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2-Isobutyl-6-(4-methoxyphenyl)-imidazo[2,1-*b*][1,3,4]thiadiazole

 Hoong-Kun Fun,^{a*‡} Chin Sing Yeap,^{a§} D. Jagadeesh Prasad,^b Prakash Anil Castelino^c and V. V. Anitha^b

^aX-ray Crystallography Unit, School of Physics, Universiti Sains Malaysia, 11800 USM, Penang, Malaysia, ^bDepartment of Chemistry, Mangalore University, Mangalore, Karnataka, India, and ^cSt. Philomena's College, Puttur, Dakshina Kannada, Karnataka, India

Correspondence e-mail: hkfun@usm.my

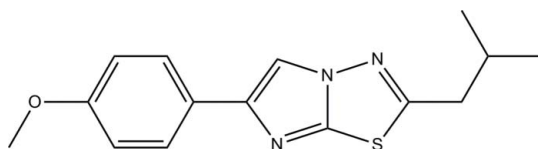
Received 17 December 2010; accepted 18 December 2010

 Key indicators: single-crystal X-ray study; $T = 296$ K; mean $\sigma(\text{C}-\text{C}) = 0.002$ Å; R factor = 0.053; wR factor = 0.163; data-to-parameter ratio = 34.3.

In the title compound, $\text{C}_{15}\text{H}_{17}\text{N}_3\text{OS}$, the dihedral angle between the statistically planar imidazo[2,1-*b*][1,3,4]thiadiazole fused-ring system (r.m.s. deviation = 0.002 Å) and the methoxybenzene ring is 4.52 (6)°. In the crystal, molecules are arranged into columns and stacked down the a axis. The crystal structure is stabilized by weak $\text{C}-\text{H}\cdots\pi$ and $\pi-\pi$ interactions [centroid-centroid separations = 3.6053 (8) and 3.7088 (7) Å].

Related literature

For a related structure and background references to imidazo[2,1-*b*]-1,3,4-thiadiazole derivatives, see: Fun *et al.* (2011).



Experimental

Crystal data

 $\text{C}_{15}\text{H}_{17}\text{N}_3\text{OS}$
 $M_r = 287.38$

 Triclinic, $P\bar{1}$
 $a = 5.7139$ (1) Å

 $b = 10.1795$ (1) Å

 $c = 12.9689$ (2) Å

 $\alpha = 85.174$ (1)°

 $\beta = 85.164$ (1)°

 $\gamma = 80.690$ (1)°

 $V = 739.84$ (2) Å³
 $Z = 2$

 Mo $K\alpha$ radiation

 $\mu = 0.22$ mm⁻¹
 $T = 296$ K

 $0.43 \times 0.31 \times 0.17$ mm

Data collection

Bruker SMART APEXII CCD diffractometer

Absorption correction: multi-scan

 (*SADABS*; Bruker, 2009)

 $T_{\min} = 0.911$, $T_{\max} = 0.964$

23503 measured reflections

6213 independent reflections

 3805 reflections with $I > 2\sigma(I)$
 $R_{\text{int}} = 0.028$

Refinement

 $R[F^2 > 2\sigma(F^2)] = 0.053$
 $wR(F^2) = 0.163$
 $S = 1.03$

6213 reflections

181 parameters

H-atom parameters constrained

 $\Delta\rho_{\text{max}} = 0.30$ e Å⁻³
 $\Delta\rho_{\text{min}} = -0.28$ e Å⁻³
Table 1

Hydrogen-bond geometry (Å, °).

 Cg3 is the centroid of the C1–C6 benzene ring.

| $D-\text{H}\cdots A$ | $D-\text{H}$ | $\text{H}\cdots A$ | $D\cdots A$ | $D-\text{H}\cdots A$ |
|--|--------------|--------------------|-------------|----------------------|
| $\text{C11}-\text{H11A}\cdots\text{Cg3}^i$ | 0.97 | 2.60 | 3.5063 (16) | 155 |

 Symmetry code: (i) $-x + 2, -y + 2, -z + 1$.

Data collection: *APEX2* (Bruker, 2009); cell refinement: *SAINT* (Bruker, 2009); data reduction: *SAINT*; program(s) used to solve structure: *SHELXTL* (Sheldrick, 2008); program(s) used to refine structure: *SHELXTL*; molecular graphics: *SHELXTL*; software used to prepare material for publication: *SHELXTL* and *PLATON* (Spek, 2009).

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Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: HB5778).

References

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[‡] Thomson Reuters ResearcherID: A-3561-2009.

[§] Thomson Reuters ResearcherID: A-5523-2009.

supporting information

Acta Cryst. (2011). E67, o255 [doi:10.1107/S1600536810053225]

2-Isobutyl-6-(4-methoxyphenyl)imidazo[2,1-*b*][1,3,4]thiadiazole

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S1. Comment

As part of our ongoing synthetic and structural studies of imidazo[2,1-*b*]-1,3,4-thiadiazole derivatives (Fun *et al.*, 2011), we now report the structure of the title compound, (I).

The mean plane through the imidazo[2,1-*b*]-1,3,4-thiadiazole ring and the methoxyphenyl moiety is essentially planar with the maximum deviation of 0.045 Å for atom C2 (Fig. 1). The isobutyl is twisted away from this mean plane with torsion angles of C9–C11–C12–C13 = 64.3 (2)° and C9–C11–C12–C14 = -172.81 (17)°. In the crystal structure, the molecules are arranged into columns and stacked down *a* axis (Fig. 2). The molecules are stabilized by the weak Cg1...Cg2ⁱ = 3.7088 (7) Å, Cg2...Cg2ⁱ = 3.6053 (8) Å and C11–H11A...Cg3ⁱ interactions [Cg1, Cg2 and Cg3 are centroids of S1/C9/N1/N2/C10, N2/C8/C7/N3/C10 and C–C6 ring respectively; (i) 2 - *x*, 2 - *y*, 1 - *z*].

S2. Experimental

5-Isobutyl-1,3,4-thiadiazol-2-amine (1 molar equivalent) and 4-methoxyphenacylbromide (1 molar equivalent) are refluxed with ethanol for 4 h. The solvent was then distilled and the reaction mass was poured onto the crushed ice. The resulting solid that separated out was filtered and dried. The compound was re-crystallized using ethanol and DMF mixture to yield colourless blocks of (I). M.P.: 118–122°C.

S3. Refinement

All hydrogen atoms were positioned geometrically [C–H = 0.93–0.98 Å] and refined using a riding model, with $U_{\text{iso}}(\text{H}) = 1.2$ or $1.5U_{\text{eq}}(\text{C})$.

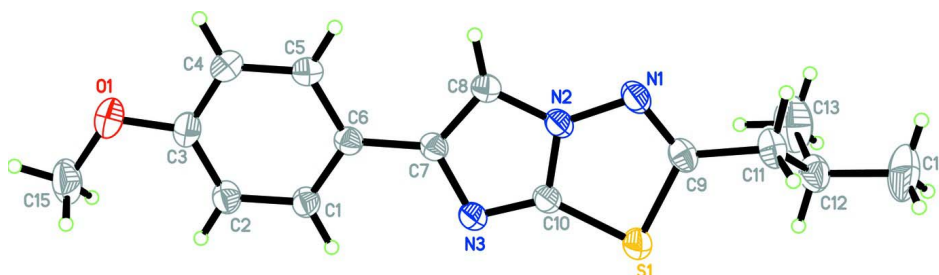
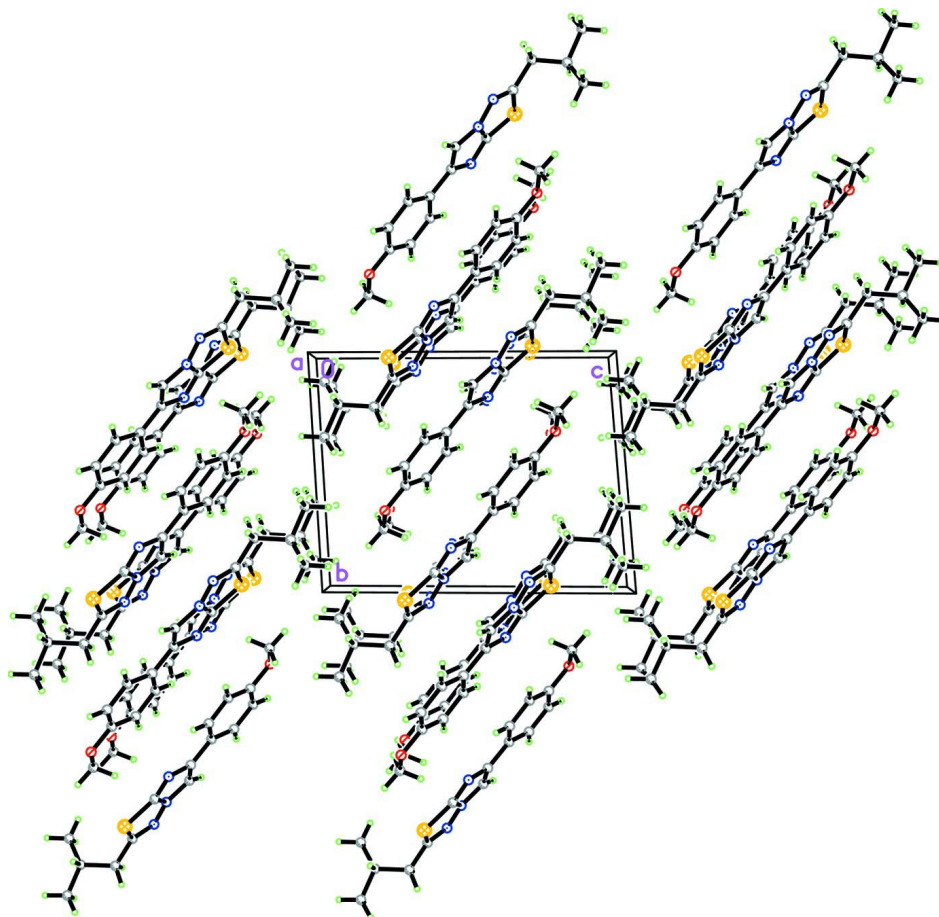


Figure 1

The molecular structure of (I) with 30% probability ellipsoids for non-H atoms.

**Figure 2**

The crystal packing of (I), viewed down the *a* axis, showing molecules stacked down *a* axis.

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Crystal data

$C_{15}H_{17}N_3OS$

$M_r = 287.38$

Triclinic, $P\bar{1}$

Hall symbol: $-P\ 1$

$a = 5.7139$ (1) Å

$b = 10.1795$ (1) Å

$c = 12.9689$ (2) Å

$\alpha = 85.174$ (1)°

$\beta = 85.164$ (1)°

$\gamma = 80.690$ (1)°

$V = 739.84$ (2) Å³

$Z = 2$

$F(000) = 304$

$D_x = 1.290$ Mg m⁻³

Mo $K\alpha$ radiation, $\lambda = 0.71073$ Å

Cell parameters from 6150 reflections

$\theta = 2.5$ – 30.1 °

$\mu = 0.22$ mm⁻¹

$T = 296$ K

Block, colourless

$0.43 \times 0.31 \times 0.17$ mm

Data collection

Bruker SMART APEXII CCD
diffractometer

Radiation source: fine-focus sealed tube

Graphite monochromator

φ and ω scans

Absorption correction: multi-scan
(*SADABS*; Bruker, 2009)

$T_{\min} = 0.911$, $T_{\max} = 0.964$

23503 measured reflections

6213 independent reflections

3805 reflections with $I > 2\sigma(I)$

$R_{\text{int}} = 0.028$
 $\theta_{\text{max}} = 34.5^\circ$, $\theta_{\text{min}} = 1.6^\circ$
 $h = -9 \rightarrow 9$

$k = -16 \rightarrow 16$
 $l = -20 \rightarrow 20$

Refinement

Refinement on F^2
 Least-squares matrix: full
 $R[F^2 > 2\sigma(F^2)] = 0.053$
 $wR(F^2) = 0.163$
 $S = 1.03$
 6213 reflections
 181 parameters
 0 restraints
 Primary atom site location: structure-invariant
 direct methods

Secondary atom site location: difference Fourier
 map
 Hydrogen site location: inferred from
 neighbouring sites
 H-atom parameters constrained
 $w = 1/[\sigma^2(F_o^2) + (0.0765P)^2 + 0.0748P]$
 where $P = (F_o^2 + 2F_c^2)/3$
 $(\Delta/\sigma)_{\text{max}} < 0.001$
 $\Delta\rho_{\text{max}} = 0.30 \text{ e } \text{\AA}^{-3}$
 $\Delta\rho_{\text{min}} = -0.28 \text{ e } \text{\AA}^{-3}$

Special details

Geometry. All e.s.d.'s (except the e.s.d. in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell e.s.d.'s are taken into account individually in the estimation of e.s.d.'s in distances, angles and torsion angles; correlations between e.s.d.'s in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell e.s.d.'s is used for estimating e.s.d.'s involving l.s. planes.

Refinement. Refinement of F^2 against ALL reflections. The weighted R -factor wR and goodness of fit S are based on F^2 , conventional R -factors R are based on F , with F set to zero for negative F^2 . The threshold expression of $F^2 > \sigma(F^2)$ is used only for calculating R -factors(gt) *etc.* and is not relevant to the choice of reflections for refinement. R -factors based on F^2 are statistically about twice as large as those based on F , and R -factors based on ALL data will be even larger.

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (\AA^2)

| | x | y | z | $U_{\text{iso}}^*/U_{\text{eq}}$ |
|------|--------------|--------------|--------------|----------------------------------|
| S1 | 1.14439 (6) | 1.02701 (4) | 0.26563 (3) | 0.05911 (14) |
| O1 | 1.1085 (2) | 0.33486 (12) | 0.78317 (9) | 0.0741 (3) |
| N1 | 0.7031 (2) | 1.05243 (12) | 0.33719 (9) | 0.0523 (3) |
| N2 | 0.82991 (18) | 0.94641 (11) | 0.39100 (8) | 0.0448 (2) |
| N3 | 1.16801 (18) | 0.81311 (11) | 0.42311 (8) | 0.0470 (2) |
| C1 | 1.2380 (2) | 0.58244 (13) | 0.57132 (10) | 0.0459 (3) |
| H1A | 1.3637 | 0.6030 | 0.5258 | 0.055* |
| C2 | 1.2769 (3) | 0.47212 (13) | 0.64264 (10) | 0.0498 (3) |
| H2A | 1.4261 | 0.4199 | 0.6442 | 0.060* |
| C3 | 1.0917 (3) | 0.44125 (14) | 0.71078 (10) | 0.0515 (3) |
| C4 | 0.8704 (3) | 0.52078 (17) | 0.70788 (12) | 0.0620 (4) |
| H4A | 0.7456 | 0.5008 | 0.7542 | 0.074* |
| C5 | 0.8345 (2) | 0.62946 (16) | 0.63658 (12) | 0.0541 (3) |
| H5A | 0.6852 | 0.6817 | 0.6354 | 0.065* |
| C6 | 1.0186 (2) | 0.66223 (12) | 0.56621 (9) | 0.0405 (2) |
| C7 | 0.9808 (2) | 0.77580 (12) | 0.48936 (9) | 0.0397 (2) |
| C8 | 0.7710 (2) | 0.85747 (13) | 0.47028 (10) | 0.0497 (3) |
| H8A | 0.6217 | 0.8533 | 0.5038 | 0.060* |
| C9 | 0.8461 (2) | 1.10357 (13) | 0.26920 (10) | 0.0473 (3) |
| C10 | 1.0665 (2) | 0.91576 (13) | 0.36590 (9) | 0.0441 (3) |
| C11 | 0.7678 (3) | 1.22647 (14) | 0.20133 (11) | 0.0571 (4) |
| H11A | 0.8370 | 1.2992 | 0.2240 | 0.069* |

| | | | | |
|------|------------|--------------|--------------|------------|
| H11B | 0.5965 | 1.2493 | 0.2123 | 0.069* |
| C12 | 0.8305 (3) | 1.21810 (16) | 0.08633 (11) | 0.0642 (4) |
| H12A | 1.0026 | 1.1905 | 0.0754 | 0.077* |
| C13 | 0.7083 (6) | 1.1178 (2) | 0.04179 (17) | 0.1097 (9) |
| H13A | 0.7482 | 1.0322 | 0.0783 | 0.165* |
| H13B | 0.7594 | 1.1119 | -0.0303 | 0.165* |
| H13C | 0.5393 | 1.1455 | 0.0489 | 0.165* |
| C14 | 0.7663 (6) | 1.3557 (2) | 0.03141 (17) | 0.1117 (9) |
| H14A | 0.8461 | 1.4183 | 0.0607 | 0.167* |
| H14B | 0.5975 | 1.3836 | 0.0400 | 0.167* |
| H14C | 0.8148 | 1.3522 | -0.0411 | 0.167* |
| C15 | 1.3354 (4) | 0.25718 (19) | 0.79377 (16) | 0.0840 (6) |
| H15A | 1.3233 | 0.1863 | 0.8467 | 0.126* |
| H15B | 1.4436 | 0.3125 | 0.8127 | 0.126* |
| H15C | 1.3931 | 0.2200 | 0.7291 | 0.126* |

Atomic displacement parameters (Å²)

| | U^{11} | U^{22} | U^{33} | U^{12} | U^{13} | U^{23} |
|-----|--------------|-------------|-------------|---------------|--------------|--------------|
| S1 | 0.04595 (19) | 0.0694 (3) | 0.0549 (2) | -0.00092 (16) | 0.00022 (14) | 0.01745 (17) |
| O1 | 0.0866 (8) | 0.0673 (7) | 0.0650 (7) | -0.0162 (6) | -0.0051 (6) | 0.0229 (5) |
| N1 | 0.0439 (6) | 0.0567 (7) | 0.0504 (6) | 0.0074 (5) | -0.0069 (5) | 0.0047 (5) |
| N2 | 0.0364 (5) | 0.0504 (6) | 0.0438 (5) | 0.0022 (4) | -0.0044 (4) | 0.0022 (4) |
| N3 | 0.0361 (5) | 0.0551 (6) | 0.0462 (6) | -0.0009 (4) | -0.0031 (4) | 0.0060 (4) |
| C1 | 0.0453 (6) | 0.0489 (7) | 0.0394 (6) | 0.0004 (5) | 0.0018 (5) | 0.0001 (5) |
| C2 | 0.0532 (7) | 0.0483 (7) | 0.0440 (6) | 0.0019 (5) | -0.0021 (5) | -0.0004 (5) |
| C3 | 0.0644 (8) | 0.0468 (7) | 0.0441 (6) | -0.0126 (6) | -0.0077 (6) | 0.0034 (5) |
| C4 | 0.0531 (8) | 0.0731 (10) | 0.0584 (8) | -0.0177 (7) | 0.0027 (6) | 0.0119 (7) |
| C5 | 0.0394 (6) | 0.0654 (8) | 0.0546 (7) | -0.0063 (6) | 0.0001 (5) | 0.0060 (6) |
| C6 | 0.0395 (6) | 0.0448 (6) | 0.0374 (5) | -0.0054 (5) | -0.0040 (4) | -0.0050 (4) |
| C7 | 0.0372 (5) | 0.0445 (6) | 0.0365 (5) | -0.0023 (4) | -0.0035 (4) | -0.0050 (4) |
| C8 | 0.0375 (6) | 0.0565 (8) | 0.0505 (7) | -0.0008 (5) | 0.0019 (5) | 0.0051 (6) |
| C9 | 0.0514 (7) | 0.0489 (7) | 0.0394 (6) | 0.0027 (5) | -0.0096 (5) | -0.0042 (5) |
| C10 | 0.0364 (6) | 0.0528 (7) | 0.0409 (6) | -0.0022 (5) | -0.0038 (4) | 0.0010 (5) |
| C11 | 0.0711 (9) | 0.0495 (7) | 0.0465 (7) | 0.0050 (6) | -0.0117 (6) | 0.0006 (5) |
| C12 | 0.0702 (10) | 0.0673 (9) | 0.0486 (8) | 0.0017 (8) | -0.0023 (7) | 0.0078 (7) |
| C13 | 0.175 (3) | 0.0923 (15) | 0.0672 (12) | -0.0145 (17) | -0.0461 (15) | -0.0119 (11) |
| C14 | 0.164 (3) | 0.0872 (14) | 0.0717 (13) | -0.0059 (16) | -0.0039 (14) | 0.0319 (11) |
| C15 | 0.1056 (16) | 0.0615 (10) | 0.0816 (12) | -0.0074 (10) | -0.0270 (11) | 0.0232 (9) |

Geometric parameters (Å, °)

| | | | |
|--------|-------------|----------|-------------|
| S1—C10 | 1.7290 (13) | C6—C7 | 1.4622 (16) |
| S1—C9 | 1.7545 (14) | C7—C8 | 1.3722 (17) |
| O1—C3 | 1.3690 (16) | C8—H8A | 0.9300 |
| O1—C15 | 1.416 (2) | C9—C11 | 1.4966 (18) |
| N1—C9 | 1.2864 (19) | C11—C12 | 1.510 (2) |
| N1—N2 | 1.3721 (14) | C11—H11A | 0.9700 |

| | | | |
|-----------|-------------|---------------|-------------|
| N2—C10 | 1.3554 (16) | C11—H11B | 0.9700 |
| N2—C8 | 1.3689 (16) | C12—C13 | 1.503 (3) |
| N3—C10 | 1.3133 (15) | C12—C14 | 1.521 (2) |
| N3—C7 | 1.3957 (16) | C12—H12A | 0.9800 |
| C1—C6 | 1.3825 (17) | C13—H13A | 0.9600 |
| C1—C2 | 1.3947 (17) | C13—H13B | 0.9600 |
| C1—H1A | 0.9300 | C13—H13C | 0.9600 |
| C2—C3 | 1.379 (2) | C14—H14A | 0.9600 |
| C2—H2A | 0.9300 | C14—H14B | 0.9600 |
| C3—C4 | 1.388 (2) | C14—H14C | 0.9600 |
| C4—C5 | 1.381 (2) | C15—H15A | 0.9600 |
| C4—H4A | 0.9300 | C15—H15B | 0.9600 |
| C5—C6 | 1.3967 (18) | C15—H15C | 0.9600 |
| C5—H5A | 0.9300 | | |
| | | | |
| C10—S1—C9 | 88.40 (6) | N3—C10—N2 | 112.76 (11) |
| C3—O1—C15 | 117.66 (14) | N3—C10—S1 | 138.85 (10) |
| C9—N1—N2 | 108.55 (11) | N2—C10—S1 | 108.39 (9) |
| C10—N2—C8 | 107.69 (10) | C9—C11—C12 | 116.36 (12) |
| C10—N2—N1 | 118.51 (11) | C9—C11—H11A | 108.2 |
| C8—N2—N1 | 133.79 (11) | C12—C11—H11A | 108.2 |
| C10—N3—C7 | 103.73 (10) | C9—C11—H11B | 108.2 |
| C6—C1—C2 | 122.09 (12) | C12—C11—H11B | 108.2 |
| C6—C1—H1A | 119.0 | H11A—C11—H11B | 107.4 |
| C2—C1—H1A | 119.0 | C13—C12—C11 | 112.03 (16) |
| C3—C2—C1 | 119.33 (13) | C13—C12—C14 | 110.67 (17) |
| C3—C2—H2A | 120.3 | C11—C12—C14 | 109.10 (15) |
| C1—C2—H2A | 120.3 | C13—C12—H12A | 108.3 |
| O1—C3—C2 | 124.39 (14) | C11—C12—H12A | 108.3 |
| O1—C3—C4 | 115.99 (14) | C14—C12—H12A | 108.3 |
| C2—C3—C4 | 119.62 (13) | C12—C13—H13A | 109.5 |
| C5—C4—C3 | 120.36 (14) | C12—C13—H13B | 109.5 |
| C5—C4—H4A | 119.8 | H13A—C13—H13B | 109.5 |
| C3—C4—H4A | 119.8 | C12—C13—H13C | 109.5 |
| C4—C5—C6 | 121.14 (14) | H13A—C13—H13C | 109.5 |
| C4—C5—H5A | 119.4 | H13B—C13—H13C | 109.5 |
| C6—C5—H5A | 119.4 | C12—C14—H14A | 109.5 |
| C1—C6—C5 | 117.46 (12) | C12—C14—H14B | 109.5 |
| C1—C6—C7 | 121.12 (11) | H14A—C14—H14B | 109.5 |
| C5—C6—C7 | 121.42 (11) | C12—C14—H14C | 109.5 |
| C8—C7—N3 | 111.03 (11) | H14A—C14—H14C | 109.5 |
| C8—C7—C6 | 127.62 (11) | H14B—C14—H14C | 109.5 |
| N3—C7—C6 | 121.35 (10) | O1—C15—H15A | 109.5 |
| N2—C8—C7 | 104.79 (11) | O1—C15—H15B | 109.5 |
| N2—C8—H8A | 127.6 | H15A—C15—H15B | 109.5 |
| C7—C8—H8A | 127.6 | O1—C15—H15C | 109.5 |
| N1—C9—C11 | 122.38 (13) | H15A—C15—H15C | 109.5 |
| N1—C9—S1 | 116.15 (10) | H15B—C15—H15C | 109.5 |

| | | | |
|--------------|--------------|----------------|--------------|
| C11—C9—S1 | 121.36 (11) | | |
| C9—N1—N2—C10 | 0.23 (17) | C10—N2—C8—C7 | 0.06 (15) |
| C9—N1—N2—C8 | 179.94 (14) | N1—N2—C8—C7 | -179.67 (13) |
| C6—C1—C2—C3 | -0.3 (2) | N3—C7—C8—N2 | 0.03 (15) |
| C15—O1—C3—C2 | 4.8 (2) | C6—C7—C8—N2 | -179.34 (11) |
| C15—O1—C3—C4 | -175.43 (15) | N2—N1—C9—C11 | -176.23 (11) |
| C1—C2—C3—O1 | 179.30 (13) | N2—N1—C9—S1 | -0.04 (15) |
| C1—C2—C3—C4 | -0.4 (2) | C10—S1—C9—N1 | -0.10 (12) |
| O1—C3—C4—C5 | -179.11 (14) | C10—S1—C9—C11 | 176.13 (11) |
| C2—C3—C4—C5 | 0.7 (2) | C7—N3—C10—N2 | 0.14 (15) |
| C3—C4—C5—C6 | -0.2 (2) | C7—N3—C10—S1 | -179.92 (13) |
| C2—C1—C6—C5 | 0.7 (2) | C8—N2—C10—N3 | -0.13 (16) |
| C2—C1—C6—C7 | -178.73 (12) | N1—N2—C10—N3 | 179.65 (11) |
| C4—C5—C6—C1 | -0.5 (2) | C8—N2—C10—S1 | 179.92 (9) |
| C4—C5—C6—C7 | 178.95 (13) | N1—N2—C10—S1 | -0.31 (15) |
| C10—N3—C7—C8 | -0.10 (14) | C9—S1—C10—N3 | -179.73 (16) |
| C10—N3—C7—C6 | 179.31 (11) | C9—S1—C10—N2 | 0.21 (10) |
| C1—C6—C7—C8 | 175.00 (13) | N1—C9—C11—C12 | -129.32 (16) |
| C5—C6—C7—C8 | -4.5 (2) | S1—C9—C11—C12 | 54.69 (18) |
| C1—C6—C7—N3 | -4.31 (18) | C9—C11—C12—C13 | 64.3 (2) |
| C5—C6—C7—N3 | 176.24 (12) | C9—C11—C12—C14 | -172.81 (17) |

Hydrogen-bond geometry (\AA , $^\circ$)

Cg3 is the centroid of the C1—C6 benzene ring.

| <i>D</i> —H \cdots <i>A</i> | <i>D</i> —H | H \cdots <i>A</i> | <i>D</i> \cdots <i>A</i> | <i>D</i> —H \cdots <i>A</i> |
|--|-------------|---------------------|----------------------------|-------------------------------|
| C11—H11 <i>A</i> \cdots Cg3 ⁱ | 0.97 | 2.60 | 3.5063 (16) | 155 |

Symmetry code: (i) $-x+2, -y+2, -z+1$.