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Bis(acetato- κ^2O,O')[2,6-bis(1*H*-pyrazol-3-yl)- κN^2]pyridine- κN]manganese(II)

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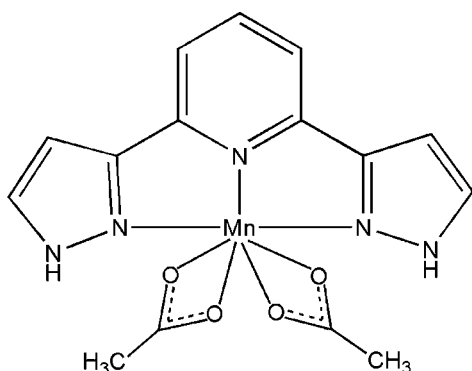
Received 1 March 2011; accepted 21 March 2011

Key indicators: single-crystal X-ray study; $T = 293$ K; mean $\sigma(C-C) = 0.004$ Å; R factor = 0.034; wR factor = 0.117; data-to-parameter ratio = 13.3.

In the title complex, $[Mn(CH_3CO_2)_2(C_{11}H_9N_5)]$, the Mn^{II} atom is coordinated by the pyridine N atom and two pyrazole N atoms from a 2,6-bis(pyrazol-3-yl)pyridine ligand and four O atoms from two bidentate acetate ligands. The complex molecules are linked by intermolecular $N-H \cdots O$ hydrogen bonds into a chain along [010]. $\pi-\pi$ interactions between the pyridine rings and between the pyrazole rings [centroid-centroid distances = 3.772 (2) and 3.546 (2) Å] connect the chains.

Related literature

For a related structure, see: Rich *et al.* (2010).



Experimental

Crystal data

 $[Mn(C_2H_3O_2)_2(C_{11}H_9N_5)]$
 $M_r = 384.26$ Triclinic, $P\bar{1}$ $a = 8.2386$ (16) Å $b = 9.4324$ (19) Å $c = 11.081$ (2) Å $\alpha = 98.32$ (3)° $\beta = 95.01$ (3)°

$\gamma = 106.11$ (3)°
 $V = 811.2$ (3) Å³
 $Z = 2$
 Mo $K\alpha$ radiation

$\mu = 0.85$ mm⁻¹
 $T = 293$ K
 $0.30 \times 0.20 \times 0.20$ mm

Data collection

Rigaku R-Axis RAPID diffractometer
 Absorption correction: multi-scan (ABSCOR; Higashi, 1995)
 $T_{min} = 0.816$, $T_{max} = 0.849$

5563 measured reflections
 3001 independent reflections
 2386 reflections with $I > 2\sigma(I)$
 $R_{int} = 0.030$

Refinement

$R[F^2 > 2\sigma(F^2)] = 0.034$
 $wR(F^2) = 0.117$
 $S = 1.10$
 3001 reflections

226 parameters
 H-atom parameters constrained
 $\Delta\rho_{max} = 0.51$ e Å⁻³
 $\Delta\rho_{min} = -0.43$ e Å⁻³

Table 1

Selected bond lengths (Å).

| | | | |
|--------|-----------|--------|-----------|
| Mn1—O1 | 2.480 (2) | Mn1—N2 | 2.262 (2) |
| Mn1—O2 | 2.192 (2) | Mn1—N3 | 2.235 (2) |
| Mn1—O3 | 2.596 (2) | Mn1—N4 | 2.270 (2) |
| Mn1—O4 | 2.160 (2) | | |

Table 2

Hydrogen-bond geometry (Å, °).

| $D-H \cdots A$ | $D-H$ | $H \cdots A$ | $D \cdots A$ | $D-H \cdots A$ |
|----------------------------------|-------|--------------|--------------|----------------|
| N1—H1B \cdots O4 ⁱ | 0.86 | 1.91 | 2.751 (3) | 165 |
| N5—H5B \cdots O2 ⁱⁱ | 0.86 | 1.85 | 2.712 (3) | 180 |

Symmetry codes: (i) $-x + 1, -y + 2, -z + 2$; (ii) $-x + 2, -y + 2, -z + 2$.

Data collection: *RAPID-AUTO* (Rigaku, 1998); cell refinement: *RAPID-AUTO*; data reduction: *CrystalStructure* (Rigaku/MS, 2002); program(s) used to solve structure: *SHELXS97* (Sheldrick, 2008); program(s) used to refine structure: *SHELXL97* (Sheldrick, 2008); molecular graphics: *SHELXTL* (Sheldrick, 2008) and *Mercury* (Macrae *et al.*, 2006); software used to prepare material for publication: *SHELXTL*.

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Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: HY2414).

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supporting information

Acta Cryst. (2011). E67, m502 [doi:10.1107/S1600536811010506]

Bis(acetato- κ^2O,O')[2,6-bis(1*H*-pyrazol-3-yl- κN^2)pyridine- κN]manganese(II)

Fan Yu and Bao Li

S1. Experimental

2,6-Bis(pyrazol-3-yl)pyridine (0.1 mmol) was dissolved in methanol (2.5 ml) with 0.2 mmol of trimethylamine. $Mn(OAc)_2$ (0.2 mmol) in methanol (2.5 ml) was added into the resulting solution. After stirring at room temperature for 1 h, the resulting yellow solution was put into a tube layered with aether. Yellow crystals were obtained in three days.

S2. Refinement

H atoms were positioned geometrically and refined as riding atoms, with C—H = 0.93 (aromatic), 0.96 (CH₃) and N—H = 0.86 Å and with $U_{iso}(H) = 1.2(1.5 \text{ for methyl})U_{eq}(C, N)$.

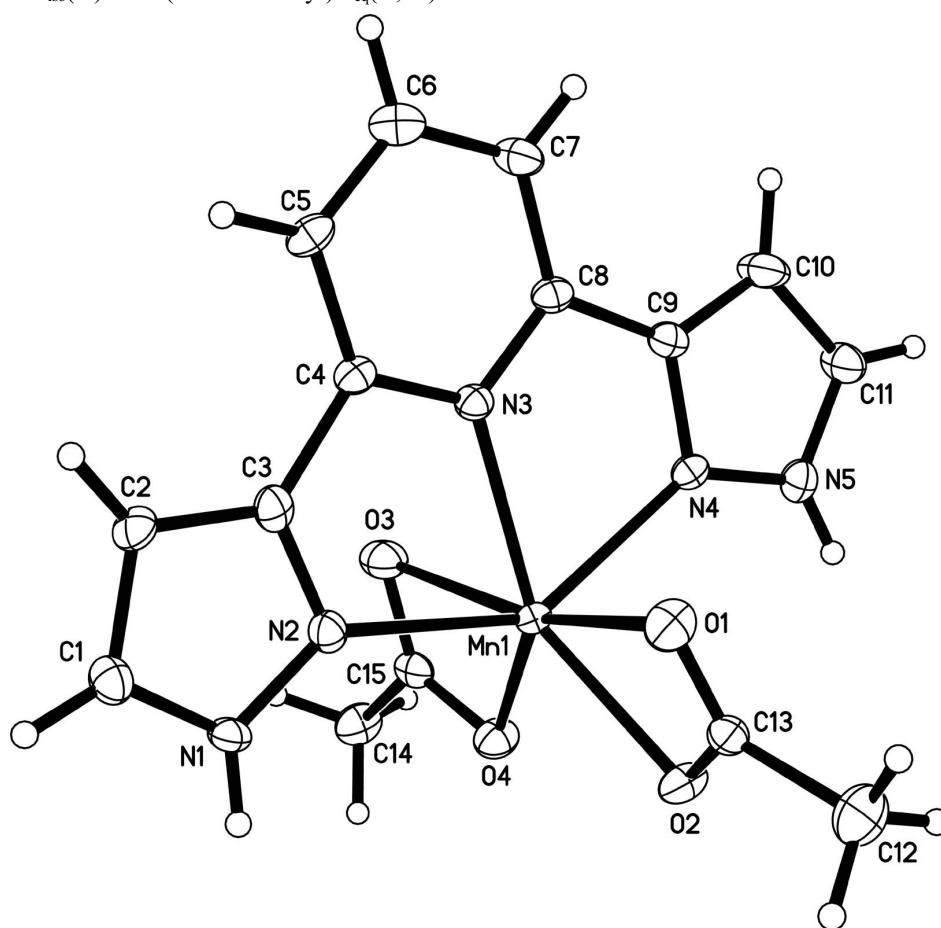
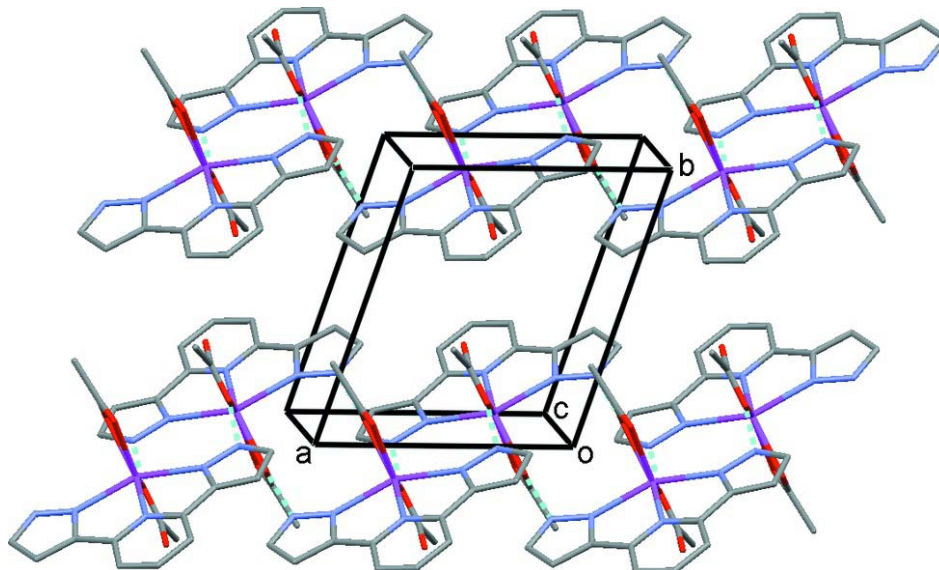


Figure 1

Molecular structure of the title compound. Displacement ellipsoids are drawn at the 50% probability level.

**Figure 2**

Crystal packing of the title compound, showing the chain structures along the *b* axis. Dashed lines denote hydrogen bonds. H atoms have been omitted for clarity.

Bis(acetato- κ^2O,O')[2,6-bis(1*H*-pyrazol-3-yl- κN^2)pyridine- κN]manganese(II)

Crystal data

[Mn(C₂H₃O₂)₂(C₁₁H₉N₅)]

M_r = 384.26

Triclinic, *P*1

Hall symbol: -P 1

a = 8.2386 (16) Å

b = 9.4324 (19) Å

c = 11.081 (2) Å

α = 98.32 (3)°

β = 95.01 (3)°

γ = 106.11 (3)°

V = 811.2 (3) Å³

Z = 2

F(000) = 394

D_x = 1.573 Mg m⁻³

Mo *K* α radiation, λ = 0.71073 Å

Cell parameters from 2386 reflections

θ = 6.1–54.9°

μ = 0.85 mm⁻¹

T = 293 K

Block, yellow

0.30 × 0.20 × 0.20 mm

Data collection

Rigaku R-AXIS RAPID
diffractometer

Radiation source: rotation anode

Graphite monochromator

ω scans

Absorption correction: multi-scan
(*ABSCOR*; Higashi, 1995)

T_{min} = 0.816, *T_{max}* = 0.849

5563 measured reflections

3001 independent reflections

2386 reflections with *I* > 2 σ (*I*)

R_{int} = 0.030

θ_{\max} = 26.0°, θ_{\min} = 3.4°

h = -10→10

k = -11→9

l = -13→13

Refinement

Refinement on F^2

Least-squares matrix: full

$R[F^2 > 2\sigma(F^2)] = 0.034$

$wR(F^2) = 0.117$

$S = 1.10$

3001 reflections

226 parameters

0 restraints

Primary atom site location: structure-invariant
direct methods

Secondary atom site location: difference Fourier
map

Hydrogen site location: inferred from
neighbouring sites

H-atom parameters constrained

$w = 1/[\sigma^2(F_o^2) + (0.0565P)^2]$

where $P = (F_o^2 + 2F_c^2)/3$

$(\Delta/\sigma)_{\max} = 0.001$

$\Delta\rho_{\max} = 0.51 \text{ e } \text{\AA}^{-3}$

$\Delta\rho_{\min} = -0.43 \text{ e } \text{\AA}^{-3}$

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (\AA^2)

| | x | y | z | $U_{\text{iso}}^*/U_{\text{eq}}$ |
|------|-------------|-------------|--------------|----------------------------------|
| Mn1 | 0.67466 (5) | 0.89985 (4) | 0.81715 (4) | 0.01323 (17) |
| N1 | 0.3285 (3) | 1.0204 (2) | 0.7643 (2) | 0.0155 (5) |
| H1B | 0.3327 | 1.0677 | 0.8375 | 0.019* |
| N2 | 0.4400 (3) | 0.9459 (2) | 0.7303 (2) | 0.0155 (5) |
| N3 | 0.6132 (3) | 0.7810 (2) | 0.6217 (2) | 0.0131 (5) |
| N4 | 0.8712 (3) | 0.7781 (2) | 0.7766 (2) | 0.0148 (5) |
| N5 | 1.0113 (3) | 0.7595 (3) | 0.8392 (2) | 0.0179 (5) |
| H5B | 1.0537 | 0.8029 | 0.9135 | 0.021* |
| C1 | 0.2104 (4) | 1.0115 (3) | 0.6700 (3) | 0.0187 (6) |
| H1A | 0.1217 | 1.0545 | 0.6732 | 0.022* |
| C2 | 0.2431 (4) | 0.9274 (3) | 0.5670 (3) | 0.0169 (6) |
| H2A | 0.1842 | 0.9025 | 0.4877 | 0.020* |
| C3 | 0.3877 (3) | 0.8887 (3) | 0.6116 (3) | 0.0156 (6) |
| C4 | 0.4818 (3) | 0.7946 (3) | 0.5476 (2) | 0.0135 (6) |
| C5 | 0.4382 (4) | 0.7223 (3) | 0.4265 (2) | 0.0181 (6) |
| H5A | 0.3462 | 0.7324 | 0.3771 | 0.022* |
| C6 | 0.5374 (4) | 0.6335 (3) | 0.3808 (3) | 0.0225 (7) |
| H6A | 0.5109 | 0.5824 | 0.2999 | 0.027* |
| C7 | 0.6738 (4) | 0.6213 (3) | 0.4548 (3) | 0.0204 (7) |
| H7A | 0.7413 | 0.5638 | 0.4244 | 0.025* |
| C8 | 0.7087 (4) | 0.6962 (3) | 0.5750 (2) | 0.0145 (6) |
| C9 | 0.8480 (4) | 0.6925 (3) | 0.6657 (3) | 0.0158 (6) |
| C10 | 0.9715 (4) | 0.6183 (4) | 0.6587 (3) | 0.0252 (7) |
| H10A | 0.9825 | 0.5508 | 0.5922 | 0.030* |
| C11 | 1.0740 (4) | 0.6645 (3) | 0.7698 (3) | 0.0213 (7) |
| H11A | 1.1696 | 0.6353 | 0.7926 | 0.026* |
| C12 | 1.0613 (4) | 1.3063 (3) | 0.8677 (3) | 0.0306 (8) |
| H12A | 1.0919 | 1.3407 | 0.7931 | 0.046* |
| H12B | 1.1568 | 1.2860 | 0.9100 | 0.046* |
| H12C | 1.0294 | 1.3823 | 0.9194 | 0.046* |
| C13 | 0.9129 (4) | 1.1649 (3) | 0.8373 (3) | 0.0166 (6) |
| O1 | 0.8498 (3) | 1.1097 (2) | 0.7299 (2) | 0.0253 (5) |
| O2 | 0.8561 (3) | 1.1034 (2) | 0.92644 (17) | 0.0196 (5) |
| C14 | 0.3975 (4) | 0.6485 (3) | 1.0567 (3) | 0.0228 (7) |

| | | | | |
|------|------------|------------|--------------|------------|
| H14A | 0.3170 | 0.5539 | 1.0197 | 0.034* |
| H14B | 0.3380 | 0.7142 | 1.0935 | 0.034* |
| H14C | 0.4766 | 0.6332 | 1.1188 | 0.034* |
| C15 | 0.4932 (4) | 0.7179 (3) | 0.9591 (3) | 0.0153 (6) |
| O3 | 0.4611 (3) | 0.6531 (2) | 0.85024 (18) | 0.0227 (5) |
| O4 | 0.6077 (3) | 0.8442 (2) | 0.99224 (18) | 0.0195 (5) |

Atomic displacement parameters (Å²)

| | U^{11} | U^{22} | U^{33} | U^{12} | U^{13} | U^{23} |
|-----|-------------|-------------|-------------|--------------|--------------|---------------|
| Mn1 | 0.0131 (3) | 0.0136 (3) | 0.0113 (2) | 0.00332 (17) | 0.00028 (18) | -0.00105 (16) |
| N1 | 0.0176 (13) | 0.0184 (12) | 0.0118 (12) | 0.0079 (10) | 0.0042 (10) | 0.0006 (9) |
| N2 | 0.0136 (13) | 0.0139 (12) | 0.0192 (13) | 0.0045 (10) | 0.0024 (10) | 0.0029 (10) |
| N3 | 0.0139 (12) | 0.0139 (12) | 0.0114 (12) | 0.0035 (10) | 0.0023 (10) | 0.0024 (9) |
| N4 | 0.0146 (12) | 0.0180 (12) | 0.0110 (12) | 0.0050 (10) | -0.0001 (10) | 0.0011 (9) |
| N5 | 0.0161 (13) | 0.0185 (13) | 0.0185 (13) | 0.0047 (10) | -0.0002 (10) | 0.0037 (10) |
| C1 | 0.0177 (16) | 0.0223 (15) | 0.0179 (15) | 0.0082 (13) | 0.0028 (13) | 0.0052 (12) |
| C2 | 0.0176 (15) | 0.0212 (15) | 0.0110 (14) | 0.0052 (12) | -0.0009 (12) | 0.0031 (11) |
| C3 | 0.0128 (14) | 0.0119 (14) | 0.0216 (16) | 0.0020 (11) | 0.0035 (12) | 0.0039 (11) |
| C4 | 0.0158 (15) | 0.0142 (14) | 0.0099 (14) | 0.0021 (11) | 0.0026 (12) | 0.0038 (11) |
| C5 | 0.0221 (16) | 0.0204 (15) | 0.0103 (14) | 0.0053 (13) | -0.0028 (12) | 0.0031 (11) |
| C6 | 0.0296 (18) | 0.0196 (16) | 0.0186 (16) | 0.0085 (14) | 0.0045 (14) | 0.0012 (12) |
| C7 | 0.0250 (17) | 0.0215 (16) | 0.0172 (16) | 0.0113 (13) | 0.0050 (13) | 0.0010 (12) |
| C8 | 0.0176 (15) | 0.0153 (14) | 0.0106 (14) | 0.0047 (12) | 0.0021 (12) | 0.0026 (11) |
| C9 | 0.0164 (15) | 0.0170 (14) | 0.0143 (15) | 0.0049 (12) | 0.0041 (12) | 0.0022 (11) |
| C10 | 0.0300 (19) | 0.0326 (18) | 0.0166 (16) | 0.0195 (15) | 0.0017 (14) | -0.0033 (13) |
| C11 | 0.0212 (17) | 0.0286 (17) | 0.0176 (16) | 0.0141 (14) | 0.0006 (13) | 0.0032 (13) |
| C12 | 0.0286 (19) | 0.0187 (16) | 0.039 (2) | -0.0018 (14) | 0.0063 (16) | 0.0045 (14) |
| C13 | 0.0159 (15) | 0.0148 (14) | 0.0201 (16) | 0.0086 (12) | 0.0002 (13) | 0.0000 (12) |
| O1 | 0.0234 (12) | 0.0237 (12) | 0.0262 (12) | 0.0056 (10) | 0.0001 (10) | 0.0003 (9) |
| O2 | 0.0236 (11) | 0.0167 (10) | 0.0145 (11) | 0.0011 (9) | 0.0007 (9) | 0.0004 (8) |
| C14 | 0.0275 (17) | 0.0185 (15) | 0.0200 (16) | 0.0025 (13) | 0.0024 (14) | 0.0044 (12) |
| C15 | 0.0186 (15) | 0.0174 (14) | 0.0144 (15) | 0.0127 (12) | 0.0007 (12) | 0.0041 (11) |
| O3 | 0.0272 (12) | 0.0271 (12) | 0.0158 (11) | 0.0123 (10) | 0.0026 (9) | 0.0017 (9) |
| O4 | 0.0219 (11) | 0.0206 (11) | 0.0147 (11) | 0.0035 (9) | 0.0039 (9) | 0.0034 (8) |

Geometric parameters (Å, °)

| | | | |
|--------|-----------|----------|-----------|
| Mn1—O1 | 2.480 (2) | C5—C6 | 1.399 (4) |
| Mn1—O2 | 2.192 (2) | C5—H5A | 0.9300 |
| Mn1—O3 | 2.596 (2) | C6—C7 | 1.373 (4) |
| Mn1—O4 | 2.160 (2) | C6—H6A | 0.9300 |
| Mn1—N2 | 2.262 (2) | C7—C8 | 1.380 (4) |
| Mn1—N3 | 2.235 (2) | C7—H7A | 0.9300 |
| Mn1—N4 | 2.270 (2) | C8—C9 | 1.469 (4) |
| N1—C1 | 1.341 (4) | C9—C10 | 1.387 (4) |
| N1—N2 | 1.349 (3) | C10—C11 | 1.370 (4) |
| N1—H1B | 0.8600 | C10—H10A | 0.9300 |

| | | | |
|-----------|-------------|---------------|------------|
| N2—C3 | 1.332 (4) | C11—H11A | 0.9300 |
| N3—C4 | 1.347 (3) | C12—C13 | 1.510 (4) |
| N3—C8 | 1.353 (3) | C12—H12A | 0.9600 |
| N4—C9 | 1.336 (3) | C12—H12B | 0.9600 |
| N4—N5 | 1.361 (3) | C12—H12C | 0.9600 |
| N5—C11 | 1.338 (4) | C13—O1 | 1.233 (3) |
| N5—H5B | 0.8600 | C13—O2 | 1.275 (3) |
| C1—C2 | 1.383 (4) | C14—C15 | 1.512 (4) |
| C1—H1A | 0.9300 | C14—H14A | 0.9600 |
| C2—C3 | 1.410 (4) | C14—H14B | 0.9600 |
| C2—H2A | 0.9300 | C14—H14C | 0.9600 |
| C3—C4 | 1.478 (4) | C15—O3 | 1.241 (3) |
| C4—C5 | 1.382 (4) | C15—O4 | 1.278 (3) |
| O4—Mn1—O2 | 85.32 (8) | N3—C4—C5 | 122.6 (2) |
| O4—Mn1—N3 | 135.59 (8) | N3—C4—C3 | 112.7 (2) |
| O2—Mn1—N3 | 138.38 (8) | C5—C4—C3 | 124.6 (3) |
| O4—Mn1—N2 | 103.71 (8) | C4—C5—C6 | 117.6 (3) |
| O2—Mn1—N2 | 111.90 (8) | C4—C5—H5A | 121.2 |
| N3—Mn1—N2 | 71.44 (8) | C6—C5—H5A | 121.2 |
| O4—Mn1—N4 | 103.09 (8) | C7—C6—C5 | 120.3 (3) |
| O2—Mn1—N4 | 95.79 (8) | C7—C6—H6A | 119.8 |
| N3—Mn1—N4 | 71.19 (8) | C5—C6—H6A | 119.8 |
| N2—Mn1—N4 | 142.63 (8) | C6—C7—C8 | 118.8 (3) |
| O4—Mn1—O1 | 140.27 (8) | C6—C7—H7A | 120.6 |
| O2—Mn1—O1 | 55.33 (7) | C8—C7—H7A | 120.6 |
| N3—Mn1—O1 | 84.13 (8) | N3—C8—C7 | 122.0 (3) |
| N2—Mn1—O1 | 88.11 (8) | N3—C8—C9 | 112.8 (2) |
| N4—Mn1—O1 | 87.66 (8) | C7—C8—C9 | 125.2 (3) |
| O4—Mn1—O3 | 54.13 (7) | N4—C9—C10 | 109.9 (3) |
| O2—Mn1—O3 | 139.25 (7) | N4—C9—C8 | 117.9 (2) |
| N3—Mn1—O3 | 81.62 (7) | C10—C9—C8 | 132.1 (3) |
| N2—Mn1—O3 | 84.46 (7) | C11—C10—C9 | 106.1 (3) |
| N4—Mn1—O3 | 90.63 (7) | C11—C10—H10A | 127.0 |
| O1—Mn1—O3 | 165.39 (6) | C9—C10—H10A | 127.0 |
| C1—N1—N2 | 111.6 (2) | N5—C11—C10 | 107.3 (3) |
| C1—N1—H1B | 124.2 | N5—C11—H11A | 126.4 |
| N2—N1—H1B | 124.2 | C10—C11—H11A | 126.4 |
| C3—N2—N1 | 104.9 (2) | C13—C12—H12A | 109.5 |
| C3—N2—Mn1 | 117.16 (17) | C13—C12—H12B | 109.5 |
| N1—N2—Mn1 | 137.93 (18) | H12A—C12—H12B | 109.5 |
| C4—N3—C8 | 118.8 (2) | C13—C12—H12C | 109.5 |
| C4—N3—Mn1 | 120.55 (17) | H12A—C12—H12C | 109.5 |
| C8—N3—Mn1 | 120.66 (18) | H12B—C12—H12C | 109.5 |
| C9—N4—N5 | 105.9 (2) | O1—C13—O2 | 121.1 (3) |
| C9—N4—Mn1 | 117.09 (18) | O1—C13—C12 | 121.2 (3) |
| N5—N4—Mn1 | 136.93 (17) | O2—C13—C12 | 117.8 (3) |
| C11—N5—N4 | 110.9 (2) | C13—O1—Mn1 | 85.49 (18) |

| | | | |
|------------|-----------|---------------|-------------|
| C11—N5—H5B | 124.6 | C13—O2—Mn1 | 97.79 (17) |
| N4—N5—H5B | 124.6 | C15—C14—H14A | 109.5 |
| N1—C1—C2 | 108.3 (2) | C15—C14—H14B | 109.5 |
| N1—C1—H1A | 125.8 | H14A—C14—H14B | 109.5 |
| C2—C1—H1A | 125.8 | C15—C14—H14C | 109.5 |
| C1—C2—C3 | 103.0 (3) | H14A—C14—H14C | 109.5 |
| C1—C2—H2A | 128.5 | H14B—C14—H14C | 109.5 |
| C3—C2—H2A | 128.5 | O3—C15—O4 | 121.3 (3) |
| N2—C3—C2 | 112.1 (2) | O3—C15—C14 | 120.6 (3) |
| N2—C3—C4 | 118.0 (2) | O4—C15—C14 | 118.1 (2) |
| C2—C3—C4 | 129.8 (3) | C15—O4—Mn1 | 101.76 (17) |

Hydrogen-bond geometry (Å, °)

| <i>D</i> —H... <i>A</i> | <i>D</i> —H | H... <i>A</i> | <i>D</i> ... <i>A</i> | <i>D</i> —H... <i>A</i> |
|---------------------------|-------------|---------------|-----------------------|-------------------------|
| N1—H1B...O4 ⁱ | 0.86 | 1.91 | 2.751 (3) | 165 |
| N5—H5B...O2 ⁱⁱ | 0.86 | 1.85 | 2.712 (3) | 180 |

Symmetry codes: (i) $-x+1, -y+2, -z+2$; (ii) $-x+2, -y+2, -z+2$.