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1-[4-(Dimethylamino)benzylidene]-4-methylthiosemicarbazide

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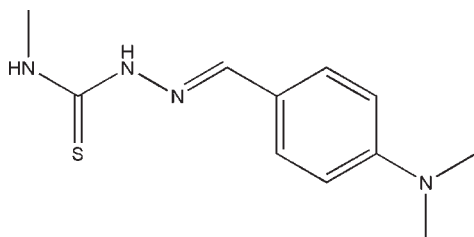
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Key indicators: single-crystal X-ray study; $T = 293$ K; mean $\sigma(\text{C}-\text{C}) = 0.003$ Å; R factor = 0.055; wR factor = 0.178; data-to-parameter ratio = 19.6.

In the title compound, $\text{C}_{11}\text{H}_{16}\text{N}_4\text{S}$, an intramolecular $\text{N}-\text{H}\cdots\text{N}$ hydrogen bond generates an $S(5)$ ring. In the crystal, inversion dimers linked by pairs of $\text{N}-\text{H}\cdots\text{S}$ bonds occur, generating an $R_2^2(8)$ loop.

Related literature

For a related structure, see: Girgis (2006).



Experimental

Crystal data

 $\text{C}_{11}\text{H}_{16}\text{N}_4\text{S}$
 $M_r = 236.34$

Monoclinic, $P2_1/n$
 $a = 10.517$ (2) Å
 $b = 12.873$ (3) Å
 $c = 10.552$ (2) Å
 $\beta = 119.19$ (3)°
 $V = 1247.3$ (4) Å³

$Z = 4$
 Mo $K\alpha$ radiation
 $\mu = 0.24$ mm⁻¹
 $T = 293$ K
 $0.22 \times 0.20 \times 0.18$ mm

Data collection

Bruker SMART CCD
 diffractometer
 11842 measured reflections

2847 independent reflections
 2391 reflections with $I > 2\sigma(I)$
 $R_{\text{int}} = 0.031$

Refinement

$R[F^2 > 2\sigma(F^2)] = 0.055$
 $wR(F^2) = 0.178$
 $S = 1.12$
 2847 reflections

145 parameters
 H-atom parameters constrained
 $\Delta\rho_{\text{max}} = 0.43$ e Å⁻³
 $\Delta\rho_{\text{min}} = -0.26$ e Å⁻³

Table 1

Hydrogen-bond geometry (Å, °).

$D-H\cdots A$	$D-H$	$H\cdots A$	$D\cdots A$	$D-H\cdots A$
$\text{N4}-\text{H4A}\cdots\text{N2}$	0.86	2.22	2.613 (2)	108
$\text{N3}-\text{H3A}\cdots\text{S1}^1$	0.86	2.59	3.3890 (16)	155

 Symmetry code: (i) $-x + 1, -y, -z$.

Data collection: *SMART* (Bruker 1997); cell refinement: *SAINTE* (Bruker 1997); data reduction: *SAINTE*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 2008); program(s) used to refine structure: *SHELXL97* (Sheldrick, 2008); molecular graphics: *SHELXTL* (Sheldrick, 2008); software used to prepare material for publication: *SHELXTL*.

Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: HB5446).

References

- Bruker (1997). *SMART* and *SAINTE*. Bruker AXS, Inc., Madison, Wisconsin, USA.
 Girgis, A. S. (2006). *J. Chem. Res.* pp. 81–85.
 Sheldrick, G. M. (2008). *Acta Cryst.* **A64**, 112–122.

supporting information

Acta Cryst. (2010). E66, o1397 [https://doi.org/10.1107/S1600536810018386]

1-[4-(Dimethylamino)benzylidene]-4-methylthiosemicarbazide

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S1. Experimental

A mixture of 4-(dimethylamino)benzaldehyde (0.10 mol), and 4-methylthiosemicarbazide (0.10 mol) was stirred in refluxing ethanol (10 ml) for 4 h to afford the title compound (0.083 mol, yield 83%). Colourless blocks of (I) were obtained by recrystallization from ethanol at room temperature.

S2. Refinement

H atoms were fixed geometrically and allowed to ride on their attached atoms, with C—H distances = 0.93–0.97 Å; N—H = 0.86 Å and with $U_{\text{iso}}(\text{H}) = 1.2U_{\text{eq}}(\text{C}, \text{N})$ or $1.5U_{\text{eq}}(\text{C}_{\text{methyl}})$.

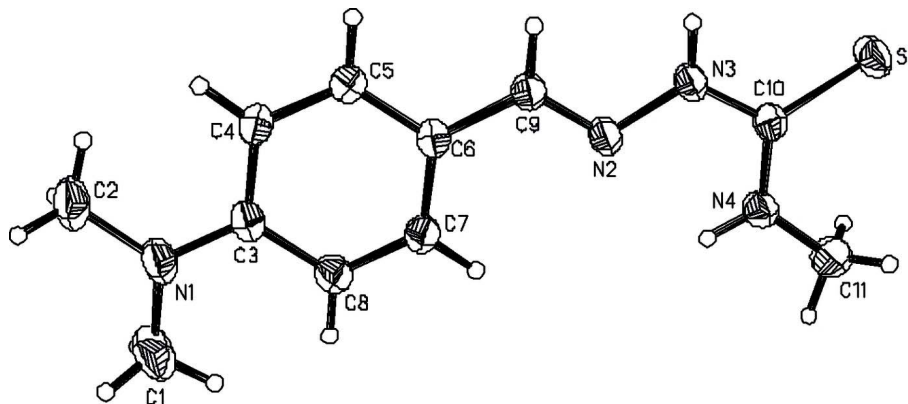


Figure 1

The structure of (I) showing 30% probability displacement ellipsoids.

1-[4-(Dimethylamino)benzylidene]-4-methylthiosemicarbazide

Crystal data

$\text{C}_{11}\text{H}_{16}\text{N}_4\text{S}$

$M_r = 236.34$

Monoclinic, $P2_1/n$

Hall symbol: $-P\ 2_1n$

$a = 10.517(2)\ \text{\AA}$

$b = 12.873(3)\ \text{\AA}$

$c = 10.552(2)\ \text{\AA}$

$\beta = 119.19(3)^\circ$

$V = 1247.3(4)\ \text{\AA}^3$

$Z = 4$

$F(000) = 504$

$D_x = 1.259\ \text{Mg m}^{-3}$

Mo $K\alpha$ radiation, $\lambda = 0.71073\ \text{\AA}$

Cell parameters from 2391 reflections

$\theta = 3.2\text{--}27.5^\circ$

$\mu = 0.24\ \text{mm}^{-1}$

$T = 293\ \text{K}$

Block, colorless

$0.22 \times 0.20 \times 0.18\ \text{mm}$

Data collection

Bruker SMART CCD diffractometer	2391 reflections with $I > 2\sigma(I)$
Radiation source: fine-focus sealed tube	$R_{\text{int}} = 0.031$
Graphite monochromator	$\theta_{\text{max}} = 27.5^\circ$, $\theta_{\text{min}} = 3.2^\circ$
phi and ω scans	$h = -13 \rightarrow 11$
11842 measured reflections	$k = -15 \rightarrow 16$
2847 independent reflections	$l = -13 \rightarrow 13$

Refinement

Refinement on F^2	Secondary atom site location: difference Fourier map
Least-squares matrix: full	Hydrogen site location: inferred from neighbouring sites
$R[F^2 > 2\sigma(F^2)] = 0.055$	H-atom parameters constrained
$wR(F^2) = 0.178$	$w = 1/[\sigma^2(F_o^2) + (0.1186P)^2 + 0.111P]$
$S = 1.12$	where $P = (F_o^2 + 2F_c^2)/3$
2847 reflections	$(\Delta/\sigma)_{\text{max}} < 0.001$
145 parameters	$\Delta\rho_{\text{max}} = 0.43 \text{ e } \text{\AA}^{-3}$
0 restraints	$\Delta\rho_{\text{min}} = -0.26 \text{ e } \text{\AA}^{-3}$
Primary atom site location: structure-invariant direct methods	

Special details

Geometry. All esds (except the esd in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell esds are taken into account individually in the estimation of esds in distances, angles and torsion angles; correlations between esds in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell esds is used for estimating esds involving l.s. planes.

Refinement. Refinement of F^2 against ALL reflections. The weighted R -factor wR and goodness of fit S are based on F^2 , conventional R -factors R are based on F , with F set to zero for negative F^2 . The threshold expression of $F^2 > \sigma(F^2)$ is used only for calculating R -factors(gt) etc. and is not relevant to the choice of reflections for refinement. R -factors based on F^2 are statistically about twice as large as those based on F , and R -factors based on ALL data will be even larger.

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (\AA^2)

	x	y	z	$U_{\text{iso}}^*/U_{\text{eq}}$
S1	0.60342 (4)	0.15360 (4)	0.05021 (6)	0.0595 (2)
N2	0.17952 (14)	0.12707 (11)	-0.13146 (16)	0.0468 (3)
N3	0.32723 (14)	0.10421 (10)	-0.05970 (16)	0.0491 (4)
H3A	0.3569	0.0426	-0.0272	0.059*
C6	-0.06129 (17)	0.06639 (12)	-0.20701 (17)	0.0435 (4)
N4	0.36809 (16)	0.27166 (11)	-0.09695 (18)	0.0537 (4)
H4A	0.2749	0.2772	-0.1463	0.064*
C3	-0.36590 (17)	0.09210 (14)	-0.32393 (17)	0.0481 (4)
C7	-0.13111 (19)	0.15789 (12)	-0.2769 (2)	0.0495 (4)
H7A	-0.0763	0.2116	-0.2848	0.059*
C10	0.42361 (17)	0.17963 (13)	-0.04141 (17)	0.0433 (4)
C9	0.09536 (17)	0.05248 (12)	-0.13907 (18)	0.0448 (4)
H9A	0.1355	-0.0119	-0.1002	0.054*
C5	-0.14729 (19)	-0.01233 (13)	-0.1990 (2)	0.0501 (4)
H5A	-0.1036	-0.0746	-0.1543	0.060*
C8	-0.27852 (19)	0.17078 (14)	-0.3344 (2)	0.0537 (4)
H8A	-0.3215	0.2327	-0.3812	0.064*

C4	-0.29596 (19)	-0.00089 (14)	-0.2553 (2)	0.0529 (4)
H4B	-0.3502	-0.0552	-0.2478	0.063*
N1	-0.51338 (16)	0.10490 (14)	-0.38125 (18)	0.0634 (4)
C2	-0.5998 (2)	0.0282 (2)	-0.3574 (3)	0.0758 (6)
H2B	-0.5423	-0.0329	-0.3152	0.114*
H2C	-0.6834	0.0107	-0.4484	0.114*
H2D	-0.6313	0.0560	-0.2929	0.114*
C11	0.4544 (2)	0.36333 (16)	-0.0797 (3)	0.0730 (6)
H12A	0.3912	0.4207	-0.1293	0.109*
H12B	0.5174	0.3507	-0.1201	0.109*
H12C	0.5120	0.3795	0.0216	0.109*
C1	-0.5822 (2)	0.2016 (2)	-0.4442 (3)	0.0929 (8)
H1A	-0.5142	0.2460	-0.4541	0.139*
H1B	-0.6138	0.2345	-0.3828	0.139*
H1C	-0.6647	0.1893	-0.5381	0.139*

Atomic displacement parameters (Å²)

	U^{11}	U^{22}	U^{33}	U^{12}	U^{13}	U^{23}
S1	0.0340 (3)	0.0526 (3)	0.0744 (4)	-0.00233 (15)	0.0127 (2)	-0.00072 (19)
N2	0.0336 (7)	0.0484 (7)	0.0547 (8)	0.0008 (5)	0.0187 (6)	0.0001 (6)
N3	0.0335 (6)	0.0439 (7)	0.0637 (8)	0.0010 (5)	0.0188 (6)	0.0032 (6)
C6	0.0375 (8)	0.0465 (8)	0.0464 (8)	-0.0018 (6)	0.0205 (6)	-0.0026 (6)
N4	0.0395 (7)	0.0477 (8)	0.0626 (9)	-0.0017 (5)	0.0160 (6)	0.0071 (6)
C3	0.0382 (8)	0.0593 (10)	0.0458 (8)	-0.0015 (6)	0.0197 (6)	-0.0050 (7)
C7	0.0438 (9)	0.0455 (9)	0.0594 (10)	-0.0032 (6)	0.0253 (8)	0.0021 (7)
C10	0.0382 (7)	0.0455 (8)	0.0436 (8)	-0.0027 (6)	0.0180 (6)	-0.0032 (6)
C9	0.0391 (8)	0.0437 (8)	0.0504 (8)	-0.0006 (6)	0.0210 (6)	-0.0012 (6)
C5	0.0446 (9)	0.0455 (8)	0.0572 (10)	-0.0021 (6)	0.0226 (8)	0.0055 (7)
C8	0.0447 (9)	0.0481 (9)	0.0653 (11)	0.0052 (7)	0.0245 (8)	0.0049 (7)
C4	0.0433 (9)	0.0567 (9)	0.0582 (10)	-0.0095 (7)	0.0245 (8)	0.0025 (7)
N1	0.0370 (8)	0.0805 (11)	0.0677 (10)	-0.0010 (7)	0.0218 (7)	0.0000 (8)
C2	0.0468 (10)	0.1131 (18)	0.0689 (12)	-0.0167 (11)	0.0293 (9)	-0.0044 (12)
C11	0.0597 (12)	0.0529 (10)	0.0951 (16)	-0.0073 (9)	0.0289 (11)	0.0121 (10)
C1	0.0477 (11)	0.0861 (17)	0.128 (2)	0.0166 (11)	0.0301 (13)	-0.0016 (16)

Geometric parameters (Å, °)

S1—C10	1.6852 (17)	C9—H9A	0.9300
N2—C9	1.282 (2)	C5—C4	1.383 (3)
N2—N3	1.3877 (19)	C5—H5A	0.9300
N3—C10	1.348 (2)	C8—H8A	0.9300
N3—H3A	0.8600	C4—H4B	0.9300
C6—C5	1.388 (2)	N1—C1	1.431 (3)
C6—C7	1.394 (2)	N1—C2	1.444 (3)
C6—C9	1.452 (2)	C2—H2B	0.9600
N4—C10	1.325 (2)	C2—H2C	0.9600
N4—C11	1.445 (2)	C2—H2D	0.9600

N4—H4A	0.8600	C11—H12A	0.9600
C3—N1	1.372 (2)	C11—H12B	0.9600
C3—C4	1.407 (2)	C11—H12C	0.9600
C3—C8	1.408 (2)	C1—H1A	0.9600
C7—C8	1.372 (2)	C1—H1B	0.9600
C7—H7A	0.9300	C1—H1C	0.9600
C9—N2—N3	115.50 (14)	C7—C8—H8A	119.4
C10—N3—N2	119.27 (14)	C3—C8—H8A	119.4
C10—N3—H3A	120.4	C5—C4—C3	120.64 (15)
N2—N3—H3A	120.4	C5—C4—H4B	119.7
C5—C6—C7	117.29 (14)	C3—C4—H4B	119.7
C5—C6—C9	119.74 (14)	C3—N1—C1	121.07 (17)
C7—C6—C9	122.92 (14)	C3—N1—C2	120.89 (18)
C10—N4—C11	124.08 (16)	C1—N1—C2	117.18 (18)
C10—N4—H4A	118.0	N1—C2—H2B	109.5
C11—N4—H4A	118.0	N1—C2—H2C	109.5
N1—C3—C4	121.59 (16)	H2B—C2—H2C	109.5
N1—C3—C8	121.31 (16)	N1—C2—H2D	109.5
C4—C3—C8	117.07 (14)	H2B—C2—H2D	109.5
C8—C7—C6	121.76 (15)	H2C—C2—H2D	109.5
C8—C7—H7A	119.1	N4—C11—H12A	109.5
C6—C7—H7A	119.1	N4—C11—H12B	109.5
N4—C10—N3	116.27 (14)	H12A—C11—H12B	109.5
N4—C10—S1	124.11 (13)	N4—C11—H12C	109.5
N3—C10—S1	119.61 (13)	H12A—C11—H12C	109.5
N2—C9—C6	121.42 (15)	H12B—C11—H12C	109.5
N2—C9—H9A	119.3	N1—C1—H1A	109.5
C6—C9—H9A	119.3	N1—C1—H1B	109.5
C4—C5—C6	121.98 (15)	H1A—C1—H1B	109.5
C4—C5—H5A	119.0	N1—C1—H1C	109.5
C6—C5—H5A	119.0	H1A—C1—H1C	109.5
C7—C8—C3	121.23 (16)	H1B—C1—H1C	109.5

Hydrogen-bond geometry (Å, °)

<i>D</i> —H \cdots <i>A</i>	<i>D</i> —H	H \cdots <i>A</i>	<i>D</i> \cdots <i>A</i>	<i>D</i> —H \cdots <i>A</i>
N4—H4A \cdots N2	0.86	2.22	2.613 (2)	108
N3—H3A \cdots S1 ⁱ	0.86	2.59	3.3890 (16)	155

Symmetry code: (i) $-x+1, -y, -z$.