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Diphenyl (benzylamido)phosphate

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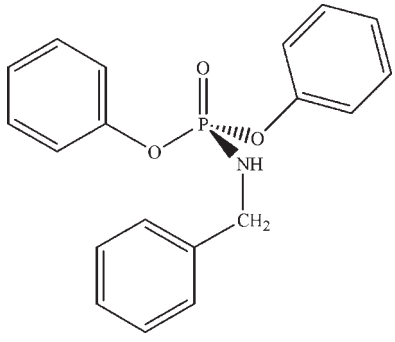
Key indicators: single-crystal X-ray study; T = 295 K; mean σ(C–C) = 0.004 Å; R factor = 0.040; wR factor = 0.102; data-to-parameter ratio = 18.8.

The title compound, $C_{19}H_{18}NO_3P$, was prepared by the reaction of diphenyl phosphorochloridate and benzylamine. In the crystal structure, molecules are linked *via* N—H···O=P hydrogen bonds into extended chains parallel to the c axis.

Related literature

For related structures, see: Bao & Wulff (1993); Gholivand *et al.* (2005); Karolak-Wojciechowska *et al.* (1979).



Experimental

Crystal data

$C_{19}H_{18}NO_3P$
M_r = 339.31
Monoclinic, $P2_1/c$
a = 10.0226 (5) Å
b = 19.2450 (8) Å
c = 10.2273 (5) Å
β = 115.375 (6)°
V = 1782.38 (17) Å^3
Z = 4
Mo $K\alpha$ radiation
μ = 0.17 mm^{-1}
T = 295 K
0.43 × 0.28 × 0.17 mm

Data collection

Oxford Diffraction Xcalibur diffractometer with a Sapphire3 (Gemini Mo) detector
Absorption correction: multi-scan *CrysAlis PRO* (Oxford Diffraction, 2009)
T_{min} = 0.977, T_{max} = 1.000
8248 measured reflections
4105 independent reflections
2568 reflections with $I > 2\sigma(I)$
R_{int} = 0.019

Refinement

$R[F^2 > 2\sigma(F^2)]$ = 0.040
$wR(F^2)$ = 0.102
S = 0.91
4105 reflections
218 parameters
H-atom parameters constrained
$\Delta\rho_{max}$ = 0.19 e Å^{-3}
$\Delta\rho_{min}$ = −0.31 e Å^{-3}

Table 1
Hydrogen-bond geometry (Å, °).

D—H···A	D—H	H···A	D···A	D—H···A
N—H···O3^{i}	0.86	1.97	2.8241 (15)	175

Symmetry code: (i) $x, -y+\frac{1}{2}, z-\frac{1}{2}$.

Data collection: *CrysAlis PRO* (Oxford Diffraction, 2009); cell refinement: *CrysAlis PRO*; data reduction: *CrysAlis PRO*; program(s) used to solve structure: *SIR92* (Altomare *et al.*, 1993); program(s) used to refine structure: *SHELXL97* (Sheldrick, 2008); molecular graphics: *Mercury* (Macrae *et al.*, 2008); software used to prepare material for publication: *SHELXL97*.

Support of this investigation by Ferdowsi University of Mashhad is gratefully acknowledged.

Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: LH2963).

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supporting information

Acta Cryst. (2010). E**66**, o204 [doi:10.1107/S1600536809053744]

Diphenyl (benzylamido)phosphate

Mehrdad Pourayoubi, Hossein Eshtiagh-Hosseini, Poorya Zargaran and Vladimir Divjakovic

S1. Comment

In previous work, the synthesis and X-ray structures of some amidophosphoric acid ester compounds, such as [(C_6H_5)(CH_3)CH—NH]P(O)(*p*—$OC_6H_4CH_3$)$_2$ (Gholivand *et al.*, 2005) and P(O)[OC_6H_5]$_2$[N($CH_2C_6H_5$)(C(S)NH$CH_2C_6H_5$)] (Karolak-Wojciechowska *et al.*, 1979) have been investigated. We report here on the synthesis and crystal structure of a new amido bis(phosphoric acid ester) compound, [C_6H_5—CH_2—NH]P(O)[O—C_6H_5]$_2$. The title compound was synthesized from the reaction of diphenyl phosphorochloridate with an excess amount of benzylamine. The P—O3 bond length of 1.4567 (10) Å and the P—N bond length of 1.5952 (14) Å are standard for this type of compound [for example for two crystallographically different [(C_6H_5)(CH_3)CH—NH]P(O)(*p*—$OC_6H_4CH_3$)$_2$ molecules (Gholivand *et al.*, 2005), P═O = 1.462 (3) Å and 1.469 (3) Å and P—N = 1.610 (5) Å and 1.614 (5) Å and for the heterocyclic phosphorus compound obtained from sequential treatment of (+)2,2′-diphenyl-3,3′-biphenanthrol with phosphorus oxychloride and (S)-(-)-*α*-methylbenzylamine (Bao & Wulff, 1993) P═O = 1.456 (6) Å, P—N = 1.612 (7)]. In the title copmound, the P—O1 and P—O2 bond lengths are slightly different (1.5844 (12) Å and 1.5880 (12) Å) and the P atom has a distorted tetrahedral configuration (Fig. 1); the bond angles around the P atom are in the range of 98.80 (6)° [for the O1—P—O2 angle] to 114.84 (7)° [for the O3—P—O1 angle]. Molecules are linked *via* N—H···O═P hydrogen bonds (N···O3 = 2.8241 (15) Å) into extended chains parallel to the *c* axis (Fig. 2).

S2. Experimental

To a solution of diphenyl phosphorochloridate (0.572 g, 2.13 mmol) in chloroform (15 ml), a solution of benzylamine (0.456 g, 4.26 mmol) in chloroform (30 ml) was added at 273K. After 4 h of stirring, the solvent was evaporated in vacuum. The solid was washed with distilled water. Single crystals were obtained from a solution of the title compound in chloroform and n-heptane (4:1) after slow evaporation at room temperature. IR (KBr, cm^{-1}): 3165 s, 2891 m, 2680 w, 2221 w, 1952 w, 1592 m, 1475 s, 1242 *vs*, 1198 *vs*, 1116 s, 1004 m, 931 *vs*, 759 s, 687 s.

S3. Refinement

H atoms were placed in the calculated positions and included in the refinement in a riding-model approximation with C-H = 0.93-0.97Å, N-H = 0.86Å and U_{iso}(H) = 1.2U_{eq}(C, N).

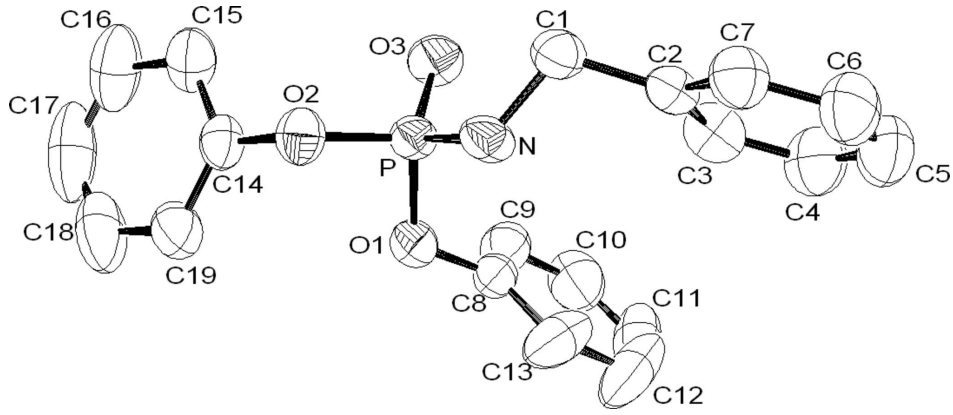

Figure 1

The molecular structure of the title compound, indicating the atom labeling scheme. Displacement ellipsoids are drawn at the 50% probability level.

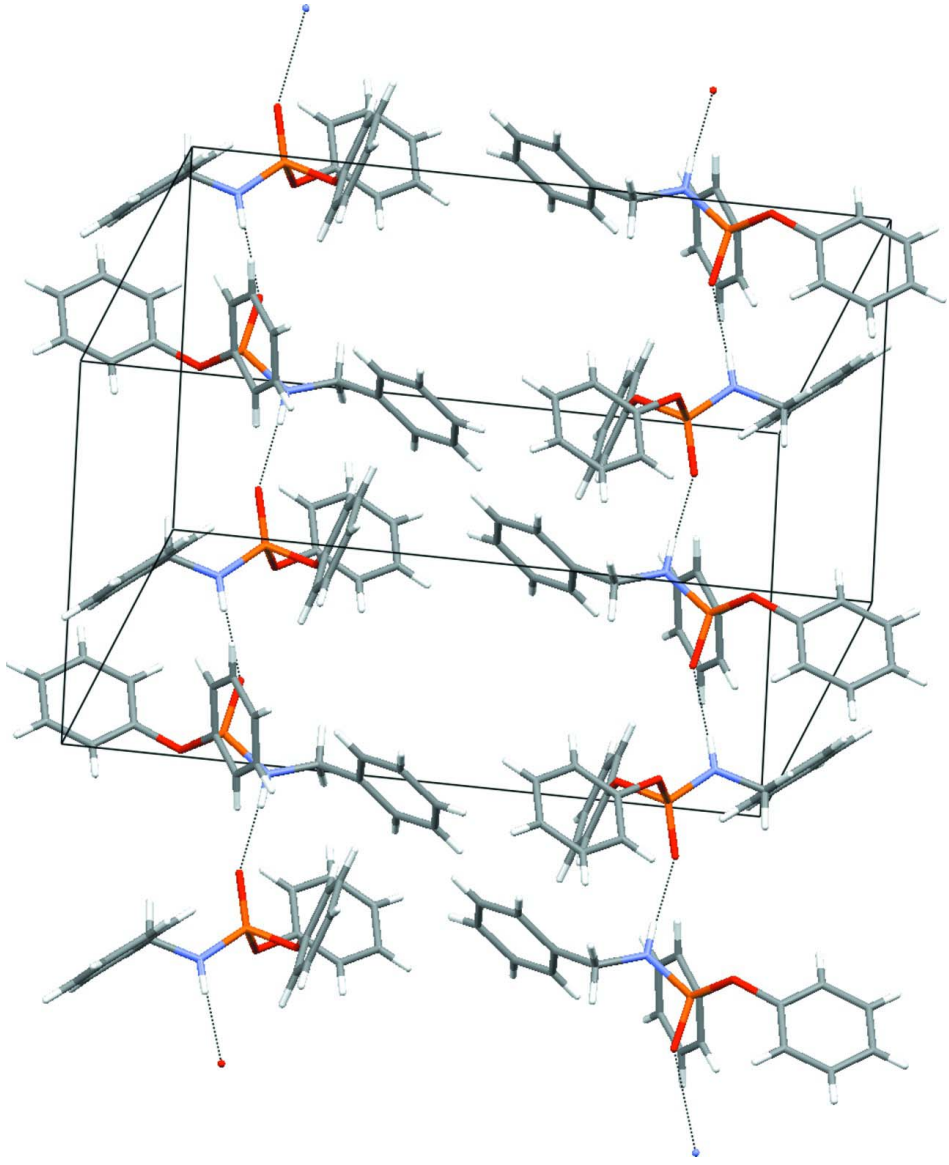

Figure 2

Part of the crystal structure with hydrogen bonds shown as dashed lines.

Diphenyl (benzylamido)phosphate

Crystal data

$C_{19}H_{18}NO_3P$
$M_r = 339.31$
Monoclinic, $P2_1/c$
Hall symbol: -P 2ybc
$a = 10.0226\ (5)$ Å
$b = 19.2450\ (8)$ Å
$c = 10.2273\ (5)$ Å
$\beta = 115.375\ (6)°$
$V = 1782.38\ (17)$ Å^3
$Z = 4$

$F(000) = 712$
$D_x = 1.264$ Mg m^{-3}
Mo $K\alpha$ radiation, $\lambda = 0.71073$ Å
Cell parameters from 3418 reflections
$\theta = 3.2–29.1°$
$\mu = 0.17$ mm^{-1}
$T = 295$ K
Prism, colorless
0.43 × 0.28 × 0.17 mm

Data collection

Oxford Diffraction Xcalibur diffractometer with a Sapphire3 (Gemini Mo) detector
Radiation source: Enhance (Mo) X-ray Source
Graphite monochromator
Detector resolution: 16.3280 pixels mm^{-1}
ω scans
Absorption correction: multi-scan CrysAlis (Oxford Diffraction, 2009)
T_{min} = 0.977, T_{max} = 1.000
8248 measured reflections
4105 independent reflections
2568 reflections with $I > 2\sigma(I)$
R_{int} = 0.019
θ_{max} = 29.2°, θ_{min} = 3.2°
h = −7→13
k = −17→24
l = −13→13

Refinement

Refinement on F^2
Least-squares matrix: full
$R[F^2 > 2\sigma(F^2)]$ = 0.040
$wR(F^2)$ = 0.102
S = 0.91
4105 reflections
218 parameters
0 restraints
Primary atom site location: structure-invariant direct methods
Secondary atom site location: difference Fourier map
Hydrogen site location: inferred from neighbouring sites
H-atom parameters constrained
$w = 1/[\sigma^2(F_o^2) + (0.0573P)^2]$ where $P = (F_o^2 + 2F_c^2)/3$
$(\Delta/\sigma)_{max} < 0.001$
$\Delta\rho_{max}$ = 0.19 e Å^{-3}
$\Delta\rho_{min}$ = −0.31 e Å^{-3}
Extinction correction: *SHELXL97* (Sheldrick, 2008), $Fc^* = kFc[1+0.001xFc^2\lambda^3/\sin(2\theta)]^{-1/4}$
Extinction coefficient: 0.0047 (10)

Special details

Experimental. #__ type_ start__ end____ width___ exp.time_ 1 omega -51.00 47.00 1.0000 19.0400 omega____ theta____ kappa____ phi_____ frames - 21.0423 - 37.0000 300.0000 98
#__ type_ start__ end____ width___ exp.time_ 2 omega 5.00 91.00 1.0000 19.0400 omega____ theta____ kappa____ phi_____ frames - 21.0423 77.0000 150.0000 86
#__ type_ start__ end____ width___ exp.time_ 3 omega -6.00 41.00 1.0000 19.0400 omega____ theta____ kappa____ phi_____ frames - 21.0423 - 77.0000 240.0000 47

Geometry. All e.s.d.'s (except the e.s.d. in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell e.s.d.'s are taken into account individually in the estimation of e.s.d.'s in distances, angles and torsion angles; correlations between e.s.d.'s in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell e.s.d.'s is used for estimating e.s.d.'s involving l.s. planes.

Refinement. Refinement of F^2 against ALL reflections. The weighted R-factor wR and goodness of fit S are based on F^2, conventional R-factors R are based on F, with F set to zero for negative F^2. The threshold expression of $F^2 > \sigma(F^2)$ is used only for calculating R-factors(gt) *etc*. and is not relevant to the choice of reflections for refinement. R-factors based on F^2 are statistically about twice as large as those based on F, and R- factors based on ALL data will be even larger.

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (Å^2)

	x	y	z	U_{iso}*/U_{eq}
P	0.97981 (4)	0.22655 (2)	0.06013 (4)	0.04171 (14)
O1	0.87224 (12)	0.16886 (6)	−0.04116 (11)	0.0509 (3)
O2	1.13289 (12)	0.18955 (6)	0.09457 (11)	0.0514 (3)
O3	0.95622 (13)	0.24405 (6)	0.18721 (10)	0.0563 (3)
N	0.97260 (16)	0.29192 (7)	−0.03860 (13)	0.0491 (4)
H	0.9660	0.2836	−0.1238	0.059*
C17	1.2659 (3)	0.00509 (17)	0.3197 (5)	0.1137 (12)
H17	1.2997	−0.0363	0.3701	0.136*
C2	0.84557 (18)	0.40475 (9)	−0.10133 (16)	0.0458 (4)

C1	0.97616 (19)	0.36437 (9)	0.00303 (17)	0.0507 (4)
H1A	1.0661	0.3856	0.0084	0.061*
H1B	0.9782	0.3667	0.0986	0.061*
C8	0.71826 (18)	0.17787 (9)	−0.09973 (17)	0.0490 (4)
C14	1.17229 (17)	0.12655 (9)	0.17145 (18)	0.0495 (4)
C13	0.6460 (2)	0.20013 (13)	−0.2392 (2)	0.0841 (7)
H13	0.6973	0.2108	−0.2939	0.101*
C3	0.7036 (2)	0.38616 (11)	−0.1264 (2)	0.0617 (5)
H3	0.6885	0.3474	−0.0802	0.074*
C7	0.8641 (2)	0.46219 (10)	−0.17277 (18)	0.0576 (5)
H7	0.9585	0.4753	−0.1589	0.069*
C9	0.6453 (2)	0.16092 (11)	−0.0180 (2)	0.0638 (5)
H9	0.6972	0.1456	0.0768	0.077*
C15	1.2344 (2)	0.12743 (12)	0.3197 (2)	0.0668 (5)
H15	1.2451	0.1690	0.3697	0.080*
C5	0.6047 (2)	0.48142 (13)	−0.2871 (2)	0.0779 (6)
H5	0.5239	0.5074	−0.3487	0.094*
C19	1.1552 (2)	0.06622 (12)	0.0974 (2)	0.0756 (6)
H19	1.1122	0.0662	−0.0032	0.091*
C6	0.7436 (2)	0.50035 (11)	−0.2647 (2)	0.0735 (6)
H6	0.7576	0.5392	−0.3116	0.088*
C10	0.4937 (2)	0.16685 (13)	−0.0781 (3)	0.0870 (7)
H10	0.4423	0.1553	−0.0240	0.104*
C18	1.2022 (3)	0.00495 (13)	0.1733 (5)	0.1065 (10)
H18	1.1901	−0.0367	0.1234	0.128*
C4	0.5843 (2)	0.42419 (14)	−0.2187 (2)	0.0765 (6)
H4	0.4893	0.4109	−0.2347	0.092*
C16	1.2806 (3)	0.06559 (18)	0.3931 (3)	0.0959 (9)
H16	1.3221	0.0652	0.4937	0.115*
C11	0.4196 (3)	0.18955 (14)	−0.2160 (4)	0.1061 (10)
H11	0.3173	0.1937	−0.2562	0.127*
C12	0.4943 (3)	0.20642 (15)	−0.2968 (3)	0.1170 (11)
H12	0.4423	0.2222	−0.3912	0.140*

Atomic displacement parameters (Å²)

	U^{11}	U^{22}	U^{33}	U^{12}	U^{13}	U^{23}
P	0.0509 (2)	0.0414 (3)	0.0336 (2)	0.0017 (2)	0.01886 (17)	0.0016 (2)
O1	0.0523 (7)	0.0403 (7)	0.0545 (6)	0.0012 (5)	0.0175 (5)	−0.0030 (5)
O2	0.0500 (6)	0.0496 (8)	0.0554 (6)	0.0035 (6)	0.0234 (5)	0.0076 (6)
O3	0.0751 (8)	0.0621 (9)	0.0367 (6)	0.0050 (7)	0.0287 (5)	0.0045 (6)
N	0.0775 (9)	0.0403 (9)	0.0341 (6)	0.0033 (7)	0.0283 (6)	−0.0005 (6)
C17	0.0716 (17)	0.080 (2)	0.198 (4)	0.0297 (16)	0.067 (2)	0.068 (3)
C2	0.0540 (9)	0.0409 (10)	0.0455 (8)	−0.0013 (8)	0.0242 (7)	−0.0057 (8)
C1	0.0614 (10)	0.0417 (11)	0.0466 (9)	−0.0041 (9)	0.0210 (8)	−0.0025 (8)
C8	0.0525 (10)	0.0325 (10)	0.0494 (9)	−0.0007 (8)	0.0097 (8)	−0.0013 (8)
C14	0.0388 (8)	0.0452 (11)	0.0628 (11)	0.0030 (8)	0.0202 (8)	0.0040 (9)
C13	0.0884 (16)	0.0824 (17)	0.0522 (11)	−0.0214 (13)	0.0022 (10)	0.0109 (11)

C3	0.0655 (12)	0.0597 (13)	0.0704 (11)	−0.0050 (10)	0.0390 (10)	0.0001 (10)
C7	0.0605 (11)	0.0512 (12)	0.0630 (11)	−0.0032 (10)	0.0282 (9)	0.0037 (10)
C9	0.0561 (11)	0.0623 (14)	0.0643 (10)	−0.0019 (10)	0.0177 (9)	−0.0007 (10)
C15	0.0664 (11)	0.0729 (15)	0.0637 (11)	0.0193 (11)	0.0305 (9)	0.0141 (11)
C5	0.0683 (14)	0.0810 (18)	0.0751 (14)	0.0245 (13)	0.0216 (11)	0.0064 (13)
C19	0.0555 (11)	0.0584 (15)	0.0963 (15)	0.0033 (11)	0.0167 (10)	−0.0175 (13)
C6	0.0884 (16)	0.0574 (14)	0.0726 (13)	0.0109 (12)	0.0325 (12)	0.0164 (11)
C10	0.0579 (13)	0.0773 (18)	0.1166 (18)	−0.0056 (12)	0.0286 (13)	−0.0169 (15)
C18	0.0680 (15)	0.0452 (16)	0.190 (3)	0.0061 (13)	0.0401 (19)	−0.005 (2)
C4	0.0524 (11)	0.0902 (18)	0.0894 (14)	0.0068 (12)	0.0326 (11)	−0.0006 (14)
C16	0.0857 (16)	0.116 (2)	0.0997 (17)	0.0436 (17)	0.0529 (14)	0.0563 (19)
C11	0.0530 (13)	0.0567 (16)	0.153 (3)	−0.0022 (12)	−0.0092 (16)	0.0056 (17)
C12	0.093 (2)	0.090 (2)	0.0937 (18)	−0.0231 (16)	−0.0313 (15)	0.0330 (15)

Geometric parameters (Å, º)

P—O3	1.4567 (10)	C3—C4	1.375 (3)
P—O1	1.5844 (12)	C3—H3	0.9300
P—O2	1.5880 (12)	C7—C6	1.381 (3)
P—N	1.5952 (14)	C7—H7	0.9300
O1—C8	1.4065 (19)	C9—C10	1.379 (3)
O2—C14	1.406 (2)	C9—H9	0.9300
N—C1	1.454 (2)	C15—C16	1.377 (3)
N—H	0.8600	C15—H15	0.9300
C17—C18	1.353 (4)	C5—C6	1.360 (3)
C17—C16	1.359 (4)	C5—C4	1.366 (3)
C17—H17	0.9300	C5—H5	0.9300
C2—C3	1.381 (2)	C19—C18	1.379 (4)
C2—C7	1.381 (2)	C19—H19	0.9300
C2—C1	1.503 (2)	C6—H6	0.9300
C1—H1A	0.9700	C10—C11	1.355 (4)
C1—H1B	0.9700	C10—H10	0.9300
C8—C13	1.363 (2)	C18—H18	0.9300
C8—C9	1.365 (3)	C4—H4	0.9300
C14—C19	1.356 (3)	C16—H16	0.9300
C14—C15	1.370 (2)	C11—C12	1.370 (4)
C13—C12	1.381 (3)	C11—H11	0.9300
C13—H13	0.9300	C12—H12	0.9300
O3—P—O1	114.84 (7)	C2—C7—H7	119.7
O3—P—O2	114.62 (6)	C6—C7—H7	119.7
O1—P—O2	98.80 (6)	C8—C9—C10	119.0 (2)
O3—P—N	113.69 (7)	C8—C9—H9	120.5
O1—P—N	107.77 (6)	C10—C9—H9	120.5
O2—P—N	105.75 (7)	C14—C15—C16	118.7 (2)
C8—O1—P	120.47 (10)	C14—C15—H15	120.7
C14—O2—P	121.58 (10)	C16—C15—H15	120.7
C1—N—P	125.61 (10)	C6—C5—C4	119.8 (2)

C1—N—H	117.2	C6—C5—H5	120.1
P—N—H	117.2	C4—C5—H5	120.1
C18—C17—C16	120.1 (3)	C14—C19—C18	119.2 (2)
C18—C17—H17	120.0	C14—C19—H19	120.4
C16—C17—H17	120.0	C18—C19—H19	120.4
C3—C2—C7	118.08 (17)	C5—C6—C7	120.3 (2)
C3—C2—C1	120.83 (16)	C5—C6—H6	119.8
C7—C2—C1	121.08 (16)	C7—C6—H6	119.8
N—C1—C2	112.54 (13)	C11—C10—C9	119.9 (2)
N—C1—H1A	109.1	C11—C10—H10	120.1
C2—C1—H1A	109.1	C9—C10—H10	120.1
N—C1—H1B	109.1	C17—C18—C19	120.4 (3)
C2—C1—H1B	109.1	C17—C18—H18	119.8
H1A—C1—H1B	107.8	C19—C18—H18	119.8
C13—C8—C9	122.16 (18)	C5—C4—C3	120.2 (2)
C13—C8—O1	118.58 (17)	C5—C4—H4	119.9
C9—C8—O1	119.17 (15)	C3—C4—H4	119.9
C19—C14—C15	121.14 (19)	C17—C16—C15	120.5 (3)
C19—C14—O2	119.20 (16)	C17—C16—H16	119.7
C15—C14—O2	119.57 (17)	C15—C16—H16	119.7
C8—C13—C12	117.9 (2)	C10—C11—C12	120.5 (2)
C8—C13—H13	121.0	C10—C11—H11	119.8
C12—C13—H13	121.0	C12—C11—H11	119.8
C4—C3—C2	120.91 (19)	C11—C12—C13	120.5 (2)
C4—C3—H3	119.5	C11—C12—H12	119.7
C2—C3—H3	119.5	C13—C12—H12	119.7
C2—C7—C6	120.63 (18)		

O3—P—O1—C8	56.68 (13)	C3—C2—C7—C6	1.0 (3)
O2—P—O1—C8	179.11 (11)	C1—C2—C7—C6	−177.78 (16)
N—P—O1—C8	−71.14 (13)	C13—C8—C9—C10	−0.4 (3)
O3—P—O2—C14	58.59 (14)	O1—C8—C9—C10	−177.10 (18)
O1—P—O2—C14	−63.99 (12)	C19—C14—C15—C16	0.5 (3)
N—P—O2—C14	−175.37 (11)	O2—C14—C15—C16	−176.02 (16)
O3—P—N—C1	13.52 (17)	C15—C14—C19—C18	−0.5 (3)
O1—P—N—C1	142.00 (14)	O2—C14—C19—C18	176.05 (17)
O2—P—N—C1	−113.09 (14)	C4—C5—C6—C7	−0.3 (3)
P—N—C1—C2	−124.33 (14)	C2—C7—C6—C5	−0.6 (3)
C3—C2—C1—N	60.3 (2)	C8—C9—C10—C11	−0.3 (4)
C7—C2—C1—N	−120.96 (17)	C16—C17—C18—C19	1.7 (4)
P—O1—C8—C13	99.80 (18)	C14—C19—C18—C17	−0.6 (3)
P—O1—C8—C9	−83.34 (19)	C6—C5—C4—C3	0.7 (3)
P—O2—C14—C19	99.71 (17)	C2—C3—C4—C5	−0.2 (3)
P—O2—C14—C15	−83.73 (17)	C18—C17—C16—C15	−1.7 (4)
C9—C8—C13—C12	1.0 (3)	C14—C15—C16—C17	0.6 (3)
O1—C8—C13—C12	177.8 (2)	C9—C10—C11—C12	0.3 (4)
C7—C2—C3—C4	−0.7 (3)	C10—C11—C12—C13	0.4 (4)
C1—C2—C3—C4	178.17 (17)	C8—C13—C12—C11	−1.1 (4)

Hydrogen-bond geometry (Å, º)

D—H···*A*	*D*—H	H···*A*	*D*···*A*	*D*—H···*A*
N—H···O3^{i}	0.86	1.97	2.8241 (15)	175

Symmetry code: (i) *x*, −*y*+1/2, *z*−1/2.