

Acta Crystallographica Section E

Structure Reports

Online

ISSN 1600-5368

N-{4-[(3,4-Dimethylphenyl)(ethyl)-sulfamoyl]phenyl}-*N*-ethylacetamide

Peter John,^a Saima Khizar,^a Islam Ullah Khan,^a ‡ Shahzad Sharif^a and Edward R. T. Tiekink^{b*}

^aMaterials Chemistry Laboratory, Department of Chemistry, Government College, University, Lahore 54000, Pakistan, and ^bDepartment of Chemistry, University of Malaya, 50603 Kuala Lumpur, Malaysia

Correspondence e-mail: edward.tiekink@gmail.com

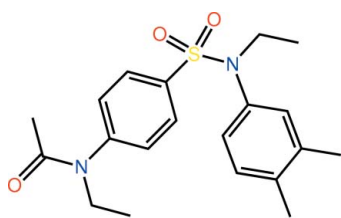
Received 4 November 2010; accepted 7 November 2010

Key indicators: single-crystal X-ray study; $T = 293$ K; mean $\sigma(\text{C}-\text{C}) = 0.002$ Å; R factor = 0.039; wR factor = 0.122; data-to-parameter ratio = 17.0.

When viewed down the central $\text{S}\cdots\text{N}$ axis of the title compound, $\text{C}_{20}\text{H}_{26}\text{N}_2\text{O}_3\text{S}$, it is apparent that the molecule adopts a *gauche* conformation with all O atoms lying to one side of the central benzene ring; the carbonyl O atom is directed away from the central ring and the *N*-bound ethyl groups lie to one side of the molecule. Supramolecular helical chains aligned along the *b* axis and sustained by $\text{C}-\text{H}\cdots\text{O}$ contacts feature in the crystal packing. These are consolidated in the three-dimensional structure by $\text{C}-\text{H}\cdots\pi$ interactions.

Related literature

For background to the pharmacological uses of sulfonamides, see: Korolkovas (1988); Mandell & Sande (1992). For related structures, see: Sharif *et al.* (2010); Khan *et al.* (2010).



Experimental

Crystal data

$\text{C}_{20}\text{H}_{26}\text{N}_2\text{O}_3\text{S}$
 $M_r = 374.51$
 Monoclinic, $P2_1/c$
 $a = 8.0882$ (2) Å

$b = 11.5978$ (3) Å
 $c = 21.2717$ (5) Å
 $\beta = 97.194$ (1)°
 $V = 1979.69$ (8) Å³

$Z = 4$
 Mo $K\alpha$ radiation
 $\mu = 0.19$ mm⁻¹

$T = 293$ K
 $0.28 \times 0.14 \times 0.08$ mm

Data collection

Bruker APEXII CCD diffractometer
 Absorption correction: multi-scan (SADABS; Sheldrick, 1996)
 $T_{\min} = 0.692$, $T_{\max} = 0.895$

16612 measured reflections
 4079 independent reflections
 3325 reflections with $I > 2\sigma(I)$
 $R_{\text{int}} = 0.029$

Refinement

$R[F^2 > 2\sigma(F^2)] = 0.039$
 $wR(F^2) = 0.122$
 $S = 1.01$
 4079 reflections

240 parameters
 H-atom parameters constrained
 $\Delta\rho_{\max} = 0.24$ e Å⁻³
 $\Delta\rho_{\min} = -0.26$ e Å⁻³

Table 1

Hydrogen-bond geometry (Å, °).

Cg1 is the centroid of the C3–C8 ring.

$D-H\cdots A$	$D-H$	$H\cdots A$	$D\cdots A$	$D-H\cdots A$
$\text{C8}-\text{H8}\cdots\text{O1}^{\text{i}}$	0.93	2.54	3.455 (2)	170
$\text{C10}-\text{H10a}\cdots\text{Cg1}^{\text{ii}}$	0.96	2.93	3.728 (2)	142

Symmetry codes: (i) $-x + 2, y - \frac{1}{2}, -z + \frac{1}{2}$; (ii) $-x + 2, -y + 1, -z + 1$.

Data collection: APEX2 (Bruker, 2007); cell refinement: SAINT (Bruker, 2007); data reduction: SAINT; program(s) used to solve structure: SHELXS97 (Sheldrick, 2008); program(s) used to refine structure: SHELXL97 (Sheldrick, 2008); molecular graphics: ORTEP-3 (Farrugia, 1997) and DIAMOND (Brandenburg, 2006); software used to prepare material for publication: publCIF (Westrip, 2010).

The authors are grateful to the Higher Education Commission of Pakistan for financial support to purchase the diffractometer.

Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: HG2743).

References

- Brandenburg, K. (2006). *DIAMOND*. Crystal Impact GbR, Bonn, Germany.
 Bruker (2007). *APEX2* and *SAINTE*. Bruker AXS Inc., Madison Wisconsin, USA.
 Farrugia, L. J. (1997). *J. Appl. Cryst.* **30**, 565.
 Khan, I. U., Sharif, S., Batool, S., Mumtaz, A. M. & Tiekink, E. R. T. (2010). *Acta Cryst.* **E66**, o2641.
 Korolkovas, A. (1988). *Essentials of Medicinal Chemistry*, 2nd ed., pp. 699–716. New York: Wiley.
 Mandell, G. L. & Sande, M. A. (1992). *Goodman and Gilman, The Pharmacological Basis of Therapeutics 2*, edited by A. Gilman, T. W. Rall, A. S. Nies & P. Taylor, 8th ed., pp. 1047–1057. Singapore: Graw-Hill.
 Sharif, S., Iqbal, H., Khan, I. U., John, P. & Tiekink, E. R. T. (2010). *Acta Cryst.* **E66**, o1288.
 Sheldrick, G. M. (1996). *SADABS*. University of Göttingen, Germany.
 Sheldrick, G. M. (2008). *Acta Cryst.* **A64**, 112–122.
 Westrip, S. P. (2010). *J. Appl. Cryst.* **43**, 920–925.

‡ Additional correspondence author, e-mail: iuklodhi@yahoo.com.

supporting information

Acta Cryst. (2010). E66, o3152 [https://doi.org/10.1107/S1600536810045708]

N*-{4-[(3,4-Dimethylphenyl)(ethyl)sulfamoyl]phenyl}-*N*-ethylacetamide*Peter John, Saima Khizar, Islam Ullah Khan, Shahzad Sharif and Edward R. T. Tiekink****S1. Comment**

In connection with on-going structural studies of sulfonamides (Sharif *et al.*, 2010; Khan *et al.*, 2010), of interest owing to their biological properties (Korolkovas, 1988; Mandell & Sande, 1992), the title compound, (I), was investigated.

With reference to the central benzene ring in (I), Fig. 1, the S1 [deviation = -0.068 (1) Å] and N2 [-0.005 (1) Å] atoms are co-planar. Both sulfonamide-O atoms lie to the same side of the plane as does the carbonyl-O atom, which is directed away from the ring, with the remaining substituents lying to the other side. When viewed down the S1...N2 vector, both N-bound ethyl groups lie to the same side of the molecule. Similarly, when viewed down the S1...N2 vector, the molecule has a *gauche* conformation.

In the crystal packing, molecules are connected into a helical supramolecular chain along the *b* axis via C—H...O contacts occurring between benzene-H and sulfonamide-O atoms, Table 1 and Fig. 2. The chains are consolidated in the crystal packing by C—H... π interactions, Table 1 and Fig. 3.

S2. Experimental

A mixture of *N*-{4-[(3,4-dimethylphenyl)sulfamoyl]phenyl}acetamide 100 mg (0.314 mmol) and sodium hydride 85 mg (0.78 mmol) in *N,N*-dimethylformamide (10 ml) was stirred at room temperature for 30 min. followed by addition of ethyl iodide 199 μ l (0.785 mmol). Stirring was continued for a further 3 h and the contents were poured over crushed ice. The precipitate that formed was isolated, washed and crystallized from methanol solution by slow evaporation; *M.pt.* 472 K.

S3. Refinement

The C-bound H atoms were geometrically placed (C—H = 0.93–0.97 Å) and refined as riding with $U_{iso}(\text{H}) = 1.2-1.5U_{eq}(\text{C})$. In the final refinement four low angle reflections evidently effected by the beam stop were omitted, *i.e.* (100), (002), (011) and ($\bar{1}$ 11).

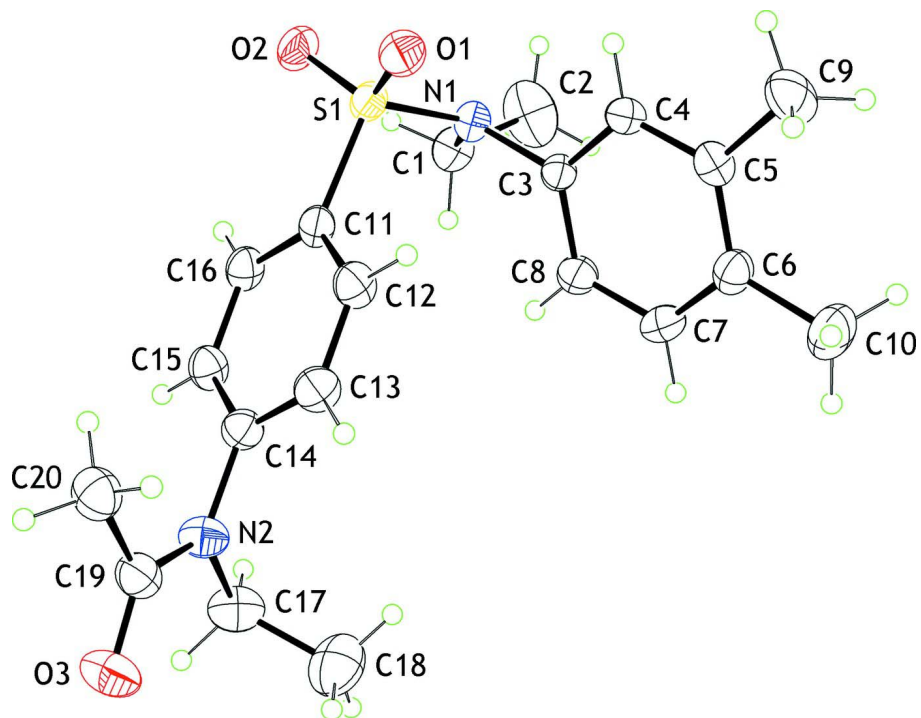


Figure 1

The molecular structure of (I) showing the atom-labelling scheme and displacement ellipsoids at the 35% probability level.

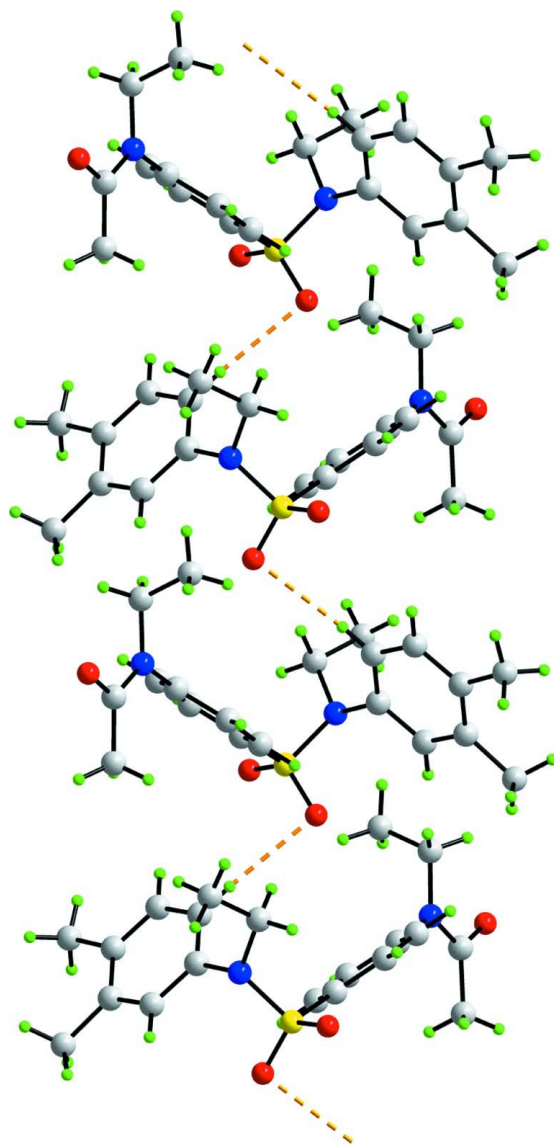


Figure 2

A view of the helical supramolecular chain along the *b* axis in (I). The C—H...O contacts sustaining this chain are shown as orange dashed lines.

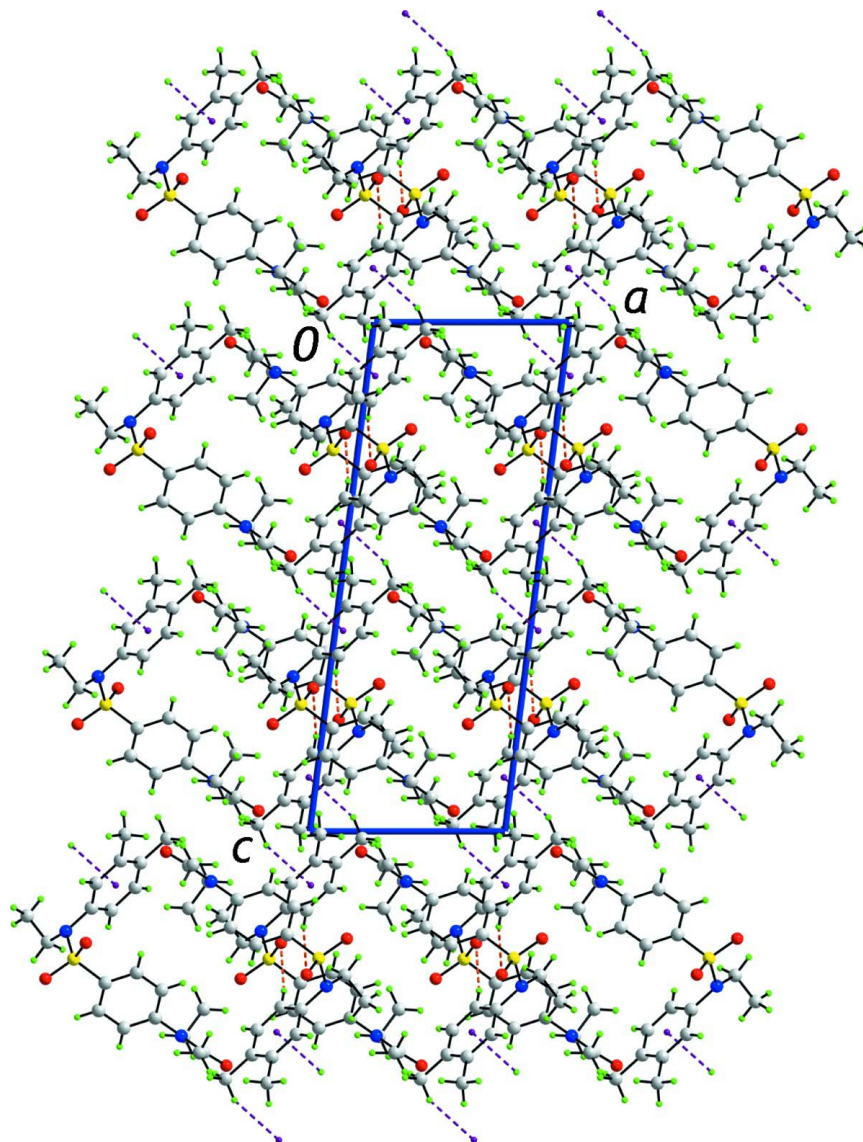


Figure 3

View in projection down the *b* axis of the unit-cell contents for (I). The C—H...O and C—H... π contacts are shown as orange and purple dashed lines, respectively.

N-{4-[(3,4-Dimethylphenyl)(ethyl)sulfamoyl]phenyl}-*N*-ethylacetamide

Crystal data

$C_{20}H_{26}N_2O_3S$

$M_r = 374.51$

Monoclinic, $P2_1/c$

Hall symbol: $-P\ 2_1/c$

$a = 8.0882\ (2)\ \text{\AA}$

$b = 11.5978\ (3)\ \text{\AA}$

$c = 21.2717\ (5)\ \text{\AA}$

$\beta = 97.194\ (1)^\circ$

$V = 1979.69\ (8)\ \text{\AA}^3$

$Z = 4$

$F(000) = 800$

$D_x = 1.257\ \text{Mg m}^{-3}$

Mo $K\alpha$ radiation, $\lambda = 0.71073\ \text{\AA}$

Cell parameters from 7015 reflections

$\theta = 3.0\text{--}27.8^\circ$

$\mu = 0.19\ \text{mm}^{-1}$

$T = 293\ \text{K}$

Block, colourless

$0.28 \times 0.14 \times 0.08\ \text{mm}$

Data collection

Bruker APEXII CCD diffractometer	16612 measured reflections
Radiation source: fine-focus sealed tube	4079 independent reflections
Graphite monochromator	3325 reflections with $I > 2\sigma(I)$
φ and ω scans	$R_{\text{int}} = 0.029$
Absorption correction: multi-scan (SADABS; Sheldrick, 1996)	$\theta_{\text{max}} = 26.5^\circ$, $\theta_{\text{min}} = 2.6^\circ$
$T_{\text{min}} = 0.692$, $T_{\text{max}} = 0.895$	$h = -9 \rightarrow 10$
	$k = -14 \rightarrow 11$
	$l = -26 \rightarrow 26$

Refinement

Refinement on F^2	Secondary atom site location: difference Fourier map
Least-squares matrix: full	Hydrogen site location: inferred from neighbouring sites
$R[F^2 > 2\sigma(F^2)] = 0.039$	H-atom parameters constrained
$wR(F^2) = 0.122$	$w = 1/[\sigma^2(F_o^2) + (0.0713P)^2 + 0.4585P]$
$S = 1.01$	where $P = (F_o^2 + 2F_c^2)/3$
4079 reflections	$(\Delta/\sigma)_{\text{max}} = 0.001$
240 parameters	$\Delta\rho_{\text{max}} = 0.24 \text{ e } \text{\AA}^{-3}$
0 restraints	$\Delta\rho_{\text{min}} = -0.26 \text{ e } \text{\AA}^{-3}$
Primary atom site location: structure-invariant direct methods	

Special details

Geometry. All s.u.'s (except the s.u. in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell s.u.'s are taken into account individually in the estimation of s.u.'s in distances, angles and torsion angles; correlations between s.u.'s in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell s.u.'s is used for estimating s.u.'s involving l.s. planes.

Refinement. Refinement of F^2 against ALL reflections. The weighted R -factor wR and goodness of fit S are based on F^2 , conventional R -factors R are based on F , with F set to zero for negative F^2 . The threshold expression of $F^2 > 2\sigma(F^2)$ is used only for calculating R -factors(gt) etc. and is not relevant to the choice of reflections for refinement. R -factors based on F^2 are statistically about twice as large as those based on F , and R -factors based on ALL data will be even larger.

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (\AA^2)

	x	y	z	$U_{\text{iso}}^*/U_{\text{eq}}$
S1	1.12740 (5)	0.72075 (4)	0.249229 (18)	0.03937 (14)
O1	1.07590 (16)	0.82013 (10)	0.28110 (6)	0.0509 (3)
O2	1.25922 (15)	0.73062 (12)	0.21059 (6)	0.0536 (3)
O3	0.27644 (17)	0.53664 (14)	0.04014 (7)	0.0694 (4)
N1	1.18549 (16)	0.62329 (11)	0.30379 (6)	0.0376 (3)
N2	0.52634 (17)	0.52003 (13)	0.09842 (7)	0.0475 (4)
C1	1.2726 (2)	0.52088 (16)	0.28259 (8)	0.0476 (4)
H1A	1.1913	0.4620	0.2683	0.057*
H1B	1.3306	0.5419	0.2470	0.057*
C2	1.3933 (3)	0.4738 (2)	0.33407 (12)	0.0874 (8)
H2A	1.4686	0.5337	0.3505	0.131*
H2B	1.4551	0.4123	0.3178	0.131*
H2C	1.3347	0.4450	0.3673	0.131*
C3	1.06969 (18)	0.60395 (13)	0.34975 (7)	0.0339 (3)
C4	1.06258 (19)	0.68223 (14)	0.39837 (7)	0.0389 (3)
H4	1.1336	0.7456	0.4015	0.047*

C5	0.9517 (2)	0.66828 (15)	0.44270 (7)	0.0422 (4)
C6	0.8482 (2)	0.57149 (16)	0.43848 (8)	0.0442 (4)
C7	0.8584 (2)	0.49348 (15)	0.39000 (8)	0.0488 (4)
H7	0.7902	0.4286	0.3873	0.059*
C8	0.9669 (2)	0.50905 (14)	0.34545 (8)	0.0440 (4)
H8	0.9704	0.4559	0.3129	0.053*
C9	0.9443 (3)	0.7571 (2)	0.49396 (9)	0.0631 (5)
H9A	1.0265	0.8155	0.4903	0.095*
H9B	0.9660	0.7207	0.5347	0.095*
H9C	0.8355	0.7916	0.4896	0.095*
C10	0.7272 (3)	0.5500 (2)	0.48570 (10)	0.0673 (6)
H10A	0.7868	0.5468	0.5276	0.101*
H10B	0.6708	0.4781	0.4761	0.101*
H10C	0.6471	0.6114	0.4834	0.101*
C11	0.95077 (19)	0.66426 (14)	0.20256 (7)	0.0377 (3)
C12	0.7959 (2)	0.68032 (16)	0.22264 (8)	0.0454 (4)
H12	0.7859	0.7231	0.2589	0.055*
C13	0.6567 (2)	0.63238 (16)	0.18843 (8)	0.0477 (4)
H13	0.5526	0.6421	0.2020	0.057*
C14	0.67146 (19)	0.56991 (14)	0.13400 (8)	0.0410 (4)
C15	0.8263 (2)	0.55379 (15)	0.11401 (8)	0.0435 (4)
H15	0.8359	0.5115	0.0775	0.052*
C16	0.96666 (19)	0.60077 (15)	0.14841 (7)	0.0422 (4)
H16	1.0710	0.5898	0.1353	0.051*
C17	0.5128 (2)	0.39364 (17)	0.09782 (10)	0.0592 (5)
H17A	0.6237	0.3608	0.1001	0.071*
H17B	0.4510	0.3697	0.0579	0.071*
C18	0.4287 (4)	0.3465 (2)	0.15098 (12)	0.0839 (7)
H18A	0.4904	0.3684	0.1907	0.126*
H18B	0.4241	0.2639	0.1480	0.126*
H18C	0.3177	0.3768	0.1484	0.126*
C19	0.4017 (2)	0.58325 (17)	0.06710 (8)	0.0485 (4)
C20	0.4223 (3)	0.71197 (17)	0.06548 (9)	0.0564 (5)
H20A	0.3588	0.7423	0.0280	0.085*
H20B	0.5379	0.7305	0.0652	0.085*
H20C	0.3834	0.7454	0.1022	0.085*

Atomic displacement parameters (\AA^2)

	U^{11}	U^{22}	U^{33}	U^{12}	U^{13}	U^{23}
S1	0.0372 (2)	0.0426 (3)	0.0387 (2)	-0.00274 (16)	0.00612 (16)	0.00737 (16)
O1	0.0619 (8)	0.0375 (6)	0.0527 (7)	0.0004 (6)	0.0049 (6)	0.0032 (5)
O2	0.0408 (6)	0.0717 (9)	0.0498 (7)	-0.0092 (6)	0.0114 (5)	0.0127 (6)
O3	0.0518 (8)	0.0747 (10)	0.0762 (9)	-0.0066 (7)	-0.0140 (7)	-0.0079 (8)
N1	0.0376 (7)	0.0394 (7)	0.0365 (6)	0.0037 (5)	0.0071 (5)	0.0036 (5)
N2	0.0397 (7)	0.0436 (8)	0.0584 (9)	0.0001 (6)	0.0028 (6)	-0.0044 (7)
C1	0.0457 (9)	0.0479 (10)	0.0514 (9)	0.0085 (8)	0.0142 (8)	0.0004 (8)
C2	0.0891 (17)	0.0869 (18)	0.0821 (16)	0.0473 (14)	-0.0048 (13)	0.0050 (14)

C3	0.0342 (7)	0.0354 (8)	0.0323 (7)	-0.0006 (6)	0.0043 (6)	0.0021 (6)
C4	0.0388 (8)	0.0380 (8)	0.0392 (8)	-0.0052 (7)	0.0018 (6)	-0.0018 (6)
C5	0.0413 (8)	0.0477 (9)	0.0367 (8)	0.0048 (7)	0.0021 (7)	-0.0035 (7)
C6	0.0393 (8)	0.0528 (10)	0.0416 (8)	0.0021 (7)	0.0086 (7)	0.0060 (7)
C7	0.0494 (10)	0.0434 (9)	0.0544 (10)	-0.0142 (8)	0.0095 (8)	0.0013 (8)
C8	0.0530 (10)	0.0379 (9)	0.0416 (8)	-0.0073 (7)	0.0082 (7)	-0.0068 (7)
C9	0.0642 (12)	0.0715 (13)	0.0544 (11)	0.0027 (10)	0.0107 (10)	-0.0209 (10)
C10	0.0596 (12)	0.0847 (16)	0.0622 (11)	-0.0031 (11)	0.0254 (10)	0.0073 (11)
C11	0.0340 (8)	0.0451 (9)	0.0343 (7)	0.0036 (7)	0.0061 (6)	0.0065 (7)
C12	0.0404 (9)	0.0559 (10)	0.0414 (8)	0.0047 (8)	0.0104 (7)	-0.0042 (8)
C13	0.0339 (8)	0.0619 (11)	0.0490 (9)	0.0029 (8)	0.0122 (7)	-0.0034 (8)
C14	0.0351 (8)	0.0438 (9)	0.0439 (8)	0.0014 (7)	0.0038 (7)	0.0027 (7)
C15	0.0428 (9)	0.0504 (10)	0.0380 (8)	0.0055 (7)	0.0080 (7)	-0.0016 (7)
C16	0.0344 (8)	0.0541 (10)	0.0396 (8)	0.0058 (7)	0.0110 (6)	0.0046 (7)
C17	0.0519 (11)	0.0488 (11)	0.0774 (13)	0.0015 (8)	0.0105 (9)	-0.0121 (10)
C18	0.112 (2)	0.0563 (13)	0.0870 (16)	-0.0039 (13)	0.0250 (15)	0.0065 (12)
C19	0.0429 (9)	0.0578 (11)	0.0445 (9)	0.0004 (8)	0.0043 (7)	-0.0042 (8)
C20	0.0589 (11)	0.0538 (11)	0.0549 (10)	0.0054 (9)	0.0002 (9)	0.0030 (9)

Geometric parameters (Å, °)

S1—O1	1.4256 (13)	C9—H9A	0.9600
S1—O2	1.4301 (12)	C9—H9B	0.9600
S1—N1	1.6458 (13)	C9—H9C	0.9600
S1—C11	1.7608 (16)	C10—H10A	0.9600
O3—C19	1.226 (2)	C10—H10B	0.9600
N1—C3	1.4534 (18)	C10—H10C	0.9600
N1—C1	1.480 (2)	C11—C12	1.386 (2)
N2—C19	1.352 (2)	C11—C16	1.386 (2)
N2—C14	1.436 (2)	C12—C13	1.379 (2)
N2—C17	1.470 (2)	C12—H12	0.9300
C1—C2	1.478 (3)	C13—C14	1.384 (2)
C1—H1A	0.9700	C13—H13	0.9300
C1—H1B	0.9700	C14—C15	1.385 (2)
C2—H2A	0.9600	C15—C16	1.383 (2)
C2—H2B	0.9600	C15—H15	0.9300
C2—H2C	0.9600	C16—H16	0.9300
C3—C8	1.376 (2)	C17—C18	1.494 (3)
C3—C4	1.383 (2)	C17—H17A	0.9700
C4—C5	1.390 (2)	C17—H17B	0.9700
C4—H4	0.9300	C18—H18A	0.9600
C5—C6	1.396 (2)	C18—H18B	0.9600
C5—C9	1.506 (2)	C18—H18C	0.9600
C6—C7	1.382 (2)	C19—C20	1.503 (3)
C6—C10	1.508 (2)	C20—H20A	0.9600
C7—C8	1.382 (2)	C20—H20B	0.9600
C7—H7	0.9300	C20—H20C	0.9600
C8—H8	0.9300		

O1—S1—O2	119.46 (8)	H9B—C9—H9C	109.5
O1—S1—N1	107.13 (7)	C6—C10—H10A	109.5
O2—S1—N1	107.03 (7)	C6—C10—H10B	109.5
O1—S1—C11	107.71 (8)	H10A—C10—H10B	109.5
O2—S1—C11	108.45 (7)	C6—C10—H10C	109.5
N1—S1—C11	106.37 (7)	H10A—C10—H10C	109.5
C3—N1—C1	116.91 (12)	H10B—C10—H10C	109.5
C3—N1—S1	115.41 (10)	C12—C11—C16	120.59 (15)
C1—N1—S1	116.17 (10)	C12—C11—S1	118.50 (12)
C19—N2—C14	123.41 (15)	C16—C11—S1	120.86 (12)
C19—N2—C17	119.09 (15)	C13—C12—C11	119.53 (15)
C14—N2—C17	117.48 (15)	C13—C12—H12	120.2
C2—C1—N1	111.47 (16)	C11—C12—H12	120.2
C2—C1—H1A	109.3	C12—C13—C14	120.17 (15)
N1—C1—H1A	109.3	C12—C13—H13	119.9
C2—C1—H1B	109.3	C14—C13—H13	119.9
N1—C1—H1B	109.3	C15—C14—C13	120.26 (15)
H1A—C1—H1B	108.0	C15—C14—N2	119.64 (15)
C1—C2—H2A	109.5	C13—C14—N2	120.09 (14)
C1—C2—H2B	109.5	C16—C15—C14	119.88 (15)
H2A—C2—H2B	109.5	C16—C15—H15	120.1
C1—C2—H2C	109.5	C14—C15—H15	120.1
H2A—C2—H2C	109.5	C15—C16—C11	119.56 (14)
H2B—C2—H2C	109.5	C15—C16—H16	120.2
C8—C3—C4	119.57 (14)	C11—C16—H16	120.2
C8—C3—N1	120.92 (13)	N2—C17—C18	113.53 (17)
C4—C3—N1	119.50 (13)	N2—C17—H17A	108.9
C3—C4—C5	121.45 (14)	C18—C17—H17A	108.9
C3—C4—H4	119.3	N2—C17—H17B	108.9
C5—C4—H4	119.3	C18—C17—H17B	108.9
C4—C5—C6	118.84 (15)	H17A—C17—H17B	107.7
C4—C5—C9	119.93 (16)	C17—C18—H18A	109.5
C6—C5—C9	121.23 (16)	C17—C18—H18B	109.5
C7—C6—C5	118.90 (15)	H18A—C18—H18B	109.5
C7—C6—C10	119.53 (17)	C17—C18—H18C	109.5
C5—C6—C10	121.57 (16)	H18A—C18—H18C	109.5
C8—C7—C6	121.90 (15)	H18B—C18—H18C	109.5
C8—C7—H7	119.0	O3—C19—N2	120.85 (18)
C6—C7—H7	119.0	O3—C19—C20	120.99 (18)
C3—C8—C7	119.31 (15)	N2—C19—C20	118.16 (16)
C3—C8—H8	120.3	C19—C20—H20A	109.5
C7—C8—H8	120.3	C19—C20—H20B	109.5
C5—C9—H9A	109.5	H20A—C20—H20B	109.5
C5—C9—H9B	109.5	C19—C20—H20C	109.5
H9A—C9—H9B	109.5	H20A—C20—H20C	109.5
C5—C9—H9C	109.5	H20B—C20—H20C	109.5
H9A—C9—H9C	109.5		

O1—S1—N1—C3	48.43 (12)	O2—S1—C11—C12	-159.96 (14)
O2—S1—N1—C3	177.67 (10)	N1—S1—C11—C12	85.22 (14)
C11—S1—N1—C3	-66.55 (12)	O1—S1—C11—C16	153.43 (13)
O1—S1—N1—C1	-169.22 (12)	O2—S1—C11—C16	22.83 (16)
O2—S1—N1—C1	-39.98 (14)	N1—S1—C11—C16	-91.99 (14)
C11—S1—N1—C1	75.80 (13)	C16—C11—C12—C13	0.2 (3)
C3—N1—C1—C2	-68.9 (2)	S1—C11—C12—C13	-177.06 (13)
S1—N1—C1—C2	149.37 (17)	C11—C12—C13—C14	-0.8 (3)
C1—N1—C3—C8	-39.7 (2)	C12—C13—C14—C15	0.9 (3)
S1—N1—C3—C8	102.32 (16)	C12—C13—C14—N2	180.00 (16)
C1—N1—C3—C4	140.40 (15)	C19—N2—C14—C15	-114.12 (19)
S1—N1—C3—C4	-77.54 (16)	C17—N2—C14—C15	67.4 (2)
C8—C3—C4—C5	-1.0 (2)	C19—N2—C14—C13	66.8 (2)
N1—C3—C4—C5	178.83 (13)	C17—N2—C14—C13	-111.71 (19)
C3—C4—C5—C6	1.4 (2)	C13—C14—C15—C16	-0.3 (3)
C3—C4—C5—C9	-178.33 (16)	N2—C14—C15—C16	-179.42 (15)
C4—C5—C6—C7	-0.7 (2)	C14—C15—C16—C11	-0.4 (3)
C9—C5—C6—C7	179.10 (17)	C12—C11—C16—C15	0.4 (2)
C4—C5—C6—C10	178.90 (16)	S1—C11—C16—C15	177.57 (13)
C9—C5—C6—C10	-1.3 (3)	C19—N2—C17—C18	-89.7 (2)
C5—C6—C7—C8	-0.5 (3)	C14—N2—C17—C18	88.9 (2)
C10—C6—C7—C8	179.92 (17)	C14—N2—C19—O3	-176.18 (16)
C4—C3—C8—C7	-0.2 (2)	C17—N2—C19—O3	2.3 (3)
N1—C3—C8—C7	179.99 (15)	C14—N2—C19—C20	4.7 (2)
C6—C7—C8—C3	0.9 (3)	C17—N2—C19—C20	-176.83 (17)
O1—S1—C11—C12	-29.36 (15)		

Hydrogen-bond geometry (Å, °)

Cg1 is the centroid of the C3—C8 ring.

<i>D</i> —H... <i>A</i>	<i>D</i> —H	H... <i>A</i>	<i>D</i> ... <i>A</i>	<i>D</i> —H... <i>A</i>
C8—H8...O1 ⁱ	0.93	2.54	3.455 (2)	170
C10—H10a...Cg1 ⁱⁱ	0.96	2.93	3.728 (2)	142

Symmetry codes: (i) $-x+2, y-1/2, -z+1/2$; (ii) $-x+2, -y+1, -z+1$.