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(E)-3-(4-Fluorophenyl)-1-phenyl-2-propen-1-one

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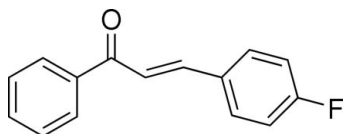
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 Key indicators: single-crystal X-ray study; $T = 93$ K; mean $\sigma(\text{C}-\text{C}) = 0.003$ Å; R factor = 0.032; wR factor = 0.066; data-to-parameter ratio = 8.2.

In the title compound, $\text{C}_{15}\text{H}_{11}\text{FO}$, the configuration of the keto group with respect to the olefinic double bond is *s-cis*. The dihedral angle between the planes of the two benzene rings is 10.61 (10)°. The crystal packing is stabilized by $\text{C}-\text{H}\cdots\pi$ interactions involving both benzene rings.

Related literature

For the synthesis, see: Chimenti *et al.* (2008). For the biological activity of chalcone derivatives, see: Dimmock *et al.* (1999).



Experimental

Crystal data

$\text{C}_{15}\text{H}_{11}\text{FO}$
 $M_r = 226.24$
Monoclinic, Cc
 $a = 24.926$ (9) Å
 $b = 5.6940$ (19) Å

$c = 7.749$ (3) Å
 $\beta = 94.747$ (5)°
 $V = 1096.0$ (6) Å³
 $Z = 4$
Mo $K\alpha$ radiation

$\mu = 0.10$ mm⁻¹
 $T = 93$ K

$0.40 \times 0.33 \times 0.30$ mm

Data collection

Rigaku SPIDER diffractometer
Absorption correction: none
4214 measured reflections

1256 independent reflections
1174 reflections with $I > 2\sigma(I)$
 $R_{\text{int}} = 0.027$

Refinement

$R[F^2 > 2\sigma(F^2)] = 0.032$
 $wR(F^2) = 0.066$
 $S = 1.05$
1256 reflections
154 parameters

2 restraints
H-atom parameters constrained
 $\Delta\rho_{\text{max}} = 0.20$ e Å⁻³
 $\Delta\rho_{\text{min}} = -0.14$ e Å⁻³

Table 1

Hydrogen-bond geometry (Å, °).

$D-H\cdots A$	$D-H$	$H\cdots A$	$D\cdots A$	$D-H\cdots A$
$\text{C1}-\text{H1}\cdots\text{Cg1}^{\text{i}}$	0.95	2.89	3.592 (3)	132
$\text{C4}-\text{H4}\cdots\text{Cg1}^{\text{ii}}$	0.95	2.93	3.646 (6)	133
$\text{C12}-\text{H12}\cdots\text{Cg2}^{\text{iii}}$	0.95	2.85	3.505 (8)	127

Symmetry codes: (i) $x + \frac{1}{2}, y + \frac{5}{2}, z$; (ii) $x + \frac{1}{2}, y + \frac{3}{2}, z - 1$; (iii) $x + \frac{1}{2}, y + \frac{1}{2}, z - 1$. Cg1 and Cg2 are the centroids of the $\text{C1}-\text{C6}$ and $\text{C10}-\text{C15}$ rings, respectively.

Data collection: *RAPID-AUTO* (Rigaku/MSC, 2004); cell refinement: *RAPID-AUTO*; data reduction: *RAPID-AUTO*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 2008); program(s) used to refine structure: *SHELXL97* (Sheldrick, 2008); molecular graphics: *XP* in *SHELXTL* (Sheldrick, 2008); software used to prepare material for publication: *SHELXL97*.

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Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: CI2914).

References

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supporting information

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(E)-3-(4-Fluorophenyl)-1-phenyl-2-propen-1-one**Lin-Hai Jing****S1. Comment**

Chalcone derivatives are a class of important compounds that possess antiprotzoal, antihelmintic, amoebicidal, anti-ulcer, antiviral, insecticidal, antibacterial, anticancer, cytotoxic and immunosuppressive activities (Dimmock *et al.*, 1999). The author reports here the crystal structure of the title compound, a chalcone derivative.

Bond lengths and angles in the title molecule (Fig.1) are normal. The configuration of the keto group with respect to the olefinic double bond is *s-cis*, with a O1—C7—C8—C9 torsion angle of $-7.1(3)^\circ$. The C1-C6 and C10-C15 benzene rings form a dihedral angle of $10.61(10)^\circ$.

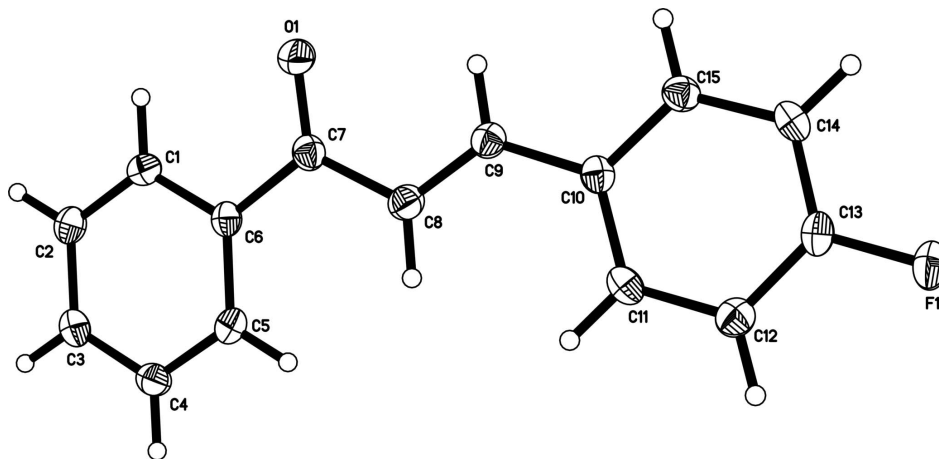
The crystal packing is stabilized by C—H \cdots π interactions involving both benzene rings (Table 1; Cg1 and Cg2 are centroids of the C1-C6 and C10-C15 rings, respectively).

S2. Experimental

The title compound was synthesized according to the method reported in the literature (Chimenti *et al.*, 2008). Colourless single crystals suitable for X-ray diffraction were obtained by slow evaporation of a acetone solution of the compound.

S3. Refinement

H atoms were placed in calculated positions, with C-H = 0.95 Å, and refined using a riding model, with $U_{\text{iso}}(\text{H}) = 1.2U_{\text{eq}}(\text{C})$. In the absence of significant anomalous scattering effects, Friedel pairs were averaged.

**Figure 1**

The molecular structure of the title compound, showing 50% probability displacement ellipsoids and the atomic numbering.

(E)-3-(4-Fluorophenyl)-1-phenyl-2-propen-1-one*Crystal data*C₁₅H₁₁FO $M_r = 226.24$ Monoclinic, *Cc*

Hall symbol: C -2yc

 $a = 24.926$ (9) Å $b = 5.6940$ (19) Å $c = 7.749$ (3) Å $\beta = 94.747$ (5)° $V = 1096.0$ (6) Å³ $Z = 4$ $F(000) = 472$ $D_x = 1.371$ Mg m⁻³Mo $K\alpha$ radiation, $\lambda = 0.71073$ Å

Cell parameters from 1803 reflections

 $\theta = 3.2$ – 27.5 ° $\mu = 0.10$ mm⁻¹ $T = 93$ K

Block, colourless

 $0.40 \times 0.33 \times 0.30$ mm*Data collection*

Rigaku SPIDER

diffractometer

Radiation source: Rotating anode

Graphite monochromator

 ω scans

4214 measured reflections

1256 independent reflections

1174 reflections with $I > 2\sigma(I)$ $R_{\text{int}} = 0.027$ $\theta_{\text{max}} = 27.5$ °, $\theta_{\text{min}} = 3.3$ ° $h = -32$ → 32 $k = -7$ → 6 $l = -10$ → 9 *Refinement*Refinement on F^2

Least-squares matrix: full

 $R[F^2 > 2\sigma(F^2)] = 0.032$ $wR(F^2) = 0.066$ $S = 1.05$

1256 reflections

154 parameters

2 restraints

Primary atom site location: structure-invariant

direct methods

Secondary atom site location: difference Fourier map

Hydrogen site location: inferred from neighbouring sites

H-atom parameters constrained

 $w = 1/[\sigma^2(F_o^2) + (0.02P)^2 + 0.6P]$ where $P = (F_o^2 + 2F_c^2)/3$ $(\Delta/\sigma)_{\text{max}} = 0.010$ $\Delta\rho_{\text{max}} = 0.20$ e Å⁻³ $\Delta\rho_{\text{min}} = -0.14$ e Å⁻³*Special details*

Geometry. All e.s.d.'s (except the e.s.d. in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell e.s.d.'s are taken into account individually in the estimation of e.s.d.'s in distances, angles and torsion angles; correlations between e.s.d.'s in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell e.s.d.'s is used for estimating e.s.d.'s involving l.s. planes.

Refinement. Refinement of F^2 against ALL reflections. The weighted R -factor wR and goodness of fit S are based on F^2 , conventional R -factors R are based on F , with F set to zero for negative F^2 . The threshold expression of $F^2 > \sigma(F^2)$ is used only for calculating R -factors(gt) etc. and is not relevant to the choice of reflections for refinement. R -factors based on F^2 are statistically about twice as large as those based on F , and R -factors based on ALL data will be even larger.

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (Å²)

	<i>x</i>	<i>y</i>	<i>z</i>	$U_{\text{iso}}^*/U_{\text{eq}}$
F1	0.78252 (5)	-0.0769 (2)	0.41272 (15)	0.0283 (3)
O1	0.54146 (6)	0.8936 (3)	0.5750 (2)	0.0298 (4)
C1	0.43870 (9)	0.9349 (4)	0.4181 (3)	0.0201 (5)
H1	0.4526	1.0584	0.4912	0.024*
C2	0.38614 (9)	0.9458 (4)	0.3436 (3)	0.0224 (5)

H2	0.3642	1.0767	0.3660	0.027*
C3	0.36564 (8)	0.7667 (4)	0.2369 (3)	0.0220 (5)
H3	0.3296	0.7749	0.1868	0.026*
C4	0.39742 (9)	0.5758 (4)	0.2026 (3)	0.0222 (5)
H4	0.3834	0.4539	0.1282	0.027*
C5	0.45022 (9)	0.5632 (4)	0.2780 (3)	0.0205 (5)
H5	0.4720	0.4315	0.2555	0.025*
C6	0.47111 (8)	0.7420 (4)	0.3855 (3)	0.0180 (4)
C7	0.52705 (8)	0.7380 (4)	0.4726 (3)	0.0205 (4)
C8	0.56390 (9)	0.5435 (4)	0.4363 (3)	0.0217 (5)
H8	0.5512	0.4142	0.3669	0.026*
C9	0.61504 (8)	0.5513 (4)	0.5016 (3)	0.0199 (4)
H9	0.6250	0.6836	0.5718	0.024*
C10	0.65773 (9)	0.3806 (4)	0.4783 (3)	0.0185 (4)
C11	0.64836 (9)	0.1657 (4)	0.3938 (3)	0.0214 (5)
H11	0.6127	0.1239	0.3526	0.026*
C12	0.69023 (9)	0.0130 (4)	0.3692 (3)	0.0225 (5)
H12	0.6838	-0.1310	0.3091	0.027*
C13	0.74170 (9)	0.0754 (4)	0.4343 (3)	0.0211 (5)
C14	0.75290 (8)	0.2834 (4)	0.5203 (3)	0.0212 (4)
H14	0.7886	0.3216	0.5635	0.025*
C15	0.71063 (9)	0.4355 (4)	0.5420 (3)	0.0207 (4)
H15	0.7176	0.5799	0.6012	0.025*

Atomic displacement parameters (Å²)

	U^{11}	U^{22}	U^{33}	U^{12}	U^{13}	U^{23}
F1	0.0222 (7)	0.0284 (7)	0.0342 (8)	0.0079 (6)	0.0022 (5)	0.0009 (6)
O1	0.0228 (8)	0.0330 (9)	0.0326 (9)	0.0047 (7)	-0.0034 (7)	-0.0121 (8)
C1	0.0228 (11)	0.0192 (11)	0.0185 (11)	-0.0002 (9)	0.0021 (8)	-0.0011 (9)
C2	0.0221 (11)	0.0227 (11)	0.0227 (11)	0.0043 (9)	0.0032 (9)	0.0017 (9)
C3	0.0177 (10)	0.0265 (11)	0.0218 (11)	0.0006 (9)	0.0017 (9)	0.0031 (9)
C4	0.0218 (11)	0.0232 (11)	0.0215 (11)	-0.0036 (9)	0.0012 (8)	-0.0011 (9)
C5	0.0218 (11)	0.0205 (11)	0.0196 (10)	0.0027 (9)	0.0046 (9)	0.0002 (9)
C6	0.0177 (10)	0.0206 (11)	0.0161 (10)	0.0007 (8)	0.0031 (8)	0.0028 (8)
C7	0.0187 (10)	0.0230 (11)	0.0199 (10)	0.0001 (9)	0.0026 (8)	0.0005 (9)
C8	0.0228 (10)	0.0215 (11)	0.0206 (10)	0.0014 (9)	0.0012 (8)	-0.0019 (9)
C9	0.0209 (10)	0.0209 (11)	0.0178 (10)	0.0023 (9)	0.0012 (8)	-0.0003 (9)
C10	0.0179 (10)	0.0210 (10)	0.0163 (10)	0.0021 (9)	0.0004 (8)	0.0033 (9)
C11	0.0177 (10)	0.0249 (11)	0.0211 (11)	-0.0018 (9)	-0.0013 (8)	0.0021 (9)
C12	0.0247 (11)	0.0222 (11)	0.0206 (11)	0.0014 (9)	0.0012 (9)	0.0005 (9)
C13	0.0191 (11)	0.0229 (11)	0.0217 (12)	0.0047 (9)	0.0040 (8)	0.0042 (9)
C14	0.0155 (10)	0.0259 (11)	0.0218 (10)	-0.0017 (9)	-0.0005 (8)	0.0030 (10)
C15	0.0214 (11)	0.0204 (10)	0.0199 (11)	-0.0005 (9)	-0.0009 (8)	0.0004 (9)

Geometric parameters (Å, °)

F1—C13	1.358 (2)	C8—C9	1.333 (3)
O1—C7	1.223 (3)	C8—H8	0.95
C1—C2	1.389 (3)	C9—C10	1.463 (3)
C1—C6	1.399 (3)	C9—H9	0.95
C1—H1	0.95	C10—C11	1.398 (3)
C2—C3	1.384 (3)	C10—C15	1.404 (3)
C2—H2	0.95	C11—C12	1.383 (3)
C3—C4	1.384 (3)	C11—H11	0.95
C3—H3	0.95	C12—C13	1.386 (3)
C4—C5	1.397 (3)	C12—H12	0.95
C4—H4	0.95	C13—C14	1.376 (3)
C5—C6	1.389 (3)	C14—C15	1.385 (3)
C5—H5	0.95	C14—H14	0.95
C6—C7	1.498 (3)	C15—H15	0.95
C7—C8	1.481 (3)		
C2—C1—C6	119.9 (2)	C7—C8—H8	120.3
C2—C1—H1	120.0	C8—C9—C10	127.8 (2)
C6—C1—H1	120.0	C8—C9—H9	116.1
C3—C2—C1	120.3 (2)	C10—C9—H9	116.1
C3—C2—H2	119.8	C11—C10—C15	118.37 (19)
C1—C2—H2	119.8	C11—C10—C9	122.95 (19)
C2—C3—C4	120.3 (2)	C15—C10—C9	118.68 (19)
C2—C3—H3	119.8	C12—C11—C10	121.0 (2)
C4—C3—H3	119.8	C12—C11—H11	119.5
C3—C4—C5	119.6 (2)	C10—C11—H11	119.5
C3—C4—H4	120.2	C11—C12—C13	118.4 (2)
C5—C4—H4	120.2	C11—C12—H12	120.8
C6—C5—C4	120.43 (19)	C13—C12—H12	120.8
C6—C5—H5	119.8	F1—C13—C14	119.03 (19)
C4—C5—H5	119.8	F1—C13—C12	118.19 (19)
C5—C6—C1	119.37 (19)	C14—C13—C12	122.8 (2)
C5—C6—C7	123.20 (18)	C13—C14—C15	118.11 (19)
C1—C6—C7	117.42 (18)	C13—C14—H14	120.9
O1—C7—C8	120.79 (19)	C15—C14—H14	120.9
O1—C7—C6	119.56 (19)	C14—C15—C10	121.3 (2)
C8—C7—C6	119.64 (18)	C14—C15—H15	119.3
C9—C8—C7	119.4 (2)	C10—C15—H15	119.3
C9—C8—H8	120.3		
C6—C1—C2—C3	−0.1 (3)	C7—C8—C9—C10	−178.7 (2)
C1—C2—C3—C4	−0.3 (3)	C8—C9—C10—C11	−6.7 (3)
C2—C3—C4—C5	0.7 (3)	C8—C9—C10—C15	172.6 (2)
C3—C4—C5—C6	−0.6 (3)	C15—C10—C11—C12	−1.7 (3)
C4—C5—C6—C1	0.2 (3)	C9—C10—C11—C12	177.7 (2)
C4—C5—C6—C7	179.0 (2)	C10—C11—C12—C13	1.6 (3)

C2—C1—C6—C5	0.2 (3)	C11—C12—C13—F1	178.78 (19)
C2—C1—C6—C7	-178.7 (2)	C11—C12—C13—C14	-0.8 (3)
C5—C6—C7—O1	-175.0 (2)	F1—C13—C14—C15	-179.54 (18)
C1—C6—C7—O1	3.9 (3)	C12—C13—C14—C15	0.0 (3)
C5—C6—C7—C8	4.0 (3)	C13—C14—C15—C10	-0.1 (3)
C1—C6—C7—C8	-177.2 (2)	C11—C10—C15—C14	0.9 (3)
O1—C7—C8—C9	-7.1 (3)	C9—C10—C15—C14	-178.5 (2)
C6—C7—C8—C9	174.0 (2)		

Hydrogen-bond geometry (Å, °)

<i>D</i> —H \cdots <i>A</i>	<i>D</i> —H	H \cdots <i>A</i>	<i>D</i> \cdots <i>A</i>	<i>D</i> —H \cdots <i>A</i>
C1—H1 \cdots Cg1 ⁱ	0.95	2.89	3.592 (3)	132
C4—H4 \cdots Cg1 ⁱⁱ	0.95	2.93	3.646 (6)	133
C12—H12 \cdots Cg2 ⁱⁱⁱ	0.95	2.85	3.505 (8)	127

Symmetry codes: (i) $x+1/2, y+5/2, z$; (ii) $x+1/2, y+3/2, z-1$; (iii) $x+1/2, y+1/2, z-1$.