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(*E*)-1-(2-Nitrobenzylidene)-2-phenylhydrazine

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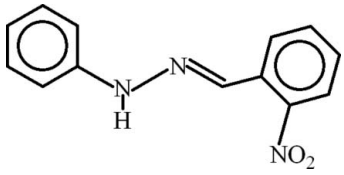
Key indicators: single-crystal X-ray study; T = 296 K; mean σ(C–C) = 0.003 Å; R factor = 0.045; wR factor = 0.117; data-to-parameter ratio = 13.1.

The asymmetric unit of the title compound, $C_{13}H_{11}N_3O_2$, contains two molecules with slightly different conformations: the dihedral angle between the aromatic rings is 13.01 (10)° in one molecule and 14.05 (10)° in the other. Both molecules feature short intramolecular C—H···O contacts, which generate S(6) rings. In the crystal, both molecules form inversion dimers linked by pairs of N—H···O hydrogen bonds, thereby generating $R_2^2(16)$ rings.

Related literature

For background information on Schiff bases and related crystal structures, see: Mufakkar *et al.* (2010); Tahir *et al.* (2010). For graph-set notation, see: Bernstein *et al.* (1995).



Experimental

Crystal data

$C_{13}H_{11}N_3O_2$
M_r = 241.25
Orthorhombic, *Pbca*
a = 19.4021 (13) Å
b = 12.1065 (7) Å
c = 20.0554 (11) Å
V = 4710.8 (5) Å^3
Z = 16
Mo $K\alpha$ radiation
μ = 0.10 mm^{-1}
T = 296 K
0.32 × 0.28 × 0.24 mm

Data collection

Bruker Kappa APEXII CCD diffractometer
Absorption correction: multi-scan (*SADABS*; Bruker, 2005)
T_{min} = 0.972, T_{max} = 0.979
18932 measured reflections
4260 independent reflections
2577 reflections with $I > 2\sigma(I)$
R_{int} = 0.041

Refinement

$R[F^2 > 2\sigma(F^2)]$ = 0.045
$wR(F^2)$ = 0.117
S = 1.01
4260 reflections
325 parameters
H-atom parameters constrained
$\Delta\rho_{max}$ = 0.16 e Å^{-3}
$\Delta\rho_{min}$ = −0.14 e Å^{-3}

Table 1
Hydrogen-bond geometry (Å, °).

D—H···A	D—H	H···A	D···A	D—H···A
C7—H7···O1	0.93	2.24	2.773 (3)	116
C20—H20···O3	0.93	2.27	2.788 (3)	115
N1—H1···O3^{i}	0.86	2.42	3.242 (2)	161
N4—H4A···O1^{ii}	0.86	2.39	3.207 (2)	158

Symmetry codes: (i) $x-\frac{1}{2}, -y+\frac{1}{2}, -z$; (ii) $x+\frac{1}{2}, -y+\frac{1}{2}, -z$.

Data collection: *APEX2* (Bruker, 2009); cell refinement: *SAINT* (Bruker, 2009); data reduction: *SAINT*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 2008); program(s) used to refine structure: *SHELXL97* (Sheldrick, 2008); molecular graphics: *ORTEP-3 for Windows* (Farrugia, 1997) and *PLATON* (Spek, 2009); software used to prepare material for publication: *WinGX* (Farrugia, 1999) and *PLATON*.

The authors acknowledge the provision of funds for the purchase of the diffractometer and encouragement by Dr Muhammad Akram Chaudhary, Vice Chancellor, University of Sargodha, Pakistan.

Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: HB5534).

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supporting information

Acta Cryst. (2010). E**66**, o1955 [https://doi.org/10.1107/S1600536810025882]

(*E*)-1-(2-Nitrobenzylidene)-2-phenylhydrazine

Hazoor Ahmad Shad, M. Nawaz Tahir, Muhammad Ilyas Tariq, Muhammad Sarfraz and Shahbaz Ahmad

S1. Comment

We have reported crystal structures of Schiff bases containing phenylhydrazine (Mufakkar *et al.*, 2010) and 2-nitrobenzaldehyde (Tahir *et al.*, 2010) and as a part of this project, we report herein the structure and synthesis of the title compound (I, Fig. 1).

The title compound consists of two molecules in the crystallographic asymmetric unit which differ from each other geometrically. In one molecule, the phenylhydrazine A (C1—C6/N1/N2) and group B (C7—C13) of 2-nitrobenzaldehyde are planar with r. m. s deviation of 0.0228 and 0.0068 Å, respectively. The nitro group C (O1/N3/O2) is of course planar. The dihedral angle between A/B, A/C and B/C is 14.05 (10)°, 13.38 (32)° and 17.41 (30)°, respectively. In second molecule, the phenylhydrazine D (C1—C6/N1/N2) and group E (C7—C13) of 2-nitrobenzaldehyde are also planar with r. m. s deviation of 0.0054 and 0.0037 Å, respectively. The dihedral angle between D/E is 13.01 (10)°. The nitro group F (O3/N6/O4) of this molecule makes dihedral angle of 25.02 (27)° with group D, whereas it is oriented at 27.12 (27)° with group E. In each molecule there exist S(5) and S(6) ring motifs (Bernstein *et al.*, 1995) due to intramolecular H-bonding of C—H···N and C—H···O type, respectively. The molecules are stabilized in the form of dimers due to N—H···O type of H-bondings with $R_2^2(16)$ ring motifs (Table 1, Fig. 2).

S2. Experimental

Equimolar quantities of phenylhydrazine and 2-nitrobenzaldehyde were refluxed in methanol for 25 min resulting in a violet solution. The solution was kept at room temperature, which afforded violet prisms of (I) after 48 h.

S3. Refinement

The H-atoms were positioned geometrically (N–H = 0.86, C–H = 0.93 Å) and refined as riding with U_{iso}(H) = xU_{eq}(C, N), where x = x = 1.2 for all H-atoms.

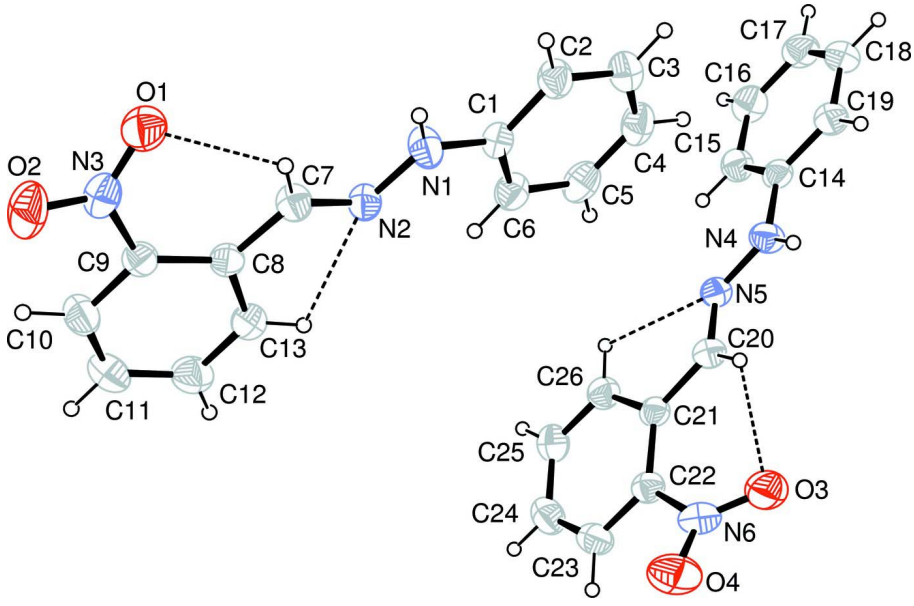

Figure 1

View of (I) with displacement ellipsoids drawn at the 50% probability level. H-atoms are shown by small circles of arbitrary radii.

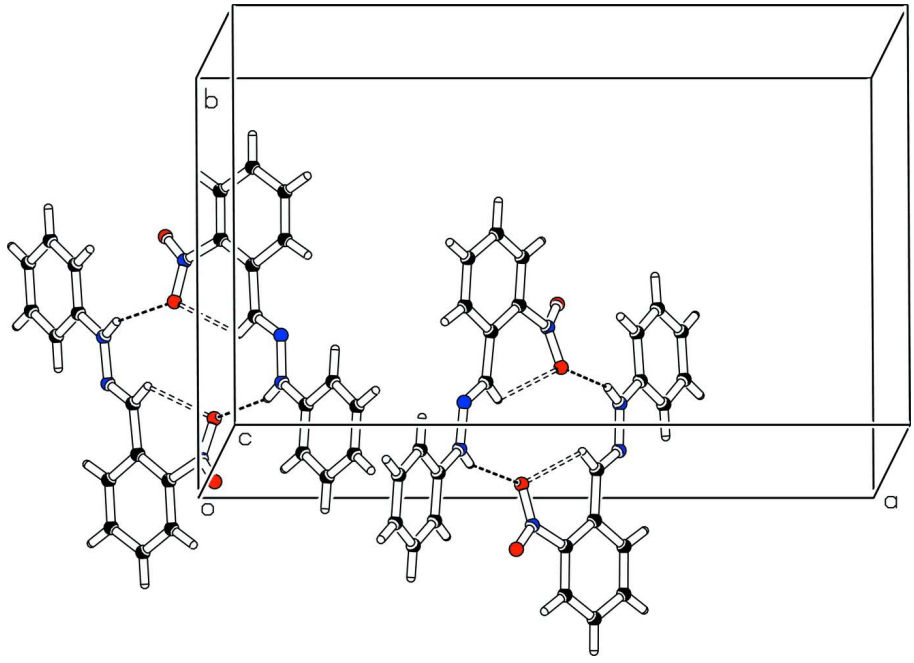

Figure 2

The partial packing of (I), which shows that molecules form dimers.

(*E*)-1-(2-Nitrobenzylidene)-2-phenylhydrazine

Crystal data

$C_{13}H_{11}N_3O_2$
$M_r = 241.25$

Orthorhombic, *Pbca*
Hall symbol: -P 2ac 2ab

a = 19.4021 (13) Å
b = 12.1065 (7) Å
c = 20.0554 (11) Å
V = 4710.8 (5) Å^3
Z = 16
F(000) = 2016
D_x = 1.361 Mg m^{-3}

Mo $K\alpha$ radiation, λ = 0.71073 Å
Cell parameters from 2577 reflections
θ = 2.2–25.3°
μ = 0.10 mm^{-1}
T = 296 K
Prism, violet
0.32 × 0.28 × 0.24 mm

Data collection

Bruker Kappa APEXII CCD diffractometer
Radiation source: fine-focus sealed tube
Graphite monochromator
Detector resolution: 8.20 pixels mm^{-1}
ω scans
Absorption correction: multi-scan (*SADABS*; Bruker, 2005)
T_{min} = 0.972, T_{max} = 0.979

18932 measured reflections
4260 independent reflections
2577 reflections with $I > 2\sigma(I)$
R_{int} = 0.041
θ_{max} = 25.3°, θ_{min} = 2.2°
h = −21→23
k = −13→14
l = −24→24

Refinement

Refinement on F^2
Least-squares matrix: full
$R[F^2 > 2\sigma(F^2)]$ = 0.045
$wR(F^2)$ = 0.117
S = 1.01
4260 reflections
325 parameters
0 restraints
Primary atom site location: structure-invariant direct methods

Secondary atom site location: difference Fourier map
Hydrogen site location: inferred from neighbouring sites
H-atom parameters constrained
$w = 1/[\sigma^2(F_o^2) + (0.0426P)^2 + 0.9743P]$ where $P = (F_o^2 + 2F_c^2)/3$
$(\Delta/\sigma)_{max} < 0.001$
$\Delta\rho_{max}$ = 0.16 e Å^{-3}
$\Delta\rho_{min}$ = −0.14 e Å^{-3}

Special details

Geometry. Bond distances, angles *etc*. have been calculated using the rounded fractional coordinates. All su's are estimated from the variances of the (full) variance-covariance matrix. The cell e.s.d.'s are taken into account in the estimation of distances, angles and torsion angles

Refinement. Refinement of F^2 against ALL reflections. The weighted R-factor wR and goodness of fit S are based on F^2, conventional R-factors R are based on F, with F set to zero for negative F^2. The threshold expression of $F^2 > \sigma(F^2)$ is used only for calculating R-factors(gt) *etc*. and is not relevant to the choice of reflections for refinement. R-factors based on F^2 are statistically about twice as large as those based on F, and R- factors based on ALL data will be even larger.

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (Å^2)

	x	y	z	U_{iso}*/U_{eq}
O1	−0.02767 (9)	0.48801 (15)	−0.11736 (8)	0.0811 (7)
O2	−0.03824 (10)	0.64989 (16)	−0.15815 (9)	0.0982 (8)
N1	0.12471 (9)	0.27574 (14)	−0.00241 (8)	0.0646 (7)
N2	0.12334 (8)	0.38691 (13)	0.00220 (8)	0.0546 (6)
N3	−0.01209 (9)	0.58595 (18)	−0.11891 (9)	0.0642 (7)
C1	0.15768 (10)	0.21311 (16)	0.04626 (10)	0.0519 (7)
C2	0.15290 (12)	0.09962 (17)	0.04246 (10)	0.0634 (8)
C3	0.18603 (12)	0.03400 (19)	0.08869 (12)	0.0709 (9)
C4	0.22378 (12)	0.0800 (2)	0.13901 (11)	0.0709 (10)
C5	0.22787 (12)	0.1929 (2)	0.14304 (11)	0.0716 (10)

C6	0.19503 (11)	0.26019 (18)	0.09740 (10)	0.0621 (8)
C7	0.08784 (11)	0.44017 (16)	−0.04126 (10)	0.0568 (8)
C8	0.08420 (10)	0.56020 (16)	−0.03663 (9)	0.0505 (7)
C9	0.03909 (11)	0.62879 (17)	−0.07170 (9)	0.0528 (7)
C10	0.03841 (13)	0.74245 (19)	−0.06341 (12)	0.0724 (9)
C11	0.08391 (15)	0.79155 (19)	−0.02040 (13)	0.0799 (10)
C12	0.12983 (13)	0.7266 (2)	0.01410 (12)	0.0764 (10)
C13	0.13014 (11)	0.61467 (18)	0.00641 (10)	0.0621 (8)
O3	0.53109 (9)	0.28838 (15)	0.13060 (9)	0.0901 (8)
O4	0.52867 (11)	0.44681 (16)	0.08338 (9)	0.1054 (8)
N4	0.37407 (9)	0.07450 (14)	0.24028 (8)	0.0580 (6)
N5	0.38069 (8)	0.18402 (13)	0.25032 (8)	0.0529 (6)
N6	0.51143 (10)	0.38409 (18)	0.12819 (9)	0.0680 (8)
C14	0.33940 (10)	0.00973 (16)	0.28709 (10)	0.0501 (7)
C15	0.30960 (11)	0.05488 (18)	0.34361 (10)	0.0608 (8)
C16	0.27540 (12)	−0.0131 (2)	0.38795 (11)	0.0713 (9)
C17	0.27064 (12)	−0.1249 (2)	0.37713 (12)	0.0723 (9)
C18	0.30009 (12)	−0.16936 (18)	0.32096 (12)	0.0706 (9)
C19	0.33432 (11)	−0.10272 (17)	0.27593 (11)	0.0610 (8)
C20	0.41511 (10)	0.23954 (16)	0.20722 (10)	0.0539 (7)
C21	0.42235 (10)	0.35828 (16)	0.21663 (9)	0.0487 (7)
C22	0.46637 (10)	0.42733 (17)	0.18061 (10)	0.0527 (7)
C23	0.47060 (12)	0.53979 (18)	0.19184 (11)	0.0642 (8)
C24	0.43038 (13)	0.58790 (19)	0.23970 (12)	0.0719 (9)
C25	0.38560 (12)	0.5228 (2)	0.27580 (11)	0.0678 (9)
C26	0.38167 (11)	0.41148 (18)	0.26445 (10)	0.0596 (8)
H1	0.10513	0.24344	−0.03557	0.0775*
H2	0.12723	0.06716	0.00856	0.0761*
H3	0.18261	−0.04244	0.08555	0.0851*
H4	0.24630	0.03560	0.16995	0.0851*
H5	0.25332	0.22489	0.17727	0.0858*
H6	0.19807	0.33661	0.10112	0.0746*
H7	0.06484	0.40278	−0.07511	0.0682*
H10	0.00707	0.78539	−0.08706	0.0868*
H11	0.08374	0.86777	−0.01461	0.0959*
H12	0.16112	0.75952	0.04309	0.0917*
H13	0.16181	0.57300	0.03047	0.0745*
H4A	0.39117	0.04455	0.20509	0.0696*
H15	0.31260	0.13038	0.35166	0.0729*
H16	0.25524	0.01739	0.42579	0.0856*
H17	0.24775	−0.16993	0.40748	0.0868*
H18	0.29697	−0.24492	0.31318	0.0848*
H19	0.35401	−0.13356	0.23798	0.0732*
H20	0.43505	0.20467	0.17068	0.0647*
H23	0.50081	0.58273	0.16684	0.0770*
H24	0.43318	0.66341	0.24777	0.0862*
H25	0.35776	0.55483	0.30822	0.0814*
H26	0.35083	0.36969	0.28943	0.0716*

Atomic displacement parameters (Å²)

	U^{11}	U^{22}	U^{33}	U^{12}	U^{13}	U^{23}
O1	0.0805 (12)	0.0809 (12)	0.0818 (12)	−0.0026 (10)	−0.0189 (9)	−0.0115 (9)
O2	0.0928 (14)	0.1218 (15)	0.0801 (12)	0.0250 (12)	−0.0258 (10)	0.0245 (11)
N1	0.0801 (14)	0.0509 (11)	0.0628 (11)	0.0090 (9)	−0.0199 (10)	−0.0048 (8)
N2	0.0537 (11)	0.0525 (11)	0.0575 (10)	0.0061 (8)	−0.0015 (9)	−0.0006 (8)
N3	0.0593 (12)	0.0829 (14)	0.0504 (11)	0.0128 (11)	0.0009 (9)	0.0032 (10)
C1	0.0478 (12)	0.0550 (13)	0.0530 (12)	0.0077 (10)	−0.0007 (10)	0.0004 (10)
C2	0.0649 (15)	0.0613 (14)	0.0640 (14)	0.0058 (12)	−0.0117 (11)	−0.0015 (11)
C3	0.0731 (16)	0.0616 (14)	0.0781 (16)	0.0093 (12)	−0.0048 (14)	0.0072 (12)
C4	0.0629 (16)	0.0837 (19)	0.0660 (15)	0.0187 (13)	−0.0016 (12)	0.0156 (12)
C5	0.0583 (15)	0.097 (2)	0.0595 (14)	0.0071 (13)	−0.0102 (11)	−0.0014 (13)
C6	0.0562 (14)	0.0670 (14)	0.0632 (14)	0.0044 (11)	−0.0053 (11)	−0.0060 (11)
C7	0.0614 (14)	0.0557 (14)	0.0533 (12)	0.0019 (11)	−0.0049 (11)	−0.0015 (10)
C8	0.0495 (13)	0.0528 (13)	0.0492 (11)	−0.0008 (10)	0.0046 (10)	0.0046 (9)
C9	0.0532 (13)	0.0572 (13)	0.0479 (12)	−0.0021 (10)	0.0013 (10)	0.0052 (9)
C10	0.0797 (18)	0.0590 (15)	0.0784 (16)	0.0070 (13)	0.0002 (14)	0.0158 (12)
C11	0.098 (2)	0.0505 (14)	0.0912 (18)	−0.0078 (14)	0.0007 (16)	0.0030 (13)
C12	0.0789 (18)	0.0650 (16)	0.0854 (17)	−0.0130 (14)	−0.0085 (14)	−0.0043 (13)
C13	0.0575 (14)	0.0613 (14)	0.0674 (14)	−0.0010 (11)	−0.0072 (12)	0.0035 (11)
O3	0.0841 (13)	0.0784 (12)	0.1079 (14)	−0.0043 (10)	0.0345 (10)	−0.0129 (10)
O4	0.1200 (16)	0.1156 (15)	0.0807 (12)	−0.0243 (12)	0.0354 (11)	0.0206 (11)
N4	0.0609 (12)	0.0537 (11)	0.0593 (10)	−0.0076 (9)	0.0082 (9)	0.0003 (8)
N5	0.0471 (10)	0.0521 (11)	0.0596 (10)	−0.0030 (8)	−0.0027 (8)	0.0050 (8)
N6	0.0617 (13)	0.0785 (14)	0.0639 (12)	−0.0201 (11)	0.0049 (10)	0.0003 (11)
C14	0.0411 (11)	0.0551 (13)	0.0540 (12)	−0.0017 (10)	−0.0054 (10)	0.0088 (10)
C15	0.0600 (14)	0.0611 (14)	0.0612 (13)	0.0017 (11)	−0.0018 (11)	0.0040 (11)
C16	0.0699 (16)	0.0852 (18)	0.0588 (14)	0.0048 (14)	0.0074 (12)	0.0083 (13)
C17	0.0656 (16)	0.0809 (18)	0.0705 (15)	−0.0069 (13)	0.0057 (13)	0.0224 (13)
C18	0.0674 (16)	0.0575 (14)	0.0870 (17)	−0.0081 (12)	0.0041 (14)	0.0106 (12)
C19	0.0576 (14)	0.0597 (14)	0.0657 (13)	−0.0019 (11)	0.0077 (11)	0.0022 (10)
C20	0.0495 (13)	0.0569 (13)	0.0554 (12)	−0.0045 (10)	0.0039 (10)	0.0017 (10)
C21	0.0440 (12)	0.0529 (12)	0.0493 (11)	−0.0018 (10)	−0.0045 (10)	0.0055 (9)
C22	0.0500 (12)	0.0585 (13)	0.0497 (12)	−0.0034 (10)	−0.0035 (10)	0.0027 (10)
C23	0.0676 (15)	0.0598 (15)	0.0653 (14)	−0.0136 (12)	−0.0092 (12)	0.0119 (11)
C24	0.0839 (18)	0.0525 (14)	0.0793 (16)	0.0009 (13)	−0.0142 (14)	−0.0009 (12)
C25	0.0714 (16)	0.0646 (16)	0.0675 (14)	0.0105 (12)	0.0007 (13)	−0.0037 (12)
C26	0.0556 (14)	0.0627 (15)	0.0606 (13)	0.0006 (11)	0.0033 (11)	0.0041 (10)

Geometric parameters (Å, º)

O1—N3	1.224 (3)	C5—H5	0.9300
O2—N3	1.215 (3)	C6—H6	0.9300
O3—N6	1.221 (3)	C7—H7	0.9300
O4—N6	1.223 (3)	C10—H10	0.9300
N1—C1	1.392 (3)	C11—H11	0.9300
N1—N2	1.349 (2)	C12—H12	0.9300

N2—C7	1.285 (3)	C13—H13	0.9300
N3—C9	1.467 (3)	C14—C19	1.383 (3)
N1—H1	0.8600	C14—C15	1.385 (3)
N4—N5	1.347 (2)	C15—C16	1.382 (3)
N4—C14	1.396 (3)	C16—C17	1.374 (3)
N5—C20	1.283 (3)	C17—C18	1.373 (3)
N6—C22	1.464 (3)	C18—C19	1.381 (3)
N4—H4A	0.8600	C20—C21	1.457 (3)
C1—C2	1.379 (3)	C21—C26	1.399 (3)
C1—C6	1.379 (3)	C21—C22	1.397 (3)
C2—C3	1.380 (3)	C22—C23	1.382 (3)
C3—C4	1.366 (3)	C23—C24	1.367 (3)
C4—C5	1.372 (3)	C24—C25	1.379 (3)
C5—C6	1.381 (3)	C25—C26	1.369 (3)
C7—C8	1.458 (3)	C15—H15	0.9300
C8—C9	1.397 (3)	C16—H16	0.9300
C8—C13	1.405 (3)	C17—H17	0.9300
C9—C10	1.386 (3)	C18—H18	0.9300
C10—C11	1.370 (4)	C19—H19	0.9300
C11—C12	1.375 (4)	C20—H20	0.9300
C12—C13	1.364 (3)	C23—H23	0.9300
C2—H2	0.9300	C24—H24	0.9300
C3—H3	0.9300	C25—H25	0.9300
C4—H4	0.9300	C26—H26	0.9300

N2—N1—C1	120.29 (16)	C11—C10—H10	120.00
N1—N2—C7	117.70 (16)	C10—C11—H11	120.00
O1—N3—O2	122.04 (19)	C12—C11—H11	120.00
O1—N3—C9	119.56 (18)	C13—C12—H12	120.00
O2—N3—C9	118.4 (2)	C11—C12—H12	119.00
N2—N1—H1	120.00	C12—C13—H13	119.00
C1—N1—H1	120.00	C8—C13—H13	119.00
N5—N4—C14	119.88 (16)	N4—C14—C15	122.00 (18)
N4—N5—C20	117.69 (16)	N4—C14—C19	118.58 (18)
O4—N6—C22	118.0 (2)	C15—C14—C19	119.42 (19)
O3—N6—C22	119.83 (19)	C14—C15—C16	119.5 (2)
O3—N6—O4	122.2 (2)	C15—C16—C17	121.2 (2)
C14—N4—H4A	120.00	C16—C17—C18	119.2 (2)
N5—N4—H4A	120.00	C17—C18—C19	120.5 (2)
N1—C1—C6	122.54 (18)	C14—C19—C18	120.2 (2)
N1—C1—C2	118.24 (18)	N5—C20—C21	118.69 (18)
C2—C1—C6	119.22 (19)	C22—C21—C26	115.09 (18)
C1—C2—C3	120.3 (2)	C20—C21—C22	125.64 (18)
C2—C3—C4	120.8 (2)	C20—C21—C26	119.26 (18)
C3—C4—C5	118.8 (2)	C21—C22—C23	122.79 (19)
C4—C5—C6	121.5 (2)	N6—C22—C21	121.51 (18)
C1—C6—C5	119.4 (2)	N6—C22—C23	115.70 (19)
N2—C7—C8	118.89 (18)	C22—C23—C24	120.0 (2)

C7—C8—C9	126.23 (18)	C23—C24—C25	119.0 (2)
C9—C8—C13	115.33 (18)	C24—C25—C26	120.7 (2)
C7—C8—C13	118.44 (18)	C21—C26—C25	122.4 (2)
C8—C9—C10	122.40 (19)	C14—C15—H15	120.00
N3—C9—C10	114.96 (19)	C16—C15—H15	120.00
N3—C9—C8	122.63 (18)	C15—C16—H16	119.00
C9—C10—C11	120.0 (2)	C17—C16—H16	119.00
C10—C11—C12	119.1 (2)	C16—C17—H17	120.00
C11—C12—C13	120.9 (2)	C18—C17—H17	120.00
C8—C13—C12	122.2 (2)	C17—C18—H18	120.00
C3—C2—H2	120.00	C19—C18—H18	120.00
C1—C2—H2	120.00	C14—C19—H19	120.00
C4—C3—H3	120.00	C18—C19—H19	120.00
C2—C3—H3	120.00	N5—C20—H20	121.00
C3—C4—H4	121.00	C21—C20—H20	121.00
C5—C4—H4	121.00	C22—C23—H23	120.00
C6—C5—H5	119.00	C24—C23—H23	120.00
C4—C5—H5	119.00	C23—C24—H24	121.00
C5—C6—H6	120.00	C25—C24—H24	120.00
C1—C6—H6	120.00	C24—C25—H25	120.00
C8—C7—H7	121.00	C26—C25—H25	120.00
N2—C7—H7	121.00	C21—C26—H26	119.00
C9—C10—H10	120.00	C25—C26—H26	119.00

C1—N1—N2—C7	−174.85 (18)	C7—C8—C13—C12	−179.5 (2)
N2—N1—C1—C2	174.72 (18)	C9—C8—C13—C12	1.0 (3)
N2—N1—C1—C6	−5.8 (3)	C13—C8—C9—C10	−1.6 (3)
N1—N2—C7—C8	178.67 (17)	N3—C9—C10—C11	179.9 (2)
O1—N3—C9—C8	17.2 (3)	C8—C9—C10—C11	1.2 (3)
O1—N3—C9—C10	−161.6 (2)	C9—C10—C11—C12	0.0 (4)
O2—N3—C9—C8	−163.5 (2)	C10—C11—C12—C13	−0.6 (4)
O2—N3—C9—C10	17.8 (3)	C11—C12—C13—C8	0.0 (4)
N5—N4—C14—C15	1.6 (3)	N4—C14—C15—C16	179.44 (19)
N5—N4—C14—C19	−179.04 (18)	C19—C14—C15—C16	0.1 (3)
C14—N4—N5—C20	177.50 (18)	N4—C14—C19—C18	−179.7 (2)
N4—N5—C20—C21	179.35 (17)	C15—C14—C19—C18	−0.3 (3)
O3—N6—C22—C23	152.6 (2)	C14—C15—C16—C17	0.3 (3)
O4—N6—C22—C21	153.5 (2)	C15—C16—C17—C18	−0.4 (4)
O3—N6—C22—C21	−27.2 (3)	C16—C17—C18—C19	0.2 (4)
O4—N6—C22—C23	−26.7 (3)	C17—C18—C19—C14	0.1 (3)
C6—C1—C2—C3	−1.0 (3)	N5—C20—C21—C22	169.55 (19)
N1—C1—C6—C5	−178.38 (19)	N5—C20—C21—C26	−11.8 (3)
N1—C1—C2—C3	178.5 (2)	C20—C21—C22—N6	−0.5 (3)
C2—C1—C6—C5	1.1 (3)	C20—C21—C22—C23	179.7 (2)
C1—C2—C3—C4	0.3 (3)	C26—C21—C22—N6	−179.12 (18)
C2—C3—C4—C5	0.4 (3)	C26—C21—C22—C23	1.1 (3)
C3—C4—C5—C6	−0.3 (3)	C20—C21—C26—C25	−179.8 (2)
C4—C5—C6—C1	−0.5 (3)	C22—C21—C26—C25	−1.0 (3)

N2—C7—C8—C9	−167.98 (19)	N6—C22—C23—C24	179.8 (2)
N2—C7—C8—C13	12.6 (3)	C21—C22—C23—C24	−0.3 (3)
C7—C8—C9—N3	0.3 (3)	C22—C23—C24—C25	−0.5 (3)
C7—C8—C9—C10	178.9 (2)	C23—C24—C25—C26	0.5 (4)
C13—C8—C9—N3	179.78 (18)	C24—C25—C26—C21	0.3 (3)

Hydrogen-bond geometry (Å, º)

D—H···*A*	*D*—H	H···*A*	*D*···*A*	*D*—H···*A*
C7—H7···O1	0.93	2.24	2.773 (3)	116
C20—H20···O3	0.93	2.27	2.788 (3)	115
N1—H1···O3^{i}	0.86	2.42	3.242 (2)	161
N4—H4*A*···O1^{ii}	0.86	2.39	3.207 (2)	158

Symmetry codes: (i) *x*−1/2, −*y*+1/2, −*z*; (ii) *x*+1/2, −*y*+1/2, −*z*.