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Phenyl 2,3,4-tri-*O*-benzyl-1-thio- α -D-mannopyranoside monohydrate

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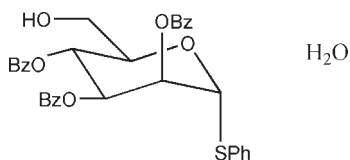
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 Key indicators: single-crystal X-ray study; $T = 150$ K; mean $\sigma(\text{C}-\text{C}) = 0.003$ Å; disorder in main residue; R factor = 0.030; wR factor = 0.062; data-to-parameter ratio = 13.0.

In the title compound, $\text{C}_{33}\text{H}_{34}\text{O}_5\text{S}\cdot\text{H}_2\text{O}$, the mannopyranoside ring adopts a chair conformation with the 2- α -thiophenyl group occupying an axial position. One of the pendant benzyl groups is disordered over two sets of sites in a 0.5:0.5 ratio. In the crystal, the water molecule makes two $\text{O}-\text{H}\cdots\text{O}$ hydrogen bonds to an adjacent sugar molecule with the O atoms of the primary alcohol and ether groups acting as acceptors. At the same time, the OH group of the sugar makes a hydrogen bond to a water molecule.

Related literature

For background to the synthesis and properties of mannopyranosides, see: Boons (1991); Szurmai *et al.* (1994); Caravano *et al.* (2003); Grizot *et al.* (2006); Dohi *et al.* (2008). For ring conformation analysis, see: Cremer & Pople (1975).



Experimental

Crystal data

 $\text{C}_{33}\text{H}_{34}\text{O}_5\text{S}\cdot\text{H}_2\text{O}$
 $M_r = 560.69$

 Monoclinic, $P2_1$
 $a = 12.628$ (1) Å

 $b = 8.084$ (1) Å

 $c = 14.832$ (2) Å

 $\beta = 101.380$ (5)°

 $V = 1484.4$ (2) Å³
 $Z = 2$

 Mo $K\alpha$ radiation

 $\mu = 0.15$ mm⁻¹
 $T = 150$ K

 $0.35 \times 0.30 \times 0.17$ mm

Data collection

 Oxford Diffraction Gemini Ruby
CCD diffractometer

 Absorption correction: multi-scan
(*CrysAlis PRO*; Oxford
Diffraction, 2009)

 $T_{\min} = 0.948$, $T_{\max} = 0.974$

 12650 measured reflections
5205 independent reflections
4333 reflections with $I > 2\sigma(I)$
 $R_{\text{int}} = 0.022$

Refinement

 $R[F^2 > 2\sigma(F^2)] = 0.030$
 $wR(F^2) = 0.062$
 $S = 0.97$

5205 reflections

400 parameters

17 restraints

 H atoms treated by a mixture of
independent and constrained
refinement

 $\Delta\rho_{\max} = 0.13$ e Å⁻³
 $\Delta\rho_{\min} = -0.17$ e Å⁻³

 Absolute structure: Flack (1983),
2381 Friedel pairs

Flack parameter: 0.01 (5)

Table 1

Hydrogen-bond geometry (Å, °).

$D-H\cdots A$	$D-H$	$H\cdots A$	$D\cdots A$	$D-H\cdots A$
OW—HW1 \cdots O2	0.88 (3)	2.00 (3)	2.861 (2)	166 (2)
OW—HW2 \cdots O6	0.78 (2)	2.07 (2)	2.827 (2)	165 (2)
O6—H6 \cdots OW ⁱ	0.87 (2)	1.89 (2)	2.745 (2)	169 (2)

 Symmetry code: (i) $-x + 1, y - \frac{1}{2}, -z + 2$.

Data collection: *CrysAlis CCD* (Oxford Diffraction, 2009); cell refinement: *CrysAlis RED* (Oxford Diffraction, 2009); data reduction: *CrysAlis RED*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 2008); program(s) used to refine structure: *SHELXL97* (Sheldrick, 2008); molecular graphics: *PLATON* (Spek, 2009); software used to prepare material for publication: *PLATON*.

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Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: HB5424).

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supporting information

Acta Cryst. (2010). E66, o1525 [doi:10.1107/S1600536810019604]

Phenyl 2,3,4-tri-*O*-benzyl-1-thio- α -D-mannopyranoside monohydrate

Maxime Durka, Bernadette Norberg, Yvain Roué, Stéphane P. Vincent and Johan Wouters

S1. Comment

Oligosaccharides are natural ligands of lectins, which are important receptors involved in key biological processes. Efficient strategies for the regioselective transformations of monosaccharides to achieve glycosylation reactions or further functionalizations are of paramount importance. Furthermore, the knowledge of the conformations of carbohydrates alone or bound to protein gives access to key informations that can be exploited to understand biocatalytic processes or stereoelectronic effects (Caravano *et al.*, 2003).

Numerous strategies of protections of mannopyranosides have been described in the litterature (*e.g.* Boons, 1991; Szirmai *et al.*, 1994) in order to derivatise the primary alcohol. The title compound is also a key intermediate for the synthesis of heptosyl transferase inhibitors (Dohi *et al.*, 2008; Grizot *et al.*, 2006).

Crystal structure of the title compound confirms the expected relative stereochemistry : C1 *R*, C2 *S*, C3 *S*, C4 *R*, C5 *R*. The mannopyranoside adopts a chair conformation with puckering amplitude (Q) = 0.522 (3) Å, Theta = 3.6 (3) °, and Phi = 60 (5) ° (Cremer & Pople, 1975).

The 2- α -thiophenyl group on C1 and the *O*-benzyl group on C2 occupy an axial position, the two other *O*-benzyl groups (on C3 and C4) and carbon atom C6 occupying an equatorial position.

Thiophenyl-2,3,4-*O*-tri-benzyl- α -D-mannopyranoside co-crystallized with one water molecule (O_w). This water molecule is part of a H bond network involving the primary alcohol O6 and also secondary alcohol O2 (Table 1). Packing is further reinforced by van der Waals interactions involving the aromatic rings.

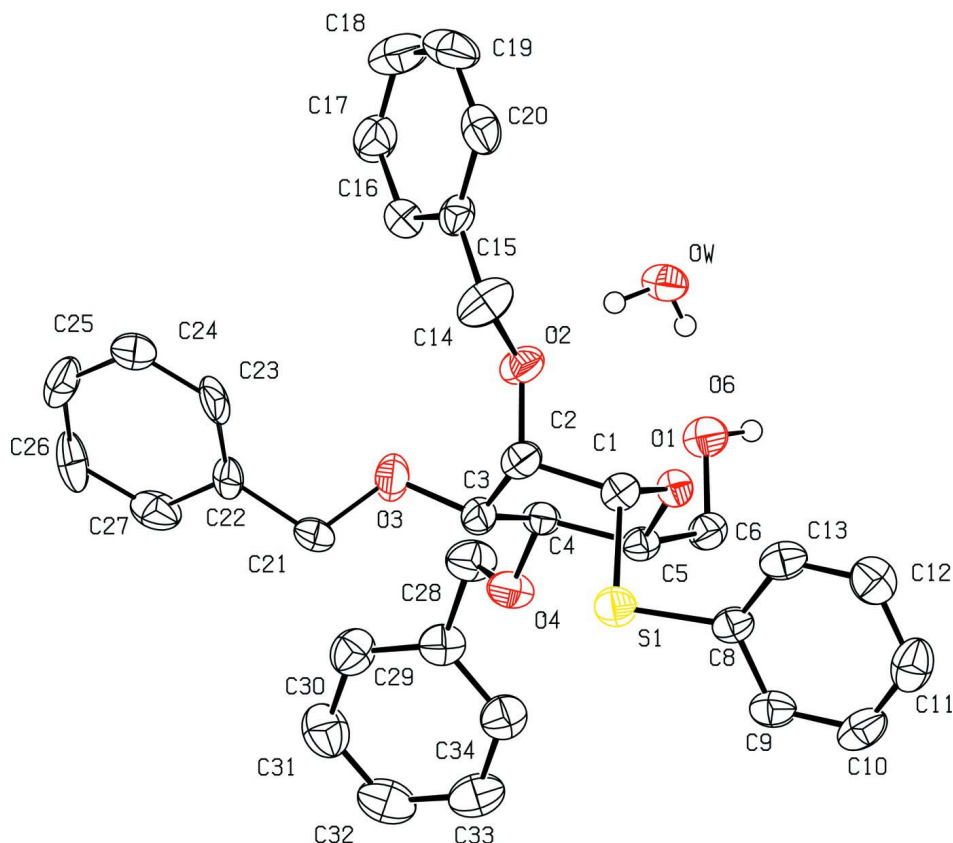
S2. Experimental

The title compound was obtained by a six-step synthetic route that will be described in details elsewhere. Overall yield is 70%. Colourless prisms of (I) were obtained by evaporation of a solution in ethyl acetate.

S3. Refinement

Disorder of the benzyl C(21) > C(27) moiety substituting oxygen O(4) was included in the refinement (0.5 occupancy for both parts that were restrained to have similar bond lengths and angles). H atoms on water oxygen atom O_w and on alcohol oxygen atom O(6) were located by Fourier difference maps and allowed to ride on their parent O atoms.

All other H atoms were placed at idealized positions and allowed to ride on their parent atoms, with C—H = 0.97 Å and U_{iso}(H) = 1.2U_{eq}(C) for methylene groups, C—H = 0.93 Å and U_{iso}(H) = 1.2U_{eq}(C) for aromatic carbons, and C—H = 0.96 Å and U_{iso}(H) = 1.5U_{eq}(C) for the methyl group.

**Figure 1**

The molecular structure of (I): C-bound H atoms are omitted and disorder is not presented for clarity. Displacement ellipsoids for non-H atoms are drawn at the 50% probability level.

Phenyl 2,3,4-tri-O-benzyl-1-thio- α -D-mannopyranoside monohydrate

Crystal data

$C_{33}H_{34}O_5S \cdot H_2O$
 $M_r = 560.69$
 Monoclinic, $P2_1$
 Hall symbol: P 2yb
 $a = 12.628$ (1) Å
 $b = 8.084$ (1) Å
 $c = 14.832$ (2) Å
 $\beta = 101.380$ (5)°
 $V = 1484.4$ (2) Å³
 $Z = 2$

$F(000) = 596.0$
 $D_x = 1.255$ Mg m⁻³
 Mo $K\alpha$ radiation, $\lambda = 0.71073$ Å
 Cell parameters from 7534 reflections
 $\theta = 3.0$ – 28.2 °
 $\mu = 0.15$ mm⁻¹
 $T = 150$ K
 Prism, colourless
 $0.35 \times 0.30 \times 0.17$ mm

Data collection

Oxford Diffraction Gemini Ruby CCD
 diffractometer
 Radiation source: Enhanced fine-focus sealed
 tube
 Graphite monochromator
 ω scans

Absorption correction: multi-scan
 (*CrysAlis PRO*; Oxford Diffraction, 2009)
 $T_{\min} = 0.948$, $T_{\max} = 0.974$
 12650 measured reflections
 5205 independent reflections
 4333 reflections with $I > 2\sigma(I)$
 $R_{\text{int}} = 0.022$

$\theta_{\max} = 25.0^\circ$, $\theta_{\min} = 3.0^\circ$
 $h = -15 \rightarrow 14$

$k = -9 \rightarrow 9$
 $l = -17 \rightarrow 17$

Refinement

Refinement on F^2
 Least-squares matrix: full
 $R[F^2 > 2\sigma(F^2)] = 0.030$
 $wR(F^2) = 0.062$
 $S = 0.97$
 5205 reflections
 400 parameters
 17 restraints
 Primary atom site location: structure-invariant
 direct methods
 Secondary atom site location: difference Fourier
 map

Hydrogen site location: inferred from
 neighbouring sites
 H atoms treated by a mixture of independent
 and constrained refinement
 $w = 1/[\sigma^2(F_o^2) + (0.0359P)^2]$
 where $P = (F_o^2 + 2F_c^2)/3$
 $(\Delta/\sigma)_{\max} = 0.001$
 $\Delta\rho_{\max} = 0.13 \text{ e } \text{\AA}^{-3}$
 $\Delta\rho_{\min} = -0.17 \text{ e } \text{\AA}^{-3}$
 Absolute structure: Flack (1983), 2381 Friedel
 pairs
 Absolute structure parameter: 0.01 (5)

Special details

Geometry. All esds (except the esd in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell esds are taken into account individually in the estimation of esds in distances, angles and torsion angles; correlations between esds in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell esds is used for estimating esds involving l.s. planes.

Refinement. Refinement of F^2 against ALL reflections. The weighted R -factor wR and goodness of fit S are based on F^2 , conventional R -factors R are based on F , with F set to zero for negative F^2 . The threshold expression of $F^2 > \sigma(F^2)$ is used only for calculating R -factors(gt) etc. and is not relevant to the choice of reflections for refinement. R -factors based on F^2 are statistically about twice as large as those based on F , and R -factors based on ALL data will be even larger.

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (\AA^2)

	<i>x</i>	<i>y</i>	<i>z</i>	$U_{\text{iso}}^*/U_{\text{eq}}$	Occ. (<1)
C1	0.28677 (14)	0.7097 (2)	0.89516 (11)	0.0300 (4)	
H1	0.3281	0.7834	0.9416	0.036*	
C2	0.32672 (13)	0.7370 (2)	0.80557 (11)	0.0334 (4)	
H2	0.3072	0.8487	0.7825	0.040*	
C3	0.27664 (14)	0.6105 (2)	0.73342 (11)	0.0337 (4)	
H3	0.1995	0.6349	0.7142	0.040*	
C4	0.28925 (14)	0.4357 (2)	0.77082 (10)	0.0297 (4)	
H4	0.3641	0.3989	0.7759	0.036*	
C5	0.25565 (14)	0.4210 (2)	0.86404 (10)	0.0276 (4)	
H5	0.1771	0.4349	0.8547	0.033*	
C6	0.28498 (13)	0.2565 (2)	0.90995 (11)	0.0323 (4)	
H6A	0.2635	0.2547	0.9692	0.039*	
H6B	0.2466	0.1688	0.8725	0.039*	
C8	0.11808 (13)	0.7620 (2)	0.98996 (11)	0.0308 (4)	
C9	0.02534 (15)	0.6816 (2)	1.00193 (13)	0.0370 (5)	
H9	-0.0181	0.6284	0.9525	0.044*	
C10	-0.00307 (16)	0.6802 (2)	1.08781 (14)	0.0448 (5)	
H10	-0.0665	0.6283	1.0954	0.054*	
C11	0.06208 (17)	0.7552 (3)	1.16149 (13)	0.0489 (5)	
H11	0.0431	0.7537	1.2190	0.059*	
C12	0.15545 (18)	0.8325 (3)	1.14995 (13)	0.0453 (5)	

H12	0.2004	0.8810	1.2002	0.054*	
C13	0.18316 (15)	0.8388 (2)	1.06442 (13)	0.0378 (5)	
H13	0.2453	0.8945	1.0568	0.045*	
C14	0.50097 (18)	0.8655 (2)	0.81855 (16)	0.0507 (6)	
H14A	0.5083	0.9246	0.8764	0.061*	
H14B	0.4635	0.9367	0.7699	0.061*	
C15	0.61013 (15)	0.8228 (2)	0.80096 (11)	0.0322 (4)	
C16	0.62073 (15)	0.7141 (2)	0.73104 (11)	0.0367 (4)	
H16	0.5595	0.6650	0.6961	0.044*	
C17	0.72045 (17)	0.6782 (3)	0.71287 (13)	0.0478 (5)	
H17	0.7264	0.6048	0.6659	0.057*	
C18	0.81128 (17)	0.7497 (3)	0.76340 (17)	0.0603 (6)	
H18	0.8787	0.7253	0.7505	0.072*	
C19	0.80278 (18)	0.8570 (3)	0.83291 (16)	0.0578 (7)	
H19	0.8647	0.9052	0.8673	0.069*	
C20	0.70225 (17)	0.8946 (2)	0.85263 (12)	0.0434 (5)	
H20	0.6968	0.9673	0.9001	0.052*	
O3	0.32582 (10)	0.61829 (18)	0.65476 (8)	0.0464 (4)	
C21	0.2578 (3)	0.6608 (6)	0.5714 (2)	0.0441 (11)	0.50
H21A	0.2135	0.7550	0.5805	0.053*	0.50
H21B	0.2102	0.5689	0.5495	0.053*	0.50
C22	0.3247 (5)	0.7026 (7)	0.5018 (4)	0.0280 (18)	0.50
C23	0.4047 (6)	0.8186 (9)	0.5201 (3)	0.043 (2)	0.50
H23	0.4207	0.8676	0.5779	0.051*	0.50
C24	0.4617 (5)	0.8638 (7)	0.4547 (5)	0.0407 (15)	0.50
H24	0.5178	0.9396	0.4692	0.049*	0.50
C25	0.4364 (5)	0.7974 (10)	0.3673 (5)	0.0456 (16)	0.50
H25	0.4748	0.8301	0.3229	0.055*	0.50
C26	0.3545 (7)	0.6830 (9)	0.3456 (3)	0.0481 (19)	0.50
H26	0.3372	0.6388	0.2866	0.058*	0.50
C27	0.2978 (6)	0.6338 (10)	0.4127 (4)	0.044 (2)	0.50
H27	0.2427	0.5562	0.3988	0.053*	0.50
C21'	0.2980 (4)	0.7526 (6)	0.5997 (2)	0.0413 (7)	0.50
H21C	0.3253	0.8515	0.6334	0.050*	0.50
H21D	0.2199	0.7608	0.5844	0.050*	0.50
C22'	0.3411 (10)	0.7456 (13)	0.5118 (7)	0.0413 (7)	0.50
C23'	0.4273 (9)	0.8369 (12)	0.4994 (6)	0.0413 (7)	0.50
H23'	0.4617	0.9043	0.5471	0.050*	0.50
C24'	0.4653 (6)	0.8333 (8)	0.4192 (7)	0.0413 (7)	0.50
H24'	0.5225	0.9015	0.4122	0.050*	0.50
C25'	0.4206 (7)	0.7315 (9)	0.3503 (6)	0.0413 (7)	0.50
H25'	0.4465	0.7287	0.2958	0.050*	0.50
C26'	0.3361 (8)	0.6316 (8)	0.3617 (6)	0.0413 (7)	0.50
H26'	0.3052	0.5596	0.3149	0.050*	0.50
C27'	0.2970 (8)	0.6377 (14)	0.4420 (8)	0.0413 (7)	0.50
H27'	0.2405	0.5687	0.4494	0.050*	0.50
C28	0.26726 (15)	0.2153 (3)	0.66005 (13)	0.0451 (5)	
H28A	0.3146	0.1422	0.7016	0.054*	

H28B	0.3101	0.2731	0.6224	0.054*
C29	0.17992 (15)	0.1171 (2)	0.60016 (12)	0.0376 (5)
C30	0.17758 (17)	0.1035 (3)	0.50676 (13)	0.0445 (5)
H30	0.2310	0.1538	0.4812	0.053*
C31	0.09579 (19)	0.0151 (3)	0.45124 (14)	0.0540 (6)
H31	0.0945	0.0064	0.3885	0.065*
C32	0.01714 (19)	-0.0592 (3)	0.48815 (15)	0.0570 (6)
H32	-0.0373	-0.1189	0.4506	0.068*
C33	0.01817 (17)	-0.0461 (3)	0.58167 (15)	0.0516 (6)
H33	-0.0359	-0.0957	0.6067	0.062*
C34	0.09947 (16)	0.0405 (3)	0.63681 (13)	0.0430 (5)
H34	0.1007	0.0480	0.6996	0.052*
O1	0.30602 (9)	0.54559 (13)	0.92705 (7)	0.0281 (3)
O2	0.44079 (9)	0.71853 (15)	0.82204 (8)	0.0383 (3)
O4	0.21835 (9)	0.32927 (16)	0.71031 (8)	0.0385 (3)
O6	0.39918 (10)	0.22905 (16)	0.92224 (9)	0.0376 (3)
OW	0.55620 (12)	0.4829 (2)	0.94836 (10)	0.0379 (3)
S1	0.14464 (4)	0.77358 (5)	0.87681 (3)	0.03636 (13)
HW1	0.528 (2)	0.551 (4)	0.9035 (18)	0.084 (9)*
HW2	0.5117 (18)	0.417 (3)	0.9508 (13)	0.047 (7)*
H6	0.4137 (16)	0.142 (3)	0.9571 (13)	0.050 (6)*

Atomic displacement parameters (Å²)

	U^{11}	U^{22}	U^{33}	U^{12}	U^{13}	U^{23}
C1	0.0308 (10)	0.0249 (9)	0.0353 (10)	0.0021 (8)	0.0090 (8)	0.0008 (7)
C2	0.0287 (10)	0.0296 (10)	0.0436 (10)	0.0051 (8)	0.0111 (8)	0.0099 (8)
C3	0.0279 (10)	0.0449 (11)	0.0295 (9)	0.0067 (9)	0.0089 (8)	0.0063 (8)
C4	0.0231 (10)	0.0371 (11)	0.0284 (9)	0.0003 (8)	0.0037 (7)	-0.0026 (8)
C5	0.0255 (10)	0.0274 (10)	0.0303 (9)	-0.0021 (8)	0.0061 (7)	-0.0032 (8)
C6	0.0313 (10)	0.0333 (10)	0.0333 (8)	-0.0040 (9)	0.0088 (7)	-0.0021 (8)
C8	0.0266 (10)	0.0275 (9)	0.0394 (9)	0.0041 (8)	0.0093 (7)	-0.0005 (9)
C9	0.0277 (11)	0.0346 (11)	0.0477 (11)	0.0026 (8)	0.0048 (8)	0.0003 (8)
C10	0.0356 (11)	0.0380 (12)	0.0662 (14)	0.0049 (9)	0.0232 (10)	0.0098 (10)
C11	0.0597 (14)	0.0457 (12)	0.0468 (11)	0.0112 (12)	0.0238 (10)	0.0057 (11)
C12	0.0497 (13)	0.0427 (12)	0.0427 (11)	0.0021 (10)	0.0073 (9)	-0.0073 (9)
C13	0.0318 (11)	0.0327 (10)	0.0501 (12)	0.0001 (8)	0.0105 (9)	-0.0028 (9)
C14	0.0534 (14)	0.0313 (12)	0.0750 (14)	-0.0061 (10)	0.0314 (11)	-0.0078 (10)
C15	0.0386 (11)	0.0282 (10)	0.0313 (9)	-0.0064 (8)	0.0103 (8)	0.0023 (8)
C16	0.0331 (11)	0.0442 (11)	0.0312 (9)	-0.0001 (9)	0.0024 (8)	0.0017 (8)
C17	0.0458 (13)	0.0565 (14)	0.0441 (11)	0.0070 (11)	0.0165 (10)	0.0045 (10)
C18	0.0344 (13)	0.0552 (15)	0.0942 (17)	0.0017 (12)	0.0198 (12)	0.0163 (14)
C19	0.0390 (14)	0.0482 (14)	0.0751 (16)	-0.0134 (11)	-0.0160 (12)	0.0203 (12)
C20	0.0600 (15)	0.0312 (10)	0.0345 (10)	-0.0123 (10)	-0.0011 (10)	0.0050 (8)
O3	0.0426 (8)	0.0664 (9)	0.0334 (7)	0.0167 (7)	0.0157 (6)	0.0184 (7)
C21	0.031 (2)	0.065 (3)	0.032 (2)	-0.020 (2)	-0.0058 (17)	0.002 (2)
C22	0.036 (3)	0.022 (4)	0.025 (3)	0.011 (3)	0.007 (3)	0.010 (3)
C23	0.049 (5)	0.065 (4)	0.011 (2)	0.006 (4)	0.001 (2)	-0.004 (2)

C24	0.033 (3)	0.046 (3)	0.041 (4)	-0.012 (2)	0.000 (3)	-0.004 (3)
C25	0.052 (4)	0.052 (4)	0.041 (4)	0.009 (3)	0.030 (3)	0.010 (3)
C26	0.080 (5)	0.047 (4)	0.017 (2)	0.012 (4)	0.008 (3)	-0.002 (2)
C27	0.038 (3)	0.036 (3)	0.052 (6)	-0.011 (2)	-0.007 (4)	-0.002 (4)
C21'	0.0536 (15)	0.0354 (13)	0.0370 (14)	0.0045 (11)	0.0144 (11)	0.0001 (11)
C22'	0.0536 (15)	0.0354 (13)	0.0370 (14)	0.0045 (11)	0.0144 (11)	0.0001 (11)
C23'	0.0536 (15)	0.0354 (13)	0.0370 (14)	0.0045 (11)	0.0144 (11)	0.0001 (11)
C24'	0.0536 (15)	0.0354 (13)	0.0370 (14)	0.0045 (11)	0.0144 (11)	0.0001 (11)
C25'	0.0536 (15)	0.0354 (13)	0.0370 (14)	0.0045 (11)	0.0144 (11)	0.0001 (11)
C26'	0.0536 (15)	0.0354 (13)	0.0370 (14)	0.0045 (11)	0.0144 (11)	0.0001 (11)
C27'	0.0536 (15)	0.0354 (13)	0.0370 (14)	0.0045 (11)	0.0144 (11)	0.0001 (11)
C28	0.0364 (11)	0.0497 (13)	0.0494 (11)	0.0013 (10)	0.0092 (9)	-0.0137 (10)
C29	0.0362 (11)	0.0329 (10)	0.0443 (12)	0.0027 (9)	0.0097 (9)	-0.0077 (9)
C30	0.0488 (13)	0.0425 (12)	0.0447 (12)	-0.0036 (10)	0.0152 (10)	-0.0046 (9)
C31	0.0657 (15)	0.0514 (14)	0.0423 (12)	-0.0035 (12)	0.0040 (11)	-0.0069 (10)
C32	0.0486 (15)	0.0532 (15)	0.0629 (14)	-0.0103 (11)	-0.0046 (12)	-0.0147 (11)
C33	0.0439 (14)	0.0417 (13)	0.0709 (15)	-0.0084 (10)	0.0157 (11)	-0.0053 (11)
C34	0.0427 (12)	0.0416 (11)	0.0464 (11)	0.0025 (10)	0.0124 (9)	-0.0043 (9)
O1	0.0309 (7)	0.0244 (6)	0.0286 (6)	-0.0002 (5)	0.0054 (5)	0.0000 (5)
O2	0.0307 (7)	0.0340 (7)	0.0521 (7)	-0.0014 (6)	0.0127 (6)	0.0108 (6)
O4	0.0283 (7)	0.0484 (8)	0.0378 (7)	0.0024 (6)	0.0042 (5)	-0.0151 (6)
O6	0.0350 (8)	0.0285 (8)	0.0494 (7)	0.0019 (6)	0.0082 (6)	0.0079 (6)
OW	0.0299 (8)	0.0340 (8)	0.0469 (8)	-0.0020 (7)	0.0005 (6)	0.0019 (6)
S1	0.0316 (3)	0.0396 (3)	0.0383 (2)	0.0082 (2)	0.00785 (19)	0.0010 (2)

Geometric parameters (Å, °)

C1—O1	1.414 (2)	O3—C21	1.403 (3)
C1—C2	1.528 (2)	C21—C22	1.496 (6)
C1—S1	1.8355 (17)	C21—H21A	0.9700
C1—H1	0.9800	C21—H21B	0.9700
C2—O2	1.421 (2)	C22—C23	1.366 (8)
C2—C3	1.524 (2)	C22—C27	1.412 (7)
C2—H2	0.9800	C23—C24	1.366 (7)
C3—O3	1.427 (2)	C23—H23	0.9300
C3—C4	1.515 (3)	C24—C25	1.382 (7)
C3—H3	0.9800	C24—H24	0.9300
C4—O4	1.425 (2)	C25—C26	1.377 (8)
C4—C5	1.529 (2)	C25—H25	0.9300
C4—H4	0.9800	C26—C27	1.393 (8)
C5—O1	1.4347 (18)	C26—H26	0.9300
C5—C6	1.506 (2)	C27—H27	0.9300
C5—H5	0.9800	C21'—C22'	1.510 (8)
C6—O6	1.435 (2)	C21'—H21C	0.9700
C6—H6A	0.9700	C21'—H21D	0.9700
C6—H6B	0.9700	C22'—C23'	1.356 (10)
C8—C9	1.381 (2)	C22'—C27'	1.384 (10)
C8—C13	1.386 (2)	C23'—C24'	1.370 (9)

C8—S1	1.7777 (17)	C23'—H23'	0.9300
C9—C10	1.390 (3)	C24'—C25'	1.346 (9)
C9—H9	0.9300	C24'—H24'	0.9300
C10—C11	1.372 (3)	C25'—C26'	1.375 (9)
C10—H10	0.9300	C25'—H25'	0.9300
C11—C12	1.375 (3)	C26'—C27'	1.377 (9)
C11—H11	0.9300	C26'—H26'	0.9300
C12—C13	1.382 (3)	C27'—H27'	0.9300
C12—H12	0.9300	C28—O4	1.403 (2)
C13—H13	0.9300	C28—C29	1.500 (3)
C14—O2	1.417 (2)	C28—H28A	0.9700
C14—C15	1.493 (3)	C28—H28B	0.9700
C14—H14A	0.9700	C29—C30	1.384 (3)
C14—H14B	0.9700	C29—C34	1.389 (3)
C15—C16	1.386 (2)	C30—C31	1.386 (3)
C15—C20	1.388 (3)	C30—H30	0.9300
C16—C17	1.370 (3)	C31—C32	1.365 (3)
C16—H16	0.9300	C31—H31	0.9300
C17—C18	1.368 (3)	C32—C33	1.389 (3)
C17—H17	0.9300	C32—H32	0.9300
C18—C19	1.368 (3)	C33—C34	1.372 (3)
C18—H18	0.9300	C33—H33	0.9300
C19—C20	1.392 (3)	C34—H34	0.9300
C19—H19	0.9300	O6—H6	0.87 (2)
C20—H20	0.9300	OW—HW1	0.88 (3)
O3—C21'	1.362 (4)	OW—HW2	0.78 (2)
O1—C1—C2	111.21 (14)	C21—O3—C3	116.2 (2)
O1—C1—S1	114.34 (12)	O3—C21—C22	109.4 (4)
C2—C1—S1	108.15 (11)	O3—C21—H21A	109.8
O1—C1—H1	107.6	C22—C21—H21A	109.8
C2—C1—H1	107.6	O3—C21—H21B	109.8
S1—C1—H1	107.6	C22—C21—H21B	109.8
O2—C2—C3	108.46 (14)	H21A—C21—H21B	108.2
O2—C2—C1	109.45 (13)	C23—C22—C27	119.2 (6)
C3—C2—C1	110.59 (14)	C23—C22—C21	120.9 (5)
O2—C2—H2	109.4	C27—C22—C21	119.5 (5)
C3—C2—H2	109.4	C22—C23—C24	121.0 (5)
C1—C2—H2	109.4	C22—C23—H23	119.5
O3—C3—C4	108.00 (14)	C24—C23—H23	119.5
O3—C3—C2	110.94 (15)	C23—C24—C25	120.4 (4)
C4—C3—C2	111.68 (13)	C23—C24—H24	119.8
O3—C3—H3	108.7	C25—C24—H24	119.8
C4—C3—H3	108.7	C26—C25—C24	120.3 (5)
C2—C3—H3	108.7	C26—C25—H25	119.9
O4—C4—C3	108.98 (12)	C24—C25—H25	119.9
O4—C4—C5	105.68 (14)	C25—C26—C27	119.5 (5)
C3—C4—C5	112.09 (14)	C25—C26—H26	120.2

O4—C4—H4	110.0	C27—C26—H26	120.2
C3—C4—H4	110.0	C26—C27—C22	119.6 (6)
C5—C4—H4	110.0	C26—C27—H27	120.2
O1—C5—C6	106.70 (11)	C22—C27—H27	120.2
O1—C5—C4	111.58 (13)	O3—C21'—C22'	113.2 (5)
C6—C5—C4	112.98 (14)	O3—C21'—H21C	108.9
O1—C5—H5	108.5	C22'—C21'—H21C	108.9
C6—C5—H5	108.5	O3—C21'—H21D	108.9
C4—C5—H5	108.5	C22'—C21'—H21D	108.9
O6—C6—C5	110.21 (14)	H21C—C21'—H21D	107.8
O6—C6—H6A	109.6	C23'—C22'—C27'	117.2 (8)
C5—C6—H6A	109.6	C23'—C22'—C21'	122.1 (10)
O6—C6—H6B	109.6	C27'—C22'—C21'	120.6 (9)
C5—C6—H6B	109.6	C22'—C23'—C24'	122.3 (7)
H6A—C6—H6B	108.1	C22'—C23'—H23'	118.9
C9—C8—C13	119.53 (16)	C24'—C23'—H23'	118.9
C9—C8—S1	117.95 (13)	C25'—C24'—C23'	120.4 (7)
C13—C8—S1	122.40 (14)	C25'—C24'—H24'	119.8
C8—C9—C10	120.02 (17)	C23'—C24'—H24'	119.8
C8—C9—H9	120.0	C24'—C25'—C26'	119.0 (7)
C10—C9—H9	120.0	C24'—C25'—H25'	120.5
C11—C10—C9	120.23 (19)	C26'—C25'—H25'	120.5
C11—C10—H10	119.9	C25'—C26'—C27'	120.3 (8)
C9—C10—H10	119.9	C25'—C26'—H26'	119.9
C10—C11—C12	119.72 (18)	C27'—C26'—H26'	119.9
C10—C11—H11	120.1	C26'—C27'—C22'	120.7 (9)
C12—C11—H11	120.1	C26'—C27'—H27'	119.7
C11—C12—C13	120.66 (17)	C22'—C27'—H27'	119.7
C11—C12—H12	119.7	O4—C28—C29	108.24 (15)
C13—C12—H12	119.7	O4—C28—H28A	110.0
C12—C13—C8	119.79 (18)	C29—C28—H28A	110.0
C12—C13—H13	120.1	O4—C28—H28B	110.0
C8—C13—H13	120.1	C29—C28—H28B	110.0
O2—C14—C15	109.44 (16)	H28A—C28—H28B	108.4
O2—C14—H14A	109.8	C30—C29—C34	118.92 (17)
C15—C14—H14A	109.8	C30—C29—C28	120.18 (18)
O2—C14—H14B	109.8	C34—C29—C28	120.90 (17)
C15—C14—H14B	109.8	C29—C30—C31	120.1 (2)
H14A—C14—H14B	108.2	C29—C30—H30	119.9
C16—C15—C20	118.96 (19)	C31—C30—H30	119.9
C16—C15—C14	120.41 (16)	C32—C31—C30	120.32 (19)
C20—C15—C14	120.61 (18)	C32—C31—H31	119.8
C17—C16—C15	120.65 (18)	C30—C31—H31	119.8
C17—C16—H16	119.7	C31—C32—C33	120.20 (19)
C15—C16—H16	119.7	C31—C32—H32	119.9
C18—C17—C16	120.4 (2)	C33—C32—H32	119.9
C18—C17—H17	119.8	C34—C33—C32	119.5 (2)
C16—C17—H17	119.8	C34—C33—H33	120.2

C19—C18—C17	119.9 (2)	C32—C33—H33	120.2
C19—C18—H18	120.0	C33—C34—C29	120.89 (18)
C17—C18—H18	120.0	C33—C34—H34	119.6
C18—C19—C20	120.51 (19)	C29—C34—H34	119.6
C18—C19—H19	119.7	C1—O1—C5	114.61 (11)
C20—C19—H19	119.7	C14—O2—C2	116.02 (14)
C15—C20—C19	119.52 (19)	C28—O4—C4	116.34 (13)
C15—C20—H20	120.2	C6—O6—H6	106.8 (14)
C19—C20—H20	120.2	HW1—OW—HW2	106 (2)
C21'—O3—C21	40.1 (2)	C8—S1—C1	101.91 (8)
C21'—O3—C3	115.02 (19)		

Hydrogen-bond geometry (Å, °)

<i>D</i> —H \cdots <i>A</i>	<i>D</i> —H	H \cdots <i>A</i>	<i>D</i> \cdots <i>A</i>	<i>D</i> —H \cdots <i>A</i>
OW—HW1 \cdots O2	0.88 (3)	2.00 (3)	2.861 (2)	166 (2)
OW—HW2 \cdots O6	0.78 (2)	2.07 (2)	2.827 (2)	165 (2)
O6—H6 \cdots OW ⁱ	0.87 (2)	1.89 (2)	2.745 (2)	169 (2)

Symmetry code: (i) $-x+1, y-1/2, -z+2$.