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# *meso*-Dimethyl 2,5-dibromohexanedioate

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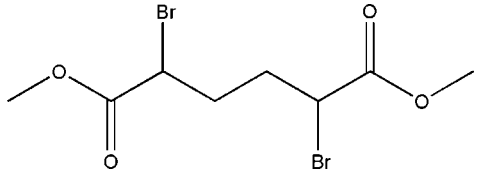
Key indicators: single-crystal X-ray study; $T$ = 293 K; mean $\sigma$(C–C) = 0.006 Å; $R$ factor = 0.040; $wR$ factor = 0.072; data-to-parameter ratio = 19.9.

The title compound, $C_8H_{12}Br_2O_4$, lies about a crystallographic center of inversion at the midpoint of the central C—C bond. The latter is also repsonsible for the observation of the *meso* form. There are no intramolecular hydrogen bonds, but molecules are connected by intermolecular C—H···O interactions, forming a three-dimensional network.

## Related literature

The title compound is an important intermediate in organic synthesis. For the synthetic procedure, see: McDonald & Reitz (1972). For bond-length data, see: Allen *et al.* (1987).




## Experimental

### *Crystal data*

$C_8H_{12}Br_2O_4$
$M_r$ = 331.98
Monoclinic, $P2_1/c$
$a$ = 4.5580 (9) Å
$b$ = 12.134 (2) Å
$c$ = 10.554 (2) Å
$\beta$ = 90.36 (3)°
$V$ = 583.7 (2) Å$^3$
$Z$ = 2
Mo $K\alpha$ radiation
$\mu$ = 6.93 mm$^{-1}$
$T$ = 293 K
0.20 × 0.10 × 0.10 mm

### *Data collection*

Enraf–Nonius CAD-4 diffractometer
Absorption correction: $\psi$ scan (North *et al.*, 1968)
$T_{min}$ = 0.338, $T_{max}$ = 0.544
1428 measured reflections
1271 independent reflections
639 reflections with $I > 2\sigma(I)$
$R_{int}$ = 0.071
3 standard reflections every 200 reflections
intensity decay: 1%

### *Refinement*

$R[F^2 > 2\sigma(F^2)]$ = 0.040
$wR(F^2)$ = 0.072
$S$ = 1.00
1271 reflections
64 parameters
3 restraints
H-atom parameters constrained
$\Delta\rho_{max}$ = 0.33 e Å$^{-3}$
$\Delta\rho_{min}$ = −0.37 e Å$^{-3}$

**Table 1**
Hydrogen-bond geometry (Å, °).

| $D$—H···$A$ | $D$—H | H···$A$ | $D$···$A$ | $D$—H···$A$ |
|---|---|---|---|---|
| C3—H3$A$···O2$^{i}$ | 0.98 | 2.59 | 3.33 (1) | 132 |

Symmetry code: (i) $x+1, y, z$.

Data collection: *CAD-4 Software* (Enraf–Nonius, 1985); cell refinement: *CAD-4 Software*; data reduction: *XCAD4* (Harms & Wocadlo, 1995); program(s) used to solve structure: *SHELXS97* (Sheldrick, 2008); program(s) used to refine structure: *SHELXL97* (Sheldrick, 2008); molecular graphics: *SHELXTL* (Sheldrick, 2008); software used to prepare material for publication: *SHELXTL*.

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Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: IM2243).

## References

Allen, F. H., Kennard, O., Watson, D. G., Brammer, L., Orpen, A. G. & Taylor, R. (1987). *J. Chem. Soc. Perkin Trans. 2*, pp. S1–19.
Enraf–Nonius (1985). *CAD-4 Software*. Enraf–Nonius, Delft, The Netherlands.
Harms, K. & Wocadlo, S. (1995). *XCAD4*. University of Marburg, Germany.
McDonald, R. N. & Reitz, R. R. (1972). *J. Org. Chem.* **37**, 2418–2423.
North, A. C. T., Phillips, D. C. & Mathews, F. S. (1968). *Acta Cryst.* A**24**, 351–359.
Sheldrick, G. M. (2008). *Acta Cryst.* A**64**, 112–122.

# supporting information

*Acta Cryst.* (2010). E**66**, o3163 [https://doi.org/10.1107/S1600536810045642]

## *meso*-Dimethyl 2,5-dibromohexanedioate

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### S1. Comment

The tittle compound, *meso*-2,5-dibromo-hexanedioic acid dimethyl ester is an important intermediate for the synthesis of dimethyl cyclobut-1-ene-1,2-dicarboxylate. We herein report the crystal structure of the title compound, (I).

The molecular structure of (I) is shown in Fig. 1. Bond lengths and angles are within normal ranges (Allen *et al.*, 1987).

The central C4—C4A bond of the title compound, $C_8H_{12}Br_2O_4$, represents a crystallographic center of inversion. The latter is also repsonsible for the observation of the *meso* form. There are no intramolecular hydrogen bonds, but molecules of the title compound are connected by C—H···O intermolecular interactions to form a three dimensional network (Table 1).

### S2. Experimental

The title compound, (I) was prepared by a method reported in literature (McDonald & Reitz, 1972). Single crystals were obtained by dissolving (I) (0.5 g, 1.5 mmol) in ethanol (25 ml) and evaporating the solvent slowly at room temperature for about 3 d.

### S3. Refinement

H atoms were positioned geometrically, with C—H = 0.96 Å for alkyl H, and constrained to ride on their parent atoms, with $U_{iso}$(H) = 1.5$U_{eq}$(C).

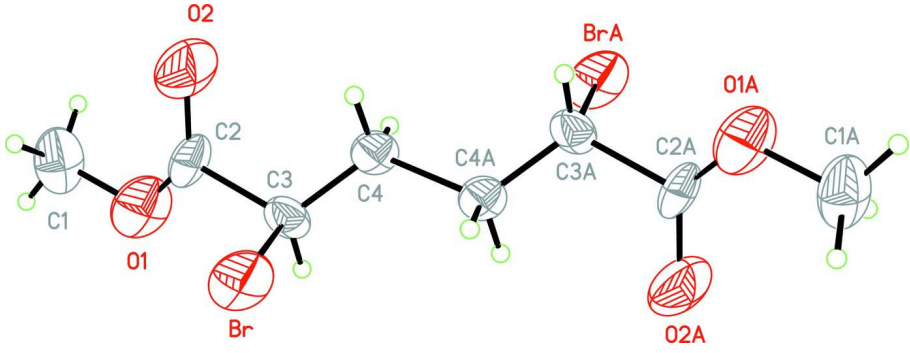


**Figure 1**

Molecular structure of the title compound showing the atom-numbering scheme. Displacement ellipsoids are drawn at the 40% probability level. Atoms labeled with the suffixes A are generated by the symmetry operation (1/2 - *x*, 3/2 - *y*, 1 - *z*). Hydrogen bonds are shown as dashed lines.

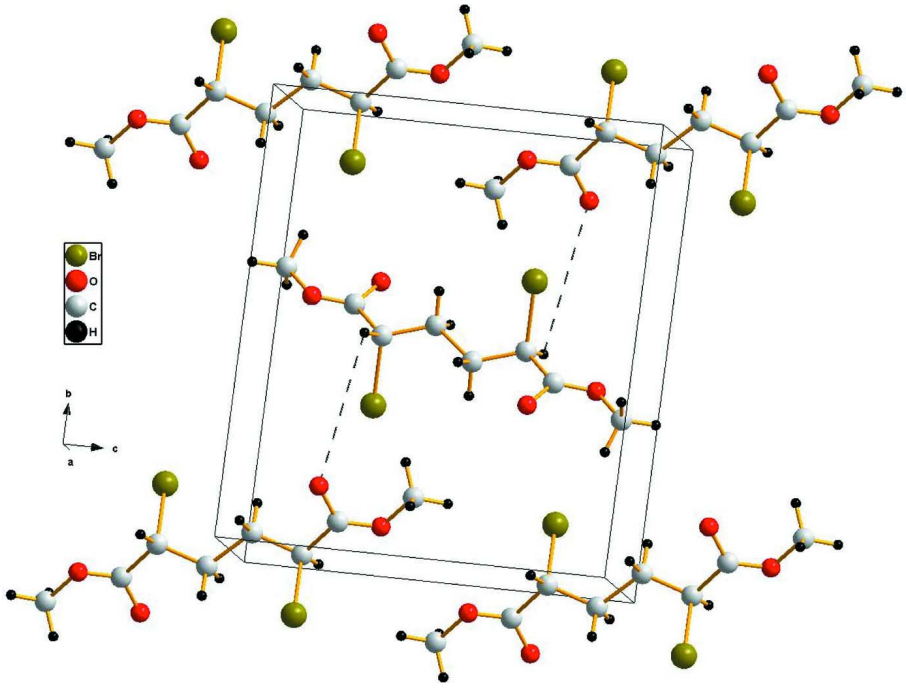


**Figure 2**

Packing diagram for (I). C—H···O hydrogen bonds are shown as dashed lines.

### *meso*-Dimethyl 2,5-dibromohexanedioate

*Crystal data*

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$M_r = 331.98$
Monoclinic, $P2_1/c$
Hall symbol: -P 2ybc
$a = 4.5580\ (9)$ Å
$b = 12.134\ (2)$ Å
$c = 10.554\ (2)$ Å
$\beta = 90.36\ (3)°$
$V = 583.7\ (2)$ Å$^3$
$Z = 2$
$F(000) = 324$
$D_x = 1.889$ Mg m$^{-3}$
Mo $K\alpha$ radiation, $\lambda = 0.71073$ Å
Cell parameters from 25 reflections
$\theta = 9–13°$
$\mu = 6.93$ mm$^{-1}$
$T = 293$ K
Block, colourless
0.20 × 0.10 × 0.10 mm

*Data collection*

Enraf–Nonius CAD-4 diffractometer
Radiation source: fine-focus sealed tube
Graphite monochromator
$\omega/2\theta$ scans
Absorption correction: $\psi$ scan (North *et al.*, 1968)
$T_{min} = 0.338$, $T_{max} = 0.544$
1428 measured reflections
1271 independent reflections
639 reflections with $I > 2\sigma(I)$
$R_{int} = 0.071$
$\theta_{max} = 27.1°$, $\theta_{min} = 2.6°$
$h = 0\rightarrow5$
$k = -15\rightarrow0$
$l = -13\rightarrow13$
3 standard reflections every 200 reflections
intensity decay: 1%

*Refinement*

Refinement on $F^2$
Least-squares matrix: full
$R[F^2 > 2\sigma(F^2)] = 0.040$
$wR(F^2) = 0.072$
$S = 1.00$
1271 reflections
64 parameters
3 restraints
Primary atom site location: structure-invariant direct methods
Secondary atom site location: difference Fourier map
Hydrogen site location: inferred from neighbouring sites
H-atom parameters constrained
$w = 1/[\sigma^2(F_o^2) + (0.022P)^2]$ where $P = (F_o^2 + 2F_c^2)/3$
$(\Delta/\sigma)_{max} < 0.001$
$\Delta\rho_{max} = 0.33$ e Å$^{-3}$
$\Delta\rho_{min} = -0.37$ e Å$^{-3}$

*Special details*

**Geometry**. All e.s.d.'s (except the e.s.d. in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell e.s.d.'s are taken into account individually in the estimation of e.s.d.'s in distances, angles and torsion angles; correlations between e.s.d.'s in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell e.s.d.'s is used for estimating e.s.d.'s involving l.s. planes.

**Refinement**. Refinement of $F^2$ against ALL reflections. The weighted *R*-factor *wR* and goodness of fit *S* are based on $F^2$, conventional *R*-factors *R* are based on *F*, with *F* set to zero for negative $F^2$. The threshold expression of $F^2 > \sigma(F^2)$ is used only for calculating *R*-factors(gt) *etc*. and is not relevant to the choice of reflections for refinement. *R*-factors based on $F^2$ are statistically about twice as large as those based on *F*, and *R*- factors based on ALL data will be even larger.

*Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (Å²)*

| | *x* | *y* | *z* | $U_{iso}$*/$U_{eq}$ |
|---|---|---|---|---|
| Br | 0.23822 (12) | 0.64074 (4) | 0.70596 (6) | 0.0813 (2) |
| O1 | 0.3762 (8) | 0.4221 (3) | 0.8857 (4) | 0.0888 (12) |
| O2 | 0.0821 (9) | 0.3620 (3) | 0.7400 (4) | 0.0980 (12) |
| C1 | 0.2255 (11) | 0.3583 (4) | 0.9715 (5) | 0.0872 (17) |
| H1A | 0.3263 | 0.3595 | 1.0516 | 0.131* |
| H1B | 0.2144 | 0.2839 | 0.9409 | 0.131* |
| H1C | 0.0310 | 0.3872 | 0.9817 | 0.131* |
| C2 | 0.2723 (12) | 0.4259 (4) | 0.7836 (5) | 0.0572 (12) |
| C3 | 0.4550 (10) | 0.4973 (3) | 0.6885 (4) | 0.0477 (11) |
| H3A | 0.6595 | 0.5044 | 0.7166 | 0.057* |
| C4 | 0.4369 (9) | 0.4614 (3) | 0.5570 (4) | 0.0475 (11) |
| H4A | 0.2315 | 0.4476 | 0.5384 | 0.057* |
| H4B | 0.5370 | 0.3911 | 0.5517 | 0.057* |

*Atomic displacement parameters (Å²)*

| | $U^{11}$ | $U^{22}$ | $U^{33}$ | $U^{12}$ | $U^{13}$ | $U^{23}$ |
|---|---|---|---|---|---|---|
| Br | 0.0997 (4) | 0.0232 (2) | 0.1214 (5) | 0.0072 (3) | 0.0337 (3) | −0.0032 (4) |
| O1 | 0.111 (3) | 0.058 (2) | 0.098 (3) | 0.002 (2) | 0.007 (2) | 0.035 (2) |
| O2 | 0.119 (3) | 0.063 (3) | 0.112 (3) | −0.028 (3) | 0.023 (2) | 0.020 (3) |
| C1 | 0.123 (5) | 0.061 (4) | 0.078 (3) | 0.014 (4) | 0.029 (3) | 0.022 (3) |
| C2 | 0.062 (3) | 0.040 (3) | 0.070 (3) | 0.006 (3) | 0.022 (2) | 0.025 (3) |
| C3 | 0.070 (3) | 0.029 (2) | 0.043 (2) | −0.002 (2) | −0.0005 (19) | −0.0068 (19) |
| C4 | 0.063 (3) | 0.029 (2) | 0.050 (3) | −0.004 (2) | −0.006 (2) | 0.007 (2) |

*Geometric parameters (Å, º)*

| | | | |
|---|---|---|---|
| Br—C3 | 2.011 (4) | C2—C3 | 1.569 (6) |
| O1—C2 | 1.175 (5) | C3—C4 | 1.456 (5) |
| O1—C1 | 1.378 (5) | C3—H3A | 0.9800 |
| O2—C2 | 1.249 (6) | C4—C4$^{i}$ | 1.632 (7) |
| C1—H1A | 0.9600 | C4—H4A | 0.9700 |
| C1—H1B | 0.9600 | C4—H4B | 0.9700 |
| C1—H1C | 0.9600 | | |
| | | | |
| C2—O1—C1 | 115.1 (5) | C4—C3—Br | 108.7 (3) |
| O1—C1—H1A | 109.5 | C2—C3—Br | 99.0 (3) |
| O1—C1—H1B | 109.5 | C4—C3—H3A | 111.3 |
| H1A—C1—H1B | 109.5 | C2—C3—H3A | 111.3 |
| O1—C1—H1C | 109.5 | Br—C3—H3A | 111.3 |
| H1A—C1—H1C | 109.5 | C3—C4—C4$^{i}$ | 120.9 (4) |
| H1B—C1—H1C | 109.5 | C3—C4—H4A | 107.1 |
| O1—C2—O2 | 126.0 (5) | C4$^{i}$—C4—H4A | 107.1 |
| O1—C2—C3 | 113.3 (5) | C3—C4—H4B | 107.1 |
| O2—C2—C3 | 118.5 (5) | C4$^{i}$—C4—H4B | 107.1 |
| C4—C3—C2 | 114.7 (4) | H4A—C4—H4B | 106.8 |
| | | | |
| C1—O1—C2—O2 | −15.8 (8) | O1—C2—C3—Br | −95.3 (4) |
| C1—O1—C2—C3 | −178.7 (4) | O2—C2—C3—Br | 100.4 (4) |
| O1—C2—C3—C4 | 149.2 (4) | C2—C3—C4—C4$^{i}$ | 168.8 (4) |
| O2—C2—C3—C4 | −15.1 (6) | Br—C3—C4—C4$^{i}$ | 59.1 (5) |

Symmetry code: (i) −*x*+1, −*y*+1, −*z*+1.

*Hydrogen-bond geometry (Å, º)*

| *D*—H···*A* | *D*—H | H···*A* | *D*···*A* | *D*—H···*A* |
|---|---|---|---|---|
| C3—H3*A*···O2$^{ii}$ | 0.98 | 2.59 | 3.33 (1) | 132 |

Symmetry code: (ii) *x*+1, *y*, *z*.