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## Structure Reports

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# N-(4,6-Dimethylpyrimidin-2-yl)-1H-benzimidazol-2-amine

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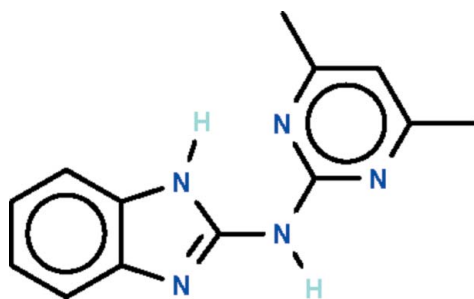
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 Key indicators: single-crystal X-ray study;  $T = 295$  K; mean  $\sigma(\text{C}-\text{C}) = 0.003$  Å;  $R$  factor = 0.044;  $wR$  factor = 0.136; data-to-parameter ratio = 16.1.

There are two independent molecules in the asymmetric unit of the title compound,  $\text{C}_{13}\text{H}_{13}\text{N}_5$ . In each molecule, an amino N atom is connected to a benzimidazole fused-ring system and a pyrimidine ring [these are aligned at  $1.3$  (1)° in one independent molecule and at  $5.4$  (1)° in the other]. The amino N atom of the fused ring forms an intramolecular  $\text{N}-\text{H}\cdots\text{O}$  hydrogen bond to a pyrimidine N atom in each molecule. The amino N atom connecting the two ring systems interacts with the other N atom of the pyrimidine ring of an adjacent molecule, generating centrosymmetric hydrogen-bonded dimers.

## Related literature

 For the synthesis, see: Bossio *et al.* (1985); Shestakov *et al.* (2006).


## Experimental

## Crystal data

$\text{C}_{13}\text{H}_{13}\text{N}_5$	$\gamma = 112.597$ (1)°
$M_r = 239.28$	$V = 1206.94$ (10) Å <sup>3</sup>
Triclinic, $P\bar{1}$	$Z = 4$
$a = 8.2836$ (4) Å	Mo $K\alpha$ radiation
$b = 9.6135$ (5) Å	$\mu = 0.09$ mm <sup>-1</sup>
$c = 16.5694$ (8) Å	$T = 295$ K
$\alpha = 92.121$ (1)°	$0.30 \times 0.20 \times 0.20$ mm
$\beta = 96.100$ (1)°	

## Data collection

Bruker APEXII diffractometer	4124 reflections with $I > 2\sigma(I)$
13373 measured reflections	$R_{\text{int}} = 0.019$
5550 independent reflections	

## Refinement

$R[F^2 > 2\sigma(F^2)] = 0.044$	H atoms treated by a mixture of independent and constrained refinement
$wR(F^2) = 0.136$	$\Delta\rho_{\text{max}} = 0.25$ e Å <sup>-3</sup>
$S = 1.01$	$\Delta\rho_{\text{min}} = -0.18$ e Å <sup>-3</sup>
5550 reflections	
345 parameters	
4 restraints	

Table 1

Hydrogen-bond geometry (Å, °).

$D-\text{H}\cdots A$	$D-\text{H}$	$\text{H}\cdots A$	$D\cdots A$	$D-\text{H}\cdots A$
$\text{N1}-\text{H1}\cdots\text{N5}$	0.86 (1)	2.05 (2)	2.664 (2)	128 (2)
$\text{N3}-\text{H3}\cdots\text{N2}^i$	0.87 (1)	2.05 (1)	2.912 (2)	170 (2)
$\text{N7}-\text{H7}\cdots\text{N10}$	0.87 (1)	2.10 (2)	2.695 (2)	125 (2)
$\text{N8}-\text{H8}\cdots\text{N6}^ii$	0.87 (1)	2.05 (1)	2.908 (2)	170 (2)

 Symmetry codes: (i)  $-x + 2, -y + 1, -z + 1$ ; (ii)  $-x + 2, -y + 2, -z + 2$ .

Data collection: *APEX2* (Bruker, 2005); cell refinement: *SAINT* (Bruker, 2005); data reduction: *SAINT*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 2008); program(s) used to refine structure: *SHELXL97* (Sheldrick, 2008); molecular graphics: *X-SEED* (Barbour, 2001); software used to prepare material for publication: *pubCIF* (Westrip, 2010).

We thank Manchester Metropolitan University, Baku State University and the University of Malaya for supporting this study.

Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: HG5003).

## References

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## supporting information

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***N*-(4,6-Dimethylpyrimidin-2-yl)-1*H*-benzimidazol-2-amine**

Shaaban Kamel Mohamed, Mahmoud A. A. El-Remaily, Atash V. Gurbanov, Ali N. Khalilov and Seik Weng Ng

**S1. Comment**

The reported synthesis involves the reaction of 4,6-dimethylpyrimidin-2-ylcyanamide with *o*-phenylenediamine, and it illustrates the type of heterocycles that are formed by the reaction of cyanamides with *N,N*-binucleophiles (Shestakov *et al.*, 2006). The unsubstituted compound was reported earlier (Bossio *et al.*, 1985). The present synthesis is a more convenient synthesis that uses acetylacetone as one of the reactants. An amino N atom in the approximately planar C<sub>13</sub>H<sub>13</sub>N<sub>5</sub> molecule is connected to a benzimidazol fused-ring and a pyrimidin ring; the amino N atom of the fused ring forms an intramolecular N–H···O hydrogen bond to a pyrimidin N atom (Scheme I, Fig. 1). There are two independent molecules; each molecule is connected to an inversion-related molecule by an N–H···O hydrogen bond.

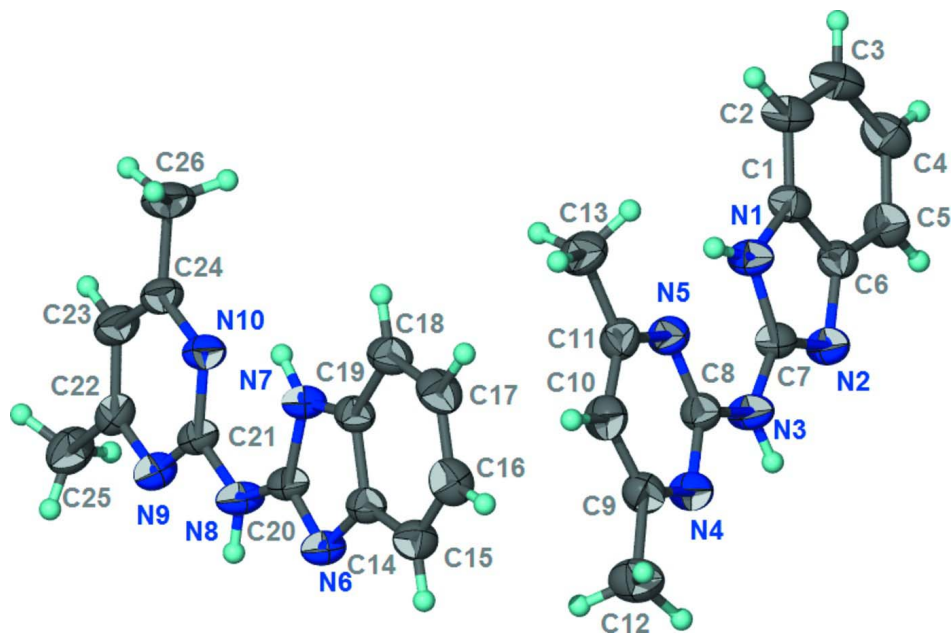
**S2. Experimental**

1*H*-Benzo[*d*]imidazol-2-yliminomethanediamine (0.05 mol) and acetylacetone (0.10 mol, approx. 10 ml) along with several drops of acetic acid were heated at 473 for 1 h. The solid that formed on cooling was collected and recrystallized from ethanol to give the title compound in 80% yield; m.p. 623 K.

**S3. Refinement**

Carbon-bound H-atoms were placed in calculated positions [C–H 0.93 to 0.96 Å; *U*(H) 1.2 to 1.5*U*(C)] and were included in the refinement in the riding model approximation.

The amino H-atoms were located in a difference Fourier map, and were refined with a distance restraint of N–H 0.86±0.01 Å; their temperature factors were refined.

**Figure 1**

Thermal ellipsoid plot (Barbour, 2001) of the two independent molecules of  $C_{13}H_{13}N_5$  at the 50% probability level; hydrogen atoms are drawn as spheres of arbitrary radius.

### *N*-(4,6-Dimethylpyrimidin-2-yl)-1*H*-benzimidazol-2-amine

#### Crystal data

$C_{13}H_{13}N_5$   
 $M_r = 239.28$   
 Triclinic,  $P\bar{1}$   
 Hall symbol:  $-P\ 1$   
 $a = 8.2836\ (4)\ \text{\AA}$   
 $b = 9.6135\ (5)\ \text{\AA}$   
 $c = 16.5694\ (8)\ \text{\AA}$   
 $\alpha = 92.121\ (1)^\circ$   
 $\beta = 96.100\ (1)^\circ$   
 $\gamma = 112.597\ (1)^\circ$   
 $V = 1206.94\ (10)\ \text{\AA}^3$

$Z = 4$   
 $F(000) = 504$   
 $D_x = 1.317\ \text{Mg m}^{-3}$   
 Mo  $K\alpha$  radiation,  $\lambda = 0.71073\ \text{\AA}$   
 Cell parameters from 4428 reflections  
 $\theta = 2.3\text{--}27.4^\circ$   
 $\mu = 0.09\ \text{mm}^{-1}$   
 $T = 295\ \text{K}$   
 Prism, colorless  
 $0.30 \times 0.20 \times 0.20\ \text{mm}$

#### Data collection

Bruker APEXII  
 diffractometer  
 Radiation source: fine-focus sealed tube  
 Graphite monochromator  
 $\varphi$  and  $\omega$  scans  
 13373 measured reflections  
 5550 independent reflections

4124 reflections with  $I > 2\sigma(I)$   
 $R_{\text{int}} = 0.019$   
 $\theta_{\text{max}} = 27.5^\circ$ ,  $\theta_{\text{min}} = 1.2^\circ$   
 $h = -10 \rightarrow 10$   
 $k = -12 \rightarrow 12$   
 $l = -21 \rightarrow 21$

#### Refinement

Refinement on  $F^2$   
 Least-squares matrix: full  
 $R[F^2 > 2\sigma(F^2)] = 0.044$   
 $wR(F^2) = 0.136$

$S = 1.01$   
 5550 reflections  
 345 parameters  
 4 restraints

Primary atom site location: structure-invariant  
direct methods  
Secondary atom site location: difference Fourier  
map  
Hydrogen site location: inferred from  
neighbouring sites

H atoms treated by a mixture of independent  
and constrained refinement  
 $w = 1/[\sigma^2(F_o^2) + (0.0736P)^2 + 0.1933P]$   
where  $P = (F_o^2 + 2F_c^2)/3$   
 $(\Delta/\sigma)_{\max} = 0.001$   
 $\Delta\rho_{\max} = 0.25 \text{ e } \text{\AA}^{-3}$   
 $\Delta\rho_{\min} = -0.18 \text{ e } \text{\AA}^{-3}$

*Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters ( $\text{\AA}^2$ )*

	<i>x</i>	<i>y</i>	<i>z</i>	$U_{\text{iso}}^*/U_{\text{eq}}$
N1	0.66435 (17)	0.61916 (15)	0.45010 (8)	0.0480 (3)
N2	0.80016 (16)	0.45944 (14)	0.43969 (8)	0.0470 (3)
N3	0.94151 (17)	0.67635 (15)	0.53228 (8)	0.0498 (3)
N4	1.11489 (17)	0.88346 (14)	0.61816 (8)	0.0491 (3)
N5	0.83527 (16)	0.86534 (14)	0.55242 (8)	0.0477 (3)
N6	0.98905 (15)	1.11808 (14)	0.92502 (8)	0.0453 (3)
N7	0.72463 (16)	1.09201 (14)	0.86230 (8)	0.0452 (3)
N8	0.74067 (16)	0.92463 (15)	0.96388 (8)	0.0485 (3)
N9	0.52809 (16)	0.73369 (14)	1.01630 (8)	0.0477 (3)
N10	0.44770 (16)	0.87329 (14)	0.91397 (8)	0.0464 (3)
C1	0.5574 (2)	0.50529 (17)	0.39234 (9)	0.0453 (3)
C2	0.3985 (2)	0.48064 (19)	0.34608 (11)	0.0570 (4)
H2	0.3415	0.5464	0.3516	0.068*
C3	0.3283 (2)	0.3542 (2)	0.29138 (11)	0.0648 (5)
H3A	0.2219	0.3343	0.2590	0.078*
C4	0.4132 (2)	0.2562 (2)	0.28378 (11)	0.0646 (5)
H4	0.3630	0.1726	0.2458	0.077*
C5	0.5706 (2)	0.27942 (19)	0.33104 (10)	0.0566 (4)
H5	0.6258	0.2122	0.3261	0.068*
C6	0.64326 (19)	0.40638 (17)	0.38613 (9)	0.0446 (3)
C7	0.80551 (19)	0.58625 (16)	0.47531 (9)	0.0433 (3)
C8	0.96246 (19)	0.81387 (17)	0.56914 (9)	0.0444 (3)
C9	1.1398 (2)	1.01836 (18)	0.65380 (10)	0.0502 (4)
C10	1.0141 (2)	1.07950 (19)	0.64206 (10)	0.0541 (4)
H10	1.0317	1.1723	0.6684	0.065*
C11	0.8619 (2)	0.99939 (18)	0.59025 (10)	0.0492 (4)
C12	1.3142 (2)	1.1021 (2)	0.70528 (13)	0.0679 (5)
H12A	1.3969	1.0598	0.6921	0.102*
H12B	1.2994	1.0930	0.7618	0.102*
H12C	1.3577	1.2069	0.6949	0.102*
C13	0.7188 (2)	1.0572 (2)	0.57189 (12)	0.0638 (5)
H13A	0.6895	1.0528	0.5140	0.096*
H13B	0.7591	1.1600	0.5946	0.096*
H13C	0.6163	0.9959	0.5953	0.096*
C14	1.01079 (19)	1.22562 (16)	0.86868 (9)	0.0423 (3)
C15	1.1636 (2)	1.33597 (18)	0.84877 (10)	0.0510 (4)
H15	1.2738	1.3468	0.8743	0.061*
C16	1.1476 (2)	1.42941 (19)	0.78990 (10)	0.0563 (4)

H16	1.2488	1.5041	0.7758	0.068*
C17	0.9839 (2)	1.41438 (19)	0.75126 (10)	0.0569 (4)
H17	0.9777	1.4789	0.7117	0.068*
C18	0.8299 (2)	1.30521 (19)	0.77048 (10)	0.0525 (4)
H18	0.7199	1.2954	0.7451	0.063*
C19	0.84684 (19)	1.21133 (16)	0.82922 (9)	0.0433 (3)
C20	0.81693 (18)	1.04241 (16)	0.91841 (9)	0.0417 (3)
C21	0.56341 (18)	0.84116 (16)	0.96416 (9)	0.0433 (3)
C22	0.3584 (2)	0.64850 (18)	1.01769 (10)	0.0489 (4)
C23	0.2285 (2)	0.67368 (19)	0.96860 (11)	0.0554 (4)
H23	0.1101	0.6147	0.9705	0.066*
C24	0.27647 (19)	0.78683 (18)	0.91699 (10)	0.0494 (4)
C25	0.3156 (2)	0.5228 (2)	1.07291 (12)	0.0647 (5)
H25A	0.4053	0.5504	1.1191	0.097*
H25B	0.3105	0.4326	1.0439	0.097*
H25C	0.2036	0.5048	1.0911	0.097*
C26	0.1440 (2)	0.8194 (2)	0.86077 (12)	0.0683 (5)
H26A	0.1696	0.9257	0.8660	0.102*
H26B	0.0280	0.7643	0.8746	0.102*
H26C	0.1495	0.7889	0.8056	0.102*
H1	0.653 (2)	0.6974 (15)	0.4721 (10)	0.062 (5)*
H3	1.0279 (19)	0.647 (2)	0.5423 (11)	0.067 (5)*
H7	0.6096 (12)	1.0539 (19)	0.8546 (11)	0.063 (5)*
H8	0.811 (2)	0.902 (2)	0.9979 (9)	0.059 (5)*

*Atomic displacement parameters (Å<sup>2</sup>)*

	$U^{11}$	$U^{22}$	$U^{33}$	$U^{12}$	$U^{13}$	$U^{23}$
N1	0.0468 (7)	0.0444 (7)	0.0547 (8)	0.0227 (6)	-0.0035 (6)	0.0018 (6)
N2	0.0440 (7)	0.0451 (7)	0.0521 (7)	0.0199 (6)	-0.0010 (6)	-0.0007 (6)
N3	0.0447 (7)	0.0479 (7)	0.0585 (8)	0.0243 (6)	-0.0064 (6)	-0.0042 (6)
N4	0.0465 (7)	0.0477 (7)	0.0525 (7)	0.0203 (6)	-0.0013 (6)	-0.0007 (6)
N5	0.0470 (7)	0.0495 (7)	0.0504 (7)	0.0239 (6)	0.0032 (6)	0.0015 (6)
N6	0.0374 (6)	0.0460 (7)	0.0527 (7)	0.0178 (5)	-0.0004 (5)	0.0092 (5)
N7	0.0379 (6)	0.0476 (7)	0.0497 (7)	0.0186 (6)	-0.0035 (5)	0.0055 (5)
N8	0.0357 (6)	0.0484 (7)	0.0606 (8)	0.0167 (5)	-0.0021 (6)	0.0146 (6)
N9	0.0400 (6)	0.0491 (7)	0.0567 (8)	0.0203 (6)	0.0054 (6)	0.0072 (6)
N10	0.0383 (6)	0.0493 (7)	0.0498 (7)	0.0179 (5)	-0.0032 (5)	-0.0003 (6)
C1	0.0457 (8)	0.0438 (8)	0.0443 (8)	0.0157 (6)	0.0015 (6)	0.0101 (6)
C2	0.0526 (9)	0.0554 (9)	0.0616 (10)	0.0228 (8)	-0.0072 (8)	0.0119 (8)
C3	0.0567 (10)	0.0670 (11)	0.0595 (10)	0.0173 (9)	-0.0133 (8)	0.0074 (9)
C4	0.0641 (11)	0.0588 (10)	0.0566 (10)	0.0128 (9)	-0.0057 (8)	-0.0046 (8)
C5	0.0571 (10)	0.0504 (9)	0.0582 (10)	0.0182 (8)	0.0030 (8)	-0.0021 (7)
C6	0.0438 (8)	0.0451 (8)	0.0438 (8)	0.0160 (6)	0.0047 (6)	0.0079 (6)
C7	0.0415 (7)	0.0440 (8)	0.0456 (8)	0.0189 (6)	0.0020 (6)	0.0058 (6)
C8	0.0442 (8)	0.0441 (8)	0.0458 (8)	0.0187 (6)	0.0041 (6)	0.0032 (6)
C9	0.0507 (9)	0.0489 (8)	0.0486 (8)	0.0178 (7)	0.0029 (7)	0.0017 (7)
C10	0.0586 (10)	0.0495 (9)	0.0560 (9)	0.0245 (8)	0.0055 (8)	-0.0047 (7)

C11	0.0523 (9)	0.0510 (8)	0.0501 (8)	0.0259 (7)	0.0095 (7)	0.0035 (7)
C12	0.0601 (11)	0.0584 (10)	0.0767 (12)	0.0205 (9)	-0.0115 (9)	-0.0104 (9)
C13	0.0617 (11)	0.0672 (11)	0.0729 (12)	0.0382 (9)	0.0051 (9)	-0.0016 (9)
C14	0.0429 (7)	0.0425 (7)	0.0427 (7)	0.0196 (6)	0.0004 (6)	0.0015 (6)
C15	0.0464 (8)	0.0503 (9)	0.0527 (9)	0.0161 (7)	0.0018 (7)	0.0046 (7)
C16	0.0619 (10)	0.0485 (9)	0.0538 (9)	0.0158 (8)	0.0094 (8)	0.0069 (7)
C17	0.0740 (11)	0.0530 (9)	0.0490 (9)	0.0308 (9)	0.0052 (8)	0.0115 (7)
C18	0.0580 (9)	0.0567 (9)	0.0482 (9)	0.0306 (8)	-0.0026 (7)	0.0052 (7)
C19	0.0472 (8)	0.0428 (8)	0.0418 (7)	0.0218 (6)	-0.0007 (6)	-0.0004 (6)
C20	0.0381 (7)	0.0419 (7)	0.0466 (8)	0.0193 (6)	-0.0016 (6)	0.0028 (6)
C21	0.0370 (7)	0.0434 (8)	0.0503 (8)	0.0181 (6)	0.0009 (6)	-0.0005 (6)
C22	0.0431 (8)	0.0494 (8)	0.0560 (9)	0.0194 (7)	0.0101 (7)	0.0004 (7)
C23	0.0370 (8)	0.0599 (10)	0.0646 (10)	0.0145 (7)	0.0051 (7)	0.0024 (8)
C24	0.0372 (7)	0.0552 (9)	0.0527 (9)	0.0176 (7)	-0.0023 (6)	-0.0058 (7)
C25	0.0547 (10)	0.0660 (11)	0.0781 (12)	0.0242 (9)	0.0201 (9)	0.0210 (9)
C26	0.0422 (9)	0.0854 (13)	0.0718 (12)	0.0228 (9)	-0.0082 (8)	0.0089 (10)

*Geometric parameters (Å, °)*

N1—C7	1.3535 (18)	C5—H5	0.9300
N1—C1	1.377 (2)	C9—C10	1.380 (2)
N1—H1	0.86 (1)	C9—C12	1.502 (2)
N2—C7	1.3176 (19)	C10—C11	1.376 (2)
N2—C6	1.3960 (19)	C10—H10	0.9300
N3—C7	1.3689 (19)	C11—C13	1.499 (2)
N3—C8	1.3756 (19)	C12—H12A	0.9600
N3—H3	0.87 (1)	C12—H12B	0.9600
N4—C9	1.336 (2)	C12—H12C	0.9600
N4—C8	1.3396 (19)	C13—H13A	0.9600
N5—C8	1.3343 (18)	C13—H13B	0.9600
N5—C11	1.340 (2)	C13—H13C	0.9600
N6—C20	1.3183 (18)	C14—C15	1.385 (2)
N6—C14	1.3917 (19)	C14—C19	1.397 (2)
N7—C20	1.3558 (18)	C15—C16	1.380 (2)
N7—C19	1.383 (2)	C15—H15	0.9300
N7—H7	0.87 (1)	C16—C17	1.388 (2)
N8—C20	1.3650 (19)	C16—H16	0.9300
N8—C21	1.3771 (19)	C17—C18	1.383 (2)
N8—H8	0.87 (1)	C17—H17	0.9300
N9—C22	1.3310 (19)	C18—C19	1.384 (2)
N9—C21	1.3367 (19)	C18—H18	0.9300
N10—C21	1.3352 (18)	C22—C23	1.380 (2)
N10—C24	1.3481 (19)	C22—C25	1.498 (2)
C1—C2	1.382 (2)	C23—C24	1.372 (2)
C1—C6	1.396 (2)	C23—H23	0.9300
C2—C3	1.378 (3)	C24—C26	1.496 (2)
C2—H2	0.9300	C25—H25A	0.9600
C3—C4	1.384 (3)	C25—H25B	0.9600

C3—H3A	0.9300	C25—H25C	0.9600
C4—C5	1.382 (2)	C26—H26A	0.9600
C4—H4	0.9300	C26—H26B	0.9600
C5—C6	1.386 (2)	C26—H26C	0.9600
C7—N1—C1	106.87 (13)	C9—C12—H12C	109.5
C7—N1—H1	120.2 (12)	H12A—C12—H12C	109.5
C1—N1—H1	132.8 (12)	H12B—C12—H12C	109.5
C7—N2—C6	104.12 (12)	C11—C13—H13A	109.5
C7—N3—C8	126.91 (13)	C11—C13—H13B	109.5
C7—N3—H3	115.8 (13)	H13A—C13—H13B	109.5
C8—N3—H3	117.0 (13)	C11—C13—H13C	109.5
C9—N4—C8	115.66 (13)	H13A—C13—H13C	109.5
C8—N5—C11	116.23 (13)	H13B—C13—H13C	109.5
C20—N6—C14	104.20 (12)	C15—C14—N6	129.94 (13)
C20—N7—C19	106.69 (12)	C15—C14—C19	119.88 (14)
C20—N7—H7	121.9 (12)	N6—C14—C19	110.18 (13)
C19—N7—H7	131.3 (12)	C16—C15—C14	118.09 (15)
C20—N8—C21	127.59 (13)	C16—C15—H15	121.0
C20—N8—H8	116.5 (13)	C14—C15—H15	121.0
C21—N8—H8	115.9 (12)	C15—C16—C17	121.51 (16)
C22—N9—C21	116.15 (13)	C15—C16—H16	119.2
C21—N10—C24	115.57 (14)	C17—C16—H16	119.2
N1—C1—C2	132.25 (15)	C18—C17—C16	121.25 (15)
N1—C1—C6	105.39 (13)	C18—C17—H17	119.4
C2—C1—C6	122.37 (15)	C16—C17—H17	119.4
C3—C2—C1	116.96 (16)	C17—C18—C19	116.95 (15)
C3—C2—H2	121.5	C17—C18—H18	121.5
C1—C2—H2	121.5	C19—C18—H18	121.5
C2—C3—C4	121.27 (16)	N7—C19—C18	132.51 (14)
C2—C3—H3A	119.4	N7—C19—C14	105.17 (12)
C4—C3—H3A	119.4	C18—C19—C14	122.32 (15)
C5—C4—C3	121.77 (16)	N6—C20—N7	113.75 (13)
C5—C4—H4	119.1	N6—C20—N8	122.52 (13)
C3—C4—H4	119.1	N7—C20—N8	123.73 (13)
C4—C5—C6	117.67 (16)	N10—C21—N9	127.34 (13)
C4—C5—H5	121.2	N10—C21—N8	118.60 (14)
C6—C5—H5	121.2	N9—C21—N8	114.06 (12)
C5—C6—C1	119.94 (14)	N9—C22—C23	120.91 (15)
C5—C6—N2	130.14 (14)	N9—C22—C25	117.17 (14)
C1—C6—N2	109.91 (13)	C23—C22—C25	121.90 (15)
N2—C7—N1	113.70 (13)	C24—C23—C22	119.11 (14)
N2—C7—N3	123.06 (13)	C24—C23—H23	120.4
N1—C7—N3	123.24 (14)	C22—C23—H23	120.4
N5—C8—N4	126.91 (14)	N10—C24—C23	120.91 (14)
N5—C8—N3	118.65 (13)	N10—C24—C26	116.69 (15)
N4—C8—N3	114.42 (13)	C23—C24—C26	122.39 (15)
N4—C9—C10	121.78 (15)	C22—C25—H25A	109.5

N4—C9—C12	116.46 (14)	C22—C25—H25B	109.5
C10—C9—C12	121.74 (15)	H25A—C25—H25B	109.5
C11—C10—C9	118.16 (15)	C22—C25—H25C	109.5
C11—C10—H10	120.9	H25A—C25—H25C	109.5
C9—C10—H10	120.9	H25B—C25—H25C	109.5
N5—C11—C10	121.22 (14)	C24—C26—H26A	109.5
N5—C11—C13	116.37 (15)	C24—C26—H26B	109.5
C10—C11—C13	122.40 (15)	H26A—C26—H26B	109.5
C9—C12—H12A	109.5	C24—C26—H26C	109.5
C9—C12—H12B	109.5	H26A—C26—H26C	109.5
H12A—C12—H12B	109.5	H26B—C26—H26C	109.5
C7—N1—C1—C2	-179.49 (17)	C20—N6—C14—C15	-179.64 (15)
C7—N1—C1—C6	0.26 (16)	C20—N6—C14—C19	0.28 (16)
N1—C1—C2—C3	178.38 (17)	N6—C14—C15—C16	-179.89 (15)
C6—C1—C2—C3	-1.3 (2)	C19—C14—C15—C16	0.2 (2)
C1—C2—C3—C4	0.4 (3)	C14—C15—C16—C17	0.0 (2)
C2—C3—C4—C5	0.7 (3)	C15—C16—C17—C18	0.2 (3)
C3—C4—C5—C6	-0.9 (3)	C16—C17—C18—C19	-0.6 (2)
C4—C5—C6—C1	0.0 (2)	C20—N7—C19—C18	-179.50 (16)
C4—C5—C6—N2	-178.51 (16)	C20—N7—C19—C14	0.38 (16)
N1—C1—C6—C5	-178.65 (14)	C17—C18—C19—N7	-179.37 (15)
C2—C1—C6—C5	1.1 (2)	C17—C18—C19—C14	0.8 (2)
N1—C1—C6—N2	0.17 (17)	C15—C14—C19—N7	179.52 (13)
C2—C1—C6—N2	179.95 (14)	N6—C14—C19—N7	-0.42 (16)
C7—N2—C6—C5	178.13 (16)	C15—C14—C19—C18	-0.6 (2)
C7—N2—C6—C1	-0.53 (16)	N6—C14—C19—C18	179.48 (14)
C6—N2—C7—N1	0.72 (17)	C14—N6—C20—N7	-0.03 (17)
C6—N2—C7—N3	-179.09 (14)	C14—N6—C20—N8	179.56 (13)
C1—N1—C7—N2	-0.64 (18)	C19—N7—C20—N6	-0.23 (17)
C1—N1—C7—N3	179.17 (14)	C19—N7—C20—N8	-179.82 (14)
C8—N3—C7—N2	176.24 (14)	C21—N8—C20—N6	177.63 (14)
C8—N3—C7—N1	-3.6 (3)	C21—N8—C20—N7	-2.8 (3)
C11—N5—C8—N4	-1.6 (2)	C24—N10—C21—N9	0.1 (2)
C11—N5—C8—N3	179.66 (14)	C24—N10—C21—N8	179.65 (13)
C9—N4—C8—N5	0.2 (2)	C22—N9—C21—N10	0.6 (2)
C9—N4—C8—N3	179.02 (13)	C22—N9—C21—N8	-178.95 (13)
C7—N3—C8—N5	3.8 (2)	C20—N8—C21—N10	1.4 (2)
C7—N3—C8—N4	-175.12 (15)	C20—N8—C21—N9	-179.03 (14)
C8—N4—C9—C10	1.5 (2)	C21—N9—C22—C23	-1.1 (2)
C8—N4—C9—C12	-176.69 (15)	C21—N9—C22—C25	177.35 (14)
N4—C9—C10—C11	-1.8 (3)	N9—C22—C23—C24	1.0 (2)
C12—C9—C10—C11	176.30 (16)	C25—C22—C23—C24	-177.42 (16)
C8—N5—C11—C10	1.2 (2)	C21—N10—C24—C23	-0.3 (2)
C8—N5—C11—C13	-179.73 (14)	C21—N10—C24—C26	-179.39 (14)
C9—C10—C11—N5	0.4 (3)	C22—C23—C24—N10	-0.2 (2)
C9—C10—C11—C13	-178.64 (16)	C22—C23—C24—C26	178.81 (16)



*Hydrogen-bond geometry (Å, °)*

<i>D</i> —H $\cdots$ <i>A</i>	<i>D</i> —H	H $\cdots$ <i>A</i>	<i>D</i> $\cdots$ <i>A</i>	<i>D</i> —H $\cdots$ <i>A</i>
N1—H1 $\cdots$ N5	0.86 (1)	2.05 (2)	2.664 (2)	128 (2)
N3—H3 $\cdots$ N2 <sup>i</sup>	0.87 (1)	2.05 (1)	2.912 (2)	170 (2)
N7—H7 $\cdots$ N10	0.87 (1)	2.10 (2)	2.695 (2)	125 (2)
N8—H8 $\cdots$ N6 <sup>ii</sup>	0.87 (1)	2.05 (1)	2.908 (2)	170 (2)

Symmetry codes: (i)  $-x+2, -y+1, -z+1$ ; (ii)  $-x+2, -y+2, -z+2$ .