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## Structure Reports

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## Bis(2-butyliminomethyl-5-methoxyphenolato- $\kappa^2N,O^1$ )zinc(II)

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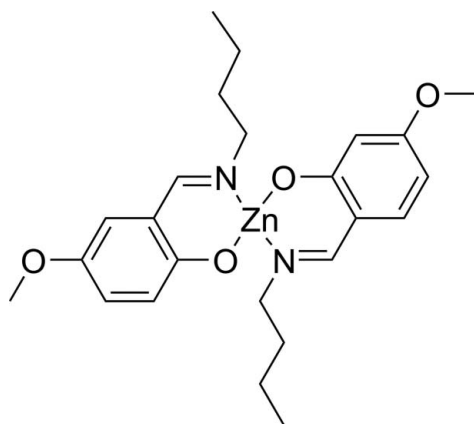
Received 4 August 2009; accepted 5 August 2009

Key indicators: single-crystal X-ray study;  $T = 200$  K; mean  $\sigma(C-C) = 0.004$  Å;  $R$  factor = 0.058;  $wR$  factor = 0.164; data-to-parameter ratio = 19.8.

In the centrosymmetric title compound,  $[Zn(C_{12}H_{16}NO_2)_2]$ , the  $Zn^{II}$  centre is coordinated by two  $O,N$ -bidentate Schiff base ligands, resulting in a slightly distorted *trans*- $ZnN_2O_2$  square-planar geometry for the metal ion. Two short intramolecular  $C-H \cdots O$  contacts occur in the molecule.

### Related literature

For related structures, see: Zhu *et al.* (2003); You *et al.* (2003).



### Experimental

#### Crystal data

$[Zn(C_{12}H_{16}NO_2)_2]$   
 $M_r = 477.89$

Monoclinic,  $P2_1/c$   
 $a = 13.250$  (4) Å

$b = 4.8845$  (15) Å  
 $c = 17.858$  (5) Å  
 $\beta = 93.555$  (5)°  
 $V = 1153.6$  (6) Å<sup>3</sup>  
 $Z = 2$

Mo  $K\alpha$  radiation  
 $\mu = 1.10$  mm<sup>-1</sup>  
 $T = 200$  K  
 $0.30 \times 0.30 \times 0.20$  mm

#### Data collection

Bruker APEX CCD diffractometer  
Absorption correction: multi-scan  
(*SADABS*; Bruker, 2000)  
 $T_{min} = 0.734$ ,  $T_{max} = 0.811$

13300 measured reflections  
2856 independent reflections  
2438 reflections with  $I > 2\sigma(I)$   
 $R_{int} = 0.155$

#### Refinement

$R[F^2 > 2\sigma(F^2)] = 0.058$   
 $wR(F^2) = 0.164$   
 $S = 1.10$   
2856 reflections

144 parameters  
H-atom parameters constrained  
 $\Delta\rho_{max} = 1.13$  e Å<sup>-3</sup>  
 $\Delta\rho_{min} = -0.67$  e Å<sup>-3</sup>

**Table 1**  
Selected bond lengths (Å).

Zn1—O1	1.8769 (17)	Zn1—N1 <sup>1</sup>	2.004 (2)
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Symmetry code: (i)  $-x + 1, -y, -z$ .

**Table 2**  
Hydrogen-bond geometry (Å, °).

$D-H \cdots A$	$D-H$	$H \cdots A$	$D \cdots A$	$D-H \cdots A$
C8—H8B $\cdots$ O1 <sup>1</sup>	0.99	2.34	2.763 (3)	105
C9—H9A $\cdots$ O1 <sup>1</sup>	0.99	2.55	3.087 (3)	114

Symmetry code: (i)  $-x + 1, -y, -z$ .

Data collection: *SMART* (Bruker, 2000); cell refinement: *SAINT* (Bruker, 2000); data reduction: *SAINT*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 2008); program(s) used to refine structure: *SHELXL97* (Sheldrick, 2008); molecular graphics: *SHELXTL* (Sheldrick, 2008); software used to prepare material for publication: *SHELXTL*.

Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: HB5031).

### References

- Bruker (2000). *SMART*, *SAINT* and *SADABS*. Bruker Analytical X-ray Instruments Inc., Madison, Wisconsin, USA.  
Sheldrick, G. M. (2008). *Acta Cryst.* **A64**, 112–122.  
You, Z.-L., Lin, Y.-S., Liu, W.-S., Tan, M.-Y. & Zhu, H.-L. (2003). *Acta Cryst.* **E59**, m1025–m1027.  
Zhu, H.-L., Meng, F.-J., Wang, Z.-D. & Yang, F. (2003). *Z. Kristallogr. New Cryst. Struct.* **218**, 321–322.

## supporting information

*Acta Cryst.* (2009). E65, m1062 [doi:10.1107/S1600536809031109]

**Bis(2-butyliminomethyl-5-methoxyphenolato- $\kappa^2N,O^1$ )zinc(II)**

**Jun Yang, Jin Li, Xian Zhang and Qiang Wang**

**S1. Comment**

As part of our ongoing studies of the structures of zinc(II) complexes (Zhu *et al.*, 2003; You *et al.*, 2003), we now report the crystal structure of the title compound, (I).

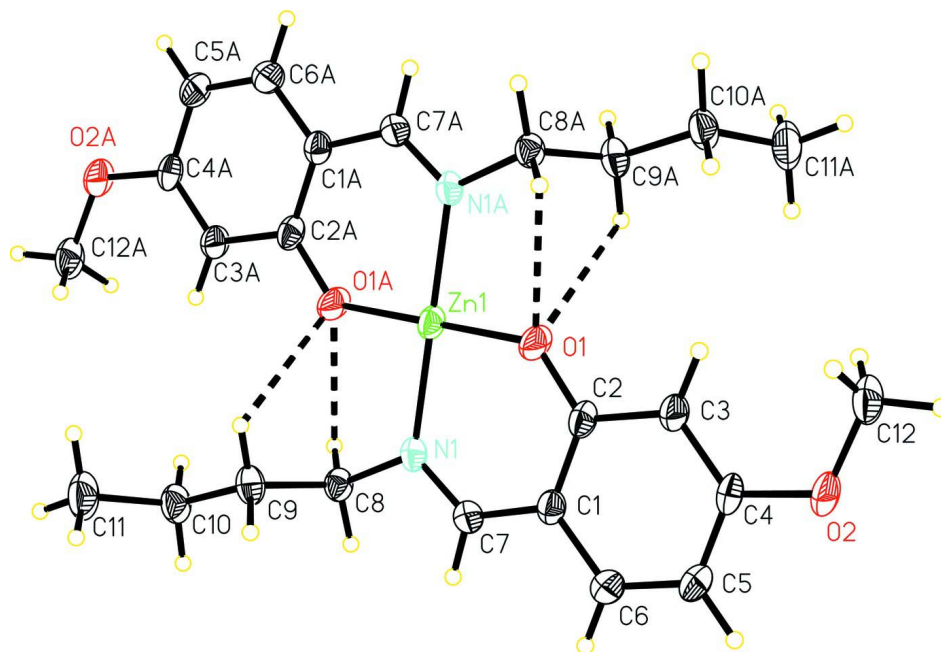
Compound (I) consists of a Zn<sup>II</sup> atom and two bidentate salicylal schiff base ligands (Fig. 1). The central Zn<sup>II</sup> atom exhibits 4-coordination by two N atoms from imine moieties and two O-anions from salicylal groups, forming a slightly distorted square planar geometry (Table 1). Ring Zn1—N1—C7—C1—C2—O1 and Ring Zn1—N1A—C7A—C1A—C2A—O1A, together with the two benzene rings, stand in a plane with mean deviation of 0.0489 °. Two short intramolecular C—H···O contacts occur in the molecule (Table 2).

**S2. Experimental**

0.5 mmol of zinc oxide, 1 mmol of 4-methoxysalicylaldehyde and 1 mmol of butylamine were dissolved in 10 ml methanol. After 3 ml ammonia was added, the result solution was then heated to 423 K for 10 h in a sealed vessel. The reactor was cooled to room temperature at a rate of 10 K h<sup>-1</sup>. The mixture was filtered and held at room temperature for 8 d. Colourless blocks of (I) were isolated in 47% yield.

**S3. Refinement**

All H atoms were placed in geometrically idealized positions (C—H = 0.95—0.99 Å) and refined as riding with  $U_{iso}(H) = 1.2U_{eq}(C)$  or  $1.5U_{eq}(\text{methyl C})$ .

**Figure 1**

The molecular structure of (I). Displacement ellipsoids are drawn at the 30% probability level and dashed lines indicate hydrogen bonds. Atoms with a label A suffix are generated by the symmetry operator (1-x, 1-y, -z).

### Bis(2-butyliminomethyl-5-methoxyphenolato- $\kappa^2N,O^1$ )zinc(II)

#### Crystal data

[Zn(C<sub>12</sub>H<sub>16</sub>NO<sub>2</sub>)<sub>2</sub>]

$M_r = 477.89$

Monoclinic,  $P2_1/c$

Hall symbol: -P 2ybc

$a = 13.250$  (4) Å

$b = 4.8845$  (15) Å

$c = 17.858$  (5) Å

$\beta = 93.555$  (5)°

$V = 1153.6$  (6) Å<sup>3</sup>

$Z = 2$

$F(000) = 504$

$D_x = 1.376$  Mg m<sup>-3</sup>

Mo  $K\alpha$  radiation,  $\lambda = 0.71073$  Å

Cell parameters from 2317 reflections

$\theta = 2.7$ – $26.5$ °

$\mu = 1.10$  mm<sup>-1</sup>

$T = 200$  K

Block, colorless

$0.30 \times 0.30 \times 0.20$  mm

#### Data collection

Bruker APEX CCD

diffractometer

Radiation source: fine-focus sealed tube

Graphite monochromator

$\varphi$  and  $\omega$  scans

Absorption correction: multi-scan

(SADABS; Bruker, 2000)

$T_{\min} = 0.734$ ,  $T_{\max} = 0.811$

13300 measured reflections

2856 independent reflections

2438 reflections with  $I > 2\sigma(I)$

$R_{\text{int}} = 0.155$

$\theta_{\max} = 28.4$ °,  $\theta_{\min} = 2.3$ °

$h = -17 \rightarrow 17$

$k = -6 \rightarrow 6$

$l = -23 \rightarrow 23$

Refinement

Refinement on  $F^2$   
 Least-squares matrix: full  
 $R[F^2 > 2\sigma(F^2)] = 0.058$   
 $wR(F^2) = 0.164$   
 $S = 1.10$   
 2856 reflections  
 144 parameters  
 0 restraints  
 Primary atom site location: structure-invariant  
 direct methods

Secondary atom site location: difference Fourier  
 map  
 Hydrogen site location: inferred from  
 neighbouring sites  
 H-atom parameters constrained  
 $w = 1/[\sigma^2(F_o^2) + (0.0665P)^2 + 0.1817P]$   
 where  $P = (F_o^2 + 2F_c^2)/3$   
 $(\Delta/\sigma)_{\max} < 0.001$   
 $\Delta\rho_{\max} = 1.13 \text{ e } \text{Å}^{-3}$   
 $\Delta\rho_{\min} = -0.67 \text{ e } \text{Å}^{-3}$

Special details

**Geometry.** All e.s.d.'s (except the e.s.d. in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell e.s.d.'s are taken into account individually in the estimation of e.s.d.'s in distances, angles and torsion angles; correlations between e.s.d.'s in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell e.s.d.'s is used for estimating e.s.d.'s involving l.s. planes.

**Refinement.** Refinement of  $F^2$  against ALL reflections. The weighted  $R$ -factor  $wR$  and goodness of fit  $S$  are based on  $F^2$ , conventional  $R$ -factors  $R$  are based on  $F$ , with  $F$  set to zero for negative  $F^2$ . The threshold expression of  $F^2 > \sigma(F^2)$  is used only for calculating  $R$ -factors(gt) *etc.* and is not relevant to the choice of reflections for refinement.  $R$ -factors based on  $F^2$  are statistically about twice as large as those based on  $F$ , and  $R$ -factors based on ALL data will be even larger.

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters ( $\text{Å}^2$ )

	$x$	$y$	$z$	$U_{\text{iso}}^*/U_{\text{eq}}$
C1	0.40796 (16)	0.2347 (5)	0.15609 (13)	0.0375 (5)
C2	0.3582 (2)	0.0191 (5)	0.11694 (18)	0.0403 (6)
C3	0.26490 (19)	-0.0732 (6)	0.14142 (16)	0.0437 (6)
H3	0.2290	-0.2136	0.1143	0.052*
C4	0.2254 (2)	0.0377 (5)	0.20399 (17)	0.0425 (6)
C5	0.27511 (19)	0.2446 (6)	0.24413 (15)	0.0488 (6)
H5	0.2475	0.3196	0.2875	0.059*
C6	0.36504 (18)	0.3396 (6)	0.22021 (13)	0.0470 (6)
H6	0.3995	0.4811	0.2479	0.056*
C7	0.49911 (17)	0.3519 (5)	0.13234 (13)	0.0392 (5)
H4	0.5281	0.4936	0.1631	0.047*
C8	0.63958 (19)	0.4507 (5)	0.06201 (16)	0.0386 (5)
H8A	0.6392	0.6192	0.0928	0.046*
H8B	0.6393	0.5060	0.0087	0.046*
C9	0.73453 (16)	0.2898 (5)	0.08264 (15)	0.0458 (6)
H9A	0.7308	0.1105	0.0568	0.055*
H9B	0.7391	0.2551	0.1374	0.055*
C10	0.8289 (2)	0.4405 (6)	0.0612 (2)	0.0523 (7)
H10A	0.8212	0.4914	0.0075	0.063*
H10B	0.8362	0.6114	0.0909	0.063*
C11	0.92355 (19)	0.2690 (8)	0.0748 (2)	0.0713 (10)
H11A	0.9303	0.2131	0.1276	0.107*
H11B	0.9828	0.3769	0.0628	0.107*
H11C	0.9188	0.1059	0.0428	0.107*
C12	0.0780 (2)	-0.2452 (6)	0.18909 (18)	0.0582 (8)

H12A	0.0683	-0.1895	0.1364	0.087*
H12B	0.0119	-0.2684	0.2101	0.087*
H12C	0.1151	-0.4187	0.1924	0.087*
N1	0.54697 (13)	0.2884 (4)	0.07401 (11)	0.0366 (4)
O1	0.39291 (14)	-0.0997 (5)	0.05842 (11)	0.0542 (5)
O2	0.13414 (17)	-0.0399 (4)	0.23031 (14)	0.0539 (6)
Zn1	0.5000	0.0000	0.0000	0.0382 (2)

*Atomic displacement parameters (Å<sup>2</sup>)*

	$U^{11}$	$U^{22}$	$U^{33}$	$U^{12}$	$U^{13}$	$U^{23}$
C1	0.0311 (11)	0.0452 (13)	0.0370 (11)	0.0012 (10)	0.0067 (9)	0.0032 (9)
C2	0.0334 (14)	0.0431 (15)	0.0463 (16)	0.0002 (9)	0.0159 (11)	0.0018 (9)
C3	0.0340 (12)	0.0495 (13)	0.0491 (15)	-0.0069 (12)	0.0153 (10)	-0.0039 (12)
C4	0.0299 (13)	0.0518 (15)	0.0474 (16)	0.0030 (10)	0.0165 (11)	0.0061 (11)
C5	0.0444 (13)	0.0631 (17)	0.0407 (13)	0.0016 (12)	0.0176 (10)	-0.0046 (11)
C6	0.0440 (13)	0.0555 (16)	0.0424 (13)	-0.0034 (12)	0.0087 (10)	-0.0067 (11)
C7	0.0335 (11)	0.0431 (13)	0.0409 (12)	-0.0017 (10)	0.0023 (9)	-0.0012 (10)
C8	0.0331 (12)	0.0362 (11)	0.0472 (14)	-0.0047 (10)	0.0089 (10)	0.0018 (10)
C9	0.0311 (11)	0.0490 (14)	0.0578 (15)	-0.0058 (10)	0.0073 (10)	0.0079 (11)
C10	0.0332 (14)	0.0565 (15)	0.0681 (19)	-0.0094 (13)	0.0101 (12)	0.0000 (15)
C11	0.0345 (14)	0.086 (2)	0.094 (2)	-0.0009 (14)	0.0141 (15)	-0.0003 (19)
C12	0.0362 (13)	0.0653 (19)	0.075 (2)	-0.0047 (13)	0.0203 (13)	0.0043 (15)
N1	0.0275 (9)	0.0381 (10)	0.0446 (10)	-0.0016 (8)	0.0068 (7)	0.0031 (8)
O1	0.0458 (10)	0.0578 (12)	0.0625 (12)	-0.0190 (10)	0.0321 (9)	-0.0203 (10)
O2	0.0384 (11)	0.0685 (13)	0.0578 (13)	-0.0067 (9)	0.0265 (9)	-0.0022 (9)
Zn1	0.0311 (3)	0.0394 (3)	0.0457 (3)	-0.00210 (13)	0.0137 (2)	-0.00106 (14)

*Geometric parameters (Å, °)*

C1—C2	1.405 (4)	C9—C10	1.521 (4)
C1—C6	1.406 (3)	C9—H9A	0.9900
C1—C7	1.425 (3)	C9—H9B	0.9900
C2—O1	1.304 (3)	C10—C11	1.515 (4)
C2—C3	1.411 (4)	C10—H10A	0.9900
C3—C4	1.374 (4)	C10—H10B	0.9900
C3—H3	0.9500	C11—H11A	0.9800
C4—O2	1.377 (3)	C11—H11B	0.9800
C4—C5	1.382 (4)	C11—H11C	0.9800
C5—C6	1.371 (3)	C12—O2	1.426 (4)
C5—H5	0.9500	C12—H12A	0.9800
C6—H6	0.9500	C12—H12B	0.9800
C7—N1	1.291 (3)	C12—H12C	0.9800
C7—H4	0.9500	N1—Zn1	2.004 (2)
C8—N1	1.488 (3)	Zn1—O1	1.8769 (17)
C8—C9	1.510 (3)	Zn1—N1 <sup>i</sup>	2.004 (2)
C8—H8A	0.9900	Zn1—O1 <sup>i</sup>	1.8769 (17)
C8—H8B	0.9900	Zn1—O1 <sup>i</sup>	1.8769 (17)

C2—C1—C6	118.5 (2)	C10—C9—H9B	109.2
C2—C1—C7	122.2 (2)	H9A—C9—H9B	107.9
C6—C1—C7	119.3 (2)	C11—C10—C9	112.2 (3)
O1—C2—C1	123.7 (3)	C11—C10—H10A	109.2
O1—C2—C3	117.7 (2)	C9—C10—H10A	109.2
C1—C2—C3	118.6 (3)	C11—C10—H10B	109.2
C4—C3—C2	120.8 (3)	C9—C10—H10B	109.2
C4—C3—H3	119.6	H10A—C10—H10B	107.9
C2—C3—H3	119.6	C10—C11—H11A	109.5
C3—C4—O2	123.8 (3)	C10—C11—H11B	109.5
C3—C4—C5	121.1 (3)	H11A—C11—H11B	109.5
O2—C4—C5	115.1 (3)	C10—C11—H11C	109.5
C6—C5—C4	118.8 (2)	H11A—C11—H11C	109.5
C6—C5—H5	120.6	H11B—C11—H11C	109.5
C4—C5—H5	120.6	O2—C12—H12A	109.5
C5—C6—C1	122.2 (2)	O2—C12—H12B	109.5
C5—C6—H6	118.9	H12A—C12—H12B	109.5
C1—C6—H6	118.9	O2—C12—H12C	109.5
N1—C7—C1	127.7 (2)	H12A—C12—H12C	109.5
N1—C7—H4	116.1	H12B—C12—H12C	109.5
C1—C7—H4	116.1	C7—N1—C8	116.0 (2)
N1—C8—C9	111.7 (2)	C7—N1—Zn1	123.59 (15)
N1—C8—H8A	109.3	C8—N1—Zn1	120.33 (16)
C9—C8—H8A	109.3	C2—O1—Zn1	130.26 (18)
N1—C8—H8B	109.3	C4—O2—C12	117.3 (2)
C9—C8—H8B	109.3	O1 <sup>i</sup> —Zn1—O1	180.0
H8A—C8—H8B	107.9	O1 <sup>i</sup> —Zn1—N1 <sup>i</sup>	91.74 (8)
C8—C9—C10	111.9 (2)	O1—Zn1—N1 <sup>i</sup>	88.26 (8)
C8—C9—H9A	109.2	O1 <sup>i</sup> —Zn1—N1	88.26 (8)
C10—C9—H9A	109.2	O1—Zn1—N1	91.74 (8)
C8—C9—H9B	109.2	N1 <sup>i</sup> —Zn1—N1	180.0
C6—C1—C2—O1	-177.6 (3)	C9—C8—N1—C7	104.5 (3)
C7—C1—C2—O1	3.3 (4)	C9—C8—N1—Zn1	-77.6 (2)
C6—C1—C2—C3	3.2 (4)	C1—C2—O1—Zn1	-10.8 (4)
C7—C1—C2—C3	-175.9 (2)	C3—C2—O1—Zn1	168.5 (2)
O1—C2—C3—C4	178.1 (3)	C3—C4—O2—C12	-0.7 (4)
C1—C2—C3—C4	-2.6 (4)	C5—C4—O2—C12	177.6 (3)
C2—C3—C4—O2	179.0 (3)	C2—O1—Zn1—O1 <sup>i</sup>	-103.5 (2)
C2—C3—C4—C5	0.8 (4)	C2—O1—Zn1—O1 <sup>i</sup>	-103.5 (2)
C3—C4—C5—C6	0.3 (4)	C2—O1—Zn1—N1 <sup>i</sup>	-169.8 (3)
O2—C4—C5—C6	-178.0 (2)	C2—O1—Zn1—N1	10.2 (3)
C4—C5—C6—C1	0.3 (4)	C7—N1—Zn1—O1 <sup>i</sup>	175.0 (2)
C2—C1—C6—C5	-2.1 (4)	C8—N1—Zn1—O1 <sup>i</sup>	-2.69 (17)
C7—C1—C6—C5	177.0 (2)	C7—N1—Zn1—O1 <sup>i</sup>	175.0 (2)
C2—C1—C7—N1	1.5 (4)	C8—N1—Zn1—O1 <sup>i</sup>	-2.69 (17)
C6—C1—C7—N1	-177.5 (2)	C7—N1—Zn1—O1	-5.0 (2)

N1—C8—C9—C10	172.7 (2)	C8—N1—Zn1—O1	177.31 (17)
C8—C9—C10—C11	-174.5 (3)	C7—N1—Zn1—N1 <sup>i</sup>	104.5 (4)
C1—C7—N1—C8	178.5 (2)	C8—N1—Zn1—N1 <sup>i</sup>	-73.1 (2)
C1—C7—N1—Zn1	0.8 (3)		

Symmetry code: (i)  $-x+1, -y, -z$ .

*Hydrogen-bond geometry (Å, °)*

<i>D</i> —H $\cdots$ <i>A</i>	<i>D</i> —H	H $\cdots$ <i>A</i>	<i>D</i> $\cdots$ <i>A</i>	<i>D</i> —H $\cdots$ <i>A</i>
C8—H8B $\cdots$ O1 <sup>i</sup>	0.99	2.34	2.763 (3)	105
C9—H9A $\cdots$ O1 <sup>i</sup>	0.99	2.55	3.087 (3)	114

Symmetry code: (i)  $-x+1, -y, -z$ .