Acta Crystallographica Section E
Structure Reports Online
ISSN 1600-5368

Sodium samarium tetrakis(polyphosphate), $NaSm(PO_3)_4$

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Received 2 June 2010; accepted 12 June 2010

Key indicators: single-crystal X-ray study; T = 293 K; mean σ(P–O) = 0.004 Å; R factor = 0.037; wR factor = 0.068; data-to-parameter ratio = 13.0.

$NaSm(PO_3)_4$ has been prepared by solid state reactions. It belongs to type II of the structural family of $M^{I}Ln^{III}(PO_3)_4$ compounds (M^{I} = alkali metal and Ln^{III} = rare earth metal) and is composed of $_{\infty}(PO_3)_n]^{n-}$ polyphosphate chains with a repeating unit of four PO_4 tetrahedra. The chains extend parallel to [100] and share O atoms with irregular SmO_8 polyhedra, forming a three-dimensional framework which delimits tunnels occupied by Na^+ cations in a distorted octahedral environment.

Related literature

The unit cell of $NaSm(PO_3)_4$, derived from X-ray powder data, was reported by Ferid *et al.* (1984). For classification of $M^{I}Ln^{III}(PO_3)_4$ structures, see: Palkina *et al.* (1981); Durif (1995). Structures, properties and applications of other members of this family were discussed by Ettis *et al.* (2003); Parreu *et al.* (2007); Zhao *et al.* (2008, 2010); Zhu *et al.* (2009).

Experimental

Crystal data

$NaSm(PO_3)_4$
M_r = 489.22
Monoclinic, $P2_1/n$
a = 7.1924 (13) Å
b = 13.091 (2) Å
c = 9.8480 (17) Å
β = 90.396 (10)°
V = 927.2 (3) Å^3
Z = 4
Mo $K\alpha$ radiation
μ = 7.14 mm^{-1}
T = 293 K
0.20 × 0.02 × 0.02 mm

Data collection

Rigaku Mercury70 CCD diffractometer
Absorption correction: multi-scan (*ABSCOR*; Higashi, 1995)
T_{min} = 0.329, T_{max} = 0.870
7036 measured reflections
2111 independent reflections
1868 reflections with $I > 2\sigma(I)$
R_{int} = 0.041

Refinement

$R[F^2 > 2\sigma(F^2)]$ = 0.037
$wR(F^2)$ = 0.068
S = 1.15
2111 reflections
163 parameters
$\Delta\rho_{max}$ = 2.38 e Å^{-3}
$\Delta\rho_{min}$ = −0.80 e Å^{-3}

Table 1
Selected bond lengths (Å).

P1—O10	1.482 (4)	P3—O1	1.482 (4)
P1—O5	1.494 (4)	P3—O12	1.486 (4)
P1—O11	1.584 (4)	P3—O6	1.591 (4)
P1—O7	1.586 (4)	P3—O3	1.596 (4)
P2—O9	1.477 (4)	P4—O8	1.479 (4)
P2—O4	1.481 (4)	P4—O2	1.487 (4)
P2—O7^{i}	1.574 (4)	P4—O11	1.594 (4)
P2—O6^{ii}	1.598 (4)	P4—O3^{i}	1.595 (4)

Symmetry codes: (i) $-x+1, -y+1, -z+1$; (ii) $x+1, y, z$.

Data collection: *CrystalClear* (Rigaku, 2004); cell refinement: *CrystalClear*; data reduction: *CrystalClear*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 2008) and *PLATON* (Spek, 2009); program(s) used to refine structure: *SHELXL97* (Sheldrick, 2008); molecular graphics: *DIAMOND* (Brandenburg, 2004); software used to prepare material for publication: *SHELXTL* (Sheldrick, 2008).

The authors acknowledge the Doctoral Foundation of Henan Polytechnic University (B2010–92, 648483).

Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: WM2360).

References

Brandenburg, K. (2004). *DIAMOND*. Crystal Impact GbR, Bonn, Germany.
Durif, A. (1995). *Crystal Chemistry of Condensed Phosphates*. New York and London: Plenum Press.
Ettis, H., Naili, H. & Mhiri, T. (2003). *Cryst. Growth Des.* **3**, 599–602.
Ferid, M., Dogguy, M., Kbir-Ariguib, N. & Trabelsi, M. (1984). *J. Solid State Chem.* **53**, 149–154.
Higashi, T. (1995). *ABSCOR*. Rigaku Corporation, Tokyo, Japan.
Palkina, K. K., Chudinova, N. N., Litvin, B. N. & Vinogradova, N. V. (1981). *Izv. Akad. Nauk SSSR Neorg. Mater.* **17**, 1501–1503.
Parreu, I., Sole, R., Massons, J., Diaz, F. & Aguilo, M. (2007). *Cryst. Growth Des.* **7**, 557–563.
Rigaku (2004). *CrystalClear*. Rigaku Corporation, Tokyo, Japan.
Sheldrick, G. M. (2008). *Acta Cryst.* A**64**, 112–122.
Spek, A. L. (2009). *Acta Cryst.* D**65**, 148–155.
Zhao, D., Li, F., Cheng, W. & Zhang, H. (2010). *Acta Cryst.* E**66**, i3.
Zhao, D., Zhang, H., Huang, S. P., Fang, M., Zhang, W. L., Yang, S. L. & Cheng, W. D. (2008). *J. Mol. Struct.* **892**, 8–12.
Zhu, J., Cheng, W.-D. & Zhang, H. (2009). *Acta Cryst.* E**65**, i70.

supporting information

Acta Cryst. (2010). E**66**, i53 [doi:10.1107/S1600536810022543]

Sodium samarium tetrakis(polyphosphate), $NaSm(PO_3)_4$

Dan Zhao, Lina Zhang and Feifei Li

S1. Comment

In recent years, alkali rare earth polyphosphates with general formula $M^{I}Ln^{III}(PO_3)_4$ (M^{I} = alkali metal, Ln^{III} = rare earth metal) have been studied mainly due to their rich structural chemistry and interesting physical and chemical properties, such as high luminescence efficiency (Durif, 1995). The nomenclature of $M^{I}Ln^{III}(PO_3)_4$ compounds has been proposed by Palkina *et al.* (1981) and is generally accepted today. Many compounds of this family have been reported, for example, Li*Ln*$(PO_3)_4$ (*Ln* =Y, Sm, Dy) (Zhao *et al.*, 2008, 2010), $KGdP_4O_{12}$ (Ettis *et al.*, 2003), $KNd(PO_3)_4$ (Parreu *et al.*, 2007) and $CsEu(PO_3)_4$ (Zhu *et al.*, 2009). For the compound $NaSm(PO_3)_4$ only unit-cell parameters have been reported on basis of refined X-ray powder diffraction data (Ferid *et al.*, 1984).

The structure of $NaSm(PO_3)_4$ is characterized by a three-dimensional framework built up of irregular SmO_8 polyhedra linked with polyphosphate chains *via* Sm–O–P bridges, as show in Fig. 2. The undulated $[(PO_3)_n]^{n-}$ chains have a repeating unit of four corner-sharing PO_4 tetrahedra and extend parallel to the *a*-axis. Furthermore, the framework delimits tunnels in which the Na^+ ions are located. They are 6-coordinated by O atoms with Na—O distances ranging from 2.386 (5)—2.741 (5) Å in a distorted octahedral arrangement.

S2. Experimental

The finely ground reagents Na_2CO_3, Sm_2O_3, and $NH_4H_2PO_4$ were mixed in a molar ratio of Na: Sm: P = 7: 1: 10, placed in a Pt crucible, and heated at 673 K for 4 h. The mixture was re-ground and heated at 1173 K for 20 h, then cooled to 673 K at a rate of 4 K h^{-1}, and finally quenched to room temperature. A few yellow crystals of the title compound with prismatic shape were obtained.

S3. Refinement

The highest peak in the difference electron density map is located 1.49 Å from atom Sm1 while the deepest hole is 1.98 Å from atom O8.

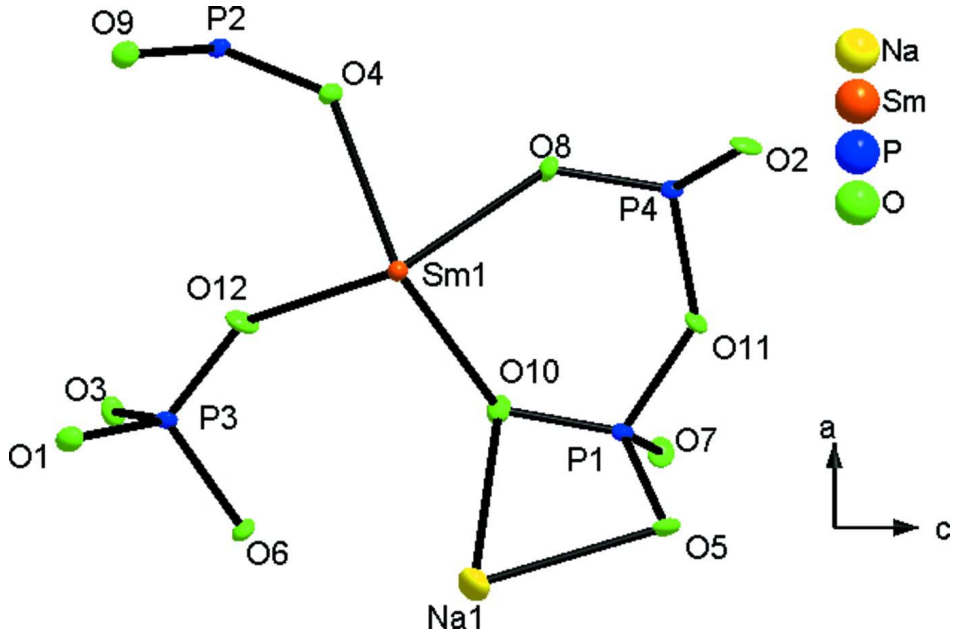

Figure 1

Asymmetric unit of the structure of $NaSm(PO_3)_4$ with the atom labelling scheme. The displacement ellipsoids are drawn at the 50% probability level.

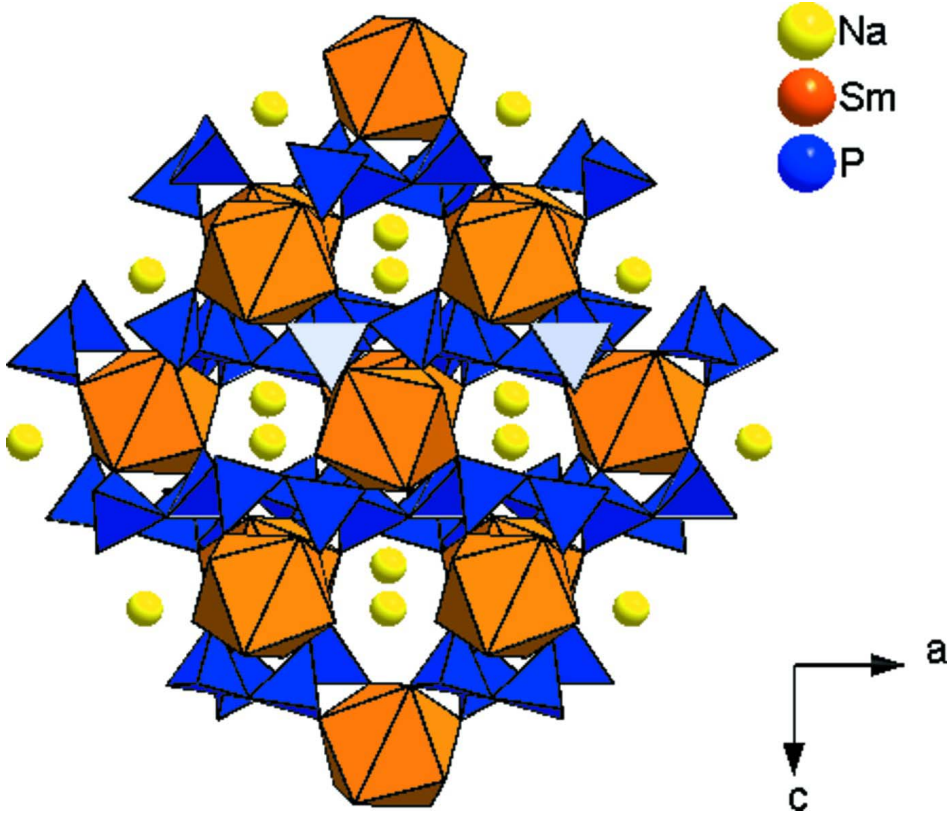

Figure 2

View of the crystal structure of $NaSm(PO_3)_4$ in a projection down [010].

Sodium samarium tetrakis(polyphosphate)

Crystal data

$NaSm(PO_3)_4$
M_r = 489.22
Monoclinic, $P2_1/n$
Hall symbol: -P 2yn
a = 7.1924 (13) Å
b = 13.091 (2) Å
c = 9.8480 (17) Å
β = 90.396 (10)°
V = 927.2 (3) Å^3
Z = 4
$F(000)$ = 916
D_x = 3.504 Mg m^{-3}
Mo $K\alpha$ radiation, λ = 0.71073 Å
Cell parameters from 2180 reflections
θ = 2.1–27.5°
μ = 7.14 mm^{-1}
T = 293 K
Prism, yellow
0.20 × 0.02 × 0.02 mm

Data collection

Rigaku Mercury70 CCD diffractometer
Radiation source: fine-focus sealed tube
Rigaku Graphite Monochromator
DIFFRACTOMETER TYPE monochromator
Detector resolution: 14.6306 pixels mm^{-1}
ω scans
Absorption correction: multi-scan (*ABSCOR*; Higashi, 1995)
T_{min} = 0.329, T_{max} = 0.870
7036 measured reflections
2111 independent reflections
1868 reflections with $I > 2\sigma(I)$
R_{int} = 0.041
θ_{max} = 27.5°, θ_{min} = 2.6°
h = −9→9
k = −15→17
l = −12→12

Refinement

Refinement on F^2
Least-squares matrix: full
$R[F^2 > 2\sigma(F^2)]$ = 0.037
$wR(F^2)$ = 0.068
S = 1.15
2111 reflections
163 parameters
0 restraints
Primary atom site location: structure-invariant direct methods
Secondary atom site location: difference Fourier map
$w = 1/[\sigma^2(F_o^2) + (0.0639P)^2]$ where $P = (F_o^2 + 2F_c^2)/3$
$(\Delta/\sigma)_{max}$ < 0.001
$\Delta\rho_{max}$ = 2.38 e Å^{-3}
$\Delta\rho_{min}$ = −0.80 e Å^{-3}

Special details

Geometry. All e.s.d.'s (except the e.s.d. in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell e.s.d.'s are taken into account individually in the estimation of e.s.d.'s in distances, angles and torsion angles; correlations between e.s.d.'s in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell e.s.d.'s is used for estimating e.s.d.'s involving l.s. planes.

Refinement. Refinement of *F*^2^ against ALL reflections. The weighted *R*-factor *wR* and goodness of fit *S* are based on *F*^2^, conventional *R*-factors *R* are based on *F*, with *F* set to zero for negative *F*^2^. The threshold expression of *F*^2^ > σ(*F*^2^) is used only for calculating *R*-factors(gt) *etc*. and is not relevant to the choice of reflections for refinement. *R*-factors based on *F*^2^ are statistically about twice as large as those based on *F*, and *R*- factors based on ALL data will be even larger.

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (Å^2)

	x	*y*	*z*	U_{iso}*/U_{eq}
Na1	−0.0002 (4)	0.7220 (2)	0.5648 (3)	0.0210 (6)

Sm1	0.51300 (4)	0.71807 (2)	0.47634 (3)	0.00716 (10)
P1	0.2493 (2)	0.60065 (11)	0.74392 (15)	0.0064 (3)
P2	0.8781 (2)	0.61480 (12)	0.26404 (15)	0.0063 (3)
P3	0.2690 (2)	0.59026 (12)	0.19783 (15)	0.0065 (3)
P4	0.6463 (2)	0.62775 (12)	0.80511 (15)	0.0065 (3)
O1	0.2374 (6)	0.6610 (3)	0.0824 (4)	0.0117 (9)
O2	0.7180 (5)	0.7105 (3)	0.8947 (4)	0.0099 (9)
O3	0.2829 (6)	0.4784 (3)	0.1343 (4)	0.0091 (9)
O4	0.8029 (6)	0.6454 (3)	0.3976 (4)	0.0095 (9)
O5	0.0936 (6)	0.6652 (3)	0.7951 (4)	0.0103 (9)
O6	0.0886 (6)	0.5804 (3)	0.2895 (4)	0.0105 (9)
O7	0.2159 (6)	0.4860 (3)	0.7898 (4)	0.0098 (9)
O8	0.6799 (6)	0.6328 (3)	0.6572 (4)	0.0100 (9)
O9	0.8672 (6)	0.6896 (3)	0.1519 (4)	0.0105 (9)
O10	0.2860 (6)	0.6073 (3)	0.5963 (4)	0.0102 (9)
O11	0.4284 (6)	0.6243 (3)	0.8331 (4)	0.0082 (9)
O12	0.4293 (6)	0.6102 (3)	0.2901 (4)	0.0135 (9)

Atomic displacement parameters (Å²)

	U^{11}	U^{22}	U^{33}	U^{12}	U^{13}	U^{23}
Na1	0.0172 (14)	0.0309 (16)	0.0149 (12)	0.0063 (13)	−0.0028 (11)	0.0029 (12)
Sm1	0.00640 (16)	0.00857 (15)	0.00651 (15)	−0.00002 (13)	0.00033 (11)	−0.00033 (13)
P1	0.0049 (7)	0.0065 (7)	0.0078 (7)	−0.0001 (6)	−0.0001 (6)	−0.0006 (6)
P2	0.0055 (7)	0.0053 (7)	0.0082 (7)	0.0002 (6)	0.0012 (6)	0.0002 (6)
P3	0.0046 (7)	0.0064 (7)	0.0084 (7)	0.0013 (6)	0.0000 (6)	−0.0001 (6)
P4	0.0053 (7)	0.0059 (7)	0.0084 (7)	−0.0007 (6)	0.0001 (6)	−0.0007 (6)
O1	0.010 (2)	0.013 (2)	0.012 (2)	0.0027 (18)	0.0012 (18)	0.0029 (18)
O2	0.005 (2)	0.009 (2)	0.016 (2)	−0.0013 (17)	−0.0032 (17)	−0.0044 (18)
O3	0.014 (2)	0.005 (2)	0.009 (2)	0.0016 (17)	−0.0019 (17)	−0.0014 (17)
O4	0.007 (2)	0.012 (2)	0.009 (2)	0.0009 (17)	0.0012 (17)	−0.0007 (17)
O5	0.005 (2)	0.013 (2)	0.013 (2)	0.0014 (17)	0.0020 (17)	−0.0007 (18)
O6	0.008 (2)	0.015 (2)	0.008 (2)	−0.0019 (18)	0.0034 (17)	0.0022 (18)
O7	0.014 (2)	0.005 (2)	0.011 (2)	−0.0042 (17)	−0.0001 (18)	−0.0013 (16)
O8	0.010 (2)	0.013 (2)	0.0064 (19)	0.0015 (18)	0.0028 (17)	0.0026 (17)
O9	0.011 (2)	0.008 (2)	0.012 (2)	0.0023 (17)	0.0018 (18)	0.0028 (17)
O10	0.012 (2)	0.010 (2)	0.009 (2)	−0.0015 (18)	0.0019 (18)	−0.0004 (17)
O11	0.007 (2)	0.010 (2)	0.0076 (19)	−0.0012 (17)	−0.0028 (17)	−0.0021 (16)
O12	0.008 (2)	0.015 (2)	0.017 (2)	−0.0020 (18)	−0.0044 (18)	−0.0047 (18)

Geometric parameters (Å, º)

Na1—O4^{i}	2.386 (5)	P2—O6^{vii}	1.598 (4)
Na1—O1^{ii}	2.438 (5)	P3—O1	1.482 (4)
Na1—O2^{iii}	2.468 (5)	P3—O12	1.486 (4)
Na1—O5	2.475 (5)	P3—O6	1.591 (4)
Na1—O10	2.565 (5)	P3—O3	1.596 (4)
Na1—O8^{i}	2.741 (5)	P4—O8	1.479 (4)

Na1—O9^{ii}	3.005 (5)	P4—O2	1.487 (4)
Sm1—O9^{ii}	2.362 (4)	P4—O11	1.594 (4)
Sm1—O12	2.389 (4)	P4—O3^{vi}	1.595 (4)
Sm1—O8	2.415 (4)	O1—Na1^{iv}	2.438 (5)
Sm1—O5^{iv}	2.423 (4)	O1—Sm1^{iii}	2.486 (4)
Sm1—O4	2.424 (4)	O2—Sm1^{v}	2.449 (4)
Sm1—O2^{iii}	2.449 (4)	O2—Na1^{v}	2.468 (5)
Sm1—O1^{v}	2.486 (4)	O3—P4^{vi}	1.595 (4)
Sm1—O10	2.488 (4)	O4—Na1^{vii}	2.386 (5)
P1—O10	1.482 (4)	O5—Sm1^{ii}	2.423 (4)
P1—O5	1.494 (4)	O6—P2^{i}	1.598 (4)
P1—O11	1.584 (4)	O7—P2^{vi}	1.574 (4)
P1—O7	1.586 (4)	O8—Na1^{vii}	2.741 (5)
P2—O9	1.477 (4)	O9—Sm1^{iv}	2.362 (4)
P2—O4	1.481 (4)	O9—Na1^{iv}	3.005 (5)
P2—O7^{vi}	1.574 (4)		

O4^{i}—Na1—O1^{ii}	81.77 (16)	O10—P1—O5	115.9 (2)
O4^{i}—Na1—O2^{iii}	93.40 (16)	O10—P1—O11	112.5 (2)
O1^{ii}—Na1—O2^{iii}	108.62 (17)	O5—P1—O11	108.1 (2)
O4^{i}—Na1—O5	131.55 (19)	O10—P1—O7	111.3 (2)
O1^{ii}—Na1—O5	109.24 (16)	O5—P1—O7	108.9 (2)
O2^{iii}—Na1—O5	123.97 (17)	O11—P1—O7	98.7 (2)
O4^{i}—Na1—O10	108.05 (17)	O9—P2—O4	117.8 (2)
O1^{ii}—Na1—O10	168.60 (18)	O9—P2—O7^{vi}	106.5 (2)
O2^{iii}—Na1—O10	77.17 (15)	O4—P2—O7^{vi}	111.6 (2)
O5—Na1—O10	60.02 (14)	O9—P2—O6^{vii}	110.5 (2)
O4^{i}—Na1—O8^{i}	63.56 (14)	O4—P2—O6^{vii}	106.8 (2)
O1^{ii}—Na1—O8^{i}	65.92 (15)	O7^{vi}—P2—O6^{vii}	102.8 (2)
O2^{iii}—Na1—O8^{i}	156.55 (17)	O1—P3—O12	118.2 (3)
O5—Na1—O8^{i}	78.01 (15)	O1—P3—O6	111.4 (2)
O10—Na1—O8^{i}	112.69 (17)	O12—P3—O6	107.4 (2)
O4^{i}—Na1—O9^{ii}	151.14 (16)	O1—P3—O3	106.4 (2)
O1^{ii}—Na1—O9^{ii}	114.66 (17)	O12—P3—O3	110.5 (2)
O2^{iii}—Na1—O9^{ii}	59.56 (14)	O6—P3—O3	101.6 (2)
O5—Na1—O9^{ii}	67.70 (14)	O8—P4—O2	119.5 (3)
O10—Na1—O9^{ii}	59.17 (13)	O8—P4—O11	109.8 (2)
O8^{i}—Na1—O9^{ii}	143.89 (15)	O2—P4—O11	104.8 (2)
P1—Na1—O9^{ii}	60.64 (10)	O8—P4—O3^{vi}	110.7 (2)
O9^{ii}—Sm1—O12	138.95 (15)	O2—P4—O3^{vi}	107.7 (2)
O9^{ii}—Sm1—O8	85.27 (14)	O11—P4—O3^{vi}	102.9 (2)
O12—Sm1—O8	114.50 (15)	P3—O1—Sm1^{iii}	144.7 (2)
O9^{ii}—Sm1—O5^{iv}	109.05 (14)	Na1^{iv}—O1—Sm1^{iii}	94.05 (16)
O12—Sm1—O5^{iv}	82.37 (15)	P4—O2—Sm1^{v}	140.3 (2)
O8—Sm1—O5^{iv}	135.61 (14)	P4—O2—Na1^{v}	116.4 (2)
O9^{ii}—Sm1—O4	145.34 (14)	Sm1^{v}—O2—Na1^{v}	101.18 (16)
O12—Sm1—O4	74.63 (14)	P4^{vi}—O3—P3	132.3 (3)
O8—Sm1—O4	68.31 (13)	P2—O4—Na1^{vii}	120.5 (2)

O5^{iv}—Sm1—O4	78.51 (14)	P2—O4—Sm1	135.3 (2)
O9^{ii}—Sm1—O2^{iii}	69.92 (14)	Na1^{vii}—O4—Sm1	97.01 (16)
O12—Sm1—O2^{iii}	76.16 (14)	P1—O5—Sm1^{ii}	142.1 (2)
O8—Sm1—O2^{iii}	147.54 (13)	P1—O5—Na1	93.5 (2)
O5^{iv}—Sm1—O2^{iii}	74.25 (14)	Sm1^{ii}—O5—Na1	114.88 (18)
O4—Sm1—O2^{iii}	142.17 (13)	P3—O6—P2^{i}	131.6 (3)
O9^{ii}—Sm1—O1^{v}	69.91 (14)	P2^{vi}—O7—P1	139.8 (3)
O12—Sm1—O1^{v}	149.17 (14)	P4—O8—Sm1	131.5 (2)
O8—Sm1—O1^{v}	70.50 (14)	P4—O8—Na1^{vii}	119.3 (2)
O5^{iv}—Sm1—O1^{v}	75.54 (14)	Sm1—O8—Na1^{vii}	88.43 (14)
O4—Sm1—O1^{v}	80.05 (14)	P2—O9—Sm1^{iv}	149.9 (3)
O2^{iii}—Sm1—O1^{v}	116.94 (14)	P2—O9—Na1^{iv}	120.7 (2)
O9^{ii}—Sm1—O10	69.79 (14)	Sm1^{iv}—O9—Na1^{iv}	89.30 (14)
O12—Sm1—O10	81.81 (14)	P1—O10—Sm1	128.5 (2)
O8—Sm1—O10	72.84 (14)	P1—O10—Na1	90.3 (2)
O5^{iv}—Sm1—O10	151.45 (14)	Sm1—O10—Na1	97.49 (15)
O4—Sm1—O10	119.46 (14)	P1—O11—P4	135.0 (3)
O2^{iii}—Sm1—O10	78.97 (14)	P3—O12—Sm1	139.8 (3)
O1^{v}—Sm1—O10	126.71 (13)		

Symmetry codes: (i) $x-1, y, z$; (ii) $x-1/2, -y+3/2, z+1/2$; (iii) $x-1/2, -y+3/2, z-1/2$; (iv) $x+1/2, -y+3/2, z-1/2$; (v) $x+1/2, -y+3/2, z+1/2$; (vi) $-x+1, -y+1, -z+1$; (vii) $x+1, y, z$.