

Acta Crystallographica Section E

Structure Reports

Online

ISSN 1600-5368

Bis(μ -9-anthracenemethanolato)bis-[dimethylaluminium(III)]

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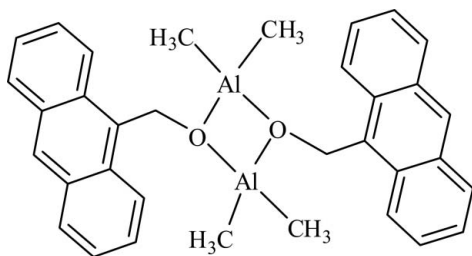
Received 29 September 2009; accepted 2 October 2009

Key indicators: single-crystal X-ray study; $T = 296$ K; mean $\sigma(\text{C}-\text{C}) = 0.004$ Å;
R factor = 0.058; wR factor = 0.234; data-to-parameter ratio = 21.3.

The title complex, $[\text{Al}_2(\text{CH}_3)_4(\text{C}_{15}\text{H}_{11}\text{O})_2]$, is dimeric bridged through the O atoms of the 9-anthracenemethanolate anions. Each Al atom is tetraordinated by two bridging O atoms from two different 9-anthracenemethanolate ligands and by two C atoms from two methyl groups, forming a distorted tetrahedral environment. The average Al—O bond distance in the Al_2O_2 core is 1.845 Å.

Related literature

For background to metal complex-catalysed ring-opening polymerization of lactones/lactides, see: Liu *et al.* (2001); Wu *et al.* (2006). For related structures, see: Lin *et al.* (1999); Lou *et al.* (2002).



Experimental

Crystal data

 $[\text{Al}_2(\text{CH}_3)_4(\text{C}_{15}\text{H}_{11}\text{O})_2]$ $M_r = 528.57$

Triclinic, $P\bar{1}$
 $a = 7.7852$ (3) Å
 $b = 11.3804$ (4) Å
 $c = 17.6749$ (6) Å
 $\alpha = 85.683$ (2)°
 $\beta = 79.883$ (2)°
 $\gamma = 74.617$ (2)°

$V = 1485.72$ (9) Å³
 $Z = 2$
Mo $K\alpha$ radiation
 $\mu = 0.13$ mm⁻¹
 $T = 296$ K
 $0.45 \times 0.38 \times 0.32$ mm

Data collection

Bruker APEXII CCD diffractometer
Absorption correction: multi-scan (SADABS; Bruker, 2008)
 $T_{\min} = 0.945$, $T_{\max} = 0.960$

31484 measured reflections
7298 independent reflections
4944 reflections with $I > 2\sigma(I)$
 $R_{\text{int}} = 0.033$

Refinement

$R[F^2 > 2\sigma(F^2)] = 0.058$
 $wR(F^2) = 0.234$
 $S = 1.01$
7298 reflections

343 parameters
H-atom parameters constrained
 $\Delta\rho_{\text{max}} = 0.35$ e Å⁻³
 $\Delta\rho_{\text{min}} = -0.27$ e Å⁻³

Data collection: APEX2 (Bruker, 2008); cell refinement: SAINT-Plus (Bruker, 2008); data reduction: SAINT-Plus; program(s) used to solve structure: SHELXS97 (Sheldrick, 2008); program(s) used to refine structure: SHELXL97 (Sheldrick, 2008); molecular graphics: SHELXTL (Sheldrick, 2008); software used to prepare material for publication: SHELXTL.

We gratefully acknowledge financial support in part from the National Science Council, Taiwan (NSC97-2113-M-033-005-MY2) and in part from the project of specific research fields in Chung Yuan Christian University, Taiwan (No. CYCU-98-CR-CH).

Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: RK2172).

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- Bruker (2008). APEX2, SADABS and SAINT-Plus. Bruker AXS Inc., Madison, Wisconsin, USA.
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Wu, J., Yu, T.-L., Chen, C.-T. & Lin, C.-C. (2006). *Coord. Chem. Rev.* **250**, 602–626.

supporting information

Acta Cryst. (2009). E65, m1320 [https://doi.org/10.1107/S1600536809040161]

Bis(μ -9-anthracenemethanolato)bis[dimethylaluminium(III)]

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S1. Comment

During the last decade, it has been of great interest to develop new catalytic/initiating systems for the preparation of aliphatic polyesters, such as *poly*(ϵ -caprolactone) and *poly*(lactide). Metal complex-catalyzed ring-opening polymerization (*ROP*) of lactones/lactides has been proven to be the most promising method to synthesize these polymers (Wu *et al.*, 2006). Among them, a variety of main group metal complexes, such as magnesium, zinc and lithium as well as aluminium complexes have been reported to be efficient initiators/catalysts. In particular, Liu *et al.* (2001) have reported the aluminium benzylalkoxide complexes supported by the bulky bisphenolate ligand and these complexes have been demonstrated as efficient initiators to catalyze *ROP* of cyclic esters. Recently, our group is interested in the synthesis and preparation of aluminium complexes derived from the 9-anthracenemethanolate ligands. The compound, 9-anthracenemethanol has been proven as a useful initiator to initiate living cationic polymerization of δ -valerolactone in the presence of $\text{HCl}\cdot\text{Et}_2\text{O}$ (Lou *et al.*, 2002). We report herein the synthesis and crystal structure of the 9-anthracenemethanolate ligand incorporated Al^{III} complex, **I**, a potential initiator for the *ROP* of ϵ -caprolactone (Fig. 2).

The solid structure of **I** reveals a dimeric Al^{III} complex (Fig. 1), doubly bridged through the O atoms of the 9-anthracenemethanolate anions. The geometry around each Al atom is four-coordinated with a distorted tetrahedral environment in which two bridging O atoms come from two different 9-anthracenemethanolate ligands and two C atoms are from two methyl groups. The average bond distance of Al-O in the Al_2O_2 core of 1.8453 (14) Å is within a normal range for an Al_2O_2 ring of four-coordinated aluminium complexes (Lin *et al.*, 1999).

S2. Experimental

The title compound **I** was synthesized by the following procedures (Fig. 2): to a rapidly stirred solution of 9-anthracenemethanol (0.21 g, 1.0 mmol) in 1,2-dichloroethane (20 ml) was slowly added AlMe_3 (0.6 ml, 2.0 M in toluene, 1.2 mmol). The mixture was further stirred at room temperature for 4 h and then dried under vacuum. The residue was extracted with 1,2-dichloroethane (10 ml), and the saturated solution was cooled to 273 K, yielding colourless crystals. Yield: 0.22 g (83%). ^1H NMR (CDCl_3 , p.p.m.): δ 7.43-8.47 (18H, m, ArH), 5.67 (4H, s, CH_2), -1.39 (12H, s, AlCH_3).

S3. Refinement

The H atoms were placed in idealized positions and constrained to ride on their parent atoms, with C-H = 0.93 Å with $U_{\text{iso}}(\text{H}) = 1.2 U_{\text{eq}}(\text{C})$ for phenyl hydrogen; 0.96 Å with $U_{\text{iso}}(\text{H}) = 1.5 U_{\text{eq}}(\text{C})$ for CH_3 group; 0.97 Å with $U_{\text{iso}}(\text{H}) = 1.2 U_{\text{eq}}(\text{C})$ for CH_2 group.

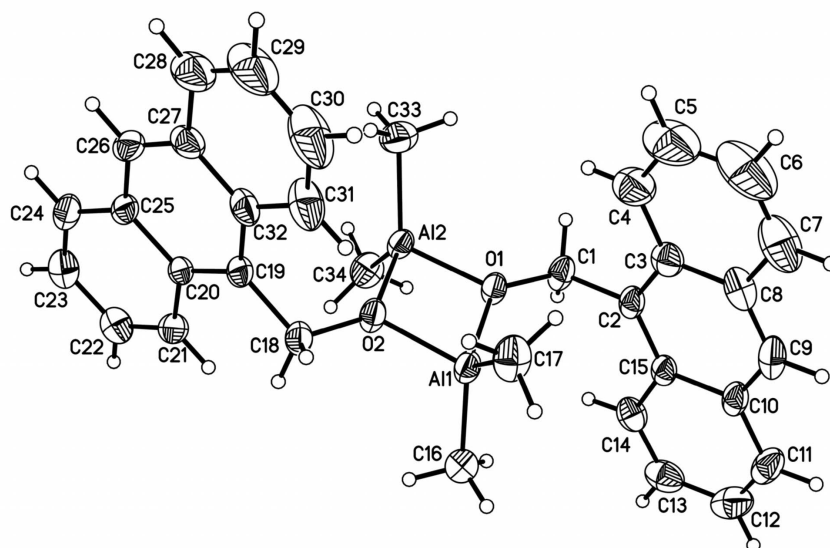


Figure 1

A view of the molecular structure of **I** with the atom numbering scheme. Displacement ellipsoids are drawn at the 30% probability level. H atoms are presented as a small spheres of arbitrary radius.

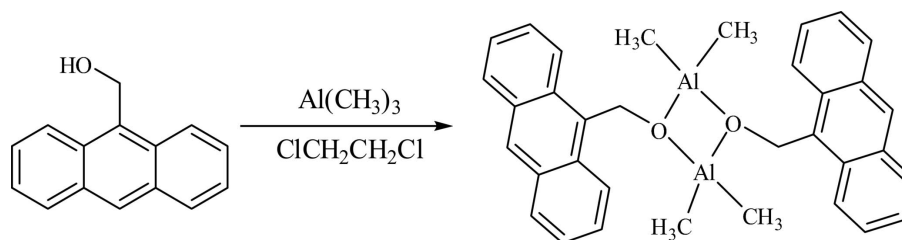


Figure 2

The title compound, **I**, (reaction path scheme).

Bis(μ -9-anthracenemethanolato)bis[$\text{dimethylaluminum(III)}$]

Crystal data

$[\text{Al}_2(\text{CH}_3)_4(\text{C}_{15}\text{H}_{11}\text{O})_2]$

$M_r = 528.57$

Triclinic, $P\bar{1}$

Hall symbol: $-P\ 1$

$a = 7.7852(3)\ \text{\AA}$

$b = 11.3804(4)\ \text{\AA}$

$c = 17.6749(6)\ \text{\AA}$

$\alpha = 85.683(2)^\circ$

$\beta = 79.883(2)^\circ$

$\gamma = 74.617(2)^\circ$

$V = 1485.72(9)\ \text{\AA}^3$

$Z = 2$

$F(000) = 560$

$D_x = 1.181\ \text{Mg m}^{-3}$

Mo $K\alpha$ radiation, $\lambda = 0.71073\ \text{\AA}$

Cell parameters from 9970 reflections

$\theta = 2.2\text{--}28.2^\circ$

$\mu = 0.13\ \text{mm}^{-1}$

$T = 296\ \text{K}$

Block, colourless

$0.45 \times 0.38 \times 0.32\ \text{mm}$

Data collection

Bruker APEXII CCD
diffractometer

Radiation source: fine-focus sealed tube

Graphite monochromator

Detector resolution: $8.3333\ \text{pixels mm}^{-1}$

φ and ω scans

Absorption correction: multi-scan
(*SADABS*; Bruker, 2008)

$T_{\min} = 0.945$, $T_{\max} = 0.960$
 31484 measured reflections
 7298 independent reflections
 4944 reflections with $I > 2\sigma(I)$
 $R_{\text{int}} = 0.033$

$\theta_{\max} = 28.4^\circ$, $\theta_{\min} = 1.9^\circ$
 $h = -10 \rightarrow 10$
 $k = -14 \rightarrow 15$
 $l = -23 \rightarrow 23$

Refinement

Refinement on F^2
 Least-squares matrix: full
 $R[F^2 > 2\sigma(F^2)] = 0.058$
 $wR(F^2) = 0.234$
 $S = 1.01$
 7298 reflections
 343 parameters
 0 restraints
 Primary atom site location: structure-invariant
 direct methods

Secondary atom site location: difference Fourier
 map
 Hydrogen site location: inferred from
 neighbouring sites
 H-atom parameters constrained
 $w = 1/[\sigma^2(F_o^2) + (0.17P)^2]$
 where $P = (F_o^2 + 2F_c^2)/3$
 $(\Delta/\sigma)_{\max} = 0.002$
 $\Delta\rho_{\max} = 0.35 \text{ e } \text{\AA}^{-3}$
 $\Delta\rho_{\min} = -0.27 \text{ e } \text{\AA}^{-3}$

Special details

Geometry. All s.u.'s (except the s.u. in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell s.u.'s are taken into account individually in the estimation of s.u.'s in distances, angles and torsion angles; correlations between s.u.'s in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell s.u.'s is used for estimating s.u.'s involving l.s. planes.

Refinement. Refinement of F^2 against ALL reflections. The weighted R -factor wR and goodness of fit S are based on F^2 , conventional R -factors R are based on F , with F set to zero for negative F^2 . The threshold expression of $F^2 > \sigma(F^2)$ is used only for calculating R -factors(gt) *etc.* and is not relevant to the choice of reflections for refinement. R -factors based on F^2 are statistically about twice as large as those based on F , and R -factors based on ALL data will be even larger.

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (\AA^2)

| | x | y | z | $U_{\text{iso}}^*/U_{\text{eq}}$ |
|-----|--------------|--------------|--------------|----------------------------------|
| Al1 | 0.71289 (7) | 0.62997 (5) | 0.30278 (4) | 0.0483 (2) |
| Al2 | 0.36490 (7) | 0.76614 (5) | 0.27971 (3) | 0.0449 (2) |
| O1 | 0.50638 (17) | 0.60755 (11) | 0.27575 (8) | 0.0466 (3) |
| O2 | 0.57693 (17) | 0.79058 (12) | 0.29925 (9) | 0.0501 (4) |
| C1 | 0.4531 (3) | 0.50037 (18) | 0.26509 (14) | 0.0550 (5) |
| H1A | 0.3752 | 0.5170 | 0.2262 | 0.066* |
| H1B | 0.3847 | 0.4779 | 0.3128 | 0.066* |
| C2 | 0.6143 (2) | 0.39518 (17) | 0.24061 (12) | 0.0460 (4) |
| C3 | 0.6999 (3) | 0.3874 (2) | 0.16343 (13) | 0.0571 (5) |
| C4 | 0.6426 (5) | 0.4791 (3) | 0.10646 (17) | 0.0878 (9) |
| H4A | 0.5446 | 0.5454 | 0.1197 | 0.105* |
| C5 | 0.7327 (7) | 0.4689 (5) | 0.0324 (2) | 0.1269 (16) |
| H5A | 0.6961 | 0.5290 | -0.0043 | 0.152* |
| C6 | 0.8791 (8) | 0.3689 (5) | 0.0114 (2) | 0.143 (2) |
| H6A | 0.9359 | 0.3626 | -0.0396 | 0.171* |
| C7 | 0.9404 (5) | 0.2806 (4) | 0.06366 (19) | 0.1063 (12) |
| H7A | 1.0406 | 0.2165 | 0.0487 | 0.128* |
| C8 | 0.8509 (3) | 0.2866 (2) | 0.14130 (14) | 0.0662 (7) |
| C9 | 0.9068 (3) | 0.1975 (2) | 0.19499 (16) | 0.0651 (6) |
| H9A | 1.0028 | 0.1310 | 0.1794 | 0.078* |

| | | | | |
|------|------------|--------------|--------------|-------------|
| C10 | 0.8260 (3) | 0.20243 (18) | 0.27155 (13) | 0.0509 (5) |
| C11 | 0.8872 (3) | 0.1102 (2) | 0.32601 (19) | 0.0709 (7) |
| H11A | 0.9810 | 0.0429 | 0.3098 | 0.085* |
| C12 | 0.8137 (4) | 0.1174 (3) | 0.40009 (19) | 0.0814 (8) |
| H12A | 0.8572 | 0.0564 | 0.4351 | 0.098* |
| C13 | 0.6709 (4) | 0.2172 (3) | 0.42484 (15) | 0.0720 (7) |
| H13A | 0.6209 | 0.2218 | 0.4767 | 0.086* |
| C14 | 0.6036 (3) | 0.3072 (2) | 0.37536 (13) | 0.0580 (6) |
| H14A | 0.5074 | 0.3718 | 0.3938 | 0.070* |
| C15 | 0.6778 (2) | 0.30507 (17) | 0.29495 (11) | 0.0435 (4) |
| C16 | 0.7421 (4) | 0.5780 (2) | 0.40803 (15) | 0.0778 (8) |
| H16A | 0.8646 | 0.5327 | 0.4093 | 0.117* |
| H16B | 0.6624 | 0.5273 | 0.4276 | 0.117* |
| H16C | 0.7136 | 0.6482 | 0.4392 | 0.117* |
| C17 | 0.9212 (3) | 0.6023 (2) | 0.22087 (17) | 0.0745 (7) |
| H17A | 1.0257 | 0.5541 | 0.2408 | 0.112* |
| H17B | 0.9420 | 0.6792 | 0.2013 | 0.112* |
| H17C | 0.8986 | 0.5600 | 0.1802 | 0.112* |
| C18 | 0.6307 (3) | 0.89486 (18) | 0.31628 (13) | 0.0550 (5) |
| H18A | 0.7572 | 0.8851 | 0.2951 | 0.066* |
| H18B | 0.6175 | 0.9003 | 0.3716 | 0.066* |
| C19 | 0.5213 (3) | 1.01085 (17) | 0.28411 (11) | 0.0450 (4) |
| C20 | 0.3866 (2) | 1.09329 (17) | 0.33176 (10) | 0.0401 (4) |
| C21 | 0.3403 (3) | 1.0730 (2) | 0.41330 (12) | 0.0514 (5) |
| H21A | 0.3966 | 1.0001 | 0.4364 | 0.062* |
| C22 | 0.2155 (3) | 1.1591 (2) | 0.45698 (14) | 0.0625 (6) |
| H22A | 0.1897 | 1.1442 | 0.5097 | 0.075* |
| C23 | 0.1249 (3) | 1.2691 (2) | 0.42521 (15) | 0.0659 (6) |
| H23A | 0.0414 | 1.3271 | 0.4565 | 0.079* |
| C24 | 0.1592 (3) | 1.2906 (2) | 0.34909 (15) | 0.0601 (6) |
| H24A | 0.0965 | 1.3632 | 0.3279 | 0.072* |
| C25 | 0.2891 (3) | 1.20518 (18) | 0.30016 (12) | 0.0460 (4) |
| C26 | 0.3239 (3) | 1.2290 (2) | 0.22149 (13) | 0.0574 (6) |
| H26A | 0.2579 | 1.3008 | 0.2007 | 0.069* |
| C27 | 0.4547 (4) | 1.1484 (2) | 0.17324 (13) | 0.0614 (6) |
| C28 | 0.4924 (5) | 1.1736 (3) | 0.09214 (15) | 0.0921 (10) |
| H28A | 0.4240 | 1.2435 | 0.0705 | 0.110* |
| C29 | 0.6261 (7) | 1.0967 (4) | 0.04722 (17) | 0.1195 (15) |
| H29A | 0.6502 | 1.1150 | -0.0051 | 0.143* |
| C30 | 0.7283 (6) | 0.9909 (4) | 0.0774 (2) | 0.1214 (15) |
| H30A | 0.8206 | 0.9396 | 0.0452 | 0.146* |
| C31 | 0.6957 (4) | 0.9609 (3) | 0.15368 (16) | 0.0860 (9) |
| H31A | 0.7643 | 0.8886 | 0.1728 | 0.103* |
| C32 | 0.5583 (3) | 1.0386 (2) | 0.20398 (12) | 0.0560 (5) |
| C33 | 0.3002 (4) | 0.8271 (3) | 0.17994 (14) | 0.0738 (7) |
| H33A | 0.1750 | 0.8706 | 0.1863 | 0.111* |
| H33B | 0.3209 | 0.7599 | 0.1469 | 0.111* |
| H33C | 0.3728 | 0.8810 | 0.1574 | 0.111* |

| | | | | |
|------|------------|------------|--------------|------------|
| C34 | 0.1772 (3) | 0.7840 (2) | 0.37059 (14) | 0.0673 (6) |
| H34A | 0.0645 | 0.8314 | 0.3570 | 0.101* |
| H34B | 0.2097 | 0.8245 | 0.4095 | 0.101* |
| H34C | 0.1652 | 0.7049 | 0.3898 | 0.101* |

Atomic displacement parameters (Å²)

| | U^{11} | U^{22} | U^{33} | U^{12} | U^{13} | U^{23} |
|-----|-------------|-------------|-------------|--------------|--------------|--------------|
| Al1 | 0.0401 (3) | 0.0352 (3) | 0.0677 (4) | -0.0034 (2) | -0.0115 (3) | -0.0059 (3) |
| Al2 | 0.0410 (3) | 0.0354 (3) | 0.0551 (4) | -0.0032 (2) | -0.0095 (2) | -0.0001 (2) |
| O1 | 0.0405 (7) | 0.0330 (7) | 0.0656 (8) | -0.0061 (5) | -0.0100 (6) | -0.0060 (6) |
| O2 | 0.0437 (7) | 0.0330 (7) | 0.0742 (9) | -0.0088 (5) | -0.0121 (6) | -0.0041 (6) |
| C1 | 0.0448 (10) | 0.0343 (11) | 0.0875 (15) | -0.0110 (8) | -0.0136 (10) | -0.0016 (10) |
| C2 | 0.0431 (9) | 0.0333 (10) | 0.0643 (12) | -0.0127 (7) | -0.0097 (8) | -0.0056 (8) |
| C3 | 0.0709 (14) | 0.0490 (12) | 0.0601 (12) | -0.0285 (10) | -0.0148 (10) | 0.0017 (10) |
| C4 | 0.124 (3) | 0.080 (2) | 0.0741 (17) | -0.0458 (18) | -0.0311 (16) | 0.0150 (14) |
| C5 | 0.195 (5) | 0.136 (4) | 0.071 (2) | -0.085 (4) | -0.028 (3) | 0.032 (2) |
| C6 | 0.201 (5) | 0.191 (6) | 0.059 (2) | -0.112 (5) | 0.021 (2) | -0.015 (3) |
| C7 | 0.121 (3) | 0.120 (3) | 0.077 (2) | -0.050 (2) | 0.0299 (19) | -0.036 (2) |
| C8 | 0.0713 (15) | 0.0648 (16) | 0.0659 (14) | -0.0290 (12) | 0.0059 (11) | -0.0211 (12) |
| C9 | 0.0531 (12) | 0.0468 (13) | 0.0918 (17) | -0.0092 (10) | 0.0008 (11) | -0.0232 (12) |
| C10 | 0.0446 (10) | 0.0341 (10) | 0.0768 (14) | -0.0114 (8) | -0.0125 (9) | -0.0082 (9) |
| C11 | 0.0621 (14) | 0.0389 (13) | 0.119 (2) | -0.0142 (10) | -0.0352 (14) | 0.0051 (13) |
| C12 | 0.092 (2) | 0.0694 (19) | 0.102 (2) | -0.0431 (16) | -0.0486 (17) | 0.0321 (16) |
| C13 | 0.0897 (18) | 0.082 (2) | 0.0616 (14) | -0.0488 (16) | -0.0242 (13) | 0.0115 (13) |
| C14 | 0.0592 (12) | 0.0580 (14) | 0.0635 (13) | -0.0270 (10) | -0.0065 (10) | -0.0096 (10) |
| C15 | 0.0408 (9) | 0.0344 (10) | 0.0598 (11) | -0.0147 (7) | -0.0107 (8) | -0.0057 (8) |
| C16 | 0.0953 (19) | 0.0590 (16) | 0.0813 (17) | -0.0070 (14) | -0.0378 (15) | -0.0061 (13) |
| C17 | 0.0463 (12) | 0.0707 (18) | 0.1021 (19) | -0.0134 (11) | 0.0024 (12) | -0.0149 (14) |
| C18 | 0.0538 (11) | 0.0385 (11) | 0.0758 (14) | -0.0087 (9) | -0.0203 (10) | -0.0096 (10) |
| C19 | 0.0500 (10) | 0.0348 (10) | 0.0559 (11) | -0.0174 (8) | -0.0141 (8) | -0.0006 (8) |
| C20 | 0.0446 (9) | 0.0337 (9) | 0.0489 (10) | -0.0186 (7) | -0.0141 (7) | 0.0017 (7) |
| C21 | 0.0567 (11) | 0.0488 (12) | 0.0544 (11) | -0.0226 (9) | -0.0135 (9) | 0.0064 (9) |
| C22 | 0.0649 (13) | 0.0680 (16) | 0.0582 (12) | -0.0289 (12) | 0.0008 (10) | -0.0070 (11) |
| C23 | 0.0548 (13) | 0.0591 (15) | 0.0837 (17) | -0.0167 (11) | -0.0018 (11) | -0.0162 (12) |
| C24 | 0.0520 (12) | 0.0388 (12) | 0.0912 (17) | -0.0103 (9) | -0.0188 (11) | -0.0012 (11) |
| C25 | 0.0467 (10) | 0.0377 (10) | 0.0603 (11) | -0.0193 (8) | -0.0165 (8) | 0.0055 (8) |
| C26 | 0.0707 (14) | 0.0456 (12) | 0.0670 (13) | -0.0274 (10) | -0.0292 (11) | 0.0166 (10) |
| C27 | 0.0900 (17) | 0.0613 (15) | 0.0494 (11) | -0.0433 (13) | -0.0216 (11) | 0.0061 (10) |
| C28 | 0.144 (3) | 0.096 (2) | 0.0536 (14) | -0.060 (2) | -0.0243 (16) | 0.0117 (15) |
| C29 | 0.201 (4) | 0.131 (4) | 0.0437 (14) | -0.086 (3) | 0.003 (2) | -0.0071 (18) |
| C30 | 0.177 (4) | 0.118 (3) | 0.073 (2) | -0.070 (3) | 0.036 (2) | -0.038 (2) |
| C31 | 0.106 (2) | 0.0741 (19) | 0.0757 (17) | -0.0322 (16) | 0.0162 (15) | -0.0268 (14) |
| C32 | 0.0733 (14) | 0.0493 (13) | 0.0544 (11) | -0.0314 (11) | -0.0076 (10) | -0.0090 (9) |
| C33 | 0.0879 (18) | 0.0706 (18) | 0.0672 (15) | -0.0211 (14) | -0.0297 (13) | 0.0142 (12) |
| C34 | 0.0529 (12) | 0.0675 (16) | 0.0729 (15) | -0.0059 (11) | -0.0001 (10) | -0.0059 (12) |

Geometric parameters (Å, °)

| | | | |
|------------|-------------|---------------|-----------|
| A11—O1 | 1.8396 (13) | C16—H16B | 0.9600 |
| A11—O2 | 1.8560 (14) | C16—H16C | 0.9600 |
| A11—C16 | 1.943 (3) | C17—H17A | 0.9600 |
| A11—C17 | 1.949 (2) | C17—H17B | 0.9600 |
| A11—A12 | 2.8236 (8) | C17—H17C | 0.9600 |
| A12—O2 | 1.8384 (14) | C18—C19 | 1.501 (3) |
| A12—O1 | 1.8473 (14) | C18—H18A | 0.9700 |
| A12—C33 | 1.946 (2) | C18—H18B | 0.9700 |
| A12—C34 | 1.956 (2) | C19—C20 | 1.406 (3) |
| O1—C1 | 1.424 (2) | C19—C32 | 1.424 (3) |
| O2—C18 | 1.427 (2) | C20—C25 | 1.428 (3) |
| C1—C2 | 1.512 (3) | C20—C21 | 1.439 (3) |
| C1—H1A | 0.9700 | C21—C22 | 1.359 (3) |
| C1—H1B | 0.9700 | C21—H21A | 0.9300 |
| C2—C15 | 1.403 (3) | C22—C23 | 1.395 (4) |
| C2—C3 | 1.407 (3) | C22—H22A | 0.9300 |
| C3—C8 | 1.429 (3) | C23—C24 | 1.342 (3) |
| C3—C4 | 1.433 (4) | C23—H23A | 0.9300 |
| C4—C5 | 1.369 (5) | C24—C25 | 1.419 (3) |
| C4—H4A | 0.9300 | C24—H24A | 0.9300 |
| C5—C6 | 1.400 (6) | C25—C26 | 1.389 (3) |
| C5—H5A | 0.9300 | C26—C27 | 1.387 (4) |
| C6—C7 | 1.360 (6) | C26—H26A | 0.9300 |
| C6—H6A | 0.9300 | C27—C32 | 1.422 (4) |
| C7—C8 | 1.423 (4) | C27—C28 | 1.435 (3) |
| C7—H7A | 0.9300 | C28—C29 | 1.345 (5) |
| C8—C9 | 1.373 (4) | C28—H28A | 0.9300 |
| C9—C10 | 1.387 (3) | C29—C30 | 1.382 (6) |
| C9—H9A | 0.9300 | C29—H29A | 0.9300 |
| C10—C11 | 1.414 (3) | C30—C31 | 1.362 (4) |
| C10—C15 | 1.434 (3) | C30—H30A | 0.9300 |
| C11—C12 | 1.333 (4) | C31—C32 | 1.414 (3) |
| C11—H11A | 0.9300 | C31—H31A | 0.9300 |
| C12—C13 | 1.396 (4) | C33—H33A | 0.9600 |
| C12—H12A | 0.9300 | C33—H33B | 0.9600 |
| C13—C14 | 1.353 (4) | C33—H33C | 0.9600 |
| C13—H13A | 0.9300 | C34—H34A | 0.9600 |
| C14—C15 | 1.437 (3) | C34—H34B | 0.9600 |
| C14—H14A | 0.9300 | C34—H34C | 0.9600 |
| C16—H16A | 0.9600 | | |
| O1—A11—O2 | 79.89 (6) | C10—C15—C14 | 115.7 (2) |
| O1—A11—C16 | 113.50 (10) | A11—C16—H16A | 109.5 |
| O2—A11—C16 | 110.32 (9) | A11—C16—H16B | 109.5 |
| O1—A11—C17 | 114.72 (10) | H16A—C16—H16B | 109.5 |
| O2—A11—C17 | 110.62 (10) | A11—C16—H16C | 109.5 |

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| C16—A11—C17 | 120.46 (13) | H16A—C16—H16C | 109.5 |
| O1—A11—A12 | 40.12 (4) | H16B—C16—H16C | 109.5 |
| O2—A11—A12 | 39.93 (4) | A11—C17—H17A | 109.5 |
| C16—A11—A12 | 116.40 (9) | A11—C17—H17B | 109.5 |
| C17—A11—A12 | 122.85 (9) | H17A—C17—H17B | 109.5 |
| O2—A12—O1 | 80.15 (6) | A11—C17—H17C | 109.5 |
| O2—A12—C33 | 115.53 (10) | H17A—C17—H17C | 109.5 |
| O1—A12—C33 | 111.86 (10) | H17B—C17—H17C | 109.5 |
| O2—A12—C34 | 112.98 (9) | O2—C18—C19 | 112.21 (16) |
| O1—A12—C34 | 109.75 (9) | O2—C18—H18A | 109.2 |
| C33—A12—C34 | 119.67 (12) | C19—C18—H18A | 109.2 |
| O2—A12—A11 | 40.39 (4) | O2—C18—H18B | 109.2 |
| O1—A12—A11 | 39.92 (4) | C19—C18—H18B | 109.2 |
| C33—A12—A11 | 124.36 (9) | H18A—C18—H18B | 107.9 |
| C34—A12—A11 | 115.69 (8) | C20—C19—C32 | 119.69 (19) |
| C1—O1—A11 | 132.01 (11) | C20—C19—C18 | 121.31 (18) |
| C1—O1—A12 | 127.49 (11) | C32—C19—C18 | 118.98 (19) |
| A11—O1—A12 | 99.97 (6) | C19—C20—C25 | 120.19 (18) |
| C18—O2—A12 | 134.10 (12) | C19—C20—C21 | 123.53 (19) |
| C18—O2—A11 | 125.88 (11) | C25—C20—C21 | 116.28 (18) |
| A12—O2—A11 | 99.69 (6) | C22—C21—C20 | 120.8 (2) |
| O1—C1—C2 | 111.61 (15) | C22—C21—H21A | 119.6 |
| O1—C1—H1A | 109.3 | C20—C21—H21A | 119.6 |
| C2—C1—H1A | 109.3 | C21—C22—C23 | 122.0 (2) |
| O1—C1—H1B | 109.3 | C21—C22—H22A | 119.0 |
| C2—C1—H1B | 109.3 | C23—C22—H22A | 119.0 |
| H1A—C1—H1B | 108.0 | C24—C23—C22 | 119.4 (2) |
| C15—C2—C3 | 120.10 (19) | C24—C23—H23A | 120.3 |
| C15—C2—C1 | 119.84 (19) | C22—C23—H23A | 120.3 |
| C3—C2—C1 | 120.1 (2) | C23—C24—C25 | 121.5 (2) |
| C2—C3—C8 | 119.2 (2) | C23—C24—H24A | 119.2 |
| C2—C3—C4 | 122.0 (2) | C25—C24—H24A | 119.2 |
| C8—C3—C4 | 118.8 (2) | C26—C25—C24 | 120.9 (2) |
| C5—C4—C3 | 119.8 (4) | C26—C25—C20 | 119.19 (19) |
| C5—C4—H4A | 120.1 | C24—C25—C20 | 119.94 (19) |
| C3—C4—H4A | 120.1 | C27—C26—C25 | 121.6 (2) |
| C4—C5—C6 | 120.7 (4) | C27—C26—H26A | 119.2 |
| C4—C5—H5A | 119.7 | C25—C26—H26A | 119.2 |
| C6—C5—H5A | 119.7 | C26—C27—C32 | 120.1 (2) |
| C7—C6—C5 | 121.6 (3) | C26—C27—C28 | 121.7 (3) |
| C7—C6—H6A | 119.2 | C32—C27—C28 | 118.2 (3) |
| C5—C6—H6A | 119.2 | C29—C28—C27 | 120.4 (3) |
| C6—C7—C8 | 119.9 (4) | C29—C28—H28A | 119.8 |
| C6—C7—H7A | 120.1 | C27—C28—H28A | 119.8 |
| C8—C7—H7A | 120.1 | C28—C29—C30 | 121.2 (3) |
| C9—C8—C7 | 121.3 (3) | C28—C29—H29A | 119.4 |
| C9—C8—C3 | 119.6 (2) | C30—C29—H29A | 119.4 |
| C7—C8—C3 | 119.1 (3) | C31—C30—C29 | 120.9 (3) |

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| C8—C9—C10 | 122.7 (2) | C31—C30—H30A | 119.6 |
| C8—C9—H9A | 118.7 | C29—C30—H30A | 119.6 |
| C10—C9—H9A | 118.7 | C30—C31—C32 | 120.6 (3) |
| C9—C10—C11 | 121.7 (2) | C30—C31—H31A | 119.7 |
| C9—C10—C15 | 118.2 (2) | C32—C31—H31A | 119.7 |
| C11—C10—C15 | 120.1 (2) | C31—C32—C27 | 118.7 (2) |
| C12—C11—C10 | 121.5 (3) | C31—C32—C19 | 122.2 (2) |
| C12—C11—H11A | 119.2 | C27—C32—C19 | 119.2 (2) |
| C10—C11—H11A | 119.2 | Al2—C33—H33A | 109.5 |
| C11—C12—C13 | 119.7 (2) | Al2—C33—H33B | 109.5 |
| C11—C12—H12A | 120.2 | H33A—C33—H33B | 109.5 |
| C13—C12—H12A | 120.2 | Al2—C33—H33C | 109.5 |
| C14—C13—C12 | 121.7 (3) | H33A—C33—H33C | 109.5 |
| C14—C13—H13A | 119.1 | H33B—C33—H33C | 109.5 |
| C12—C13—H13A | 119.1 | Al2—C34—H34A | 109.5 |
| C13—C14—C15 | 121.3 (2) | Al2—C34—H34B | 109.5 |
| C13—C14—H14A | 119.4 | H34A—C34—H34B | 109.5 |
| C15—C14—H14A | 119.4 | Al2—C34—H34C | 109.5 |
| C2—C15—C10 | 120.14 (19) | H34A—C34—H34C | 109.5 |
| C2—C15—C14 | 124.20 (19) | H34B—C34—H34C | 109.5 |
| | | | |
| O1—Al1—Al2—O2 | -173.54 (10) | C2—C3—C8—C7 | -178.3 (2) |
| C16—Al1—Al2—O2 | 90.62 (12) | C4—C3—C8—C7 | 0.3 (3) |
| C17—Al1—Al2—O2 | -83.19 (12) | C7—C8—C9—C10 | 178.0 (2) |
| O2—Al1—Al2—O1 | 173.54 (10) | C3—C8—C9—C10 | -2.1 (3) |
| C16—Al1—Al2—O1 | -95.83 (12) | C8—C9—C10—C11 | -179.3 (2) |
| C17—Al1—Al2—O1 | 90.35 (12) | C8—C9—C10—C15 | 0.2 (3) |
| O1—Al1—Al2—C33 | -83.44 (12) | C9—C10—C11—C12 | 177.5 (2) |
| O2—Al1—Al2—C33 | 90.10 (12) | C15—C10—C11—C12 | -2.0 (3) |
| C16—Al1—Al2—C33 | -179.27 (12) | C10—C11—C12—C13 | 1.0 (4) |
| C17—Al1—Al2—C33 | 6.91 (14) | C11—C12—C13—C14 | 0.4 (4) |
| O1—Al1—Al2—C34 | 90.53 (11) | C12—C13—C14—C15 | -0.8 (3) |
| O2—Al1—Al2—C34 | -95.92 (12) | C3—C2—C15—C10 | -2.2 (3) |
| C16—Al1—Al2—C34 | -5.30 (13) | C1—C2—C15—C10 | 177.45 (16) |
| C17—Al1—Al2—C34 | -179.11 (12) | C3—C2—C15—C14 | 177.77 (17) |
| O2—Al1—O1—C1 | -176.03 (18) | C1—C2—C15—C14 | -2.6 (3) |
| C16—Al1—O1—C1 | -68.15 (19) | C9—C10—C15—C2 | 2.0 (3) |
| C17—Al1—O1—C1 | 75.81 (19) | C11—C10—C15—C2 | -178.52 (18) |
| Al2—Al1—O1—C1 | -171.8 (2) | C9—C10—C15—C14 | -177.99 (17) |
| O2—Al1—O1—Al2 | -4.20 (6) | C11—C10—C15—C14 | 1.5 (3) |
| C16—Al1—O1—Al2 | 103.68 (10) | C13—C14—C15—C2 | 179.87 (19) |
| C17—Al1—O1—Al2 | -112.36 (10) | C13—C14—C15—C10 | -0.2 (3) |
| O2—Al2—O1—C1 | 176.59 (17) | Al2—O2—C18—C19 | -26.6 (3) |
| C33—Al2—O1—C1 | -69.73 (19) | Al1—O2—C18—C19 | 161.38 (14) |
| C34—Al2—O1—C1 | 65.57 (18) | O2—C18—C19—C20 | 105.0 (2) |
| Al1—Al2—O1—C1 | 172.35 (19) | O2—C18—C19—C32 | -76.5 (2) |
| O2—Al2—O1—Al1 | 4.24 (6) | C32—C19—C20—C25 | -0.9 (3) |
| C33—Al2—O1—Al1 | 117.92 (11) | C18—C19—C20—C25 | 177.62 (16) |

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| C34—A12—O1—A11 | -106.78 (10) | C32—C19—C20—C21 | -179.83 (17) |
| O1—A12—O2—C18 | -177.66 (19) | C18—C19—C20—C21 | -1.3 (3) |
| C33—A12—O2—C18 | 72.7 (2) | C19—C20—C21—C22 | 176.45 (18) |
| C34—A12—O2—C18 | -70.3 (2) | C25—C20—C21—C22 | -2.5 (3) |
| A11—A12—O2—C18 | -173.5 (2) | C20—C21—C22—C23 | 1.1 (3) |
| O1—A12—O2—A11 | -4.20 (6) | C21—C22—C23—C24 | 1.0 (3) |
| C33—A12—O2—A11 | -113.83 (11) | C22—C23—C24—C25 | -1.6 (3) |
| C34—A12—O2—A11 | 103.19 (10) | C23—C24—C25—C26 | -179.6 (2) |
| O1—A11—O2—C18 | 178.43 (17) | C23—C24—C25—C20 | 0.0 (3) |
| C16—A11—O2—C18 | 66.97 (19) | C19—C20—C25—C26 | 2.6 (3) |
| C17—A11—O2—C18 | -68.82 (18) | C21—C20—C25—C26 | -178.35 (16) |
| A12—A11—O2—C18 | 174.21 (19) | C19—C20—C25—C24 | -177.04 (16) |
| O1—A11—O2—A12 | 4.22 (6) | C21—C20—C25—C24 | 2.0 (3) |
| C16—A11—O2—A12 | -107.24 (11) | C24—C25—C26—C27 | 177.8 (2) |
| C17—A11—O2—A12 | 116.97 (10) | C20—C25—C26—C27 | -1.9 (3) |
| A11—O1—C1—C2 | -27.9 (3) | C25—C26—C27—C32 | -0.6 (3) |
| A12—O1—C1—C2 | 162.27 (14) | C25—C26—C27—C28 | -179.2 (2) |
| O1—C1—C2—C15 | 100.4 (2) | C26—C27—C28—C29 | 176.7 (3) |
| O1—C1—C2—C3 | -79.9 (2) | C32—C27—C28—C29 | -2.0 (4) |
| C15—C2—C3—C8 | 0.3 (3) | C27—C28—C29—C30 | 1.1 (6) |
| C1—C2—C3—C8 | -179.36 (18) | C28—C29—C30—C31 | 0.5 (6) |
| C15—C2—C3—C4 | -178.3 (2) | C29—C30—C31—C32 | -1.2 (5) |
| C1—C2—C3—C4 | 2.1 (3) | C30—C31—C32—C27 | 0.3 (4) |
| C2—C3—C4—C5 | 178.5 (3) | C30—C31—C32—C19 | -179.4 (3) |
| C8—C3—C4—C5 | 0.0 (4) | C26—C27—C32—C31 | -177.4 (2) |
| C3—C4—C5—C6 | 0.8 (6) | C28—C27—C32—C31 | 1.3 (3) |
| C4—C5—C6—C7 | -1.9 (7) | C26—C27—C32—C19 | 2.3 (3) |
| C5—C6—C7—C8 | 2.3 (6) | C28—C27—C32—C19 | -179.02 (19) |
| C6—C7—C8—C9 | 178.5 (3) | C20—C19—C32—C31 | 178.1 (2) |
| C6—C7—C8—C3 | -1.4 (5) | C18—C19—C32—C31 | -0.4 (3) |
| C2—C3—C8—C9 | 1.8 (3) | C20—C19—C32—C27 | -1.6 (3) |
| C4—C3—C8—C9 | -179.6 (2) | C18—C19—C32—C27 | 179.90 (18) |
