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(E)-N-[(1,3-Dihydronaphtho[2,3-c]furan-4-yl)phenylmethylene]aniline

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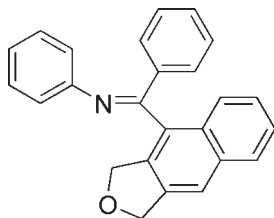
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 Key indicators: single-crystal X-ray study; $T = 293$ K; mean $\sigma(\text{C}-\text{C}) = 0.006$ Å; R factor = 0.105; wR factor = 0.278; data-to-parameter ratio = 13.2.

The title compound, $\text{C}_{25}\text{H}_{19}\text{NO}$ was synthesized by a Pd-catalysed intramolecular Diels–Alders reaction. The dihedral angle between the two benzene rings is $82.33(5)^\circ$ and the dihedral angles between the hydronaphtho[2,3-*c*]furan plane and the two benzene rings are $89.50(3)$ and $77.64(2)^\circ$. The O atom is displaced by $0.5929(3)$ Å from the hydronaphtho[2,3-*c*]furan plane.

Related literature

For Pd-catalysed [2 + 2 + 2] cocyclization of diynes and arynes, see: Sato *et al.* (2004, 2007). For the biological activity of hydronaphtho[2,3-*c*]furan derivatives, see: Baldwin *et al.* (1995); Takadoi *et al.* (1999). For bond-length data, see: Allen *et al.* (1987).



Experimental

Crystal data

$\text{C}_{25}\text{H}_{19}\text{NO}$	$\gamma = 86.299(12)^\circ$
$M_r = 349.41$	$V = 915.9(3) \text{ \AA}^3$
Triclinic, $P\bar{1}$	$Z = 2$
$a = 9.326(2) \text{ \AA}$	Mo $K\alpha$ radiation
$b = 10.198(2) \text{ \AA}$	$\mu = 0.08 \text{ mm}^{-1}$
$c = 10.878(2) \text{ \AA}$	$T = 293 \text{ K}$
$\alpha = 64.410(10)^\circ$	$0.27 \times 0.25 \times 0.19 \text{ mm}$
$\beta = 79.037(11)^\circ$	

Data collection

Bruker SMART CCD area-detector diffractometer	4805 measured reflections
Absorption correction: multi-scan (SADABS; Bruker, 2000)	3227 independent reflections
$T_{\min} = 0.979$, $T_{\max} = 0.985$	1706 reflections with $I > 2\sigma(I)$
	$R_{\text{int}} = 0.046$

Refinement

$R[F^2 > 2\sigma(F^2)] = 0.105$	244 parameters
$wR(F^2) = 0.278$	H-atom parameters constrained
$S = 1.04$	$\Delta\rho_{\text{max}} = 0.59 \text{ e \AA}^{-3}$
3227 reflections	$\Delta\rho_{\text{min}} = -0.48 \text{ e \AA}^{-3}$

Data collection: SMART (Bruker, 2000); cell refinement: SAINT (Bruker, 2000); data reduction: SAINT; program(s) used to solve structure: SHELXS97 (Sheldrick, 2008); program(s) used to refine structure: SHELXL97 (Sheldrick, 2008); molecular graphics: SHELXTL (Sheldrick, 2008); software used to prepare material for publication: SHELXTL.

Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: HG2580).

References

- Allen, F. H., Kennard, O., Watson, D. G., Brammer, L., Orpen, A. G. & Taylor, R. (1987). *J. Chem. Soc., Perkin Trans. 2*, pp. S1–19.
- Baldwin, J. E., Chesworth, R. A., Parker, J. S. & Russell, A. T. (1995). *Tetrahedron Lett.* **36**, 9551–9554.
- Bruker (2000). *SADABS, SMART* and *SAINTE*. Bruker AXS Inc., Madison, Wisconsin, USA.
- Sato, Y., Tamura, T., Kinbara, A. & Mori, M. (2007). *Adv. Synth. Catal.* **349**, 647–661.
- Sato, Y., Tamura, T. & Mori, M. (2004). *Angew. Chem. Int. Ed.* **43**, 2436–2440.
- Sheldrick, G. M. (2008). *Acta Cryst.* **A64**, 112–122.
- Takadoi, M., Katoh, T., Ishiwata, A. & Terashima, S. (1999). *Tetrahedron Lett.* **40**, 3399–3402.

supporting information

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(E)-N-[(1,3-Dihydronaphtho[2,3-*c*]furan-4-yl)phenylmethylene]aniline**Ning-De He and Wei Wang****S1. Comment**

Hydronaphtho [2,3-*c*]-furan derivatives exhibit potent and selective antagonism against muscarine M₂ receptor and are expected for the use as a therapeutic drug of Alzheimer's disease for example himbacine (Baldwin *et al.*, 1995; Takadoi *et al.*, 1999). These compounds were synthesized by Pd Catalyzed [2 + 2+2] cocyclization of diynes and arynes (Sato *et al.*, 2004, 2007). We report here the synthesis and crystal structure of the title compound. In the structure of title compound (Fig1), the values of the geometric parameters in (I) are normal (Allen *et al.*, 1987) (Table 1). The intramolecular dihedral angle between the two benzene rings is 82.33 (5)°. The dihedral angles between the hydronaphtho[2,3-*c*]furan plane and the two benzene planes are 89.50 (3)° and 77.64 (2)°. The distance of O1 to the hydronaphtho[2,3-*c*]furan plane is 0.5929 (3) Å

S2. Experimental

To a solution of Pd(dba)₃.CHCl₃ (20.8 mg, 0.02 mmol) in 2.0 ml anhydrous DMF under argon was added 1a (139.6 mg, 0.4 mmol), triethylamine (60.6 mg, 0.6 mmol). The mixture was stirred at 120 for 8 h. The reaction was quenched with a saturated aqueous solution of ammonium chloride, and the mixture was extracted with Et₂O. The combined organic extracts were washed with water and saturated brine. The organic layer was dried (Na₂SO₄) and concentrated *in vacuo*. The residue was purified by chromatography on silica gel. The resulting solution was vapor at room temperature for 6 d, after which block-shaped crystals of the title compound suitable for X-ray diffraction analysis were obtained, yield 80%.

S3. Refinement

The H atoms were fixed geometrically and were treated as riding on their parent C atoms, with C—H distances in the range of 0.93–0.97 Å (methanol hydroxyl) and with $U_{iso}(H) = 1.2U_{eq}(parent\ atom)$.

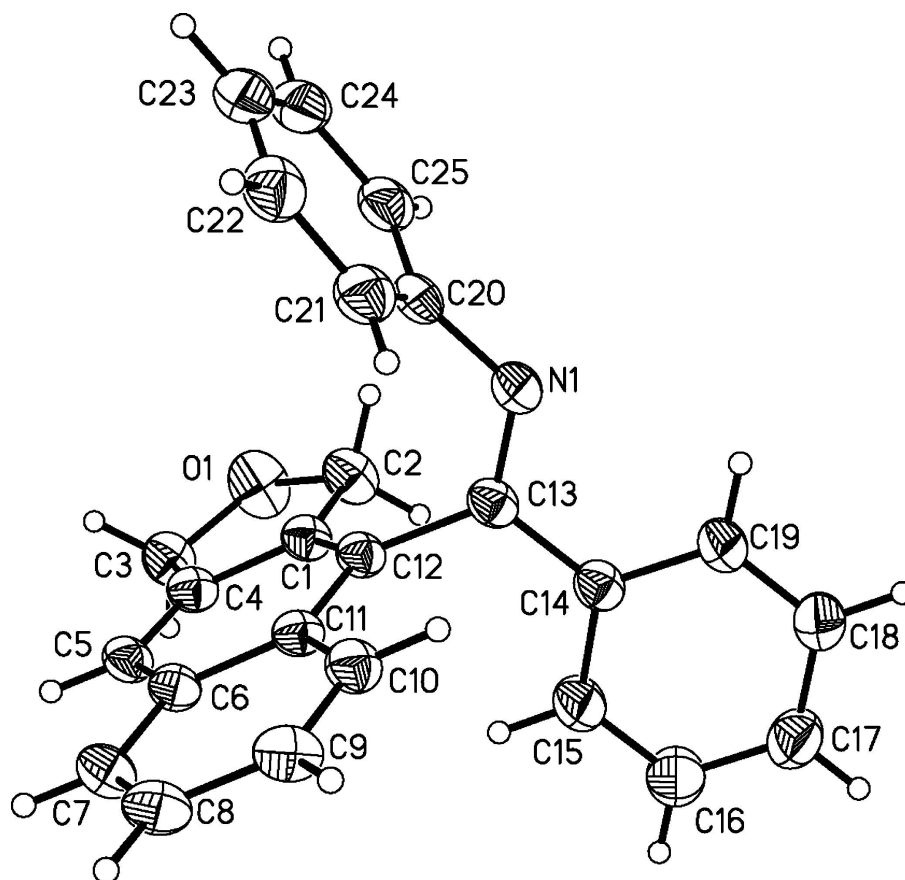


Figure 1

The independent molecules of the title compound, showing 30% probability displacement ellipsoids and the atom-numbering scheme.

(E)-N-[(1,3-Dihydronaphtho[2,3-c]furan-4-yl)phenylmethylene]aniline

Crystal data

$C_{25}H_{19}NO$

$M_r = 349.41$

Triclinic, $P\bar{1}$

Hall symbol: $-P\ 1$

$a = 9.326\ (2)\ \text{\AA}$

$b = 10.198\ (2)\ \text{\AA}$

$c = 10.878\ (2)\ \text{\AA}$

$\alpha = 64.41\ (1)^\circ$

$\beta = 79.037\ (11)^\circ$

$\gamma = 86.299\ (12)^\circ$

$V = 915.9\ (3)\ \text{\AA}^3$

$Z = 2$

$F(000) = 368$

$D_x = 1.267\ \text{Mg m}^{-3}$

Mo $K\alpha$ radiation, $\lambda = 0.71073\ \text{\AA}$

Cell parameters from 831 reflections

$\theta = 2.7\text{--}28.6^\circ$

$\mu = 0.08\ \text{mm}^{-1}$

$T = 293\ \text{K}$

Block, yellow

$0.27 \times 0.25 \times 0.19\ \text{mm}$

Data collection

Bruker SMART CCD area-detector
diffractometer

Radiation source: fine-focus sealed tube

Graphite monochromator

phi and ω scans

Absorption correction: multi-scan
(*SADABS*; Bruker, 2000)

$T_{\min} = 0.979$, $T_{\max} = 0.985$

4805 measured reflections

3227 independent reflections

1706 reflections with $I > 2\sigma(I)$
 $R_{\text{int}} = 0.046$
 $\theta_{\text{max}} = 25.1^\circ$, $\theta_{\text{min}} = 2.1^\circ$

$h = -10 \rightarrow 11$
 $k = -12 \rightarrow 11$
 $l = -12 \rightarrow 12$

Refinement

Refinement on F^2
 Least-squares matrix: full
 $R[F^2 > 2\sigma(F^2)] = 0.105$
 $wR(F^2) = 0.278$
 $S = 1.04$
 3227 reflections
 244 parameters
 0 restraints
 Primary atom site location: structure-invariant
 direct methods

Secondary atom site location: difference Fourier
 map
 Hydrogen site location: inferred from
 neighbouring sites
 H-atom parameters constrained
 $w = 1/[\sigma^2(F_o^2) + (0.150P)^2]$
 where $P = (F_o^2 + 2F_c^2)/3$
 $(\Delta/\sigma)_{\text{max}} < 0.001$
 $\Delta\rho_{\text{max}} = 0.59 \text{ e } \text{\AA}^{-3}$
 $\Delta\rho_{\text{min}} = -0.48 \text{ e } \text{\AA}^{-3}$

Special details

Geometry. All e.s.d.'s (except the e.s.d. in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell e.s.d.'s are taken into account individually in the estimation of e.s.d.'s in distances, angles and torsion angles; correlations between e.s.d.'s in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell e.s.d.'s is used for estimating e.s.d.'s involving l.s. planes.

Refinement. Refinement of F^2 against ALL reflections. The weighted R -factor wR and goodness of fit S are based on F^2 , conventional R -factors R are based on F , with F set to zero for negative F^2 . The threshold expression of $F^2 > \sigma(F^2)$ is used only for calculating R -factors(gt) *etc.* and is not relevant to the choice of reflections for refinement. R -factors based on F^2 are statistically about twice as large as those based on F , and R -factors based on ALL data will be even larger.

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (\AA^2)

	x	y	z	$U_{\text{iso}}^*/U_{\text{eq}}$
C11	0.1356 (4)	0.2470 (4)	0.7731 (4)	0.0433 (9)
C12	0.1319 (4)	0.1578 (4)	0.7038 (4)	0.0435 (9)
C6	0.1621 (4)	0.1824 (4)	0.9125 (4)	0.0446 (9)
C5	0.1846 (4)	0.0328 (4)	0.9765 (4)	0.0486 (10)
H5	0.1995	-0.0101	1.0677	0.058*
C4	0.1851 (4)	-0.0506 (4)	0.9077 (4)	0.0471 (9)
N1	-0.0090 (4)	0.2261 (4)	0.5151 (3)	0.0583 (10)
C1	0.1585 (4)	0.0116 (4)	0.7703 (4)	0.0464 (10)
C10	0.1162 (4)	0.3989 (4)	0.7104 (4)	0.0529 (10)
H10	0.0978	0.4427	0.6204	0.064*
C13	0.1141 (4)	0.2222 (4)	0.5537 (4)	0.0473 (10)
C7	0.1645 (4)	0.2734 (4)	0.9793 (4)	0.0553 (11)
H7	0.1788	0.2320	1.0707	0.066*
C14	0.2449 (4)	0.2815 (4)	0.4463 (4)	0.0495 (10)
C9	0.1235 (4)	0.4827 (4)	0.7779 (4)	0.0585 (11)
H9	0.1131	0.5828	0.7328	0.070*
C20	-0.1432 (4)	0.1792 (5)	0.6102 (4)	0.0527 (11)
C8	0.1467 (5)	0.4190 (5)	0.9153 (5)	0.0605 (11)
H8	0.1499	0.4762	0.9621	0.073*
C19	0.2363 (5)	0.3438 (5)	0.3073 (4)	0.0600 (11)
H19	0.1462	0.3476	0.2813	0.072*
C18	0.3576 (5)	0.3999 (5)	0.2072 (5)	0.0708 (13)

H18	0.3490	0.4425	0.1142	0.085*
C21	-0.2194 (5)	0.2719 (5)	0.6612 (4)	0.0632 (12)
H21	-0.1773	0.3600	0.6434	0.076*
C2	0.1712 (5)	-0.1041 (4)	0.7202 (4)	0.0645 (12)
H2A	0.0832	-0.1104	0.6875	0.077*
H2B	0.2538	-0.0840	0.6450	0.077*
C25	-0.2089 (5)	0.0486 (5)	0.6375 (4)	0.0626 (12)
H25	-0.1595	-0.0133	0.6021	0.075*
O1	0.1914 (4)	-0.2359 (3)	0.8365 (3)	0.0819 (11)
C22	-0.3594 (5)	0.2322 (5)	0.7394 (5)	0.0699 (13)
H22	-0.4112	0.2944	0.7732	0.084*
C3	0.2152 (5)	-0.2078 (4)	0.9480 (4)	0.0608 (11)
H3A	0.3153	-0.2286	0.9625	0.073*
H3B	0.1500	-0.2676	1.0328	0.073*
C24	-0.3477 (5)	0.0100 (6)	0.7172 (5)	0.0705 (13)
H24	-0.3902	-0.0784	0.7367	0.085*
C23	-0.4213 (5)	0.1018 (6)	0.7669 (5)	0.0718 (14)
H23	-0.5145	0.0758	0.8199	0.086*
C15	0.3807 (5)	0.2772 (5)	0.4806 (5)	0.0739 (14)
H15	0.3896	0.2365	0.5734	0.089*
C16	0.5024 (5)	0.3318 (6)	0.3804 (5)	0.0892 (17)
H16	0.5928	0.3263	0.4064	0.107*
C17	0.4933 (5)	0.3938 (5)	0.2436 (5)	0.0753 (14)
H17	0.5763	0.4313	0.1760	0.090*

Atomic displacement parameters (Å²)

	U^{11}	U^{22}	U^{33}	U^{12}	U^{13}	U^{23}
C11	0.041 (2)	0.038 (2)	0.050 (2)	0.0026 (15)	-0.0135 (17)	-0.0164 (17)
C12	0.041 (2)	0.043 (2)	0.046 (2)	0.0045 (16)	-0.0145 (16)	-0.0168 (17)
C6	0.040 (2)	0.046 (2)	0.050 (2)	0.0062 (16)	-0.0150 (17)	-0.0201 (17)
C5	0.048 (2)	0.050 (2)	0.048 (2)	0.0031 (17)	-0.0200 (17)	-0.0163 (18)
C4	0.044 (2)	0.041 (2)	0.054 (2)	0.0055 (16)	-0.0168 (17)	-0.0169 (17)
N1	0.052 (2)	0.069 (2)	0.0467 (19)	0.0011 (17)	-0.0183 (16)	-0.0149 (16)
C1	0.046 (2)	0.042 (2)	0.051 (2)	0.0049 (17)	-0.0155 (18)	-0.0181 (17)
C10	0.054 (3)	0.048 (2)	0.054 (2)	0.0031 (18)	-0.0186 (19)	-0.0156 (19)
C13	0.054 (2)	0.040 (2)	0.049 (2)	0.0058 (17)	-0.0199 (19)	-0.0169 (17)
C7	0.058 (3)	0.056 (3)	0.059 (2)	0.008 (2)	-0.024 (2)	-0.027 (2)
C14	0.055 (3)	0.044 (2)	0.050 (2)	0.0060 (18)	-0.0173 (19)	-0.0176 (17)
C9	0.057 (3)	0.044 (2)	0.077 (3)	0.0072 (19)	-0.026 (2)	-0.024 (2)
C20	0.052 (3)	0.058 (3)	0.043 (2)	0.007 (2)	-0.0249 (19)	-0.0111 (18)
C8	0.062 (3)	0.053 (3)	0.077 (3)	0.001 (2)	-0.022 (2)	-0.034 (2)
C19	0.058 (3)	0.068 (3)	0.048 (2)	0.002 (2)	-0.018 (2)	-0.016 (2)
C18	0.066 (3)	0.080 (3)	0.054 (3)	-0.005 (2)	-0.012 (2)	-0.016 (2)
C21	0.062 (3)	0.062 (3)	0.063 (3)	0.007 (2)	-0.023 (2)	-0.019 (2)
C2	0.079 (3)	0.052 (3)	0.067 (3)	0.012 (2)	-0.028 (2)	-0.025 (2)
C25	0.065 (3)	0.063 (3)	0.057 (3)	0.007 (2)	-0.028 (2)	-0.017 (2)
O1	0.120 (3)	0.0447 (17)	0.085 (2)	0.0196 (17)	-0.0338 (19)	-0.0285 (16)

C22	0.065 (3)	0.076 (3)	0.064 (3)	0.019 (3)	-0.021 (2)	-0.024 (2)
C3	0.071 (3)	0.049 (2)	0.061 (3)	0.010 (2)	-0.023 (2)	-0.019 (2)
C24	0.065 (3)	0.076 (3)	0.067 (3)	-0.005 (3)	-0.025 (2)	-0.021 (2)
C23	0.055 (3)	0.089 (4)	0.062 (3)	0.001 (3)	-0.023 (2)	-0.019 (3)
C15	0.057 (3)	0.098 (4)	0.055 (3)	0.000 (2)	-0.021 (2)	-0.017 (2)
C16	0.049 (3)	0.129 (5)	0.071 (3)	-0.002 (3)	-0.015 (2)	-0.024 (3)
C17	0.061 (3)	0.084 (3)	0.066 (3)	-0.006 (2)	-0.002 (2)	-0.021 (3)

Geometric parameters (Å, °)

C11—C10	1.413 (5)	C8—H8	0.9300
C11—C12	1.416 (5)	C19—C18	1.367 (6)
C11—C6	1.432 (5)	C19—H19	0.9300
C12—C1	1.377 (5)	C18—C17	1.387 (6)
C12—C13	1.513 (5)	C18—H18	0.9300
C6—C5	1.398 (5)	C21—C22	1.393 (6)
C6—C7	1.407 (5)	C21—H21	0.9300
C5—C4	1.353 (5)	C2—O1	1.426 (5)
C5—H5	0.9300	C2—H2A	0.9700
C4—C1	1.414 (5)	C2—H2B	0.9700
C4—C3	1.491 (5)	C25—C24	1.388 (6)
N1—C13	1.288 (5)	C25—H25	0.9300
N1—C20	1.422 (5)	O1—C3	1.419 (5)
C1—C2	1.489 (5)	C22—C23	1.372 (7)
C10—C9	1.357 (5)	C22—H22	0.9300
C10—H10	0.9300	C3—H3A	0.9700
C13—C14	1.470 (5)	C3—H3B	0.9700
C7—C8	1.355 (6)	C24—C23	1.361 (6)
C7—H7	0.9300	C24—H24	0.9300
C14—C15	1.379 (5)	C23—H23	0.9300
C14—C19	1.381 (5)	C15—C16	1.368 (6)
C9—C8	1.404 (6)	C15—H15	0.9300
C9—H9	0.9300	C16—C17	1.360 (6)
C20—C21	1.387 (6)	C16—H16	0.9300
C20—C25	1.390 (6)	C17—H17	0.9300
C10—C11—C12	123.0 (3)	C14—C19—H19	119.3
C10—C11—C6	117.8 (3)	C19—C18—C17	120.3 (4)
C12—C11—C6	119.3 (3)	C19—C18—H18	119.8
C1—C12—C11	119.3 (3)	C17—C18—H18	119.8
C1—C12—C13	119.3 (3)	C20—C21—C22	119.5 (4)
C11—C12—C13	121.2 (3)	C20—C21—H21	120.2
C5—C6—C7	122.7 (4)	C22—C21—H21	120.2
C5—C6—C11	119.2 (3)	O1—C2—C1	105.7 (3)
C7—C6—C11	118.2 (3)	O1—C2—H2A	110.6
C4—C5—C6	120.9 (4)	C1—C2—H2A	110.6
C4—C5—H5	119.5	O1—C2—H2B	110.6
C6—C5—H5	119.5	C1—C2—H2B	110.6

C5—C4—C1	120.5 (3)	H2A—C2—H2B	108.7
C5—C4—C3	131.5 (4)	C24—C25—C20	120.5 (4)
C1—C4—C3	108.0 (3)	C24—C25—H25	119.8
C13—N1—C20	122.9 (3)	C20—C25—H25	119.8
C12—C1—C4	120.8 (3)	C3—O1—C2	111.0 (3)
C12—C1—C2	130.7 (3)	C23—C22—C21	120.4 (5)
C4—C1—C2	108.4 (3)	C23—C22—H22	119.8
C9—C10—C11	121.9 (4)	C21—C22—H22	119.8
C9—C10—H10	119.1	O1—C3—C4	106.0 (3)
C11—C10—H10	119.1	O1—C3—H3A	110.5
N1—C13—C14	118.4 (3)	C4—C3—H3A	110.5
N1—C13—C12	123.5 (3)	O1—C3—H3B	110.5
C14—C13—C12	118.1 (3)	C4—C3—H3B	110.5
C8—C7—C6	122.4 (4)	H3A—C3—H3B	108.7
C8—C7—H7	118.8	C23—C24—C25	119.9 (5)
C6—C7—H7	118.8	C23—C24—H24	120.1
C15—C14—C19	117.5 (4)	C25—C24—H24	120.1
C15—C14—C13	121.3 (4)	C24—C23—C22	120.6 (5)
C19—C14—C13	121.2 (4)	C24—C23—H23	119.7
C10—C9—C8	120.3 (4)	C22—C23—H23	119.7
C10—C9—H9	119.8	C16—C15—C14	121.2 (4)
C8—C9—H9	119.8	C16—C15—H15	119.4
C21—C20—C25	119.1 (4)	C14—C15—H15	119.4
C21—C20—N1	120.2 (4)	C17—C16—C15	121.1 (5)
C25—C20—N1	120.0 (4)	C17—C16—H16	119.4
C7—C8—C9	119.4 (4)	C15—C16—H16	119.4
C7—C8—H8	120.3	C16—C17—C18	118.5 (4)
C9—C8—H8	120.3	C16—C17—H17	120.7
C18—C19—C14	121.3 (4)	C18—C17—H17	120.7
C18—C19—H19	119.3		
C10—C11—C12—C1	177.0 (3)	C12—C13—C14—C15	0.8 (5)
C6—C11—C12—C1	-2.0 (5)	N1—C13—C14—C19	1.9 (6)
C10—C11—C12—C13	3.1 (5)	C12—C13—C14—C19	-179.1 (3)
C6—C11—C12—C13	-176.0 (3)	C11—C10—C9—C8	-1.9 (6)
C10—C11—C6—C5	-178.7 (3)	C13—N1—C20—C21	82.6 (5)
C12—C11—C6—C5	0.4 (5)	C13—N1—C20—C25	-106.4 (5)
C10—C11—C6—C7	1.2 (5)	C6—C7—C8—C9	0.6 (6)
C12—C11—C6—C7	-179.7 (3)	C10—C9—C8—C7	1.2 (6)
C7—C6—C5—C4	-178.3 (3)	C15—C14—C19—C18	-0.4 (6)
C11—C6—C5—C4	1.5 (5)	C13—C14—C19—C18	179.5 (4)
C6—C5—C4—C1	-1.8 (6)	C14—C19—C18—C17	0.8 (7)
C6—C5—C4—C3	175.7 (4)	C25—C20—C21—C22	0.4 (6)
C11—C12—C1—C4	1.8 (5)	N1—C20—C21—C22	171.5 (4)
C13—C12—C1—C4	175.9 (3)	C12—C1—C2—O1	-176.9 (4)
C11—C12—C1—C2	-175.5 (4)	C4—C1—C2—O1	5.5 (4)
C13—C12—C1—C2	-1.4 (6)	C21—C20—C25—C24	-1.3 (6)
C5—C4—C1—C12	0.1 (6)	N1—C20—C25—C24	-172.4 (4)

C3—C4—C1—C12	-177.9 (3)	C1—C2—O1—C3	-9.2 (5)
C5—C4—C1—C2	178.0 (3)	C20—C21—C22—C23	0.5 (6)
C3—C4—C1—C2	-0.1 (4)	C2—O1—C3—C4	9.2 (5)
C12—C11—C10—C9	-178.4 (3)	C5—C4—C3—O1	176.8 (4)
C6—C11—C10—C9	0.7 (6)	C1—C4—C3—O1	-5.5 (4)
C20—N1—C13—C14	-176.4 (3)	C20—C25—C24—C23	1.3 (6)
C20—N1—C13—C12	4.6 (6)	C25—C24—C23—C22	-0.3 (7)
C1—C12—C13—N1	87.9 (5)	C21—C22—C23—C24	-0.6 (7)
C11—C12—C13—N1	-98.2 (4)	C19—C14—C15—C16	-0.4 (7)
C1—C12—C13—C14	-91.2 (4)	C13—C14—C15—C16	179.7 (4)
C11—C12—C13—C14	82.8 (4)	C14—C15—C16—C17	0.9 (8)
C5—C6—C7—C8	178.0 (4)	C15—C16—C17—C18	-0.4 (8)
C11—C6—C7—C8	-1.8 (6)	C19—C18—C17—C16	-0.4 (7)
N1—C13—C14—C15	-178.3 (4)		
