Acta Crystallographica Section E
Structure Reports Online
ISSN 1600-5368

1-(9-Methyl-11-sulfanylidene-8-oxa-10,12-diazatricyclo[7.3.1.0^{2,7}]trideca-2,4,6-trien-13-yl)ethanone

Malahat M. Kurbanova,[a] Abel M. Maharramov,[a] Aysel B. Novruzova,[a] Atash V. Gurbanov[a] and Seik Weng Ng[b]*

[a]Department of Organic Chemistry, Baku State University, Baku, Azerbaijan, and [b]Department of Chemistry, University of Malaya, 50603 Kuala Lumpur, Malaysia
Correspondence e-mail: seikweng@um.edu.my

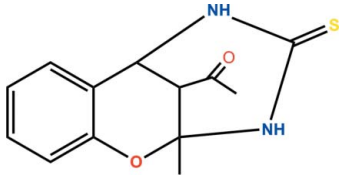
Received 11 April 2011; accepted 12 April 2011

Key indicators: single-crystal X-ray study; T = 296 K; mean σ(C–C) = 0.007 Å; R factor = 0.079; wR factor = 0.211; data-to-parameter ratio = 14.0.

The six-membered oxacyclohexene ring of the title compound, $C_{13}H_{14}N_2O_2S$, is fused with the benzene ring and the quarternary C atom lies above the plane of the benzene ring by 0.229 (8) Å, whereas the methine C atom (which bears the acetyl substituent) lies below this plane by 0.595 (8) Å. The oxacyclohexene ring is also fused with the sofa-shaped 2,6-diazacyclohexanone ring. The methine C atom that belongs to both six-membered rings lies above the mean plane of the other five atoms (r.m.s. deviation = 0.077 Å) by 0.759 (5) Å. In the crystal, N—H···S hydrogen bonds link adjacent molecules into a linear chain.

Related literature

For related structures, see: Kettmann & Svetlík (1996, 1997); Kurbanova *et al.* (2009).



Experimental

Crystal data

$C_{13}H_{14}N_2O_2S$
M_r = 262.32
Monoclinic, $P2_1/n$
a = 8.2382 (5) Å
b = 19.1223 (12) Å
c = 9.2209 (6) Å
β = 114.623 (1)°
V = 1320.51 (14) Å^3
Z = 4
Mo $K\alpha$ radiation
μ = 0.24 mm^{-1}
T = 296 K
0.40 × 0.30 × 0.20 mm

Data collection

Bruker APEXII diffractometer
Absorption correction: multi-scan (*SADABS*; Sheldrick, 1996)
T_{min} = 0.589, T_{max} = 1.000
8830 measured reflections
2292 independent reflections
1523 reflections with $I > 2\sigma(I)$
R_{int} = 0.119

Refinement

$R[F^2 > 2\sigma(F^2)]$ = 0.079
$wR(F^2)$ = 0.211
S = 0.99
2292 reflections
164 parameters
H-atom parameters constrained
$\Delta\rho_{max}$ = 0.65 e Å^{-3}
$\Delta\rho_{min}$ = −0.41 e Å^{-3}

Table 1
Hydrogen-bond geometry (Å, °).

D—H···A	D—H	H···A	D···A	D—H···A
N1—H1···S1^{i}	0.88	2.49	3.324 (3)	158
N2—H2···S1^{ii}	0.88	2.43	3.259 (3)	158

Symmetry codes: (i) $-x+1, -y+1, -z+1$; (ii) $-x+2, -y+1, -z+1$.

Data collection: *APEX2* (Bruker, 2005); cell refinement: *SAINT* (Bruker, 2005); data reduction: *SAINT*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 2008); program(s) used to refine structure: *SHELXL97* (Sheldrick, 2008); molecular graphics: *X-SEED* (Barbour, 2001); software used to prepare material for publication: *publCIF* (Westrip, 2010).

We thank Baku State University and the University of Malaya for supporting this study.

Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: XU5193).

References

Barbour, L. J. (2001). *J. Supramol. Chem.* **1**, 189–191.
Bruker (2005). *APEX2* and *SAINT*. Bruker AXS Inc., Madison, Wisconsin, USA.
Kettmann, V. & Svetlík, J. (1996). *Acta Cryst.* C**52**, 1496–1499.
Kettmann, V. & Svetlík, J. (1997). *Acta Cryst.* C**53**, 1493–1495.
Kurbanova, M. M., Kurbanov, A. V., Askerov, R. K., Allakhverdiev, M. A., Khrustalev, V. N. & Magerramov, A. M. (2009). *J. Struct. Chem.* **50**, 505–509.
Sheldrick, G. M. (1996). *SADABS*. University of Göttingen, Germany.
Sheldrick, G. M. (2008). *Acta Cryst.* A**64**, 112–122.
Westrip, S. P. (2010). *J. Appl. Cryst.* **43**, 920–925.

supporting information

Acta Cryst. (2011). E**67**, o1156 [doi:10.1107/S1600536811013699]

1-(9-Methyl-11-sulfanylidene-8-oxa-10,12-diazatricyclo[7.3.1.$0^{2,7}$]trideca-2,4,6-trien-13-yl)ethanone

Malahat M. Kurbanova, Abel M. Maharramov, Aysel B. Novruzova, Atash V. Gurbanov and Seik Weng Ng

S1. Comment

The title compound, $C_{13}H_{14}N_2O_2S$ (Scheme I, Fig. 1), is a conformationally restricted dihydropyrimidine analogue of 1,4-dihydropyridine-type calcium antagonists; the crystal structures of similar compounds have been reported (Kurbanova *et al.*, 2009; Kettmann & Svetlík, 1996; Kettmann & Svetlík, 1997). Tthe six-membered oxacyclohexene ring that is fused with the benzene ring has the quarternary C atom lying above the plane of the benzene ring and the methine C (which bears the acetyl substituent) lying below this plane. The oxacyclohexene ring is also fused with the sofa-shaped diaza-cyclcohexane ring; the methine C that belongs to both six-membered rings lies above the mean plane of the other five atoms. Hydrogen bonds of the type N–H···S link adjacent molecules to form a linear chain (Fig. 2).

S2. Experimental

In round-bottom flask that was fitted with a reflux condenser and a mechanical stirrer, salicylaldehyde (1.25 mol), acetyl-acetone (1.50 mol), thiocarbamide (1.25 mol), trichloroacetic acid (25 mg) and ethanol (10 ml) were reacted for 3 h. The solid that formed was collected and recrystallized from ethanol, m.p. 514–515 K; yield 80%.

S3. Refinement

Hydrogen atoms were placed in calculated positions [C–H 0.93 to 0.9 and N–H 0.88 7 Å; U(H) 1.2 to 1.5U(C,N)] and were included in the refinement in the riding model approximation.

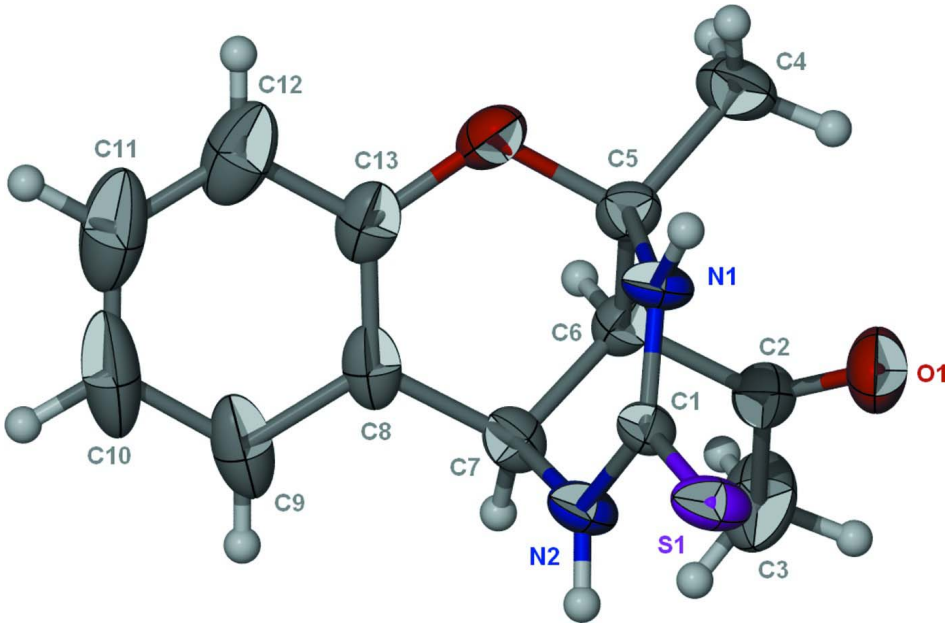


Figure 1

Thermal ellipsoid plot (Barbour, 2001) of the hydrogen-bonded dimeric structure of $C_{13}H_{14}N_2O_2$ at the 50% probability level; hydrogen atoms are drawn as spheres of arbitrary radius.

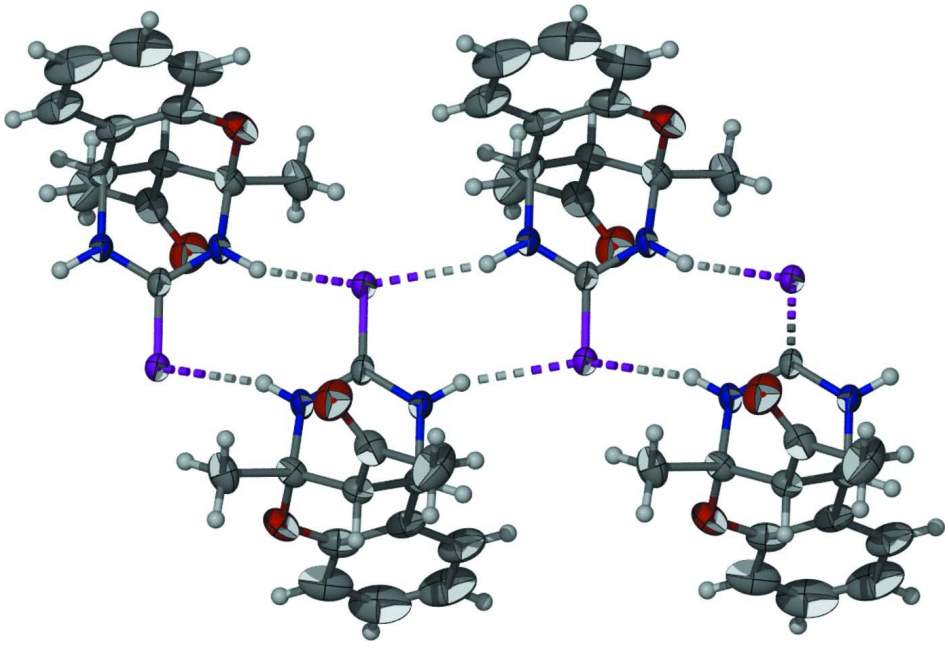


Figure 2

Hydrogen-bonded chain structure.

1-(9-Methyl-11-sulfanylidene-8-oxa-10,12-diazatricyclo[7.3.1.0^{2,7}]trideca- 2,4,6-trien-13-yl)ethan-1-one

Crystal data

$C_{13}H_{14}N_2O_2S$
M_r = 262.32
Monoclinic, $P2_1/n$
Hall symbol: -P 2yn
a = 8.2382 (5) Å
b = 19.1223 (12) Å
c = 9.2209 (6) Å
β = 114.623 (1)°
V = 1320.51 (14) Å^3
Z = 4
$F(000)$ = 552
D_x = 1.319 Mg m^{-3}
Mo $K\alpha$ radiation, λ = 0.71073 Å
Cell parameters from 2832 reflections
θ = 2.7–28.3°
μ = 0.24 mm^{-1}
T = 296 K
Irregular block, colorless
0.40 × 0.30 × 0.20 mm

Data collection

Bruker APEXII diffractometer
Radiation source: fine-focus sealed tube
Graphite monochromator
φ and ω scans
Absorption correction: multi-scan (*SADABS*; Sheldrick, 1996)
T_{min} = 0.589, T_{max} = 1.000
8830 measured reflections
2292 independent reflections
1523 reflections with $I > 2\sigma(I)$
R_{int} = 0.119
θ_{max} = 25.0°, θ_{min} = 2.1°
h = −9→9
k = −22→21
l = −10→10

Refinement

Refinement on F^2
Least-squares matrix: full
$R[F^2 > 2\sigma(F^2)]$ = 0.079
$wR(F^2)$ = 0.211
S = 0.99
2292 reflections
164 parameters
0 restraints
Primary atom site location: structure-invariant direct methods
Secondary atom site location: difference Fourier map
Hydrogen site location: inferred from neighbouring sites
H-atom parameters constrained
$w = 1/[\sigma^2(F_o^2) + (0.1282P)^2]$ where $P = (F_o^2 + 2F_c^2)/3$
$(\Delta/\sigma)_{max}$ = 0.001
$\Delta\rho_{max}$ = 0.65 e Å^{-3}
$\Delta\rho_{min}$ = −0.41 e Å^{-3}

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (Å^2)

	x	y	z	U_{iso}*/U_{eq}
S1	0.70755 (12)	0.51631 (6)	0.41553 (12)	0.0471 (4)
O1	0.6767 (4)	0.29345 (19)	0.3421 (3)	0.0690 (10)
O2	0.6966 (4)	0.34703 (14)	0.7965 (3)	0.0491 (8)
N1	0.6447 (4)	0.40871 (16)	0.5610 (3)	0.0381 (8)
H1	0.5416	0.4296	0.5371	0.046*
N2	0.9280 (4)	0.41746 (16)	0.5844 (3)	0.0398 (8)
H2	1.0075	0.4362	0.5556	0.048*
C1	0.7642 (4)	0.4431 (2)	0.5272 (4)	0.0358 (9)
C2	0.8091 (5)	0.2822 (2)	0.4611 (5)	0.0489 (10)
C3	0.9636 (7)	0.2461 (4)	0.4534 (6)	0.0855 (18)
H3A	0.9226	0.2139	0.3655	0.128*
H3B	1.0271	0.2210	0.5510	0.128*
H3C	1.0417	0.2799	0.4388	0.128*

C4	0.4941 (5)	0.3002 (2)	0.5599 (5)	0.0584 (12)
H4A	0.4010	0.3264	0.5719	0.088*
H4B	0.5078	0.2559	0.6126	0.088*
H4C	0.4633	0.2927	0.4486	0.088*
C5	0.6667 (5)	0.3403 (2)	0.6330 (4)	0.0406 (9)
C6	0.8282 (5)	0.3056 (2)	0.6238 (4)	0.0383 (9)
H6	0.8597	0.2647	0.6940	0.046*
C7	0.9783 (5)	0.35889 (19)	0.6938 (4)	0.0405 (9)
H7	1.0902	0.3386	0.7000	0.049*
C8	0.9989 (6)	0.3797 (2)	0.8581 (4)	0.0472 (11)
C9	1.1601 (7)	0.4052 (2)	0.9698 (5)	0.0716 (15)
H9	1.2577	0.4100	0.9445	0.086*
C10	1.1735 (10)	0.4238 (3)	1.1220 (6)	0.094 (2)
H10	1.2805	0.4410	1.1985	0.113*
C11	1.0297 (11)	0.4164 (3)	1.1571 (6)	0.096 (2)
H11	1.0396	0.4288	1.2580	0.115*
C12	0.8736 (8)	0.3917 (2)	1.0492 (5)	0.0747 (16)
H12	0.7766	0.3869	1.0752	0.090*
C13	0.8582 (6)	0.3734 (2)	0.8995 (4)	0.0500 (11)

Atomic displacement parameters (Å²)

	U^{11}	U^{22}	U^{33}	U^{12}	U^{13}	U^{23}
S1	0.0296 (6)	0.0510 (7)	0.0610 (7)	0.0056 (4)	0.0192 (5)	0.0222 (5)
O1	0.0578 (18)	0.099 (3)	0.0400 (16)	−0.0015 (17)	0.0104 (14)	−0.0118 (16)
O2	0.0670 (18)	0.0456 (18)	0.0456 (15)	−0.0074 (14)	0.0341 (13)	−0.0029 (12)
N1	0.0286 (15)	0.0385 (19)	0.0522 (18)	0.0001 (13)	0.0218 (13)	0.0067 (14)
N2	0.0265 (16)	0.0402 (19)	0.0509 (18)	0.0005 (13)	0.0144 (13)	0.0105 (14)
C1	0.0246 (18)	0.045 (2)	0.0375 (19)	−0.0003 (16)	0.0130 (14)	−0.0014 (16)
C2	0.051 (2)	0.053 (3)	0.046 (2)	−0.010 (2)	0.0228 (19)	−0.0091 (19)
C3	0.072 (3)	0.122 (5)	0.063 (3)	0.025 (3)	0.029 (2)	−0.022 (3)
C4	0.053 (3)	0.055 (3)	0.074 (3)	−0.018 (2)	0.033 (2)	−0.010 (2)
C5	0.047 (2)	0.037 (2)	0.043 (2)	−0.0050 (17)	0.0232 (17)	−0.0016 (16)
C6	0.043 (2)	0.035 (2)	0.0365 (18)	0.0000 (16)	0.0171 (15)	0.0004 (16)
C7	0.036 (2)	0.036 (2)	0.044 (2)	0.0025 (16)	0.0102 (15)	0.0059 (16)
C8	0.058 (3)	0.029 (2)	0.039 (2)	−0.0001 (18)	0.0048 (18)	0.0003 (16)
C9	0.076 (3)	0.050 (3)	0.055 (3)	−0.008 (2)	−0.007 (2)	0.000 (2)
C10	0.115 (5)	0.065 (4)	0.050 (3)	−0.010 (4)	−0.016 (3)	−0.011 (3)
C11	0.156 (6)	0.061 (4)	0.047 (3)	−0.012 (4)	0.019 (4)	−0.015 (3)
C12	0.127 (5)	0.045 (3)	0.054 (3)	0.003 (3)	0.038 (3)	−0.005 (2)
C13	0.075 (3)	0.029 (2)	0.042 (2)	−0.001 (2)	0.020 (2)	−0.0014 (17)

Geometric parameters (Å, º)

S1—C1	1.685 (4)	C4—H4B	0.9600
O1—C2	1.201 (5)	C4—H4C	0.9600
O2—C13	1.369 (5)	C5—C6	1.520 (5)
O2—C5	1.429 (4)	C6—C7	1.523 (5)

N1—C1	1.324 (5)	C6—H6	0.9800
N1—C5	1.443 (5)	C7—C8	1.505 (6)
N1—H1	0.8800	C7—H7	0.9800
N2—C1	1.322 (4)	C8—C13	1.369 (6)
N2—C7	1.447 (5)	C8—C9	1.387 (6)
N2—H2	0.8800	C9—C10	1.407 (9)
C2—C3	1.475 (6)	C9—H9	0.9300
C2—C6	1.509 (5)	C10—C11	1.359 (10)
C3—H3A	0.9600	C10—H10	0.9300
C3—H3B	0.9600	C11—C12	1.342 (8)
C3—H3C	0.9600	C11—H11	0.9300
C4—C5	1.506 (5)	C12—C13	1.377 (6)
C4—H4A	0.9600	C12—H12	0.9300

C13—O2—C5	117.1 (3)	C2—C6—C5	117.0 (3)
C1—N1—C5	126.5 (3)	C2—C6—C7	110.5 (3)
C1—N1—H1	116.8	C5—C6—C7	105.0 (3)
C5—N1—H1	116.8	C2—C6—H6	108.0
C1—N2—C7	120.9 (3)	C5—C6—H6	108.0
C1—N2—H2	119.5	C7—C6—H6	108.0
C7—N2—H2	119.5	N2—C7—C8	112.1 (3)
N2—C1—N1	117.2 (3)	N2—C7—C6	106.0 (3)
N2—C1—S1	121.9 (3)	C8—C7—C6	109.5 (3)
N1—C1—S1	120.9 (2)	N2—C7—H7	109.7
O1—C2—C3	120.9 (4)	C8—C7—H7	109.7
O1—C2—C6	122.2 (4)	C6—C7—H7	109.7
C3—C2—C6	116.9 (4)	C13—C8—C9	118.9 (4)
C2—C3—H3A	109.5	C13—C8—C7	120.3 (3)
C2—C3—H3B	109.5	C9—C8—C7	120.8 (5)
H3A—C3—H3B	109.5	C8—C9—C10	118.9 (6)
C2—C3—H3C	109.5	C8—C9—H9	120.6
H3A—C3—H3C	109.5	C10—C9—H9	120.6
H3B—C3—H3C	109.5	C11—C10—C9	119.9 (5)
C5—C4—H4A	109.5	C11—C10—H10	120.1
C5—C4—H4B	109.5	C9—C10—H10	120.1
H4A—C4—H4B	109.5	C12—C11—C10	121.3 (6)
C5—C4—H4C	109.5	C12—C11—H11	119.3
H4A—C4—H4C	109.5	C10—C11—H11	119.3
H4B—C4—H4C	109.5	C11—C12—C13	119.5 (6)
O2—C5—N1	109.7 (3)	C11—C12—H12	120.2
O2—C5—C4	103.5 (3)	C13—C12—H12	120.2
N1—C5—C4	110.0 (3)	C8—C13—O2	122.0 (3)
O2—C5—C6	109.2 (3)	C8—C13—C12	121.5 (4)
N1—C5—C6	108.4 (3)	O2—C13—C12	116.5 (5)
C4—C5—C6	116.0 (3)		

C7—N2—C1—N1	−6.7 (5)	C2—C6—C7—N2	60.7 (4)
C7—N2—C1—S1	173.4 (3)	C5—C6—C7—N2	−66.3 (4)

C5—N1—C1—N2	−10.3 (5)	C2—C6—C7—C8	−178.1 (3)
C5—N1—C1—S1	169.6 (3)	C5—C6—C7—C8	54.9 (4)
C13—O2—C5—N1	−71.3 (4)	N2—C7—C8—C13	93.0 (4)
C13—O2—C5—C4	171.3 (3)	C6—C7—C8—C13	−24.3 (5)
C13—O2—C5—C6	47.3 (4)	N2—C7—C8—C9	−87.1 (4)
C1—N1—C5—O2	104.1 (4)	C6—C7—C8—C9	155.5 (4)
C1—N1—C5—C4	−142.7 (4)	C13—C8—C9—C10	0.1 (6)
C1—N1—C5—C6	−15.0 (5)	C7—C8—C9—C10	−179.8 (4)
O1—C2—C6—C5	3.9 (6)	C8—C9—C10—C11	0.0 (8)
C3—C2—C6—C5	−177.7 (4)	C9—C10—C11—C12	0.0 (10)
O1—C2—C6—C7	−116.2 (4)	C10—C11—C12—C13	−0.1 (9)
C3—C2—C6—C7	62.3 (5)	C9—C8—C13—O2	−178.5 (4)
O2—C5—C6—C2	169.3 (3)	C7—C8—C13—O2	1.4 (6)
N1—C5—C6—C2	−71.2 (4)	C9—C8—C13—C12	−0.2 (6)
C4—C5—C6—C2	53.0 (5)	C7—C8—C13—C12	179.7 (4)
O2—C5—C6—C7	−67.8 (3)	C5—O2—C13—C8	−13.2 (5)
N1—C5—C6—C7	51.7 (4)	C5—O2—C13—C12	168.4 (4)
C4—C5—C6—C7	175.9 (3)	C11—C12—C13—C8	0.2 (7)
C1—N2—C7—C8	−73.8 (4)	C11—C12—C13—O2	178.6 (4)
C1—N2—C7—C6	45.6 (4)		

Hydrogen-bond geometry (Å, º)

D—H···*A*	*D*—H	H···*A*	*D*···*A*	*D*—H···*A*
N1—H1···S1^{i}	0.88	2.49	3.324 (3)	158
N2—H2···S1^{ii}	0.88	2.43	3.259 (3)	158

Symmetry codes: (i) −*x*+1, −*y*+1, −*z*+1; (ii) −*x*+2, −*y*+1, −*z*+1.