

Acta Crystallographica Section E

Structure Reports

Online

ISSN 1600-5368

1-Methyl-3-phenylquinoxalin-2(1H)-one

 Hanane Benzeid,^a El Mokhtar Essassi,^a Nathalie Saffon,^b
 Bernard Garrigues^c and Seik Weng Ng^{d*}

^aLaboratoire de Chimie Organique Hétérocyclique, Pôle de compétences Pharmacochimie, Université Mohammed V-Agdal, BP 1014 Avenue Ibn Batout, Rabat, Morocco, ^bService commun Rayons X, Université Paul Sabatier, Bâtiment 2R1, 118 route de Narbonne, 31062 Toulouse, France, ^cHétérochimie Fondamentale et Appliquée, Université Paul Sabatier, UMR 5069, 118 Route de Narbonne, 31062 Toulouse, France, and ^dDepartment of Chemistry, University of Malaya, 50603 Kuala Lumpur, Malaysia
 Correspondence e-mail: seikweng@um.edu.my

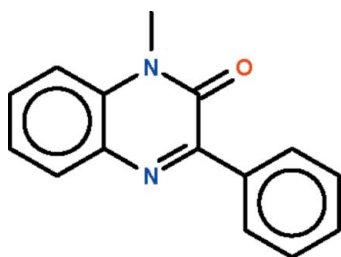
Received 24 August 2009; accepted 26 August 2009

 Key indicators: single-crystal X-ray study; $T = 193$ K; mean $\sigma(\text{C}-\text{C}) = 0.003$ Å; R factor = 0.044; wR factor = 0.114; data-to-parameter ratio = 14.3.

The phenyl substituents in both independent molecules of the title compound, $\text{C}_{15}\text{H}_{12}\text{N}_2\text{O}$, are twisted with respect to the quinoxaline system [dihedral angles = 19.3 (1) and 30.4 (1)°].

Related literature

For the structure of 1-ethyl-3-methylquinoxalin-2(1H)-one, see: Benzeid *et al.* (2008).



Experimental

Crystal data

$\text{C}_{15}\text{H}_{12}\text{N}_2\text{O}$	$V = 2312.32$ (12) Å ³
$M_r = 236.27$	$Z = 8$
Monoclinic, $P2_1/n$	Mo $K\alpha$ radiation
$a = 16.3919$ (5) Å	$\mu = 0.09$ mm ⁻¹
$b = 7.0775$ (2) Å	$T = 193$ K
$c = 20.0214$ (6) Å	$0.60 \times 0.20 \times 0.10$ mm
$\beta = 95.434$ (2)°	

Data collection

Bruker APEX2 diffractometer	4682 independent reflections
Absorption correction: none	3068 reflections with $I > 2\sigma(I)$
13360 measured reflections	$R_{\text{int}} = 0.054$

Refinement

$R[F^2 > 2\sigma(F^2)] = 0.044$	327 parameters
$wR(F^2) = 0.114$	H-atom parameters constrained
$S = 1.01$	$\Delta\rho_{\text{max}} = 0.23$ e Å ⁻³
4682 reflections	$\Delta\rho_{\text{min}} = -0.18$ e Å ⁻³

Data collection: APEX2 (Bruker, 2005); cell refinement: SAINT (Bruker, 2005); data reduction: SAINT; program(s) used to solve structure: SHELXS97 (Sheldrick, 2008); program(s) used to refine structure: SHELXL97 (Sheldrick, 2008); molecular graphics: XSEED (Barbour, 2001); software used to prepare material for publication: publCIF (Westrip, 2009).

We thank Université Mohammed V-Agdal and the University of Malaya for supporting this study.

Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: SJ2638).

References

- Barbour, L. J. (2001). *J. Supramol. Chem.* **1**, 189–191.
 Benzeid, H., Vendier, L., Ramli, Y., Garrigues, B. & Essassi, E. M. (2008). *Acta Cryst. E* **64**, o2234.
 Bruker (2005). APEX2 and SAINT. Bruker AXS Inc., Madison, Wisconsin, USA.
 Sheldrick, G. M. (2008). *Acta Cryst. A* **64**, 112–122.
 Westrip, S. P. (2009). *publCIF*. In preparation.

supporting information

Acta Cryst. (2009). E65, o2323 [doi:10.1107/S160053680903414X]

1-Methyl-3-phenylquinoxalin-2(1*H*)-one

Hanane Benzeid, El Mokhtar Essassi, Nathalie Saffon, Bernard Garrigues and Seik Weng Ng

S1. Experimental

3-Phenylquinoxalin-2(1*H*)-one (1 g, 4.5 mmol), methyl iodide (0.31 ml, 5.0 mmol), potassium carbonate (0.7 g, 5.0 mmol) and a catalytic quantity of tetra-*n*-butylammonium bromide were stirred in *N,N*-dimethylformamide (20 ml) for 24 h. The mixture was filtered to remove the salts and the solvent removed under reduce pressure. The residue was recrystallized in ethanol to afford the pure compound in 80% yield. The formulation was established by proton and carbon-13 NMR spectroscopy in CDCl₃.

S2. Refinement

Carbon-bound H-atoms were placed in calculated positions (C—H 0.95 to 0.98 Å) and were included in the refinement in the riding model approximation, with $U(\text{H})$ set to 1.2–1.5 $U(\text{C})$.

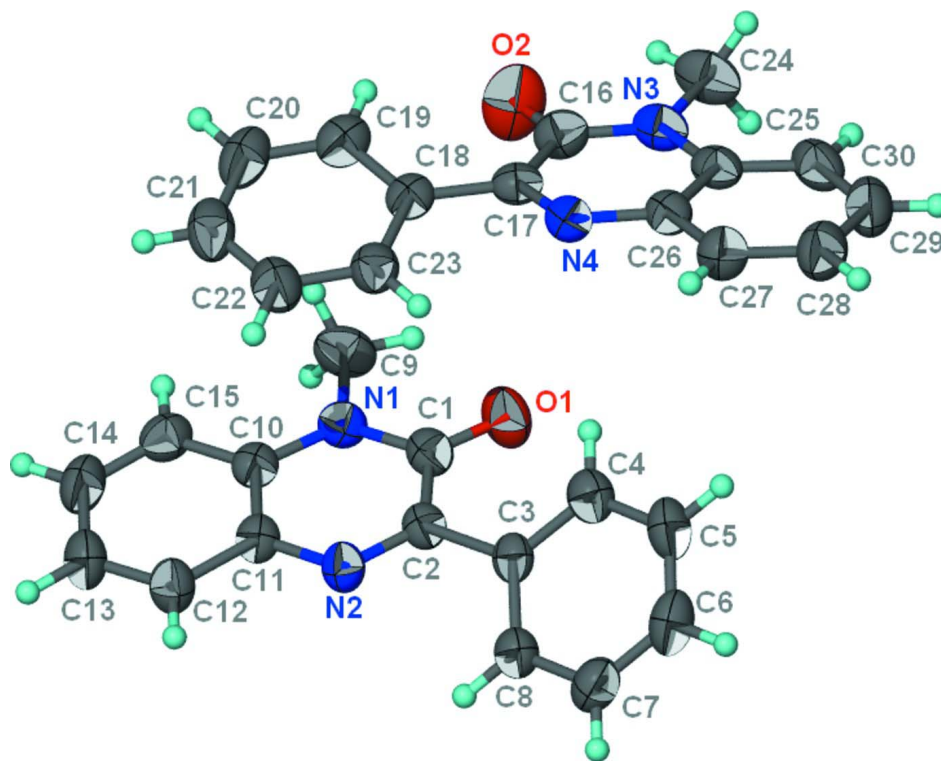


Figure 1

A thermal ellipsoid plot (Barbour, 2001) of the two unique molecules of C₁₅H₁₆N₂O₂ at the 70% probability level; hydrogen atoms are drawn as spheres of arbitrary radius.

1-Methyl-3-phenylquinoxalin-2(1H)-one

Crystal data

C₁₅H₁₂N₂O $M_r = 236.27$ Monoclinic, $P2_1/n$

Hall symbol: -P 2yn

 $a = 16.3919$ (5) Å $b = 7.0775$ (2) Å $c = 20.0214$ (6) Å $\beta = 95.434$ (2)° $V = 2312.32$ (12) Å³ $Z = 8$ $F(000) = 992$ $D_x = 1.357$ Mg m⁻³Mo $K\alpha$ radiation, $\lambda = 0.71073$ Å

Cell parameters from 4338 reflections

 $\theta = 2.5$ – 24.9 ° $\mu = 0.09$ mm⁻¹ $T = 193$ K

Plate, yellow

 $0.60 \times 0.20 \times 0.10$ mm

Data collection

Bruker APEX2

diffractometer

Radiation source: fine-focus sealed tube

Graphite monochromator

 φ and ω scans

31360 measured reflections

4682 independent reflections

3068 reflections with $I > 2\sigma(I)$ $R_{\text{int}} = 0.054$ $\theta_{\text{max}} = 26.4$ °, $\theta_{\text{min}} = 5.2$ ° $h = -20$ → 20 $k = -8$ → 8 $l = -25$ → 24

Refinement

Refinement on F^2

Least-squares matrix: full

 $R[F^2 > 2\sigma(F^2)] = 0.044$ $wR(F^2) = 0.114$ $S = 1.01$

4682 reflections

327 parameters

0 restraints

Primary atom site location: structure-invariant

direct methods

Secondary atom site location: difference Fourier map

Hydrogen site location: inferred from neighbouring sites

H-atom parameters constrained

 $w = 1/[\sigma^2(F_o^2) + (0.0498P)^2 + 0.4004P]$ where $P = (F_o^2 + 2F_c^2)/3$ $(\Delta/\sigma)_{\text{max}} = 0.001$ $\Delta\rho_{\text{max}} = 0.23$ e Å⁻³ $\Delta\rho_{\text{min}} = -0.18$ e Å⁻³Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (Å²)

	<i>x</i>	<i>y</i>	<i>z</i>	$U_{\text{iso}}^*/U_{\text{eq}}$
O1	0.53417 (7)	0.34744 (19)	0.19948 (6)	0.0472 (4)
O2	0.62711 (8)	0.8355 (2)	0.09382 (7)	0.0632 (4)
N1	0.42018 (8)	0.3190 (2)	0.12599 (7)	0.0323 (3)
N2	0.33162 (8)	0.2202 (2)	0.23283 (6)	0.0326 (3)
N3	0.69753 (8)	0.8664 (2)	0.19620 (7)	0.0358 (4)
N4	0.54955 (8)	0.86008 (19)	0.25385 (7)	0.0317 (3)
C1	0.46011 (10)	0.3149 (2)	0.18965 (9)	0.0331 (4)
C2	0.40856 (10)	0.2669 (2)	0.24408 (8)	0.0300 (4)
C3	0.44373 (10)	0.2670 (2)	0.31523 (8)	0.0309 (4)
C4	0.50477 (10)	0.3928 (3)	0.34009 (9)	0.0389 (4)
H4	0.5281	0.4779	0.3105	0.047*
C5	0.53143 (11)	0.3938 (3)	0.40790 (9)	0.0435 (5)
H5	0.5728	0.4802	0.4246	0.052*
C6	0.49836 (11)	0.2705 (3)	0.45122 (9)	0.0453 (5)

H6	0.5169	0.2722	0.4976	0.054*
C7	0.43825 (11)	0.1445 (3)	0.42723 (9)	0.0418 (5)
H7	0.4157	0.0590	0.4571	0.050*
C8	0.41091 (10)	0.1428 (2)	0.35977 (8)	0.0364 (4)
H8	0.3694	0.0562	0.3435	0.044*
C9	0.46906 (12)	0.3662 (3)	0.07079 (9)	0.0464 (5)
H9A	0.4507	0.4876	0.0512	0.070*
H9B	0.4622	0.2674	0.0364	0.070*
H9C	0.5270	0.3751	0.0878	0.070*
C10	0.33732 (10)	0.2762 (2)	0.11315 (8)	0.0306 (4)
C11	0.29496 (10)	0.2221 (2)	0.16767 (8)	0.0309 (4)
C12	0.21273 (10)	0.1659 (3)	0.15667 (9)	0.0382 (4)
H12	0.1844	0.1259	0.1935	0.046*
C13	0.17287 (11)	0.1683 (3)	0.09346 (9)	0.0428 (5)
H13	0.1173	0.1291	0.0862	0.051*
C14	0.21443 (12)	0.2287 (3)	0.03988 (9)	0.0438 (5)
H14	0.1863	0.2334	-0.0038	0.053*
C15	0.29569 (11)	0.2818 (2)	0.04900 (8)	0.0380 (4)
H15	0.3232	0.3221	0.0118	0.046*
C16	0.62597 (11)	0.8499 (3)	0.15482 (9)	0.0382 (4)
C17	0.54917 (10)	0.8488 (2)	0.18917 (8)	0.0309 (4)
C18	0.46772 (10)	0.8295 (2)	0.14974 (8)	0.0312 (4)
C19	0.45321 (12)	0.8707 (3)	0.08144 (9)	0.0407 (4)
H19	0.4969	0.9137	0.0573	0.049*
C20	0.37535 (13)	0.8491 (3)	0.04856 (10)	0.0472 (5)
H20	0.3661	0.8780	0.0021	0.057*
C21	0.31130 (12)	0.7860 (3)	0.08274 (10)	0.0469 (5)
H21	0.2584	0.7694	0.0597	0.056*
C22	0.32428 (11)	0.7471 (3)	0.15040 (10)	0.0441 (5)
H22	0.2802	0.7051	0.1742	0.053*
C23	0.40153 (10)	0.7693 (2)	0.18359 (9)	0.0369 (4)
H23	0.4098	0.7432	0.2303	0.044*
C24	0.77511 (11)	0.8735 (3)	0.16470 (10)	0.0487 (5)
H24A	0.7635	0.8699	0.1158	0.073*
H24B	0.8091	0.7647	0.1795	0.073*
H24C	0.8043	0.9905	0.1778	0.073*
C25	0.69876 (10)	0.8736 (2)	0.26575 (9)	0.0341 (4)
C26	0.62318 (10)	0.8714 (2)	0.29362 (8)	0.0314 (4)
C27	0.62170 (11)	0.8786 (3)	0.36323 (9)	0.0402 (4)
H27	0.5707	0.8792	0.3821	0.048*
C28	0.69355 (12)	0.8848 (3)	0.40451 (10)	0.0479 (5)
H28	0.6923	0.8890	0.4518	0.057*
C29	0.76810 (12)	0.8850 (3)	0.37677 (10)	0.0499 (5)
H29	0.8176	0.8885	0.4055	0.060*
C30	0.77140 (11)	0.8802 (3)	0.30840 (10)	0.0437 (5)
H30	0.8229	0.8813	0.2902	0.052*

Atomic displacement parameters (Å²)

	U^{11}	U^{22}	U^{33}	U^{12}	U^{13}	U^{23}
O1	0.0283 (7)	0.0654 (9)	0.0483 (8)	-0.0051 (6)	0.0063 (6)	-0.0005 (7)
O2	0.0531 (9)	0.1003 (13)	0.0387 (8)	0.0090 (8)	0.0178 (7)	-0.0003 (8)
N1	0.0342 (8)	0.0341 (8)	0.0294 (8)	-0.0007 (6)	0.0075 (6)	-0.0014 (6)
N2	0.0297 (8)	0.0389 (8)	0.0288 (8)	-0.0005 (7)	0.0010 (6)	-0.0027 (6)
N3	0.0315 (8)	0.0342 (8)	0.0439 (9)	0.0005 (7)	0.0150 (7)	-0.0008 (7)
N4	0.0310 (7)	0.0315 (8)	0.0330 (8)	0.0001 (6)	0.0052 (6)	-0.0016 (6)
C1	0.0288 (9)	0.0313 (9)	0.0396 (10)	0.0018 (7)	0.0047 (8)	-0.0034 (8)
C2	0.0278 (9)	0.0288 (9)	0.0334 (9)	0.0018 (7)	0.0023 (7)	-0.0033 (7)
C3	0.0241 (8)	0.0348 (9)	0.0334 (9)	0.0041 (7)	0.0007 (7)	-0.0022 (8)
C4	0.0324 (9)	0.0425 (11)	0.0415 (11)	0.0004 (8)	0.0018 (8)	-0.0024 (8)
C5	0.0340 (10)	0.0506 (12)	0.0439 (11)	0.0010 (9)	-0.0061 (8)	-0.0103 (9)
C6	0.0403 (11)	0.0583 (13)	0.0354 (10)	0.0099 (10)	-0.0061 (8)	-0.0055 (10)
C7	0.0387 (10)	0.0496 (12)	0.0366 (10)	0.0058 (9)	0.0004 (8)	0.0066 (9)
C8	0.0311 (9)	0.0396 (10)	0.0376 (10)	-0.0010 (8)	-0.0009 (8)	0.0008 (8)
C9	0.0509 (11)	0.0492 (12)	0.0417 (11)	-0.0096 (10)	0.0174 (9)	-0.0008 (9)
C10	0.0336 (9)	0.0266 (9)	0.0315 (9)	0.0030 (7)	0.0029 (7)	-0.0033 (7)
C11	0.0303 (9)	0.0321 (9)	0.0298 (9)	0.0032 (7)	0.0014 (7)	-0.0035 (7)
C12	0.0308 (9)	0.0485 (11)	0.0350 (10)	0.0004 (8)	0.0015 (8)	-0.0013 (8)
C13	0.0319 (10)	0.0520 (12)	0.0427 (11)	0.0014 (9)	-0.0059 (8)	-0.0030 (9)
C14	0.0457 (11)	0.0479 (12)	0.0353 (10)	0.0051 (9)	-0.0089 (9)	-0.0018 (9)
C15	0.0479 (11)	0.0362 (10)	0.0300 (9)	0.0010 (9)	0.0036 (8)	0.0008 (8)
C16	0.0401 (10)	0.0398 (11)	0.0361 (10)	0.0026 (8)	0.0110 (8)	0.0025 (8)
C17	0.0362 (9)	0.0241 (9)	0.0332 (10)	0.0025 (7)	0.0082 (7)	-0.0006 (7)
C18	0.0353 (9)	0.0259 (9)	0.0326 (9)	0.0056 (7)	0.0045 (7)	-0.0030 (7)
C19	0.0489 (11)	0.0371 (10)	0.0363 (10)	0.0060 (9)	0.0046 (9)	-0.0018 (8)
C20	0.0572 (13)	0.0453 (12)	0.0367 (10)	0.0127 (10)	-0.0077 (10)	-0.0053 (9)
C21	0.0431 (11)	0.0407 (11)	0.0538 (13)	0.0071 (9)	-0.0109 (10)	-0.0133 (9)
C22	0.0370 (10)	0.0414 (11)	0.0533 (12)	-0.0002 (9)	0.0004 (9)	-0.0040 (9)
C23	0.0359 (10)	0.0364 (10)	0.0381 (10)	0.0020 (8)	0.0017 (8)	-0.0015 (8)
C24	0.0368 (10)	0.0467 (12)	0.0666 (13)	-0.0015 (9)	0.0262 (10)	-0.0030 (10)
C25	0.0331 (9)	0.0252 (9)	0.0445 (10)	0.0003 (7)	0.0066 (8)	-0.0002 (8)
C26	0.0293 (9)	0.0280 (9)	0.0372 (10)	0.0016 (7)	0.0044 (7)	-0.0008 (7)
C27	0.0383 (10)	0.0439 (11)	0.0384 (10)	0.0012 (9)	0.0040 (8)	-0.0013 (8)
C28	0.0502 (12)	0.0499 (12)	0.0417 (11)	0.0009 (10)	-0.0058 (9)	-0.0030 (9)
C29	0.0405 (11)	0.0478 (12)	0.0582 (13)	0.0006 (9)	-0.0116 (10)	-0.0024 (10)
C30	0.0301 (10)	0.0380 (11)	0.0631 (13)	-0.0019 (8)	0.0049 (9)	-0.0008 (9)

Geometric parameters (Å, °)

O1—C1	1.2331 (19)	C12—H12	0.9500
O2—C16	1.228 (2)	C13—C14	1.391 (3)
N1—C1	1.377 (2)	C13—H13	0.9500
N1—C10	1.392 (2)	C14—C15	1.380 (2)
N1—C9	1.464 (2)	C14—H14	0.9500
N2—C2	1.3027 (19)	C15—H15	0.9500

N2—C11	1.384 (2)	C16—C17	1.491 (2)
N3—C16	1.376 (2)	C17—C18	1.491 (2)
N3—C25	1.392 (2)	C18—C19	1.396 (2)
N3—C24	1.473 (2)	C18—C23	1.399 (2)
N4—C17	1.297 (2)	C19—C20	1.388 (3)
N4—C26	1.384 (2)	C19—H19	0.9500
C1—C2	1.480 (2)	C20—C21	1.381 (3)
C2—C3	1.485 (2)	C20—H20	0.9500
C3—C4	1.395 (2)	C21—C22	1.379 (3)
C3—C8	1.396 (2)	C21—H21	0.9500
C4—C5	1.386 (2)	C22—C23	1.382 (2)
C4—H4	0.9500	C22—H22	0.9500
C5—C6	1.377 (3)	C23—H23	0.9500
C5—H5	0.9500	C24—H24A	0.9800
C6—C7	1.381 (3)	C24—H24B	0.9800
C6—H6	0.9500	C24—H24C	0.9800
C7—C8	1.382 (2)	C25—C30	1.399 (2)
C7—H7	0.9500	C25—C26	1.406 (2)
C8—H8	0.9500	C26—C27	1.397 (2)
C9—H9A	0.9800	C27—C28	1.374 (2)
C9—H9B	0.9800	C27—H27	0.9500
C9—H9C	0.9800	C28—C29	1.389 (3)
C10—C15	1.397 (2)	C28—H28	0.9500
C10—C11	1.401 (2)	C29—C30	1.376 (3)
C11—C12	1.402 (2)	C29—H29	0.9500
C12—C13	1.368 (2)	C30—H30	0.9500
C1—N1—C10	122.44 (14)	C13—C14—H14	119.4
C1—N1—C9	117.23 (14)	C14—C15—C10	119.70 (17)
C10—N1—C9	120.31 (14)	C14—C15—H15	120.1
C2—N2—C11	119.18 (14)	C10—C15—H15	120.1
C16—N3—C25	122.45 (14)	O2—C16—N3	120.92 (16)
C16—N3—C24	117.81 (15)	O2—C16—C17	123.49 (16)
C25—N3—C24	119.73 (15)	N3—C16—C17	115.58 (15)
C17—N4—C26	119.98 (14)	N4—C17—C16	122.41 (15)
O1—C1—N1	121.22 (15)	N4—C17—C18	116.98 (15)
O1—C1—C2	123.17 (16)	C16—C17—C18	120.59 (15)
N1—C1—C2	115.60 (14)	C19—C18—C23	118.03 (16)
N2—C2—C1	122.79 (15)	C19—C18—C17	124.07 (16)
N2—C2—C3	116.58 (14)	C23—C18—C17	117.89 (15)
C1—C2—C3	120.62 (14)	C20—C19—C18	120.34 (18)
C4—C3—C8	118.73 (15)	C20—C19—H19	119.8
C4—C3—C2	122.86 (15)	C18—C19—H19	119.8
C8—C3—C2	118.31 (15)	C21—C20—C19	120.59 (18)
C5—C4—C3	120.11 (17)	C21—C20—H20	119.7
C5—C4—H4	119.9	C19—C20—H20	119.7
C3—C4—H4	119.9	C22—C21—C20	119.84 (18)
C6—C5—C4	120.47 (18)	C22—C21—H21	120.1

C6—C5—H5	119.8	C20—C21—H21	120.1
C4—C5—H5	119.8	C21—C22—C23	119.92 (19)
C5—C6—C7	120.04 (17)	C21—C22—H22	120.0
C5—C6—H6	120.0	C23—C22—H22	120.0
C7—C6—H6	120.0	C22—C23—C18	121.26 (17)
C6—C7—C8	120.00 (18)	C22—C23—H23	119.4
C6—C7—H7	120.0	C18—C23—H23	119.4
C8—C7—H7	120.0	N3—C24—H24A	109.5
C7—C8—C3	120.65 (16)	N3—C24—H24B	109.5
C7—C8—H8	119.7	H24A—C24—H24B	109.5
C3—C8—H8	119.7	N3—C24—H24C	109.5
N1—C9—H9A	109.5	H24A—C24—H24C	109.5
N1—C9—H9B	109.5	H24B—C24—H24C	109.5
H9A—C9—H9B	109.5	N3—C25—C30	122.88 (16)
N1—C9—H9C	109.5	N3—C25—C26	117.83 (15)
H9A—C9—H9C	109.5	C30—C25—C26	119.28 (17)
H9B—C9—H9C	109.5	N4—C26—C27	118.65 (15)
N1—C10—C15	123.16 (15)	N4—C26—C25	121.70 (15)
N1—C10—C11	117.56 (14)	C27—C26—C25	119.65 (16)
C15—C10—C11	119.28 (16)	C28—C27—C26	120.42 (17)
N2—C11—C10	122.27 (15)	C28—C27—H27	119.8
N2—C11—C12	118.07 (15)	C26—C27—H27	119.8
C10—C11—C12	119.66 (15)	C27—C28—C29	119.73 (18)
C13—C12—C11	120.63 (17)	C27—C28—H28	120.1
C13—C12—H12	119.7	C29—C28—H28	120.1
C11—C12—H12	119.7	C30—C29—C28	121.12 (18)
C12—C13—C14	119.40 (17)	C30—C29—H29	119.4
C12—C13—H13	120.3	C28—C29—H29	119.4
C14—C13—H13	120.3	C29—C30—C25	119.79 (17)
C15—C14—C13	121.25 (17)	C29—C30—H30	120.1
C15—C14—H14	119.4	C25—C30—H30	120.1
C10—N1—C1—O1	177.65 (15)	C25—N3—C16—O2	-176.73 (17)
C9—N1—C1—O1	-0.8 (2)	C24—N3—C16—O2	2.3 (3)
C10—N1—C1—C2	-1.6 (2)	C25—N3—C16—C17	2.6 (2)
C9—N1—C1—C2	179.90 (15)	C24—N3—C16—C17	-178.34 (15)
C11—N2—C2—C1	-2.6 (2)	C26—N4—C17—C16	-0.8 (2)
C11—N2—C2—C3	178.43 (14)	C26—N4—C17—C18	177.68 (14)
O1—C1—C2—N2	-175.20 (16)	O2—C16—C17—N4	178.43 (17)
N1—C1—C2—N2	4.1 (2)	N3—C16—C17—N4	-0.9 (2)
O1—C1—C2—C3	3.7 (3)	O2—C16—C17—C18	0.0 (3)
N1—C1—C2—C3	-177.04 (14)	N3—C16—C17—C18	-179.32 (14)
N2—C2—C3—C4	-147.10 (16)	N4—C17—C18—C19	160.36 (16)
C1—C2—C3—C4	33.9 (2)	C16—C17—C18—C19	-21.2 (2)
N2—C2—C3—C8	29.2 (2)	N4—C17—C18—C23	-18.6 (2)
C1—C2—C3—C8	-149.75 (16)	C16—C17—C18—C23	159.91 (15)
C8—C3—C4—C5	-0.4 (3)	C23—C18—C19—C20	-0.9 (2)
C2—C3—C4—C5	175.90 (16)	C17—C18—C19—C20	-179.88 (16)

C3—C4—C5—C6	0.3 (3)	C18—C19—C20—C21	-0.3 (3)
C4—C5—C6—C7	0.1 (3)	C19—C20—C21—C22	1.2 (3)
C5—C6—C7—C8	-0.4 (3)	C20—C21—C22—C23	-0.8 (3)
C6—C7—C8—C3	0.3 (3)	C21—C22—C23—C18	-0.5 (3)
C4—C3—C8—C7	0.1 (3)	C19—C18—C23—C22	1.3 (3)
C2—C3—C8—C7	-176.37 (15)	C17—C18—C23—C22	-179.67 (15)
C1—N1—C10—C15	179.25 (15)	C16—N3—C25—C30	176.49 (16)
C9—N1—C10—C15	-2.3 (2)	C24—N3—C25—C30	-2.5 (2)
C1—N1—C10—C11	-1.9 (2)	C16—N3—C25—C26	-2.6 (2)
C9—N1—C10—C11	176.53 (15)	C24—N3—C25—C26	178.38 (15)
C2—N2—C11—C10	-1.3 (2)	C17—N4—C26—C27	-178.43 (16)
C2—N2—C11—C12	178.21 (15)	C17—N4—C26—C25	0.9 (2)
N1—C10—C11—N2	3.5 (2)	N3—C25—C26—N4	0.8 (2)
C15—C10—C11—N2	-177.55 (15)	C30—C25—C26—N4	-178.34 (15)
N1—C10—C11—C12	-175.94 (15)	N3—C25—C26—C27	-179.93 (16)
C15—C10—C11—C12	3.0 (2)	C30—C25—C26—C27	0.9 (2)
N2—C11—C12—C13	178.77 (16)	N4—C26—C27—C28	178.28 (16)
C10—C11—C12—C13	-1.7 (3)	C25—C26—C27—C28	-1.0 (3)
C11—C12—C13—C14	-0.6 (3)	C26—C27—C28—C29	0.4 (3)
C12—C13—C14—C15	1.6 (3)	C27—C28—C29—C30	0.4 (3)
C13—C14—C15—C10	-0.3 (3)	C28—C29—C30—C25	-0.5 (3)
N1—C10—C15—C14	176.88 (16)	N3—C25—C30—C29	-179.29 (17)
C11—C10—C15—C14	-2.0 (2)	C26—C25—C30—C29	-0.2 (3)
