[F^2 > 2\sigma(F^2)] = 0.059$	4 restraints
$wR(F^2) = 0.170$	H-atom parameters constrained
$S = 0.90$	$\Delta\rho_{\text{max}} = 0.16 \text{ e \AA}^{-3}$
2605 reflections	$\Delta\rho_{\text{min}} = -0.12 \text{ e \AA}^{-3}$
237 parameters	

Data collection: SMART (Bruker, 1997); cell refinement: SAINT (Bruker, 1997); data reduction: SAINT; program(s) used to solve structure: SHELXS97 (Sheldrick, 2008); program(s) used to refine structure: SHELXL97 (Sheldrick, 2008); molecular graphics: SHELXTL (Sheldrick, 2008); software used to prepare material for publication: SHELXTL.

This project was supported by the Foundation of Tianshui Normal University.

Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: B12339).

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supporting information

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(E)-1-(4-Chlorophenyl)-3-[4-(dimethylamino)phenyl]prop-2-en-1-one

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S1. Comment

This paper discloses a user-friendly, solvent-free protocol for the synthesis of chalcones, starting from the fragrant aldehydes and fragrant ketones in the presence of NaOH. The method can be considered to be a general route for chalcone synthesis, and the title compound was prepared in this way.

The bond lengths and angles of the molecule (Fig. 1) are normal and comparable to those observed in the related compound (Li *et al.*, 1992). The (dimethylamino)phenyl group exhibits rotational disorder, with one orientation including atoms C10, C11, C12, C13, C14 and C15, and another orientation including C10, C11', C12', C13, C14' and C15'. The refined site occupancy factors for the two orientations is 0.500 (5).

S2. Experimental

4-(Dimethylamino)benzaldehyde (0.5 mmol) and 4-chloroacetophenone (0.5 mmol), NaOH (0.5 mmol) were mixed in 50 ml flask under solvent-free conditions. After stirring for 6 min at 373 K, the mixture was slowly solidified to give the title compound. Recrystallization from ethanol gave a yellow crystalline solid. Elemental analysis calculated: C 71.45, H 5.64, N 4.90%; found: C 71.53, H 5.56, N 4.95%.

S3. Refinement

All H atoms were positioned geometrically and refined using a riding model with C—H = 0.93–0.96 Å and $U_{\text{iso}}(\text{H}) = 1.2$ or $1.5U_{\text{eq}}(\text{C})$. The five-membered ring was treated as disordered between two orientations with site occupancy factors refined to 0.500 (5). The bonds N1—C16, N1—C16', N1—C17 and N1—C17' were restrained to be 1.47 (1)Å.

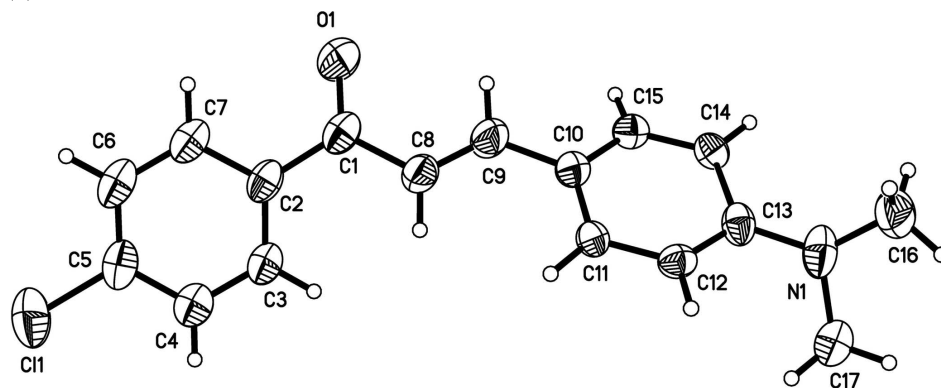


Figure 1

The molecular structure showing 30% probability displacement ellipsoids for non-H atoms.

(E)-1-(4-Chlorophenyl)-3-[4-(dimethylamino)phenyl]prop-2-en-1-one*Crystal data*C₁₇H₁₆ClNO $M_r = 285.76$ Monoclinic, $P2_1/c$

Hall symbol: -P 2ybc

 $a = 16.792$ (2) Å $b = 14.5602$ (16) Å $c = 6.1160$ (8) Å $\beta = 98.333$ (2)° $V = 1479.5$ (3) Å³ $Z = 4$ $F(000) = 600$ $D_x = 1.283$ Mg m⁻³Mo $K\alpha$ radiation, $\lambda = 0.71073$ Å

Cell parameters from 987 reflections

 $\theta = 2.5$ – 19.9 ° $\mu = 0.25$ mm⁻¹ $T = 298$ K

Needle, yellow

 $0.42 \times 0.20 \times 0.13$ mm*Data collection*

Bruker SMART CCD

diffractometer

Radiation source: fine-focus sealed tube

Graphite monochromator

 φ and ω scansAbsorption correction: multi-scan
(SADABS; Sheldrick, 2003) $T_{\min} = 0.901$, $T_{\max} = 0.968$

7357 measured reflections

2605 independent reflections

1066 reflections with $I > 2\sigma(I)$ $R_{\text{int}} = 0.068$ $\theta_{\text{max}} = 25.0$ °, $\theta_{\text{min}} = 1.9$ ° $h = -19 \rightarrow 19$ $k = -17 \rightarrow 12$ $l = -7 \rightarrow 7$ *Refinement*Refinement on F^2

Least-squares matrix: full

 $R[F^2 > 2\sigma(F^2)] = 0.059$ $wR(F^2) = 0.170$ $S = 0.90$

2605 reflections

237 parameters

4 restraints

Primary atom site location: structure-invariant
direct methodsSecondary atom site location: difference Fourier
mapHydrogen site location: inferred from
neighbouring sites

H-atom parameters constrained

 $w = 1/[\sigma^2(F_o^2) + (0.0749P)^2]$ where $P = (F_o^2 + 2F_c^2)/3$ $(\Delta/\sigma)_{\text{max}} < 0.001$ $\Delta\rho_{\text{max}} = 0.16$ e Å⁻³ $\Delta\rho_{\text{min}} = -0.12$ e Å⁻³*Special details*

Geometry. All e.s.d.'s (except the e.s.d. in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell e.s.d.'s are taken into account individually in the estimation of e.s.d.'s in distances, angles and torsion angles; correlations between e.s.d.'s in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell e.s.d.'s is used for estimating e.s.d.'s involving l.s. planes.

Refinement. Refinement of F^2 against ALL reflections. The weighted R -factor wR and goodness of fit S are based on F^2 , conventional R -factors R are based on F , with F set to zero for negative F^2 . The threshold expression of $F^2 > \sigma(F^2)$ is used only for calculating R -factors(gt) etc. and is not relevant to the choice of reflections for refinement. R -factors based on F^2 are statistically about twice as large as those based on F , and R -factors based on ALL data will be even larger.

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (Å²)

	x	y	z	$U_{\text{iso}}^*/U_{\text{eq}}$	Occ. (<1)
Cl1	1.12683 (6)	0.14642 (9)	1.2060 (3)	0.1285 (7)	
N1	0.34724 (18)	0.1225 (2)	0.8419 (7)	0.0891 (11)	
O1	0.80650 (17)	0.1009 (2)	0.4377 (6)	0.1225 (12)	
C1	0.8071 (2)	0.1135 (2)	0.6344 (8)	0.0739 (11)	

C2	0.8853 (2)	0.1192 (2)	0.7831 (7)	0.0683 (10)	
C3	0.8921 (2)	0.1538 (2)	0.9959 (7)	0.0742 (11)	
H3	0.8460	0.1712	1.0530	0.089*	
C4	0.9663 (2)	0.1627 (2)	1.1247 (7)	0.0801 (11)	
H4	0.9701	0.1868	1.2666	0.096*	
C5	1.0345 (2)	0.1356 (3)	1.0416 (9)	0.0831 (12)	
C6	1.0292 (3)	0.0990 (3)	0.8348 (9)	0.0889 (13)	
H6	1.0755	0.0792	0.7816	0.107*	
C7	0.9552 (2)	0.0914 (2)	0.7048 (7)	0.0792 (11)	
H7	0.9520	0.0673	0.5629	0.095*	
C8	0.7314 (2)	0.1221 (2)	0.7236 (7)	0.0754 (11)	
H8	0.7343	0.1272	0.8762	0.090*	
C9	0.6615 (2)	0.1233 (3)	0.6088 (7)	0.0872 (12)	
H9	0.6617	0.1225	0.4568	0.105*	
C10	0.5811 (2)	0.1255 (3)	0.6741 (6)	0.0639 (9)	
C11	0.5655 (5)	0.0979 (6)	0.8724 (14)	0.055 (2)	0.500 (5)
H11	0.6076	0.0821	0.9821	0.065*	0.500 (5)
C12	0.4860 (5)	0.0928 (6)	0.9147 (13)	0.059 (2)	0.500 (5)
H12	0.4773	0.0651	1.0463	0.070*	0.500 (5)
C13	0.4234 (2)	0.1232 (2)	0.7863 (7)	0.0639 (10)	
C14	0.4419 (4)	0.1658 (5)	0.5679 (11)	0.057 (2)	0.500 (5)
H14	0.4007	0.1885	0.4641	0.068*	0.500 (5)
C15	0.5208 (4)	0.1695 (5)	0.5295 (12)	0.059 (2)	0.500 (5)
H15	0.5339	0.2012	0.4075	0.070*	0.500 (5)
C16	0.2783 (5)	0.0794 (8)	0.7432 (18)	0.103 (4)	0.500 (5)
H16A	0.2583	0.1102	0.6074	0.154*	0.510 (7)
H16B	0.2383	0.0815	0.8404	0.154*	0.510 (7)
H16C	0.2902	0.0166	0.7128	0.154*	0.510 (7)
C17	0.3364 (5)	0.1683 (7)	1.0661 (13)	0.106 (4)	0.500 (5)
H17A	0.3811	0.1526	1.1758	0.158*	0.510 (7)
H17B	0.2874	0.1466	1.1123	0.158*	0.510 (7)
H17C	0.3338	0.2337	1.0483	0.158*	0.510 (7)
C11'	0.5658 (5)	0.1596 (6)	0.8637 (16)	0.060 (2)	0.500 (5)
H11'	0.6089	0.1842	0.9581	0.073*	0.500 (5)
C12'	0.4923 (4)	0.1618 (6)	0.9327 (12)	0.060 (2)	0.500 (5)
H12'	0.4859	0.1870	1.0688	0.072*	0.500 (5)
C14'	0.4368 (4)	0.0774 (5)	0.6145 (12)	0.059 (2)	0.500 (5)
H14'	0.3949	0.0445	0.5344	0.070*	0.500 (5)
C15'	0.5112 (4)	0.0760 (5)	0.5472 (12)	0.064 (2)	0.500 (5)
H15'	0.5184	0.0436	0.4203	0.076*	0.500 (5)
C16'	0.2804 (4)	0.1498 (7)	0.6472 (14)	0.090 (3)	0.500 (5)
H16D	0.2919	0.2095	0.5931	0.135*	0.490 (7)
H16E	0.2290	0.1509	0.6988	0.135*	0.490 (7)
H16F	0.2791	0.1056	0.5302	0.135*	0.490 (7)
C17'	0.3231 (5)	0.0619 (6)	0.9866 (13)	0.079 (3)	0.500 (5)
H17D	0.3431	0.0019	0.9590	0.119*	0.490 (7)
H17E	0.2654	0.0604	0.9688	0.119*	0.490 (7)
H17F	0.3437	0.0804	1.1347	0.119*	0.490 (7)

Atomic displacement parameters (Å²)

	U^{11}	U^{22}	U^{33}	U^{12}	U^{13}	U^{23}
C11	0.0640 (8)	0.1472 (12)	0.1697 (14)	-0.0010 (6)	0.0012 (8)	-0.0022 (9)
N1	0.053 (2)	0.084 (2)	0.132 (3)	0.0039 (18)	0.021 (2)	-0.008 (3)
O1	0.085 (2)	0.211 (4)	0.078 (2)	0.001 (2)	0.0324 (18)	0.006 (2)
C1	0.069 (3)	0.072 (3)	0.087 (3)	-0.0025 (19)	0.033 (2)	0.000 (2)
C2	0.061 (2)	0.061 (2)	0.089 (3)	-0.0045 (18)	0.032 (2)	0.003 (2)
C3	0.059 (3)	0.080 (3)	0.087 (3)	0.0014 (18)	0.024 (2)	-0.003 (2)
C4	0.074 (3)	0.075 (3)	0.094 (3)	-0.001 (2)	0.022 (2)	-0.003 (2)
C5	0.060 (3)	0.074 (3)	0.118 (4)	0.003 (2)	0.023 (3)	0.007 (3)
C6	0.071 (3)	0.084 (3)	0.120 (4)	0.003 (2)	0.043 (3)	-0.005 (3)
C7	0.074 (3)	0.074 (3)	0.097 (3)	0.001 (2)	0.037 (3)	-0.007 (2)
C8	0.062 (2)	0.090 (3)	0.077 (3)	-0.010 (2)	0.021 (2)	-0.015 (2)
C9	0.071 (3)	0.130 (3)	0.063 (3)	0.021 (2)	0.018 (2)	0.019 (2)
C10	0.058 (2)	0.075 (2)	0.059 (3)	0.007 (2)	0.0076 (19)	0.001 (2)
C11	0.050 (5)	0.057 (5)	0.056 (6)	0.007 (4)	0.003 (4)	0.004 (5)
C12	0.069 (6)	0.057 (5)	0.052 (5)	-0.006 (5)	0.013 (4)	0.000 (4)
C13	0.050 (2)	0.054 (2)	0.087 (3)	0.0039 (19)	0.005 (2)	-0.008 (2)
C14	0.057 (5)	0.051 (5)	0.058 (5)	0.002 (3)	0.000 (3)	0.008 (3)
C15	0.066 (5)	0.053 (5)	0.058 (5)	-0.007 (4)	0.012 (4)	0.003 (3)
C16	0.076 (6)	0.114 (8)	0.118 (10)	-0.024 (6)	0.015 (6)	-0.030 (7)
C17	0.071 (6)	0.151 (9)	0.098 (8)	-0.002 (5)	0.025 (5)	-0.002 (6)
C11'	0.053 (5)	0.058 (6)	0.067 (6)	-0.005 (5)	-0.001 (4)	-0.012 (5)
C12'	0.058 (5)	0.072 (6)	0.051 (5)	-0.008 (5)	0.011 (4)	-0.008 (4)
C14'	0.047 (4)	0.060 (6)	0.067 (5)	-0.005 (3)	0.000 (4)	-0.011 (4)
C15'	0.067 (5)	0.064 (6)	0.060 (5)	0.005 (4)	0.008 (4)	-0.006 (4)
C16'	0.049 (5)	0.111 (8)	0.105 (8)	-0.001 (5)	-0.001 (5)	0.014 (6)
C17'	0.076 (5)	0.094 (7)	0.071 (7)	0.011 (5)	0.023 (5)	0.011 (5)

Geometric parameters (Å, °)

C11—C5	1.729 (4)	C11—H11	0.930
N1—C17'	1.353 (7)	C12—C13	1.296 (8)
N1—C13	1.370 (4)	C12—H12	0.930
N1—C16	1.377 (7)	C13—C14'	1.292 (7)
N1—C17	1.559 (7)	C13—C12'	1.468 (8)
N1—C16'	1.565 (7)	C13—C14	1.545 (8)
O1—C1	1.215 (4)	C14—C15	1.379 (8)
C1—C8	1.460 (5)	C14—H14	0.930
C1—C2	1.486 (5)	C15—H15	0.930
C2—C3	1.385 (5)	C16—H16A	0.960
C2—C7	1.391 (4)	C16—H16B	0.960
C3—C4	1.379 (5)	C16—H16C	0.960
C3—H3	0.930	C17—H17A	0.960
C4—C5	1.376 (5)	C17—H17B	0.960
C4—H4	0.930	C17—H17C	0.960
C5—C6	1.364 (5)	C11'—C12'	1.362 (10)

C6—C7	1.380 (5)	C11'—H11'	0.930
C6—H6	0.930	C12'—H12'	0.930
C7—H7	0.930	C14'—C15'	1.372 (9)
C8—C9	1.278 (5)	C14'—H14'	0.930
C8—H8	0.930	C15'—H15'	0.930
C9—C10	1.463 (5)	C16'—H16D	0.960
C9—H9	0.930	C16'—H16E	0.960
C10—C11'	1.320 (9)	C16'—H16F	0.960
C10—C11	1.339 (9)	C17'—H17D	0.960
C10—C15	1.399 (8)	C17'—H17E	0.960
C10—C15'	1.495 (8)	C17'—H17F	0.960
C11—C12	1.398 (10)		
C17'—N1—C13	123.1 (5)	N1—C13—C12'	122.2 (5)
C13—N1—C16	130.6 (5)	C12—C13—C14	114.4 (5)
C13—N1—C17	116.6 (4)	N1—C13—C14	122.0 (4)
C16—N1—C17	112.5 (5)	C15—C14—C13	118.8 (6)
C17'—N1—C16'	114.4 (6)	C15—C14—H14	120.6
C13—N1—C16'	113.7 (5)	C13—C14—H14	120.6
O1—C1—C8	120.0 (4)	C14—C15—C10	119.9 (6)
O1—C1—C2	119.6 (3)	C14—C15—H15	120.0
C8—C1—C2	120.4 (4)	C10—C15—H15	120.0
C3—C2—C7	118.0 (4)	N1—C16—H16A	109.5
C3—C2—C1	122.8 (3)	N1—C16—H16B	109.5
C7—C2—C1	119.2 (4)	H16A—C16—H16B	109.5
C4—C3—C2	121.1 (4)	N1—C16—H16C	109.5
C4—C3—H3	119.5	H16A—C16—H16C	109.5
C2—C3—H3	119.5	H16B—C16—H16C	109.5
C5—C4—C3	119.5 (4)	N1—C17—H17A	109.5
C5—C4—H4	120.2	N1—C17—H17B	109.5
C3—C4—H4	120.2	H17A—C17—H17B	109.5
C6—C5—C4	120.6 (4)	N1—C17—H17C	109.5
C6—C5—C11	120.5 (3)	H17A—C17—H17C	109.5
C4—C5—C11	118.9 (4)	H17B—C17—H17C	109.5
C5—C6—C7	119.8 (4)	C10—C11'—C12'	125.5 (7)
C5—C6—H6	120.1	C10—C11'—H11'	117.2
C7—C6—H6	120.1	C12'—C11'—H11'	117.2
C6—C7—C2	120.9 (4)	C11'—C12'—C13	118.2 (7)
C6—C7—H7	119.6	C11'—C12'—H12'	120.9
C2—C7—H7	119.6	C13—C12'—H12'	120.9
C9—C8—C1	125.2 (4)	C13—C14'—C15'	121.8 (6)
C9—C8—H8	117.4	C13—C14'—H14'	119.1
C1—C8—H8	117.4	C15'—C14'—H14'	119.1
C8—C9—C10	131.4 (4)	C14'—C15'—C10	121.1 (6)
C8—C9—H9	114.3	C14'—C15'—H15'	119.4
C10—C9—H9	114.3	C10—C15'—H15'	119.4
C11—C10—C15	119.4 (5)	N1—C16'—H16D	109.5
C11—C10—C9	123.4 (4)	N1—C16'—H16E	109.5

C15—C10—C9	116.7 (4)	H16D—C16'—H16E	109.5
C9—C10—C15'	122.2 (4)	N1—C16'—H16F	109.5
C10—C11—C12	120.1 (6)	H16D—C16'—H16F	109.5
C10—C11—H11	120.0	H16E—C16'—H16F	109.5
C12—C11—H11	120.0	N1—C17'—H17D	109.5
C13—C12—C11	125.6 (7)	N1—C17'—H17E	109.5
C13—C12—H12	117.2	H17D—C17'—H17E	109.5
C11—C12—H12	117.2	N1—C17'—H17F	109.5
C14'—C13—N1	118.5 (4)	H17D—C17'—H17F	109.5
C12—C13—N1	123.5 (5)	H17E—C17'—H17F	109.5
C14'—C13—C12'	118.6 (5)		
