

Acta Crystallographica Section E

## Structure Reports

Online

ISSN 1600-5368

## N-(2-Methoxyphenyl)phthalimide

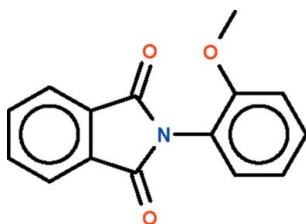
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Received 17 August 2009; accepted 18 August 2009

Key indicators: single-crystal X-ray study;  $T = 123$  K; mean  $\sigma(\text{C}-\text{C}) = 0.002$  Å;  
 $R$  factor = 0.036;  $wR$  factor = 0.098; data-to-parameter ratio = 15.5.The phthalimide fused-ring system and the phenylene ring in the title compound,  $\text{C}_{15}\text{H}_{11}\text{NO}_3$ , are inclined at an angle of  $54.2$  (1)°.

## Related literature

For the crystal structures of *N*-(phenyl)phthalimides, see: Izotova *et al.* (2009); Magomedova *et al.* (1980). For that of the 2-ethyl-substituted derivative, see: Fan *et al.* (2008).

## Experimental

## Crystal data

 $\text{C}_{15}\text{H}_{11}\text{NO}_3$  $M_r = 253.25$ Monoclinic,  $P2_1/n$   
 $a = 11.8505$  (2) Å  
 $b = 6.6903$  (1) Å  
 $c = 15.3264$  (3) Å  
 $\beta = 106.258$  (1)°  
 $V = 1166.54$  (3) Å<sup>3</sup> $Z = 4$   
Mo  $K\alpha$  radiation  
 $\mu = 0.10$  mm<sup>-1</sup>  
 $T = 123$  K  
 $0.28 \times 0.16 \times 0.04$  mm

## Data collection

Bruker SMART APEX  
diffractometer  
Absorption correction: none  
10672 measured reflections2682 independent reflections  
2190 reflections with  $I > 2\sigma(I)$   
 $R_{\text{int}} = 0.028$ 

## Refinement

 $R[F^2 > 2\sigma(F^2)] = 0.036$   
 $wR(F^2) = 0.098$   
 $S = 1.03$   
2682 reflections173 parameters  
H-atom parameters constrained  
 $\Delta\rho_{\text{max}} = 0.27$  e Å<sup>-3</sup>  
 $\Delta\rho_{\text{min}} = -0.22$  e Å<sup>-3</sup>

Data collection: APEX2 (Bruker, 2008); cell refinement: SAINT (Bruker, 2008); data reduction: SAINT; program(s) used to solve structure: SHELXS97 (Sheldrick, 2008); program(s) used to refine structure: SHELXL97 (Sheldrick, 2008); molecular graphics: X-SEED (Barbour, 2001); software used to prepare material for publication: publCIF (Westrip, 2009).

We thank the University of Malaya for supporting this study.

Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: BT5039).

## References

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## supporting information

*Acta Cryst.* (2009). E65, o2218 [doi:10.1107/S1600536809032826]

***N*-(2-Methoxyphenyl)phthalimide**

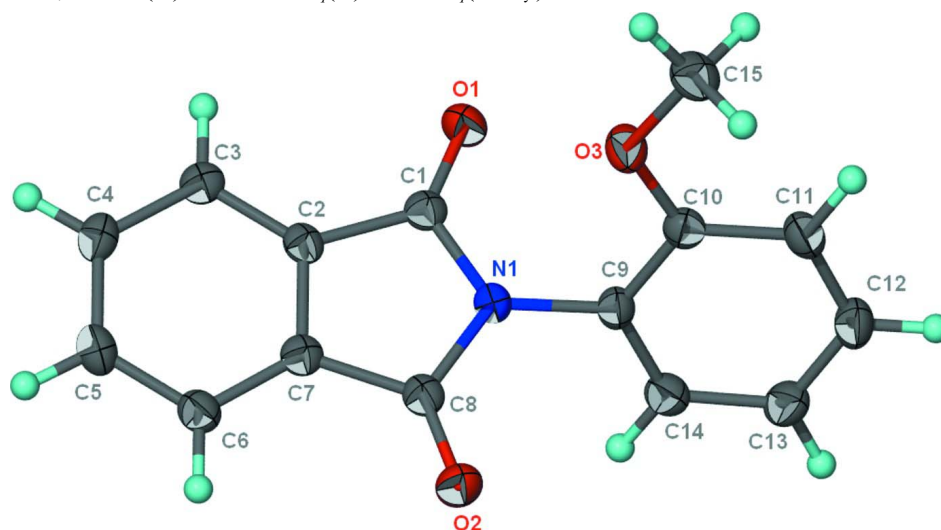
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**S1. Experimental**

Phthalic anhydride (5.00 g, 33.8 mmol) and 4-methoxyaniline (4.99 g, 40.5 mmol) were heated in acetic acid (15 ml) for 4 h. The mixture was cooled and then poured into water. The solid that separated was collected and recrystallized from ethanol in 90% yield.

**S2. Refinement**

H-atoms were placed in calculated positions (C—H 0.95 or 0.98 Å) and were included in the refinement in the riding model approximation, with  $U(\text{H})$  set to 1.2  $U_{eq}(\text{C})$  or 1.5  $U_{eq}(\text{C}_{\text{methyl}})$ .



**Figure 1**

Anisotropic displacement ellipsoid plot (Barbour, 2001) of  $\text{C}_{15}\text{H}_{11}\text{NO}_3$  at the 70% probability level; hydrogen atoms are drawn as spheres of arbitrary radius.

***N*-(2-Methoxyphenyl)phthalimide***Crystal data*

$\text{C}_{15}\text{H}_{11}\text{NO}_3$

$M_r = 253.25$

Monoclinic,  $P2_1/n$

Hall symbol:  $-P 2_1/n$

$a = 11.8505(2) \text{ \AA}$

$b = 6.6903(1) \text{ \AA}$

$c = 15.3264(3) \text{ \AA}$

$\beta = 106.258(1)^\circ$

$V = 1166.54(3) \text{ \AA}^3$

$Z = 4$

$F(000) = 528$

$D_x = 1.442 \text{ Mg m}^{-3}$

Mo  $K\alpha$  radiation,  $\lambda = 0.71073 \text{ \AA}$

Cell parameters from 3112 reflections

$\theta = 2.5\text{--}28.3^\circ$

$\mu = 0.10 \text{ mm}^{-1}$

$T = 123$  K  
Irregular, colorless

$0.28 \times 0.16 \times 0.04$  mm

*Data collection*

Bruker SMART APEX  
diffractometer  
Radiation source: fine-focus sealed tube  
Graphite monochromator  
 $\omega$  scans  
10672 measured reflections  
2682 independent reflections

2190 reflections with  $I > 2\sigma(I)$   
 $R_{\text{int}} = 0.028$   
 $\theta_{\text{max}} = 27.5^\circ$ ,  $\theta_{\text{min}} = 1.9^\circ$   
 $h = -14 \rightarrow 15$   
 $k = -8 \rightarrow 8$   
 $l = -19 \rightarrow 19$

*Refinement*

Refinement on  $F^2$   
Least-squares matrix: full  
 $R[F^2 > 2\sigma(F^2)] = 0.036$   
 $wR(F^2) = 0.098$   
 $S = 1.03$   
2682 reflections  
173 parameters  
0 restraints  
Primary atom site location: structure-invariant  
direct methods

Secondary atom site location: difference Fourier  
map  
Hydrogen site location: inferred from  
neighbouring sites  
H-atom parameters constrained  
 $w = 1/[\sigma^2(F_o^2) + (0.0506P)^2 + 0.3213P]$   
where  $P = (F_o^2 + 2F_c^2)/3$   
 $(\Delta/\sigma)_{\text{max}} = 0.001$   
 $\Delta\rho_{\text{max}} = 0.27 \text{ e } \text{\AA}^{-3}$   
 $\Delta\rho_{\text{min}} = -0.22 \text{ e } \text{\AA}^{-3}$

*Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters ( $\text{\AA}^2$ )*

	<i>x</i>	<i>y</i>	<i>z</i>	$U_{\text{iso}}^*/U_{\text{eq}}$
O1	0.76144 (8)	0.26041 (14)	0.47272 (6)	0.0232 (2)
O2	0.74087 (8)	0.35172 (14)	0.76449 (6)	0.0242 (2)
O3	0.89049 (8)	0.63238 (14)	0.56474 (6)	0.0240 (2)
N1	0.78134 (9)	0.32357 (16)	0.62507 (7)	0.0183 (2)
C1	0.71795 (11)	0.28292 (18)	0.53464 (8)	0.0177 (3)
C2	0.59283 (11)	0.26801 (17)	0.53474 (8)	0.0168 (3)
C3	0.49248 (11)	0.23766 (18)	0.46394 (9)	0.0187 (3)
H3	0.4962	0.2240	0.4031	0.022*
C4	0.38571 (11)	0.22777 (19)	0.48501 (9)	0.0205 (3)
H4	0.3149	0.2111	0.4376	0.025*
C5	0.38127 (11)	0.24202 (19)	0.57471 (9)	0.0227 (3)
H5	0.3075	0.2333	0.5875	0.027*
C6	0.48287 (11)	0.26884 (19)	0.64603 (9)	0.0203 (3)
H6	0.4801	0.2758	0.7073	0.024*
C7	0.58786 (11)	0.28491 (18)	0.62405 (8)	0.0178 (3)
C8	0.70848 (11)	0.32285 (18)	0.68350 (8)	0.0180 (3)
C9	0.90625 (10)	0.34589 (19)	0.65589 (8)	0.0183 (3)
C10	0.96165 (11)	0.50402 (19)	0.62429 (8)	0.0187 (3)
C11	1.08374 (11)	0.5197 (2)	0.65517 (8)	0.0226 (3)
H11	1.1233	0.6222	0.6324	0.027*
C12	1.14739 (12)	0.3852 (2)	0.71924 (9)	0.0251 (3)
H12	1.2304	0.3983	0.7409	0.030*
C13	1.09199 (12)	0.2325 (2)	0.75213 (9)	0.0242 (3)
H13	1.1362	0.1429	0.7968	0.029*

C14	0.97095 (11)	0.2121 (2)	0.71899 (9)	0.0217 (3)
H14	0.9323	0.1054	0.7398	0.026*
C15	0.94409 (12)	0.8082 (2)	0.54130 (10)	0.0254 (3)
H15A	0.9847	0.8806	0.5968	0.038*
H15B	0.8835	0.8942	0.5026	0.038*
H15C	1.0009	0.7703	0.5084	0.038*

*Atomic displacement parameters (Å<sup>2</sup>)*

	$U^{11}$	$U^{22}$	$U^{33}$	$U^{12}$	$U^{13}$	$U^{23}$
O1	0.0218 (5)	0.0299 (5)	0.0200 (5)	−0.0019 (4)	0.0091 (4)	−0.0012 (4)
O2	0.0228 (5)	0.0319 (5)	0.0176 (5)	−0.0004 (4)	0.0051 (4)	−0.0019 (4)
O3	0.0189 (5)	0.0231 (5)	0.0292 (5)	−0.0019 (4)	0.0052 (4)	0.0068 (4)
N1	0.0151 (5)	0.0223 (6)	0.0175 (5)	−0.0013 (4)	0.0047 (4)	0.0012 (4)
C1	0.0196 (6)	0.0152 (6)	0.0183 (6)	−0.0002 (4)	0.0054 (5)	0.0018 (5)
C2	0.0174 (6)	0.0138 (6)	0.0199 (6)	0.0000 (4)	0.0066 (5)	0.0019 (4)
C3	0.0211 (6)	0.0160 (6)	0.0187 (6)	0.0010 (5)	0.0049 (5)	0.0006 (5)
C4	0.0173 (6)	0.0180 (6)	0.0240 (6)	0.0000 (5)	0.0019 (5)	0.0002 (5)
C5	0.0176 (6)	0.0235 (7)	0.0282 (7)	−0.0013 (5)	0.0084 (5)	−0.0013 (5)
C6	0.0198 (6)	0.0220 (7)	0.0205 (6)	−0.0007 (5)	0.0079 (5)	−0.0001 (5)
C7	0.0196 (6)	0.0151 (6)	0.0186 (6)	−0.0006 (5)	0.0053 (5)	0.0005 (5)
C8	0.0182 (6)	0.0161 (6)	0.0201 (6)	0.0010 (5)	0.0063 (5)	0.0017 (5)
C9	0.0148 (6)	0.0226 (6)	0.0177 (6)	−0.0009 (5)	0.0047 (5)	−0.0025 (5)
C10	0.0186 (6)	0.0209 (6)	0.0174 (6)	0.0003 (5)	0.0061 (5)	−0.0016 (5)
C11	0.0193 (6)	0.0259 (7)	0.0242 (6)	−0.0044 (5)	0.0089 (5)	−0.0016 (5)
C12	0.0155 (6)	0.0337 (8)	0.0256 (7)	−0.0005 (5)	0.0050 (5)	−0.0022 (6)
C13	0.0206 (7)	0.0296 (7)	0.0217 (6)	0.0042 (5)	0.0045 (5)	0.0029 (5)
C14	0.0214 (7)	0.0237 (7)	0.0212 (6)	0.0004 (5)	0.0080 (5)	0.0016 (5)
C15	0.0270 (7)	0.0205 (7)	0.0296 (7)	−0.0032 (5)	0.0097 (6)	0.0032 (5)

*Geometric parameters (Å, °)*

O1—C1	1.2095 (15)	C6—C7	1.3812 (18)
O2—C8	1.2077 (15)	C6—H6	0.9500
O3—C10	1.3604 (15)	C7—C8	1.4862 (17)
O3—C15	1.4296 (16)	C9—C14	1.3818 (18)
N1—C1	1.4057 (16)	C9—C10	1.4010 (18)
N1—C8	1.4073 (16)	C10—C11	1.3946 (17)
N1—C9	1.4301 (15)	C11—C12	1.3885 (19)
C1—C2	1.4865 (17)	C11—H11	0.9500
C2—C3	1.3821 (17)	C12—C13	1.383 (2)
C2—C7	1.3910 (17)	C12—H12	0.9500
C3—C4	1.3923 (18)	C13—C14	1.3879 (18)
C3—H3	0.9500	C13—H13	0.9500
C4—C5	1.3937 (18)	C14—H14	0.9500
C4—H4	0.9500	C15—H15A	0.9800
C5—C6	1.3927 (18)	C15—H15B	0.9800
C5—H5	0.9500	C15—H15C	0.9800

C10—O3—C15	116.90 (10)	O2—C8—C7	129.18 (12)
C1—N1—C8	111.90 (10)	N1—C8—C7	105.46 (10)
C1—N1—C9	124.33 (10)	C14—C9—C10	120.61 (11)
C8—N1—C9	123.56 (10)	C14—C9—N1	118.93 (11)
O1—C1—N1	124.77 (12)	C10—C9—N1	120.43 (11)
O1—C1—C2	129.54 (11)	O3—C10—C11	124.71 (11)
N1—C1—C2	105.66 (10)	O3—C10—C9	116.50 (11)
C3—C2—C7	121.35 (11)	C11—C10—C9	118.79 (11)
C3—C2—C1	130.42 (11)	C12—C11—C10	119.85 (12)
C7—C2—C1	108.21 (11)	C12—C11—H11	120.1
C2—C3—C4	117.57 (12)	C10—C11—H11	120.1
C2—C3—H3	121.2	C13—C12—C11	121.12 (12)
C4—C3—H3	121.2	C13—C12—H12	119.4
C3—C4—C5	120.89 (12)	C11—C12—H12	119.4
C3—C4—H4	119.6	C12—C13—C14	119.12 (12)
C5—C4—H4	119.6	C12—C13—H13	120.4
C6—C5—C4	121.32 (12)	C14—C13—H13	120.4
C6—C5—H5	119.3	C9—C14—C13	120.42 (12)
C4—C5—H5	119.3	C9—C14—H14	119.8
C7—C6—C5	117.29 (12)	C13—C14—H14	119.8
C7—C6—H6	121.4	O3—C15—H15A	109.5
C5—C6—H6	121.4	O3—C15—H15B	109.5
C6—C7—C2	121.53 (11)	H15A—C15—H15B	109.5
C6—C7—C8	129.88 (12)	O3—C15—H15C	109.5
C2—C7—C8	108.59 (11)	H15A—C15—H15C	109.5
O2—C8—N1	125.34 (11)	H15B—C15—H15C	109.5
C8—N1—C1—O1	-174.23 (12)	C9—N1—C8—C7	-176.91 (11)
C9—N1—C1—O1	0.68 (19)	C6—C7—C8—O2	-1.5 (2)
C8—N1—C1—C2	3.86 (13)	C2—C7—C8—O2	177.74 (13)
C9—N1—C1—C2	178.77 (11)	C6—C7—C8—N1	179.86 (12)
O1—C1—C2—C3	-4.7 (2)	C2—C7—C8—N1	-0.90 (13)
N1—C1—C2—C3	177.31 (12)	C1—N1—C9—C14	-117.21 (13)
O1—C1—C2—C7	173.64 (12)	C8—N1—C9—C14	57.11 (17)
N1—C1—C2—C7	-4.33 (13)	C1—N1—C9—C10	64.41 (16)
C7—C2—C3—C4	1.02 (18)	C8—N1—C9—C10	-121.27 (13)
C1—C2—C3—C4	179.20 (12)	C15—O3—C10—C11	-7.96 (18)
C2—C3—C4—C5	-1.90 (18)	C15—O3—C10—C9	172.06 (11)
C3—C4—C5—C6	0.75 (19)	C14—C9—C10—O3	-177.71 (11)
C4—C5—C6—C7	1.31 (19)	N1—C9—C10—O3	0.64 (17)
C5—C6—C7—C2	-2.21 (19)	C14—C9—C10—C11	2.31 (18)
C5—C6—C7—C8	176.94 (12)	N1—C9—C10—C11	-179.34 (11)
C3—C2—C7—C6	1.07 (19)	O3—C10—C11—C12	177.10 (12)
C1—C2—C7—C6	-177.47 (11)	C9—C10—C11—C12	-2.92 (19)
C3—C2—C7—C8	-178.24 (11)	C10—C11—C12—C13	1.2 (2)
C1—C2—C7—C8	3.22 (13)	C11—C12—C13—C14	1.2 (2)
C1—N1—C8—O2	179.33 (12)	C10—C9—C14—C13	0.07 (19)

## supporting information

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C9—N1—C8—O2	4.38 (19)	N1—C9—C14—C13	-178.31 (11)
C1—N1—C8—C7	-1.96 (13)	C12—C13—C14—C9	-1.8 (2)

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