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7-Benzenesulfonamido-3-methyl-8-oxo-5-thia-1-azabicyclo[4.2.0]octa-2-ene-2-carboxylic acid monohydrate

 Shahzad Sharif,^a Mehmet Akkurt,^{b*} Islam Ullah Khan,^{a,‡} Manan Ayub Salariya^c and Sarfraz Ahmad^d

^aMaterials Chemistry Laboratory, Department of Chemistry, Government College University, Lahore 54000, Pakistan, ^bDepartment of Physics, Faculty of Arts and Sciences, Erciyes University, 38039 Kayseri, Turkey, ^cPharmagen Ltd, Lahore 54000, Pakistan, and ^dPharmagen Ltd., Lahore 54000, Pakistan
Correspondence e-mail: akkurt@erciyes.edu.tr

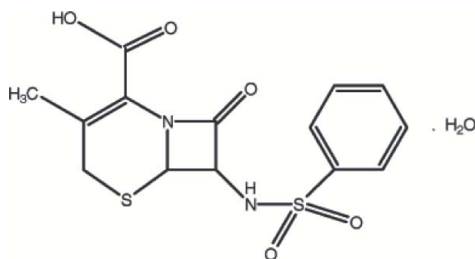
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Key indicators: single-crystal X-ray study; $T = 296$ K; mean $\sigma(\text{C}-\text{C}) = 0.007$ Å; R factor = 0.055; wR factor = 0.109; data-to-parameter ratio = 15.8.

In the title compound, $\text{C}_{14}\text{H}_{14}\text{N}_2\text{O}_5\text{S}_2 \cdot \text{H}_2\text{O}$, the six-membered ring fused to the β -lactam unit has a twisted conformation. Weak intramolecular $\text{N}-\text{H} \cdots \text{S}$ and $\text{C}-\text{H} \cdots \text{O}$ interactions occur. Intermolecular $\text{C}-\text{H} \cdots \text{S}$, $\text{N}-\text{H} \cdots \text{O}$, $\text{C}-\text{H} \cdots \text{O}$ and $\text{O}-\text{H} \cdots \text{O}$ hydrogen-bonding interactions stabilize the crystal structure, forming a three-dimensional network. Weak $\text{C}-\text{H} \cdots \pi$ interactions are also present.

Related literature

For the production of 7-amino-deacetoxycephalosporanic acid-like components by direct fermentation, see: Schroen *et al.* (2000). For 7-benzenesulfonamido-3-ethenyl-8-oxo-5-thia-1-azabicyclo[4.2.0]octa-2-ene-2-carboxylic acid methanol solvate, see: Mariam *et al.* (2009). For structures with the β -lactam unit, see: Akkurt *et al.* (2008*a,b,c*); Baktır *et al.* (2009); Pınar *et al.* (2006); Yalçın *et al.* (2009); Çelik *et al.* (2009*a,b*). For puckering parameters, see: Cremer & Pople (1975).



‡ Additional correspondence author, e-mail: iukhan@gcu.edu.pk.

Experimental

Crystal data

$\text{C}_{14}\text{H}_{14}\text{N}_2\text{O}_5\text{S}_2 \cdot \text{H}_2\text{O}$
 $M_r = 372.43$
 Orthorhombic, $P2_12_12_1$
 $a = 5.9535$ (7) Å
 $b = 15.8248$ (19) Å
 $c = 18.411$ (2) Å

$V = 1734.6$ (3) Å³
 $Z = 4$
 Mo $K\alpha$ radiation
 $\mu = 0.34$ mm⁻¹
 $T = 296$ K
 $0.20 \times 0.20 \times 0.20$ mm

Data collection

Bruker Kappa APEXII CCD area-detector diffractometer
 10847 measured reflections

3572 independent reflections
 1890 reflections with $I > 2\sigma(I)$
 $R_{\text{int}} = 0.064$

Refinement

$R[F^2 > 2\sigma(F^2)] = 0.055$
 $wR(F^2) = 0.109$
 $S = 0.97$
 3572 reflections
 226 parameters
 3 restraints

H atoms treated by a mixture of independent and constrained refinement
 $\Delta\rho_{\text{max}} = 0.20$ e Å⁻³
 $\Delta\rho_{\text{min}} = -0.25$ e Å⁻³
 Absolute structure: Flack (1983), 1425 Friedel pairs
 Flack parameter: -0.07 (11)

Table 1

Hydrogen-bond geometry (Å, °).

| $D-H \cdots A$ | $D-H$ | $H \cdots A$ | $D \cdots A$ | $D-H \cdots A$ |
|--|----------|--------------|--------------|----------------|
| $\text{O1}-\text{H1} \cdots \text{OW1}^{\text{i}}$ | 0.82 | 1.86 | 2.663 (6) | 166 |
| $\text{OW1}-\text{HW1} \cdots \text{O5}^{\text{ii}}$ | 0.83 (4) | 2.14 (5) | 2.799 (6) | 136 (5) |
| $\text{N2}-\text{H2} \cdots \text{S1}$ | 0.86 | 2.82 | 3.136 (3) | 103 |
| $\text{N2}-\text{H2} \cdots \text{O2}^{\text{iii}}$ | 0.86 | 2.30 | 2.846 (4) | 122 |
| $\text{OW1}-\text{HW2} \cdots \text{O3}$ | 0.84 (5) | 2.54 (5) | 3.135 (6) | 129 (5) |
| $\text{C3}-\text{H3A} \cdots \text{O1}$ | 0.96 | 2.22 | 2.902 (6) | 127 |
| $\text{C6}-\text{H6} \cdots \text{S1}^{\text{iv}}$ | 0.98 | 2.80 | 3.756 (4) | 165 |
| $\text{C8}-\text{H8} \cdots \text{O4}^{\text{i}}$ | 0.98 | 2.30 | 3.100 (5) | 138 |
| $\text{C11}-\text{H11} \cdots \text{O3}^{\text{v}}$ | 0.93 | 2.51 | 3.192 (6) | 130 |
| $\text{C3}-\text{H3B} \cdots \text{Cg3}^{\text{ii}}$ | 0.96 | 2.72 | 3.640 (5) | 161 |

Symmetry codes: (i) $x-1, y, z$; (ii) $-x+\frac{3}{2}, -y+1, z+\frac{1}{2}$; (iii) $x+1, y, z$; (iv) $x-\frac{1}{2}, -y+\frac{1}{2}, -z$; (v) $x+\frac{1}{2}, -y+\frac{3}{2}, -z$. Cg3 is the centroid of the C9–C14 ring.

Data collection: APEX2 (Bruker, 2007); cell refinement: SAINT (Bruker, 2007); data reduction: SAINT; program(s) used to solve structure: SIR97 (Altomare *et al.*, 1999); program(s) used to refine structure: SHELXL97 (Sheldrick, 2008); molecular graphics: ORTEP-3 for Windows (Farrugia, 1997); software used to prepare material for publication: WinGX (Farrugia, 1999) and PLATON (Spek, 2009).

Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: VM2014).

References

- Akkurt, M., Jarrahpour, A., Ebrahimi, E., Gençaslan, M. & Büyükgüngör, O. (2008*a*). *Acta Cryst.* E64, o2466–o2467.
 Akkurt, M., Karaca, S., Jarrahpour, A., Ebrahimi, E. & Büyükgüngör, O. (2008*b*). *Acta Cryst.* E64, o902–o903.
 Akkurt, M., Karaca, S., Jarrahpour, A. A., Zarei, M. & Büyükgüngör, O. (2008*c*). *Acta Cryst.* E64, o924.
 Altomare, A., Burla, M. C., Camalli, M., Cascarano, G. L., Giacovazzo, C., Guagliardi, A., Moliterni, A. G. G., Polidori, G. & Spagna, R. (1999). *J. Appl. Cryst.* 32, 115–119.
 Baktır, Z., Akkurt, M., Jarrahpour, A., Ebrahimi, E. & Büyükgüngör, O. (2009). *Acta Cryst.* E65, o1623–o1624.

- Bruker (2007). *APEX2* and *SAINT*. Bruker AXS Inc., Madison, Wisconsin, USA.
- Çelik, İ., Akkurt, M., Jarrahpour, A., Ebrahimi, E. & Büyükgüngör, O. (2009a). *Acta Cryst.* **E65**, o2522–o2523.
- Çelik, İ., Akkurt, M., Jarrahpour, A., Ebrahimi, E. & Büyükgüngör, O. (2009b). *Acta Cryst.* **E65**, o501–o502.
- Cremer, D. & Pople, J. A. (1975). *J. Am. Chem. Soc.* **97**, 1354–1358.
- Farrugia, L. J. (1997). *J. Appl. Cryst.* **30**, 565.
- Farrugia, L. J. (1999). *J. Appl. Cryst.* **32**, 837–838.
- Flack, H. D. (1983). *Acta Cryst.* **A39**, 876–881.
- Mariam, I., Akkurt, M., Sharif, S., Haider, S. K. & Khan, I. U. (2009). *Acta Cryst.* **E65**, o1737.
- Pınar, S., Akkurt, M., Jarrahpour, A. A., Khalili, D. & Büyükgüngör, O. (2006). *Acta Cryst.* **E62**, o804–o806.
- Schroen, C. G. P. H., VandeWiel, S., Kroon, P. J., DeVroom, E., Janssen, A. E. M. & Tramper, J. (2000). *Biotechnol. Bioeng.* **70**, 654–661.
- Sheldrick, G. M. (2008). *Acta Cryst.* **A64**, 112–122.
- Spek, A. L. (2009). *Acta Cryst.* **D65**, 148–155.
- Yalçın, Ş. P., Akkurt, M., Jarrahpour, A., Ebrahimi, E. & Büyükgüngör, O. (2009). *Acta Cryst.* **E65**, o626–o627.

supporting information

Acta Cryst. (2010). E66, o73–o74 [doi:10.1107/S1600536809051769]

7-Benzenesulfonamido-3-methyl-8-oxo-5-thia-1-azabicyclo[4.2.0]octa-2-ene-2-carboxylic acid monohydrate

Shahzad Sharif, Mehmet Akkurt, Islam Ullah Khan, Manan Ayub Salariya and Sarfraz Ahmad

S1. Comment

One of the building blocks of cephalosporin antibiotics is 7-amino-deacetoxycephalosporanic acid (7-ADCA). It is currently produced from penicillin G using an elaborate chemical ring-expansion step followed by an enzyme-catalyzed hydrolysis. However, 7-ADCA-like components can also be produced by direct fermentation (Schroen *et al.* 2000).

In the title molecule (I) shown in Fig. 1, the β -lactam unit (N1/C6–C8) has a twisted conformation with the dihedral angles of $164.7(4)^\circ$ and $164.7(4)^\circ$ between the planes N1 C6 C7 and C6 C8 C7 and the planes N1 C6 C8 and N1 C8 C7, respectively. The six-membered ring fused to the β -lactam unit, (N1/S1/C1/C2/C4/C6) is puckered with the puckering parameters (Cremer & Pople, 1975): $Q_T = 0.618(3) \text{ \AA}$, $\theta = 54.2(4)^\circ$, $\varphi = 340.4(5)^\circ$, respectively.

The crystal structure is stabilized by intermolecular C—H \cdots S, N—H \cdots O, C—H \cdots O and O—H \cdots O hydrogen bonding interactions between symmetry-related molecules, forming a network in three dimensions (Table 1, Fig. 2). Furthermore, there is a weak C—H \cdots π interaction [C3—H3B \cdots Cg3(-x + 3/2, -y + 1, z + 1/2); H3B \cdots Cg3 = 2.72 \text{ \AA}, C3 \cdots Cg3 = 3.640(5) \text{ \AA}, C3—H3B \cdots Cg3 = 161° , where Cg3 is a centroid of the phenyl ring C9–C14].

S2. Experimental

7-ADCA (1 g, 4.7 mmol) was dissolved in distilled water (20 ml) in a round bottom flask (50 ml). Na₂CO₃ (3M) solution was added to maintain the solution at pH 8–9, then benzene sulfonyl chloride (0.82 g, 4.7 mmol) was added dropwise to the solution and stirred at room temperature. As all benzene sulfonyl chloride was consumed, pH becomes stable at 8–9, which confirms the completion of reaction. Then pH was adjusted to 1–2, by using 3 N HCl. The precipitate obtained was filtered, washed with distilled water, dried and recrystallized in ethyl acetate. Light yellow prisms of (I) appeared after two days.

S3. Refinement

The H atoms of the water molecule were located in difference Fourier maps and were refined with O—H distances restrained to 0.83(1) \text{ \AA} and H \cdots H distances restrained to 1.30(1) \text{ \AA}, with displacement parameters fixed at 1.5 times U_{eq} of the parent O atoms. H atom on O1 was calculated and refined with a riding model [O—H = 0.82 \text{ \AA} and $U_{iso}(H) = 1.5U_{eq}(O)$]. The other H atoms were placed geometrically, with N—H = 0.86 \text{ \AA}, C—H = 0.96–0.98 \text{ \AA}, and included in the refinement using a riding model, with $U_{iso}(H) = 1.2$ or $1.5U_{eq}(\text{parent atom})$.

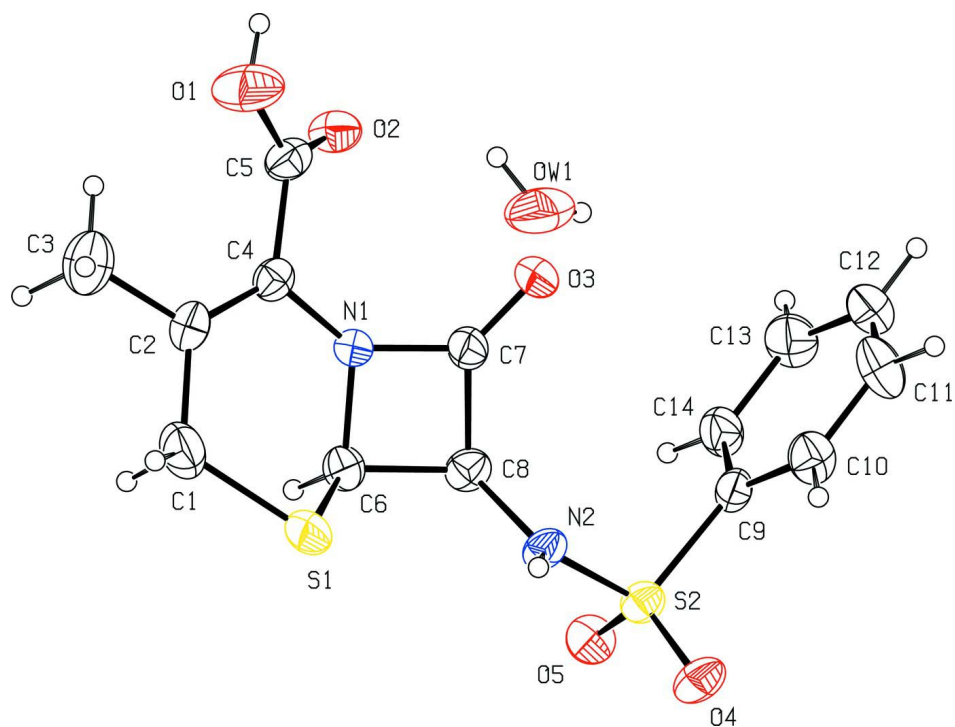


Figure 1

The title molecule with the atom numbering scheme. Displacement ellipsoids for non-H atoms are drawn at the 30% probability level.

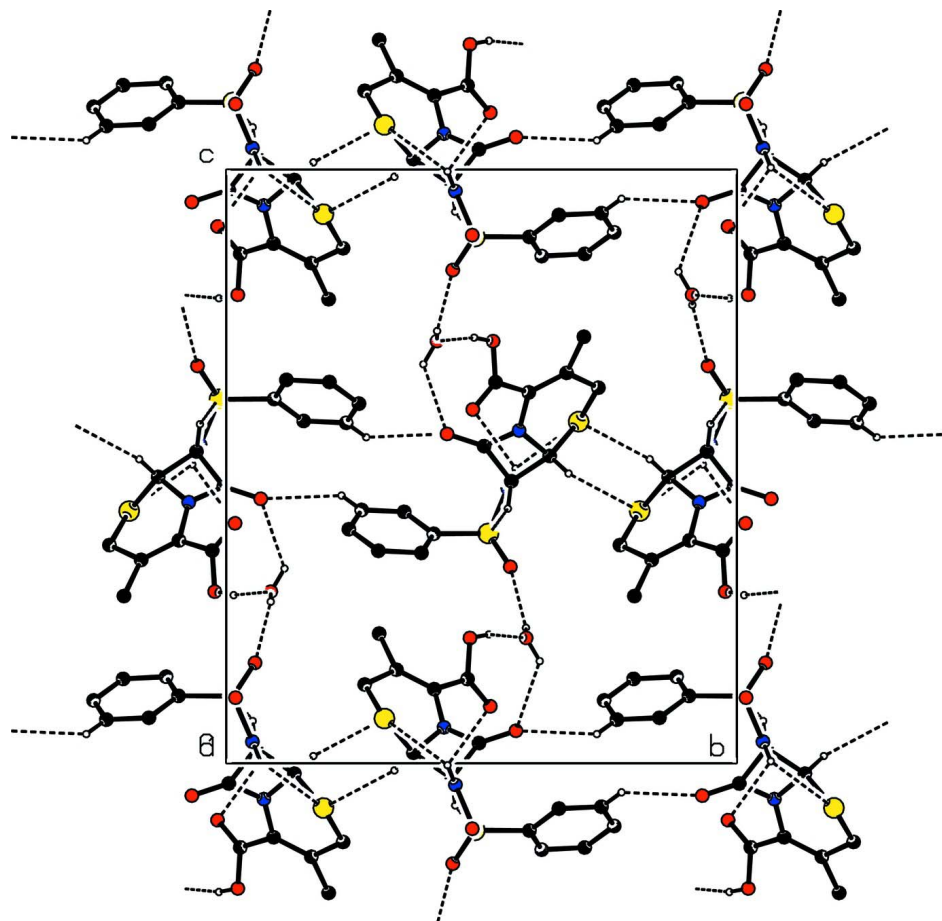


Figure 2

The packing and hydrogen bonding of the title compound viewed down *a* axis. Hydrogen atoms not involved in hydrogen bonding have been omitted for clarity.

7-Benzenesulfonamido-3-methyl-8-oxo-5-thia-1-azabicyclo[4.2.0]octa-2-ene-2-carboxylic acid monohydrate

Crystal data

$C_{14}H_{14}N_2O_5S_2 \cdot H_2O$

$M_r = 372.43$

Orthorhombic, $P2_12_12_1$

Hall symbol: P 2ac 2ab

$a = 5.9535$ (7) Å

$b = 15.8248$ (19) Å

$c = 18.411$ (2) Å

$V = 1734.6$ (3) Å³

$Z = 4$

$F(000) = 776$

$D_x = 1.426$ Mg m⁻³

Mo $K\alpha$ radiation, $\lambda = 0.71073$ Å

Cell parameters from 1402 reflections

$\theta = 2.8$ – 17.3°

$\mu = 0.34$ mm⁻¹

$T = 296$ K

Prismatic, light yellow

$0.20 \times 0.20 \times 0.20$ mm

Data collection

Bruker Kappa APEXII CCD area-detector
diffractometer

Radiation source: sealed tube

Graphite monochromator

φ and ω scans

10847 measured reflections

3572 independent reflections

1890 reflections with $I > 2\sigma(I)$

$R_{int} = 0.064$

$\theta_{max} = 26.8^\circ$, $\theta_{min} = 2.6^\circ$

$h = -4 \rightarrow 7$
 $k = -12 \rightarrow 20$

$l = -15 \rightarrow 23$

Refinement

Refinement on F^2
 Least-squares matrix: full
 $R[F^2 > 2\sigma(F^2)] = 0.055$
 $wR(F^2) = 0.109$
 $S = 0.97$
 3572 reflections
 226 parameters
 3 restraints
 Primary atom site location: structure-invariant
 direct methods
 Secondary atom site location: difference Fourier
 map
 Hydrogen site location: inferred from
 neighbouring sites

H atoms treated by a mixture of independent
 and constrained refinement
 $w = 1/[\sigma^2(F_o^2) + (0.0407P)^2]$
 where $P = (F_o^2 + 2F_c^2)/3$
 $(\Delta/\sigma)_{\max} < 0.001$
 $\Delta\rho_{\max} = 0.20 \text{ e } \text{\AA}^{-3}$
 $\Delta\rho_{\min} = -0.25 \text{ e } \text{\AA}^{-3}$
 Extinction correction: *SHELXL97* (Sheldrick,
 2008), $F_c^* = kF_c[1 + 0.001x F_c^2 \lambda^3 / \sin(2\theta)]^{-1/4}$
 Extinction coefficient: 0.0083 (10)
 Absolute structure: Flack (1983), 1425 Friedel
 pairs
 Absolute structure parameter: -0.07 (11)

Special details

Geometry. Bond distances, angles *etc.* have been calculated using the rounded fractional coordinates. All su's are estimated from the variances of the (full) variance-covariance matrix. The cell e.s.d.'s are taken into account in the estimation of distances, angles and torsion angles

Refinement. Refinement on F^2 for ALL reflections except those flagged by the user for potential systematic errors. Weighted R -factors wR and all goodnesses of fit S are based on F^2 , conventional R -factors R are based on F , with F set to zero for negative F^2 . The observed criterion of $F^2 > \sigma(F^2)$ is used only for calculating $-R$ -factor-obs *etc.* and is not relevant to the choice of reflections for refinement. R -factors based on F^2 are statistically about twice as large as those based on F , and R -factors based on ALL data will be even larger.

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (\AA^2)

| | x | y | z | $U_{\text{iso}}^*/U_{\text{eq}}$ |
|-----|--------------|--------------|---------------|----------------------------------|
| S1 | 0.8308 (2) | 0.30975 (7) | 0.07531 (7) | 0.0628 (5) |
| S2 | 1.06325 (17) | 0.48679 (7) | -0.11299 (6) | 0.0452 (4) |
| O1 | 0.1740 (7) | 0.4774 (2) | 0.21094 (17) | 0.0880 (16) |
| O2 | 0.1461 (5) | 0.51706 (18) | 0.09585 (15) | 0.0573 (11) |
| O3 | 0.6341 (5) | 0.56743 (17) | 0.05463 (18) | 0.0719 (13) |
| O4 | 1.3013 (4) | 0.48188 (19) | -0.11039 (15) | 0.0595 (11) |
| O5 | 0.9420 (5) | 0.44363 (17) | -0.16919 (15) | 0.0554 (10) |
| N1 | 0.5130 (5) | 0.42642 (19) | 0.06042 (18) | 0.0410 (11) |
| N2 | 0.9775 (5) | 0.44881 (19) | -0.03638 (18) | 0.0447 (12) |
| C1 | 0.6086 (8) | 0.2712 (3) | 0.1326 (3) | 0.072 (2) |
| C2 | 0.4335 (7) | 0.3333 (3) | 0.1575 (2) | 0.0500 (17) |
| C3 | 0.2901 (9) | 0.2995 (3) | 0.2184 (2) | 0.077 (2) |
| C4 | 0.3988 (6) | 0.4074 (3) | 0.1247 (2) | 0.0427 (16) |
| C5 | 0.2262 (7) | 0.4725 (3) | 0.1418 (3) | 0.0507 (17) |
| C6 | 0.6407 (7) | 0.3665 (2) | 0.0176 (2) | 0.0500 (17) |
| C7 | 0.6267 (6) | 0.4950 (3) | 0.0341 (2) | 0.0480 (16) |
| C8 | 0.7372 (7) | 0.4416 (3) | -0.0250 (2) | 0.0470 (17) |
| C9 | 0.9814 (6) | 0.5938 (2) | -0.1135 (2) | 0.0430 (14) |
| C10 | 1.1108 (7) | 0.6515 (3) | -0.0765 (2) | 0.0587 (17) |

| | | | | |
|-----|-------------|------------|-------------|-------------|
| C11 | 1.0450 (10) | 0.7349 (3) | -0.0734 (3) | 0.074 (2) |
| C12 | 0.8513 (10) | 0.7598 (3) | -0.1058 (3) | 0.069 (2) |
| C13 | 0.7206 (8) | 0.7021 (3) | -0.1421 (3) | 0.0683 (19) |
| C14 | 0.7853 (7) | 0.6178 (3) | -0.1457 (3) | 0.0577 (19) |
| OW1 | 0.8355 (9) | 0.5874 (3) | 0.2110 (3) | 0.128 (2) |
| H1 | 0.07960 | 0.51450 | 0.21690 | 0.1320* |
| H1A | 0.53260 | 0.22610 | 0.10690 | 0.0860* |
| H1B | 0.67690 | 0.24650 | 0.17550 | 0.0860* |
| H2 | 1.07130 | 0.43340 | -0.00340 | 0.0530* |
| H3A | 0.17150 | 0.33880 | 0.22870 | 0.1160* |
| H3B | 0.38080 | 0.29200 | 0.26100 | 0.1160* |
| H3C | 0.22680 | 0.24620 | 0.20430 | 0.1160* |
| H6 | 0.54600 | 0.32940 | -0.01210 | 0.0600* |
| H8 | 0.65710 | 0.44840 | -0.07110 | 0.0560* |
| H10 | 1.24240 | 0.63430 | -0.05370 | 0.0710* |
| H11 | 1.13350 | 0.77410 | -0.04910 | 0.0890* |
| H12 | 0.80690 | 0.81610 | -0.10330 | 0.0830* |
| H13 | 0.58820 | 0.71950 | -0.16430 | 0.0820* |
| H14 | 0.69620 | 0.57840 | -0.16970 | 0.0690* |
| HW1 | 0.706 (6) | 0.587 (5) | 0.228 (3) | 0.1920* |
| HW2 | 0.819 (12) | 0.614 (4) | 0.172 (2) | 0.1920* |

Atomic displacement parameters (Å²)

| | U^{11} | U^{22} | U^{33} | U^{12} | U^{13} | U^{23} |
|-----|-------------|-------------|-------------|--------------|--------------|--------------|
| S1 | 0.0590 (8) | 0.0401 (6) | 0.0892 (10) | 0.0106 (6) | 0.0078 (8) | 0.0044 (6) |
| S2 | 0.0366 (6) | 0.0512 (7) | 0.0479 (7) | 0.0031 (5) | 0.0001 (6) | 0.0012 (6) |
| O1 | 0.104 (3) | 0.110 (3) | 0.050 (2) | 0.032 (3) | 0.014 (2) | -0.012 (2) |
| O2 | 0.0538 (19) | 0.0611 (19) | 0.057 (2) | 0.0140 (18) | 0.0046 (17) | -0.0017 (17) |
| O3 | 0.062 (2) | 0.0368 (17) | 0.117 (3) | 0.0035 (17) | 0.029 (2) | -0.0053 (18) |
| O4 | 0.0328 (16) | 0.080 (2) | 0.0656 (19) | 0.0071 (16) | 0.0052 (15) | 0.0055 (18) |
| O5 | 0.0591 (18) | 0.0585 (17) | 0.0485 (19) | -0.0021 (16) | -0.0076 (17) | -0.0051 (15) |
| N1 | 0.0398 (19) | 0.0353 (18) | 0.048 (2) | 0.0017 (17) | 0.0067 (18) | -0.0016 (17) |
| N2 | 0.029 (2) | 0.054 (2) | 0.051 (2) | 0.0090 (16) | 0.0011 (16) | 0.0119 (18) |
| C1 | 0.077 (4) | 0.048 (3) | 0.091 (4) | -0.003 (3) | -0.002 (3) | 0.012 (3) |
| C2 | 0.043 (3) | 0.057 (3) | 0.050 (3) | -0.010 (3) | 0.000 (2) | 0.007 (2) |
| C3 | 0.080 (4) | 0.086 (4) | 0.066 (3) | -0.013 (3) | -0.005 (3) | 0.017 (3) |
| C4 | 0.038 (2) | 0.050 (3) | 0.040 (3) | -0.005 (2) | -0.001 (2) | -0.002 (2) |
| C5 | 0.040 (3) | 0.063 (3) | 0.049 (3) | -0.005 (2) | 0.004 (2) | -0.012 (3) |
| C6 | 0.048 (3) | 0.046 (3) | 0.056 (3) | -0.010 (2) | 0.000 (2) | -0.011 (2) |
| C7 | 0.041 (3) | 0.039 (2) | 0.064 (3) | 0.005 (2) | 0.003 (2) | 0.010 (2) |
| C8 | 0.037 (3) | 0.054 (3) | 0.050 (3) | 0.006 (2) | -0.003 (2) | 0.004 (2) |
| C9 | 0.037 (2) | 0.044 (2) | 0.048 (3) | -0.005 (2) | 0.004 (2) | 0.007 (2) |
| C10 | 0.052 (3) | 0.058 (3) | 0.066 (3) | -0.005 (3) | -0.012 (3) | 0.001 (3) |
| C11 | 0.094 (4) | 0.046 (3) | 0.083 (4) | -0.015 (3) | -0.020 (4) | -0.004 (3) |
| C12 | 0.082 (4) | 0.044 (3) | 0.081 (4) | 0.001 (3) | 0.009 (4) | 0.000 (3) |
| C13 | 0.053 (3) | 0.062 (3) | 0.090 (4) | 0.012 (3) | -0.012 (3) | 0.013 (3) |
| C14 | 0.047 (3) | 0.048 (3) | 0.078 (4) | -0.004 (2) | -0.004 (3) | 0.002 (2) |

| | | | | | | |
|-----|-----------|-----------|-----------|-----------|-----------|-----------|
| OW1 | 0.152 (4) | 0.104 (3) | 0.129 (4) | 0.035 (3) | 0.085 (4) | 0.000 (3) |
|-----|-----------|-----------|-----------|-----------|-----------|-----------|

Geometric parameters (Å, °)

| | | | |
|-----------------------|-----------|-----------------------|-----------|
| S1—C1 | 1.799 (5) | C6—C8 | 1.535 (6) |
| S1—C6 | 1.793 (4) | C7—C8 | 1.527 (6) |
| S2—O4 | 1.420 (3) | C9—C14 | 1.363 (6) |
| S2—O5 | 1.435 (3) | C9—C10 | 1.375 (6) |
| S2—N2 | 1.616 (3) | C10—C11 | 1.378 (7) |
| S2—C9 | 1.762 (3) | C11—C12 | 1.357 (8) |
| O1—C5 | 1.313 (6) | C12—C13 | 1.373 (7) |
| O2—C5 | 1.200 (6) | C13—C14 | 1.390 (7) |
| O3—C7 | 1.208 (5) | C1—H1B | 0.9700 |
| O1—H1 | 0.8200 | C1—H1A | 0.9700 |
| OW1—HW2 | 0.84 (5) | C3—H3C | 0.9600 |
| OW1—HW1 | 0.83 (4) | C3—H3A | 0.9600 |
| N1—C4 | 1.398 (5) | C3—H3B | 0.9600 |
| N1—C7 | 1.368 (5) | C6—H6 | 0.9800 |
| N1—C6 | 1.449 (5) | C8—H8 | 0.9800 |
| N2—C8 | 1.450 (5) | C10—H10 | 0.9300 |
| N2—H2 | 0.8600 | C11—H11 | 0.9300 |
| C1—C2 | 1.504 (7) | C12—H12 | 0.9300 |
| C2—C4 | 1.335 (6) | C13—H13 | 0.9300 |
| C2—C3 | 1.507 (6) | C14—H14 | 0.9300 |
| C4—C5 | 1.489 (6) | | |
| S1…N2 | 3.136 (3) | C5…OW1 ⁱⁱⁱ | 3.216 (7) |
| S1…C5 ⁱ | 3.698 (5) | C7…O4 ⁱⁱⁱ | 3.297 (5) |
| S1…H2 | 2.8200 | C7…O2 ⁱ | 3.313 (5) |
| S1…H6 ⁱⁱ | 2.8000 | C7…O2 | 3.099 (5) |
| O1…OW1 ⁱⁱⁱ | 2.663 (6) | C8…C14 | 3.577 (7) |
| O1…C3 | 2.902 (6) | C8…O4 ⁱⁱⁱ | 3.100 (5) |
| OW1…O2 ⁱ | 3.025 (6) | C11…O3 ^{vii} | 3.192 (6) |
| OW1…O3 | 3.135 (6) | C12…O3 ^{vii} | 3.346 (6) |
| OW1…C5 ⁱ | 3.216 (7) | C14…C8 | 3.577 (7) |
| OW1…O5 ^{iv} | 2.799 (6) | C4…H2 ⁱⁱⁱ | 3.0900 |
| OW1…O1 ⁱ | 2.663 (6) | C5…H3A | 2.6700 |
| O2…N1 | 2.693 (4) | C5…H2 ⁱⁱⁱ | 2.8900 |
| O2…C7 ⁱⁱⁱ | 3.313 (5) | C9…H3B ^{vi} | 3.0500 |
| O2…C7 | 3.099 (5) | C12…H3B ^{vi} | 3.0400 |
| O2…OW1 ⁱⁱⁱ | 3.025 (6) | C13…H3B ^{vi} | 2.9700 |
| O2…O3 | 3.107 (4) | C14…H3B ^{vi} | 2.9900 |
| O2…N2 ⁱⁱⁱ | 2.846 (4) | H1…HW2 ⁱⁱⁱ | 2.3600 |
| O3…O2 | 3.107 (4) | H1…OW1 ⁱⁱⁱ | 1.8600 |
| O3…OW1 | 3.135 (6) | H1…HW1 ⁱⁱⁱ | 2.5100 |
| O3…C5 | 3.276 (6) | HW1…O5 ^{iv} | 2.14 (5) |
| O3…C12 ^v | 3.346 (6) | HW1…H1 ⁱ | 2.5100 |
| O3…N2 | 3.242 (4) | H1A…H3C | 2.5800 |

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|-------------------------|-------------|-------------------------|-----------|
| O3...C11 ^v | 3.192 (6) | H1B...H3B | 2.4700 |
| O4...C7 ⁱ | 3.297 (5) | H2...O2 ⁱ | 2.3000 |
| O4...C8 ⁱ | 3.100 (5) | H2...N1 ⁱ | 2.8800 |
| O5...OW1 ^{vi} | 2.799 (6) | H2...S1 | 2.8200 |
| O1...H3A | 2.2200 | H2...C5 ⁱ | 2.8900 |
| OW1...H1 ⁱ | 1.8600 | H2...C4 ⁱ | 3.0900 |
| O2...HW2 ⁱⁱⁱ | 2.85 (6) | HW2...O3 | 2.54 (5) |
| O2...H2 ⁱⁱⁱ | 2.3000 | HW2...O2 ⁱ | 2.85 (6) |
| O2...H12 ^v | 2.8100 | HW2...H1 ⁱ | 2.3600 |
| O3...HW2 | 2.54 (5) | H3A...O1 | 2.2200 |
| O3...H12 ^v | 2.8300 | H3A...C5 | 2.6700 |
| O3...H11 ^v | 2.5100 | H3B...C12 ^{iv} | 3.0400 |
| O4...H10 | 2.6500 | H3B...C14 ^{iv} | 2.9900 |
| O4...H8 ⁱ | 2.3000 | H3B...C9 ^{iv} | 3.0500 |
| O5...HW1 ^{vi} | 2.14 (5) | H3B...C13 ^{iv} | 2.9700 |
| O5...H14 | 2.5900 | H3B...H1B | 2.4700 |
| O5...H8 | 2.4800 | H3C...H1A | 2.5800 |
| N1...O2 | 2.693 (4) | H6...S1 ^{viii} | 2.8000 |
| N2...O2 ⁱ | 2.846 (4) | H8...O5 | 2.4800 |
| N2...S1 | 3.136 (3) | H8...O4 ⁱⁱⁱ | 2.3000 |
| N2...O3 | 3.242 (4) | H10...O4 | 2.6500 |
| N1...H2 ⁱⁱⁱ | 2.8800 | H11...O3 ^{vii} | 2.5100 |
| C3...O1 | 2.902 (6) | H12...O3 ^{vii} | 2.8300 |
| C5...O3 | 3.276 (6) | H12...O2 ^{vii} | 2.8100 |
| C5...S1 ⁱⁱⁱ | 3.698 (5) | H14...O5 | 2.5900 |
| | | | |
| C1—S1—C6 | 93.1 (2) | S2—C9—C14 | 120.5 (3) |
| O4—S2—O5 | 120.03 (18) | C10—C9—C14 | 120.7 (4) |
| O4—S2—N2 | 105.40 (17) | C9—C10—C11 | 119.8 (4) |
| O4—S2—C9 | 109.19 (18) | C10—C11—C12 | 120.1 (5) |
| O5—S2—N2 | 107.07 (17) | C11—C12—C13 | 120.2 (5) |
| O5—S2—C9 | 108.33 (17) | C12—C13—C14 | 120.3 (5) |
| N2—S2—C9 | 105.95 (17) | C9—C14—C13 | 118.9 (4) |
| C5—O1—H1 | 110.00 | S1—C1—H1A | 108.00 |
| HW1—OW1—HW2 | 103 (6) | S1—C1—H1B | 108.00 |
| C6—N1—C7 | 93.8 (3) | C2—C1—H1B | 108.00 |
| C4—N1—C6 | 125.1 (3) | H1A—C1—H1B | 107.00 |
| C4—N1—C7 | 135.4 (3) | C2—C1—H1A | 108.00 |
| S2—N2—C8 | 117.8 (3) | C2—C3—H3B | 109.00 |
| C8—N2—H2 | 121.00 | C2—C3—H3C | 109.00 |
| S2—N2—H2 | 121.00 | H3A—C3—H3B | 109.00 |
| S1—C1—C2 | 117.8 (3) | H3A—C3—H3C | 110.00 |
| C1—C2—C4 | 122.9 (4) | H3B—C3—H3C | 109.00 |
| C3—C2—C4 | 124.1 (4) | C2—C3—H3A | 109.00 |
| C1—C2—C3 | 112.8 (4) | S1—C6—H6 | 113.00 |
| N1—C4—C2 | 119.8 (4) | C8—C6—H6 | 113.00 |
| N1—C4—C5 | 111.5 (4) | N1—C6—H6 | 113.00 |
| C2—C4—C5 | 128.2 (4) | C6—C8—H8 | 110.00 |

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|--------------|------------|-----------------|------------|
| O1—C5—C4 | 114.2 (4) | C7—C8—H8 | 110.00 |
| O2—C5—C4 | 122.1 (5) | N2—C8—H8 | 110.00 |
| O1—C5—O2 | 123.7 (4) | C11—C10—H10 | 120.00 |
| S1—C6—N1 | 109.7 (3) | C9—C10—H10 | 120.00 |
| S1—C6—C8 | 117.0 (3) | C10—C11—H11 | 120.00 |
| N1—C6—C8 | 88.2 (3) | C12—C11—H11 | 120.00 |
| N1—C7—C8 | 91.5 (3) | C13—C12—H12 | 120.00 |
| O3—C7—N1 | 131.3 (4) | C11—C12—H12 | 120.00 |
| O3—C7—C8 | 137.1 (4) | C12—C13—H13 | 120.00 |
| N2—C8—C7 | 119.0 (3) | C14—C13—H13 | 120.00 |
| C6—C8—C7 | 84.4 (3) | C9—C14—H14 | 121.00 |
| N2—C8—C6 | 120.3 (4) | C13—C14—H14 | 121.00 |
| S2—C9—C10 | 118.7 (3) | | |
| | | | |
| C6—S1—C1—C2 | -46.4 (4) | S1—C1—C2—C3 | -166.4 (3) |
| C1—S1—C6—N1 | 57.8 (3) | S1—C1—C2—C4 | 18.8 (6) |
| C1—S1—C6—C8 | 156.1 (3) | C1—C2—C4—N1 | 6.2 (6) |
| O4—S2—N2—C8 | -176.8 (3) | C1—C2—C4—C5 | 177.2 (4) |
| O5—S2—N2—C8 | -47.9 (3) | C3—C2—C4—N1 | -168.1 (4) |
| C9—S2—N2—C8 | 67.6 (3) | C3—C2—C4—C5 | 3.0 (7) |
| O4—S2—C9—C10 | -32.7 (4) | N1—C4—C5—O1 | -155.7 (4) |
| O4—S2—C9—C14 | 151.9 (4) | N1—C4—C5—O2 | 23.2 (6) |
| O5—S2—C9—C10 | -165.0 (3) | C2—C4—C5—O1 | 32.6 (6) |
| O5—S2—C9—C14 | 19.5 (4) | C2—C4—C5—O2 | -148.5 (5) |
| N2—S2—C9—C10 | 80.4 (3) | S1—C6—C8—N2 | 19.3 (5) |
| N2—S2—C9—C14 | -95.1 (4) | S1—C6—C8—C7 | -101.1 (3) |
| C6—N1—C4—C2 | 11.5 (6) | N1—C6—C8—N2 | 130.5 (4) |
| C6—N1—C4—C5 | -161.0 (4) | N1—C6—C8—C7 | 10.1 (3) |
| C7—N1—C4—C2 | -134.5 (5) | O3—C7—C8—N2 | 45.1 (7) |
| C7—N1—C4—C5 | 53.1 (6) | O3—C7—C8—C6 | 166.7 (5) |
| C4—N1—C6—S1 | -49.9 (4) | N1—C7—C8—N2 | -132.3 (4) |
| C4—N1—C6—C8 | -168.0 (3) | N1—C7—C8—C6 | -10.7 (3) |
| C7—N1—C6—S1 | 106.9 (3) | S2—C9—C10—C11 | -177.2 (4) |
| C7—N1—C6—C8 | -11.2 (3) | C14—C9—C10—C11 | -1.7 (6) |
| C4—N1—C7—O3 | -13.7 (7) | S2—C9—C14—C13 | 177.0 (4) |
| C4—N1—C7—C8 | 164.0 (4) | C10—C9—C14—C13 | 1.6 (7) |
| C6—N1—C7—O3 | -166.4 (4) | C9—C10—C11—C12 | 1.0 (7) |
| C6—N1—C7—C8 | 11.3 (3) | C10—C11—C12—C13 | -0.3 (8) |
| S2—N2—C8—C6 | 143.8 (3) | C11—C12—C13—C14 | 0.2 (8) |
| S2—N2—C8—C7 | -114.9 (3) | C12—C13—C14—C9 | -0.8 (8) |

Symmetry codes: (i) $x+1, y, z$; (ii) $x+1/2, -y+1/2, -z$; (iii) $x-1, y, z$; (iv) $-x+3/2, -y+1, z+1/2$; (v) $x-1/2, -y+3/2, -z$; (vi) $-x+3/2, -y+1, z-1/2$; (vii) $x+1/2, -y+3/2, -z$; (viii) $x-1/2, -y+1/2, -z$.

Hydrogen-bond geometry ($\text{\AA}, ^\circ$)

| $D-H\cdots A$ | $D-H$ | $H\cdots A$ | $D\cdots A$ | $D-H\cdots A$ |
|-----------------------------------|----------|-------------|-------------|---------------|
| O1—H1 \cdots OW1 ⁱⁱⁱ | 0.82 | 1.86 | 2.663 (6) | 166 |
| OW1—HW1 \cdots O5 ^{iv} | 0.83 (4) | 2.14 (5) | 2.799 (6) | 136 (5) |

| | | | | |
|--|----------|----------|-----------|---------|
| N2—H2…S1 | 0.86 | 2.82 | 3.136 (3) | 103 |
| N2—H2…O2 ⁱ | 0.86 | 2.30 | 2.846 (4) | 122 |
| O <i>W</i> 1—H <i>W</i> 2…O3 | 0.84 (5) | 2.54 (5) | 3.135 (6) | 129 (5) |
| C3—H3 <i>A</i> …O1 | 0.96 | 2.22 | 2.902 (6) | 127 |
| C6—H6…S1 ^{viii} | 0.98 | 2.80 | 3.756 (4) | 165 |
| C8—H8…O4 ⁱⁱⁱ | 0.98 | 2.30 | 3.100 (5) | 138 |
| C8—H8…O5 | 0.98 | 2.48 | 2.922 (5) | 107 |
| C11—H11…O3 ^{vii} | 0.93 | 2.51 | 3.192 (6) | 130 |
| C14—H14…O5 | 0.93 | 2.59 | 2.942 (5) | 103 |
| C3—H3 <i>B</i> …C <i>g</i> 3 ^{iv} | 0.96 | 2.72 | 3.640 (5) | 161 |

Symmetry codes: (i) $x+1, y, z$; (iii) $x-1, y, z$; (iv) $-x+3/2, -y+1, z+1/2$; (vii) $x+1/2, -y+3/2, -z$; (viii) $x-1/2, -y+1/2, -z$.