

Acta Crystallographica Section E

## Structure Reports

Online

ISSN 1600-5368

## 3,4-Dimethyl-1-phenylpyrano[2,3-*c*]-pyrazol-6(1*H*)-one

 Neman Ahmad,<sup>a</sup> M. Nawaz Tahir,<sup>b\*</sup> Misbahul Ain Khan,<sup>a</sup> Abdul Qayyum Ather<sup>c</sup> and Muhammad Naeem Khan<sup>c</sup>
<sup>a</sup>Department of Chemistry, Islamia University, Bahawalpur, Pakistan, <sup>b</sup>Department of Physics, University of Sargodha, Sargodha, Pakistan, and <sup>c</sup>Applied Chemistry Research Center, PCSIR Laboratories Complex, Lahore 54600, Pakistan  
Correspondence e-mail: dmntahir\_uos@yahoo.com

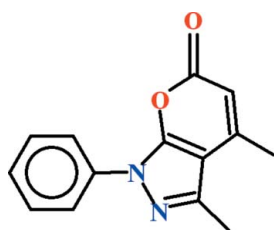
Received 26 March 2011; accepted 27 March 2011

 Key indicators: single-crystal X-ray study;  $T = 296$  K; mean  $\sigma(\text{C}-\text{C}) = 0.003$  Å;  $R$  factor = 0.046;  $wR$  factor = 0.140; data-to-parameter ratio = 13.2.

In the title compound,  $\text{C}_{14}\text{H}_{12}\text{N}_2\text{O}_2$ , the dihedral angle between the phenyl ring and the 3,4-dimethylpyrano[2,3-*c*]-pyrazol-6(1*H*)-one system is  $7.28(6)^\circ$ . An intramolecular  $\text{C}-\text{H}\cdots\text{O}$  interaction generates an  $S(6)$  ring. In the crystal, the molecules are linked by  $\text{C}-\text{H}\cdots\text{O}$  hydrogen bonds, forming  $C(8)$  chains.  $\text{C}-\text{H}\cdots\pi$  and  $\pi-\pi$  interactions [centroid-centroid separation =  $3.6374(12)$  Å] further consolidate the packing.

### Related literature

For a related structure, see: Ramsay & Steel (1985). For background to the pyrano[2,3-*c*]pyrazol-6-one ring system, see: Abdallah & Zaki (1999); Huang *et al.* (1992); Khan *et al.* (1982); Kuo *et al.* (1984); Ramsay & Steel (1985); Samaritoni *et al.* (2007). For graph-set notation, see: Bernstein *et al.* (1995).



### Experimental

#### Crystal data

$\text{C}_{14}\text{H}_{12}\text{N}_2\text{O}_2$	$V = 2401.9(3)$ Å <sup>3</sup>
$M_r = 240.26$	$Z = 8$
Monoclinic, $C2/c$	Mo $K\alpha$ radiation
$a = 15.1231(9)$ Å	$\mu = 0.09$ mm <sup>-1</sup>
$b = 13.3558(8)$ Å	$T = 296$ K
$c = 13.8684(8)$ Å	$0.35 \times 0.25 \times 0.25$ mm
$\beta = 120.965(2)^\circ$	

#### Data collection

Bruker Kappa APEXII CCD diffractometer	9214 measured reflections
Absorption correction: multi-scan ( <i>SADABS</i> ; Bruker, 2005)	2171 independent reflections
$T_{\min} = 0.975$ , $T_{\max} = 0.982$	1382 reflections with $I > 2\sigma(I)$
	$R_{\text{int}} = 0.032$

#### Refinement

$R[F^2 > 2\sigma(F^2)] = 0.046$	165 parameters
$wR(F^2) = 0.140$	H-atom parameters constrained
$S = 1.02$	$\Delta\rho_{\text{max}} = 0.15$ e Å <sup>-3</sup>
2171 reflections	$\Delta\rho_{\text{min}} = -0.19$ e Å <sup>-3</sup>

#### Table 1

Hydrogen-bond geometry (Å, °).

Cg3 is the centroid of the C1–C6 phenyl ring.

$D-H\cdots A$	$D-H$	$H\cdots A$	$D\cdots A$	$D-H\cdots A$
$\text{C3}-\text{H3}\cdots\text{O2}^i$	0.93	2.51	3.407 (3)	163
$\text{C6}-\text{H6}\cdots\text{O1}$	0.93	2.29	2.938 (3)	126
$\text{C14}-\text{H14C}\cdots\text{Cg3}^{ii}$	0.96	2.75	3.506 (2)	136

 Symmetry codes: (i)  $-x + \frac{1}{2}, y - \frac{1}{2}, -z + \frac{1}{2}$ ; (ii)  $-x, -y + 1, -z$ .

Data collection: *APEX2* (Bruker, 2009); cell refinement: *SAINT* (Bruker, 2009); data reduction: *SAINT*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 2008); program(s) used to refine structure: *SHELXL97* (Sheldrick, 2008); molecular graphics: *ORTEP-3 for Windows* (Farrugia, 1997) and *PLATON* (Spek, 2009); software used to prepare material for publication: *WinGX* (Farrugia, 1999) and *PLATON*.

The authors acknowledge the provision of funds for the purchase of the diffractometer and encouragement by Dr Muhammad Akram Chaudhary, Vice Chancellor, University of Sargodha, Pakistan. The authors also acknowledge the technical support provided by Bana International, Karachi, Pakistan.

Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: HB5827).

### References

- Abdallah, N. A. & Zaki, M. E. A. (1999). *Acta Pharm.* **49**, 159–170.  
 Bernstein, J., Davis, R. E., Shimon, L. & Chang, N.-L. (1995). *Angew. Chem. Int. Ed. Engl.* **34**, 1555–1573.  
 Bruker (2005). *SADABS*. Bruker AXS Inc., Madison, Wisconsin, USA.  
 Bruker (2009). *APEX2* and *SAINT*. Bruker AXS Inc., Madison, Wisconsin, USA.  
 Farrugia, L. J. (1997). *J. Appl. Cryst.* **30**, 565.  
 Farrugia, L. J. (1999). *J. Appl. Cryst.* **32**, 837–838.  
 Huang, L.-J., Hour, M.-J., Teng, C.-M. & Kuo, S.-C. (1992). *Chem. Pharm. Bull.* **40**, 2547–2551.  
 Khan, M. A., Cosenza, A. G. & Ellis, G. P. (1982). *J. Heterocycl. Chem.* **19**, 1077–1085.  
 Kuo, S.-C., Huang, L.-J. & Nakamura, H. (1984). *J. Med. Chem.* **27**, 539–544.  
 Ramsay, C. G. & Steel, P. J. (1985). *Acta Cryst.* **C41**, 135–136.  
 Samaritoni, G. J., Thornburgh, S., Graupner, P. R. & Cooper, D. H. (2007). *J. Heterocycl. Chem.* **44**, 1389–1393.  
 Sheldrick, G. M. (2008). *Acta Cryst.* **A64**, 112–122.  
 Spek, A. L. (2009). *Acta Cryst.* **D65**, 148–155.

## supporting information

*Acta Cryst.* (2011). E67, o1021 [doi:10.1107/S1600536811011317]

## 3,4-Dimethyl-1-phenylpyrano[2,3-*c*]pyrazol-6(1*H*)-one

Neman Ahmad, M. Nawaz Tahir, Misbahul Ain Khan, Abdul Qayyum Ather and Muhammad Naeem Khan

### S1. Comment

The parent ring system pyrano[2,3-*c*]pyrazol-6-one is an isostere of coumarin. A number of its derivatives have been prepared from corresponding hydrazines and beta ketoesters (Khan *et al.*, 1982, Samaritoni *et al.*, 2007). It has been shown that these contain analgesic and anti-inflammatory activities (Kuo *et al.*, 1984, Abdallah & Zaki, 1999) and while others are tested for their antiplatelet activity (Huang *et al.*, 1992). The title compound (I, Fig. 1) has been synthesized and its crystal structure is being reported here.

The crystal structure of (II) *i.e.*, 3,4-dimethyl-1-(2-pyridyl)pyrano(2,3 - *c*)pyrazol-6(1*H*)-one (Ramsay & Steel, 1983) has been published which differs from (I) due to pyridal attachment instead of phenyl and hence is closely related.

In (I) the phenyl ring A (C1–C6) and 3,4-dimethylpyrano[2,3-*c*]pyrazol-6(1*H*)-one moiety are almost planar with r. m. s. deviations of 0.003 and 0.023 Å, respectively. The dihedral angle between A/B is 7.28 (6)°. The molecules are linked by C(8) polymeric chains (Bernstein *et al.*, 1995) due to H-bonding of C—H···O type (Table 1, Fig. 2). An intramolecular H-bonding and a C—H··· $\pi$  interaction (Table 1) also play an important role in stabilizing the molecules. There also exist  $\pi$ ··· $\pi$  interactions between the pyrazole rings at a distance of 3.6374 (12) Å.

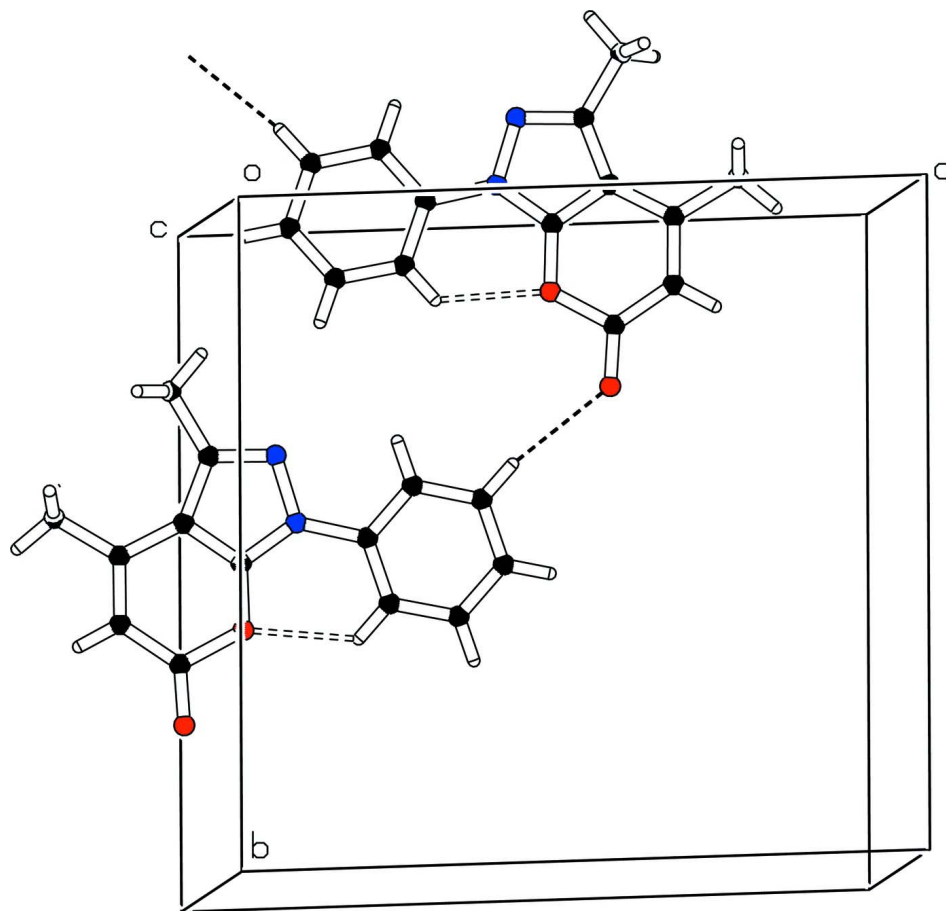
### S2. Experimental

A mixture of 3-methyl-1-phenylpyrazol-5-one (17.4 g, 0.1 mol) and ethyl acetoacetate (13 g, 0.1 mol) was heated at 413 K (oil bath) for an hour, cooled and triturated with 200 ml of pet. ether (bp. 313–333 K) and filtered to give the title compound. Light brown rods of (I) were obtained on recrystallization from ethanol. Yield 10.8 g, 45%; m.p. 415 K

### S3. Refinement

The H-atoms were positioned geometrically (C—H = 0.93–0.96 Å) and refined as riding with  $U_{\text{iso}}(\text{H}) = xU_{\text{eq}}(\text{C})$ , where  $x = 1.5$  for methyl and  $x = 1.2$  for aryl H-atoms.



**Figure 2**

Packing diagram of the title compound showing that polymeric chains are formed.

### 3,4-Dimethyl-1-phenylpyrano[2,3-c]pyrazol-6(1*H*)-one

#### Crystal data

$C_{14}H_{12}N_2O_2$   
 $M_r = 240.26$   
 Monoclinic,  $C2/c$   
 Hall symbol:  $-C 2yc$   
 $a = 15.1231 (9) \text{ \AA}$   
 $b = 13.3558 (8) \text{ \AA}$   
 $c = 13.8684 (8) \text{ \AA}$   
 $\beta = 120.965 (2)^\circ$   
 $V = 2401.9 (3) \text{ \AA}^3$   
 $Z = 8$

$F(000) = 1008$   
 $D_x = 1.329 \text{ Mg m}^{-3}$   
 Mo  $K\alpha$  radiation,  $\lambda = 0.71073 \text{ \AA}$   
 Cell parameters from 1382 reflections  
 $\theta = 2.2\text{--}25.3^\circ$   
 $\mu = 0.09 \text{ mm}^{-1}$   
 $T = 296 \text{ K}$   
 Rod, light brown  
 $0.35 \times 0.25 \times 0.25 \text{ mm}$

#### Data collection

Bruker Kappa APEXII CCD  
 diffractometer  
 Radiation source: fine-focus sealed tube  
 Graphite monochromator  
 Detector resolution:  $8.10 \text{ pixels mm}^{-1}$   
 $\omega$  scans

Absorption correction: multi-scan  
 (*SADABS*; Bruker, 2005)  
 $T_{\min} = 0.975$ ,  $T_{\max} = 0.982$   
 9214 measured reflections  
 2171 independent reflections  
 1382 reflections with  $I > 2\sigma(I)$   
 $R_{\text{int}} = 0.032$

$\theta_{\max} = 25.3^\circ$ ,  $\theta_{\min} = 2.2^\circ$   
 $h = -18 \rightarrow 14$

$k = -15 \rightarrow 16$   
 $l = -9 \rightarrow 16$

*Refinement*

Refinement on  $F^2$   
 Least-squares matrix: full  
 $R[F^2 > 2\sigma(F^2)] = 0.046$   
 $wR(F^2) = 0.140$   
 $S = 1.02$   
 2171 reflections  
 165 parameters  
 0 restraints  
 Primary atom site location: structure-invariant  
 direct methods

Secondary atom site location: difference Fourier  
 map  
 Hydrogen site location: inferred from  
 neighbouring sites  
 H-atom parameters constrained  
 $w = 1/[\sigma^2(F_o^2) + (0.0723P)^2 + 0.5405P]$   
 where  $P = (F_o^2 + 2F_c^2)/3$   
 $(\Delta/\sigma)_{\max} < 0.001$   
 $\Delta\rho_{\max} = 0.15 \text{ e } \text{\AA}^{-3}$   
 $\Delta\rho_{\min} = -0.19 \text{ e } \text{\AA}^{-3}$

*Special details*

**Geometry.** Bond distances, angles *etc.* have been calculated using the rounded fractional coordinates. All su's are estimated from the variances of the (full) variance-covariance matrix. The cell e.s.d.'s are taken into account in the estimation of distances, angles and torsion angles

**Refinement.** Refinement of  $F^2$  against ALL reflections. The weighted  $R$ -factor  $wR$  and goodness of fit  $S$  are based on  $F^2$ , conventional  $R$ -factors  $R$  are based on  $F$ , with  $F$  set to zero for negative  $F^2$ . The threshold expression of  $F^2 > \sigma(F^2)$  is used only for calculating  $R$ -factors(gt) *etc.* and is not relevant to the choice of reflections for refinement.  $R$ -factors based on  $F^2$  are statistically about twice as large as those based on  $F$ , and  $R$ -factors based on ALL data will be even larger.

*Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters ( $\text{\AA}^2$ )*

	$x$	$y$	$z$	$U_{\text{iso}}^*/U_{\text{eq}}$
O1	0.01582 (9)	0.63464 (9)	0.13521 (10)	0.0584 (5)
O2	-0.07464 (12)	0.77525 (12)	0.09062 (16)	0.0994 (7)
N1	0.09619 (12)	0.47598 (11)	0.16813 (13)	0.0526 (6)
N2	0.06513 (14)	0.37672 (12)	0.14972 (16)	0.0689 (7)
C1	0.20381 (14)	0.49922 (15)	0.22105 (15)	0.0526 (7)
C2	0.27219 (16)	0.42241 (17)	0.24254 (18)	0.0661 (8)
C3	0.37688 (16)	0.44318 (19)	0.2968 (2)	0.0794 (9)
C4	0.41168 (17)	0.5391 (2)	0.3279 (2)	0.0825 (9)
C5	0.34341 (16)	0.61512 (19)	0.30539 (19)	0.0770 (9)
C6	0.23861 (15)	0.59574 (16)	0.25142 (17)	0.0613 (8)
C7	0.01255 (14)	0.53448 (14)	0.13093 (15)	0.0484 (7)
C8	-0.08080 (15)	0.68528 (17)	0.08827 (18)	0.0659 (8)
C9	-0.17132 (15)	0.62522 (17)	0.04570 (17)	0.0638 (8)
C10	-0.17218 (14)	0.52456 (16)	0.04351 (15)	0.0537 (7)
C11	-0.07438 (14)	0.47628 (14)	0.08801 (16)	0.0519 (7)
C12	-0.03607 (17)	0.37748 (16)	0.10228 (19)	0.0657 (8)
C13	-0.0944 (2)	0.28099 (17)	0.0706 (3)	0.1025 (13)
C14	-0.27056 (15)	0.46714 (18)	-0.00325 (19)	0.0709 (8)
H2	0.24847	0.35727	0.22090	0.0793*
H3	0.42364	0.39162	0.31212	0.0953*
H4	0.48198	0.55248	0.36445	0.0990*
H5	0.36739	0.68028	0.32639	0.0924*
H6	0.19215	0.64766	0.23586	0.0736*
H9	-0.23416	0.65797	0.01749	0.0766*

H13A	-0.04834	0.22680	0.08183	0.1536*
H13B	-0.14832	0.28336	-0.00699	0.1536*
H13C	-0.12397	0.27081	0.11674	0.1536*
H14A	-0.32644	0.51265	-0.02276	0.1064*
H14B	-0.26542	0.42052	0.05210	0.1064*
H14C	-0.28305	0.43135	-0.06914	0.1064*

*Atomic displacement parameters (Å<sup>2</sup>)*

	$U^{11}$	$U^{22}$	$U^{33}$	$U^{12}$	$U^{13}$	$U^{23}$
O1	0.0380 (8)	0.0474 (9)	0.0769 (9)	0.0008 (6)	0.0203 (7)	-0.0016 (6)
O2	0.0515 (10)	0.0514 (11)	0.1565 (16)	0.0033 (8)	0.0259 (10)	-0.0053 (10)
N1	0.0409 (9)	0.0431 (10)	0.0687 (10)	0.0025 (8)	0.0245 (8)	0.0023 (7)
N2	0.0509 (12)	0.0454 (11)	0.1012 (13)	-0.0024 (8)	0.0325 (10)	0.0020 (9)
C1	0.0375 (11)	0.0555 (13)	0.0584 (11)	0.0036 (9)	0.0201 (9)	0.0015 (9)
C2	0.0503 (13)	0.0581 (14)	0.0807 (14)	0.0098 (11)	0.0272 (11)	0.0040 (10)
C3	0.0478 (14)	0.0753 (17)	0.1009 (17)	0.0200 (13)	0.0282 (13)	0.0069 (13)
C4	0.0374 (12)	0.0844 (18)	0.1021 (18)	0.0021 (13)	0.0191 (12)	-0.0092 (14)
C5	0.0428 (13)	0.0728 (16)	0.0992 (17)	-0.0039 (12)	0.0249 (12)	-0.0147 (13)
C6	0.0401 (12)	0.0550 (14)	0.0797 (13)	0.0014 (10)	0.0243 (10)	-0.0064 (10)
C7	0.0422 (11)	0.0434 (12)	0.0572 (11)	0.0013 (9)	0.0238 (9)	0.0009 (9)
C8	0.0431 (12)	0.0519 (14)	0.0856 (14)	0.0050 (11)	0.0208 (11)	-0.0010 (11)
C9	0.0357 (11)	0.0631 (15)	0.0793 (14)	0.0022 (10)	0.0201 (10)	-0.0039 (11)
C10	0.0411 (11)	0.0572 (13)	0.0587 (11)	-0.0032 (9)	0.0228 (9)	-0.0021 (9)
C11	0.0410 (11)	0.0496 (12)	0.0620 (12)	-0.0029 (9)	0.0244 (9)	-0.0002 (9)
C12	0.0525 (13)	0.0489 (14)	0.0905 (15)	-0.0057 (10)	0.0331 (12)	0.0013 (10)
C13	0.0688 (17)	0.0521 (16)	0.166 (3)	-0.0132 (13)	0.0457 (17)	-0.0009 (15)
C14	0.0444 (12)	0.0789 (16)	0.0838 (14)	-0.0161 (11)	0.0289 (11)	-0.0101 (12)

*Geometric parameters (Å, °)*

O1—C7	1.339 (2)	C10—C11	1.430 (3)
O1—C8	1.427 (3)	C10—C14	1.493 (3)
O2—C8	1.204 (3)	C11—C12	1.414 (3)
N1—N2	1.386 (2)	C12—C13	1.494 (4)
N1—C1	1.433 (3)	C2—H2	0.9300
N1—C7	1.344 (3)	C3—H3	0.9300
N2—C12	1.319 (4)	C4—H4	0.9300
C1—C2	1.376 (3)	C5—H5	0.9300
C1—C6	1.374 (3)	C6—H6	0.9300
C2—C3	1.387 (4)	C9—H9	0.9300
C3—C4	1.368 (4)	C13—H13A	0.9600
C4—C5	1.365 (4)	C13—H13B	0.9600
C5—C6	1.385 (4)	C13—H13C	0.9600
C7—C11	1.372 (3)	C14—H14A	0.9600
C8—C9	1.426 (3)	C14—H14B	0.9600
C9—C10	1.345 (3)	C14—H14C	0.9600

C7—O1—C8	116.58 (17)	N2—C12—C13	119.9 (2)
N2—N1—C1	119.31 (17)	C11—C12—C13	128.8 (3)
N2—N1—C7	108.83 (18)	C1—C2—H2	120.00
C1—N1—C7	131.86 (16)	C3—C2—H2	120.00
N1—N2—C12	106.32 (18)	C2—C3—H3	120.00
N1—C1—C2	118.57 (18)	C4—C3—H3	120.00
N1—C1—C6	121.0 (2)	C3—C4—H4	120.00
C2—C1—C6	120.4 (2)	C5—C4—H4	120.00
C1—C2—C3	119.3 (2)	C4—C5—H5	120.00
C2—C3—C4	120.3 (2)	C6—C5—H5	120.00
C3—C4—C5	120.1 (3)	C1—C6—H6	120.00
C4—C5—C6	120.4 (2)	C5—C6—H6	120.00
C1—C6—C5	119.5 (2)	C8—C9—H9	118.00
O1—C7—N1	123.92 (19)	C10—C9—H9	118.00
O1—C7—C11	126.2 (2)	C12—C13—H13A	109.00
N1—C7—C11	109.87 (17)	C12—C13—H13B	109.00
O1—C8—O2	114.4 (2)	C12—C13—H13C	109.00
O1—C8—C9	117.49 (19)	H13A—C13—H13B	109.00
O2—C8—C9	128.1 (2)	H13A—C13—H13C	109.00
C8—C9—C10	124.6 (2)	H13B—C13—H13C	109.00
C9—C10—C11	116.4 (2)	C10—C14—H14A	109.00
C9—C10—C14	121.3 (2)	C10—C14—H14B	109.00
C11—C10—C14	122.28 (19)	C10—C14—H14C	109.00
C7—C11—C10	118.64 (18)	H14A—C14—H14B	109.00
C7—C11—C12	103.7 (2)	H14A—C14—H14C	109.00
C10—C11—C12	137.7 (2)	H14B—C14—H14C	109.00
N2—C12—C11	111.3 (2)		
C8—O1—C7—N1	177.83 (18)	C1—C2—C3—C4	-0.5 (4)
C8—O1—C7—C11	-1.5 (3)	C2—C3—C4—C5	-0.2 (4)
C7—O1—C8—O2	-177.76 (19)	C3—C4—C5—C6	0.3 (4)
C7—O1—C8—C9	2.7 (3)	C4—C5—C6—C1	0.3 (3)
C1—N1—N2—C12	179.03 (18)	O1—C7—C11—C10	-0.4 (3)
C7—N1—N2—C12	-0.1 (2)	O1—C7—C11—C12	179.28 (19)
N2—N1—C1—C2	5.9 (3)	N1—C7—C11—C10	-179.82 (17)
N2—N1—C1—C6	-173.10 (19)	N1—C7—C11—C12	-0.1 (2)
C7—N1—C1—C2	-175.1 (2)	O1—C8—C9—C10	-2.2 (3)
C7—N1—C1—C6	5.8 (3)	O2—C8—C9—C10	178.3 (2)
N2—N1—C7—O1	-179.24 (17)	C8—C9—C10—C11	0.3 (3)
N2—N1—C7—C11	0.2 (2)	C8—C9—C10—C14	-179.6 (2)
C1—N1—C7—O1	1.7 (3)	C9—C10—C11—C7	1.0 (3)
C1—N1—C7—C11	-178.87 (19)	C9—C10—C11—C12	-178.6 (2)
N1—N2—C12—C11	0.1 (2)	C14—C10—C11—C7	-179.04 (19)
N1—N2—C12—C13	179.6 (2)	C14—C10—C11—C12	1.4 (4)
N1—C1—C2—C3	-178.0 (2)	C7—C11—C12—N2	0.0 (3)
C6—C1—C2—C3	1.0 (3)	C7—C11—C12—C13	-179.5 (3)
N1—C1—C6—C5	178.11 (19)	C10—C11—C12—N2	179.6 (2)
C2—C1—C6—C5	-0.9 (3)	C10—C11—C12—C13	0.1 (5)

*Hydrogen-bond geometry (Å, °)*

Cg3 is the centroid of the C1–C6 phenyl ring.

<i>D</i> —H··· <i>A</i>	<i>D</i> —H	H··· <i>A</i>	<i>D</i> ··· <i>A</i>	<i>D</i> —H··· <i>A</i>
C3—H3···O2 <sup>i</sup>	0.93	2.51	3.407 (3)	163
C6—H6···O1	0.93	2.29	2.938 (3)	126
C14—H14C···Cg3 <sup>ii</sup>	0.96	2.75	3.506 (2)	136

Symmetry codes: (i)  $-x+1/2, y-1/2, -z+1/2$ ; (ii)  $-x, -y+1, -z$ .