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3-(2-Aminophenylsulfanyl)-1,3-diphenylpropan-1-one

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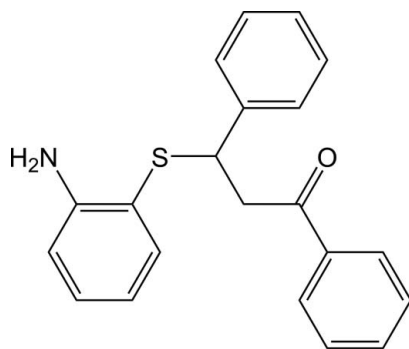
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 Key indicators: single-crystal X-ray study; $T = 296$ K; mean $\sigma(\text{C}-\text{C}) = 0.004$ Å; R factor = 0.057; wR factor = 0.101; data-to-parameter ratio = 14.3.

In the title compound, $\text{C}_{21}\text{H}_{19}\text{NOS}$, the three aromatic rings are not coplanar, the dihedral angles between them being 84.75 (7), 88.01 (8) and 8.36 (16)°. In the crystal, two types of $\text{C}-\text{H}\cdots\pi$ interactions, one of which is weak, and $\text{N}-\text{H}\cdots\pi$ interactions link the molecules into layers parallel to the ab plane.

Related literature

 For a similar structure, see: Morgant *et al.* (1996).


Experimental

Crystal data

 $\text{C}_{21}\text{H}_{19}\text{NOS}$
 $M_r = 333.43$
 Monoclinic, $P2_1/n$
 $a = 11.1741$ (16) Å
 $b = 5.6788$ (8) Å
 $c = 27.308$ (4) Å

 $\beta = 95.266$ (2)°
 $V = 1725.5$ (4) Å³
 $Z = 4$
 Mo $K\alpha$ radiation

 $\mu = 0.19$ mm⁻¹
 $T = 296$ K
 $0.18 \times 0.07 \times 0.02$ mm

Data collection

 Bruker APEX-II CCD diffractometer
 Absorption correction: multi-scan (SADABS; Sheldrick, 1996)
 $T_{\min} = 0.966$, $T_{\max} = 0.996$

 7303 measured reflections
 3195 independent reflections
 1784 reflections with $I > 2\sigma(I)$
 $R_{\text{int}} = 0.065$

Refinement

 $R[F^2 > 2\sigma(F^2)] = 0.057$
 $wR(F^2) = 0.101$
 $S = 0.95$
 3195 reflections
 223 parameters

 H atoms treated by a mixture of independent and constrained refinement
 $\Delta\rho_{\max} = 0.17$ e Å⁻³
 $\Delta\rho_{\min} = -0.23$ e Å⁻³
Table 1

Hydrogen-bond geometry (Å, °).

 $Cg1$ and $Cg2$ are the centroids of the $C10-C15$ and $C16-C21$ rings, respectively.

$D-H\cdots A$	$D-H$	$H\cdots A$	$D\cdots A$	$D-H\cdots A$
$N1-H1B\cdots Cg1^i$	0.91 (3)	2.55 (3)	3.400 (3)	155 (2)
$C12-H12\cdots Cg1^{ii}$	0.93	2.86	3.581 (4)	135
$C14-H14\cdots Cg2^i$	0.93	2.93	3.587 (3)	128

 Symmetry codes: (i) $-x + \frac{3}{2}, y - \frac{1}{2}, -z + \frac{1}{2}$; (ii) $-x + \frac{3}{2}, y + \frac{1}{2}, -z + \frac{1}{2}$.

Data collection: APEX2 (Bruker, 2007); cell refinement: SAINT (Bruker, 2007); data reduction: SAINT; program(s) used to solve structure: SHELXS97 (Sheldrick, 2008); program(s) used to refine structure: SHELXL97 (Sheldrick, 2008); molecular graphics: X-SEED (Barbour, 2001); software used to prepare material for publication: SHELXL97 and PUBLICIF (Westrip, 2010).

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Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: GO2015).

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supporting information

Acta Cryst. (2011). E67, o1693 [doi:10.1107/S1600536811022112]

3-(2-Aminophenylsulfanyl)-1,3-diphenylpropan-1-one

Marzieh Yaeghoobi, Hamid Khaledi, Zanariah Abdullah and Noorsaadah Abd.Rahman

S1. Comment

The title compound (Fig. 1) was obtained during the synthesis of benzothiazepines **II** in an ionic liquid media (Fig. 2) and is suggested to be the intermediate compound which upon subsequent cyclization would form the seven-membered thiazepine ring. The three aromatic rings of the molecule are non-planar as are in the structure of a related compound (Morgant *et al.*, 1996). The dihedral angles between the phenyl rings plane are 84.75 (7)° (between C1/C6 and C10/C15), 88.01 (8)° (between C1/C6 and C16/C21) and 8.36 (16)° (between C10/C15 and C16/C21). The amino hydrogen, H1B, is involved in an N—H··· π interaction (Table 1). In the crystal, the N—H··· π interactions connect the adjacent molecules into infinite chains along the *b* axis. The one-dimensional link is supplemented by weak C14—H14··· π interactions. The chains are connected into two-dimensional-arrays parallel to the *ab* plane via C12—H12··· π interactions (Table 1).

S2. Experimental

The title compound was synthesized as illustrated in Fig. 2. A mixture of the chalcone (0.208 g, 1 mmol), *o*-aminothiophenol (0.088 ml, 1.1 mmol) and ionic liquid, IL, (0.1 g) was heated at 80°C for 80 min. The two products, **I** & **II**, were extracted with diethylether and separated by column chromatography (hexane: ethylacetate, 8:2). The second fraction, containing (**I**), was evaporated and the resulting solid was recrystallized from ethanol at room temperature to give the colorless crystals of the title compound.

S3. Refinement

The C-bound hydrogen atoms were placed at calculated positions and refined as riding atoms with C—H distances of 0.93 (*phenyl*), 0.97 (*methylene*) and 0.98 (*methine*) Å. The N-bound hydrogen atoms were located in a difference Fourier map and refined freely. For all hydrogen atoms $U_{iso}(H)$ were set to 1.2 $U_{eq}(\text{carrier atom})$.

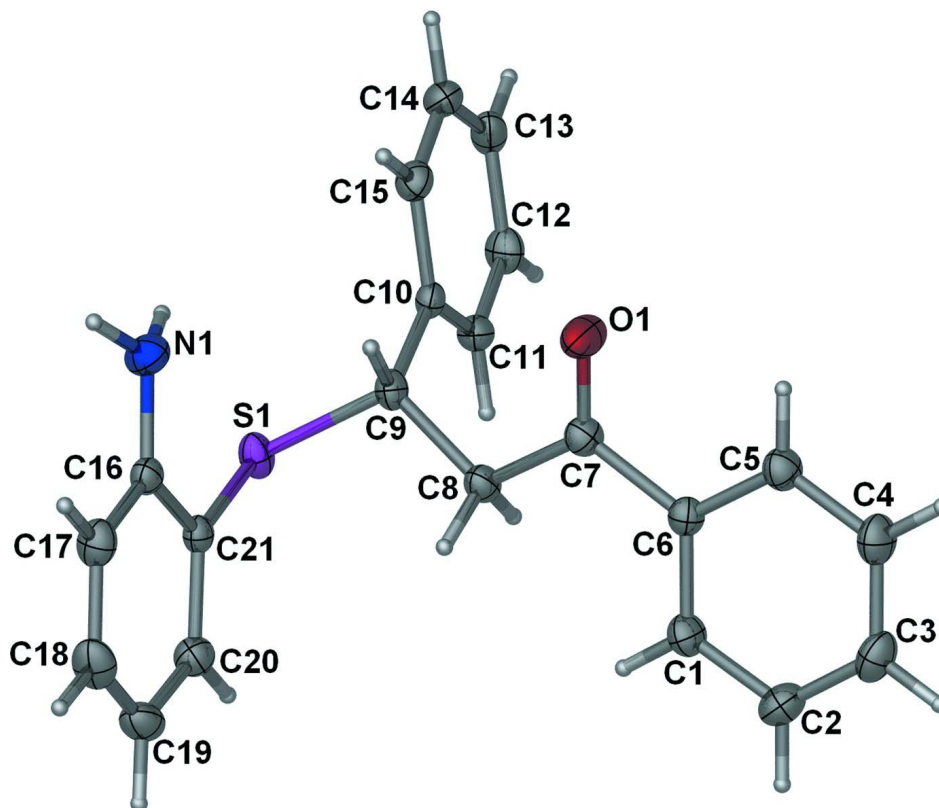


Figure 1

The molecular structure of the title compound with displacement ellipsoids drawn at 30% probability level. Hydrogen atoms are drawn as spheres of arbitrary radius.

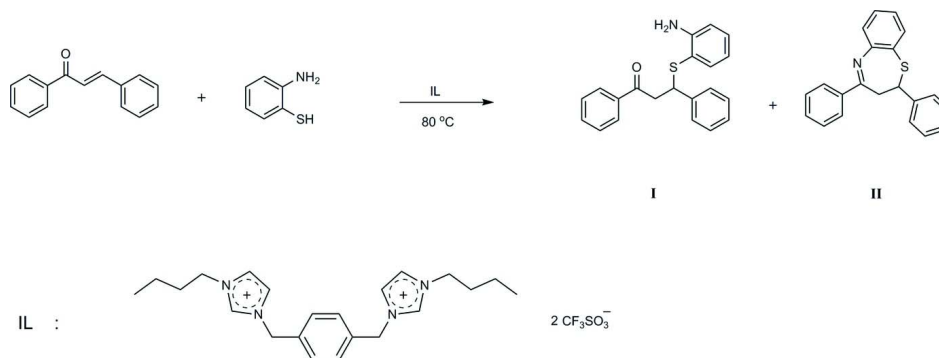


Figure 2

Synthetic route to compounds **I** and **II**.

3-(2-Aminophenylsulfanyl)-1,3-diphenylpropan-1-one

Crystal data

$C_{21}H_{19}NO$

$M_r = 333.43$

Monoclinic, $P2_1/n$

Hall symbol: $-P\ 2_1n$

$a = 11.1741$ (16) Å

$b = 5.6788$ (8) Å

$c = 27.308$ (4) Å

$\beta = 95.266$ (2)°

$V = 1725.5$ (4) Å³

$Z = 4$

$F(000) = 704$
 $D_x = 1.284 \text{ Mg m}^{-3}$
 Mo $K\alpha$ radiation, $\lambda = 0.71073 \text{ \AA}$
 Cell parameters from 713 reflections
 $\theta = 3.0\text{--}20.6^\circ$

$\mu = 0.19 \text{ mm}^{-1}$
 $T = 296 \text{ K}$
 Lath, colorless
 $0.18 \times 0.07 \times 0.02 \text{ mm}$

Data collection

Bruker APEX-II CCD
 diffractometer
 Radiation source: fine-focus sealed tube
 Graphite monochromator
 φ and ω scans
 Absorption correction: multi-scan
 (SADABS; Sheldrick, 1996)
 $T_{\min} = 0.966$, $T_{\max} = 0.996$

7303 measured reflections
 3195 independent reflections
 1784 reflections with $I > 2\sigma(I)$
 $R_{\text{int}} = 0.065$
 $\theta_{\text{max}} = 25.5^\circ$, $\theta_{\text{min}} = 2.0^\circ$
 $h = -12 \rightarrow 13$
 $k = -6 \rightarrow 5$
 $l = -29 \rightarrow 33$

Refinement

Refinement on F^2
 Least-squares matrix: full
 $R[F^2 > 2\sigma(F^2)] = 0.057$
 $wR(F^2) = 0.101$
 $S = 0.95$
 3195 reflections
 223 parameters
 0 restraints
 Primary atom site location: structure-invariant
 direct methods

Secondary atom site location: difference Fourier
 map
 Hydrogen site location: inferred from
 neighbouring sites
 H atoms treated by a mixture of independent
 and constrained refinement
 $w = 1/[\sigma^2(F_o^2) + (0.0336P)^2]$
 where $P = (F_o^2 + 2F_c^2)/3$
 $(\Delta/\sigma)_{\text{max}} < 0.001$
 $\Delta\rho_{\text{max}} = 0.17 \text{ e \AA}^{-3}$
 $\Delta\rho_{\text{min}} = -0.23 \text{ e \AA}^{-3}$

Special details

Geometry. All e.s.d.'s (except the e.s.d. in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell e.s.d.'s are taken into account individually in the estimation of e.s.d.'s in distances, angles and torsion angles; correlations between e.s.d.'s in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell e.s.d.'s is used for estimating e.s.d.'s involving l.s. planes.

Refinement. Refinement of F^2 against ALL reflections. The weighted R -factor wR and goodness of fit S are based on F^2 , conventional R -factors R are based on F , with F set to zero for negative F^2 . The threshold expression of $F^2 > \sigma(F^2)$ is used only for calculating R -factors(gt) etc. and is not relevant to the choice of reflections for refinement. R -factors based on F^2 are statistically about twice as large as those based on F , and R -factors based on ALL data will be even larger.

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (\AA^2)

	<i>x</i>	<i>y</i>	<i>z</i>	$U_{\text{iso}}^*/U_{\text{eq}}$
S1	0.80339 (6)	0.87951 (14)	0.18814 (3)	0.0447 (2)
O1	1.0501 (2)	0.3126 (4)	0.13496 (8)	0.0647 (7)
N1	0.6375 (2)	0.4530 (5)	0.18086 (10)	0.0544 (8)
H1A	0.684 (3)	0.492 (5)	0.2078 (11)	0.065*
H1B	0.576 (3)	0.353 (5)	0.1859 (10)	0.065*
C1	1.0977 (3)	0.7137 (6)	0.03620 (11)	0.0486 (8)
H1	1.0485	0.8384	0.0436	0.058*
C2	1.1576 (3)	0.7203 (6)	-0.00593 (11)	0.0583 (9)
H2	1.1483	0.8497	-0.0268	0.070*
C3	1.2307 (3)	0.5382 (7)	-0.01713 (11)	0.0576 (10)
H3	1.2698	0.5429	-0.0457	0.069*

C4	1.2458 (3)	0.3503 (6)	0.01372 (11)	0.0567 (9)
H4	1.2970	0.2285	0.0065	0.068*
C5	1.1859 (2)	0.3390 (5)	0.05554 (10)	0.0479 (8)
H5	1.1957	0.2084	0.0761	0.057*
C6	1.1110 (2)	0.5215 (5)	0.06716 (10)	0.0372 (7)
C7	1.0446 (3)	0.4961 (6)	0.11240 (11)	0.0421 (8)
C8	0.9699 (2)	0.7008 (5)	0.12804 (9)	0.0407 (8)
H8A	0.9049	0.7301	0.1027	0.049*
H8B	1.0198	0.8409	0.1310	0.049*
C9	0.9174 (2)	0.6565 (5)	0.17669 (9)	0.0389 (7)
H9	0.8789	0.5013	0.1753	0.047*
C10	1.0071 (2)	0.6649 (5)	0.22161 (9)	0.0330 (7)
C11	1.0841 (2)	0.8534 (5)	0.23096 (10)	0.0397 (7)
H11	1.0841	0.9755	0.2083	0.048*
C12	1.1609 (2)	0.8630 (6)	0.27341 (11)	0.0484 (8)
H12	1.2120	0.9912	0.2792	0.058*
C13	1.1620 (3)	0.6835 (6)	0.30718 (11)	0.0511 (9)
H13	1.2142	0.6889	0.3357	0.061*
C14	1.0852 (3)	0.4954 (6)	0.29848 (11)	0.0502 (9)
H14	1.0850	0.3743	0.3214	0.060*
C15	1.0089 (2)	0.4857 (5)	0.25623 (10)	0.0436 (8)
H15	0.9577	0.3574	0.2507	0.052*
C16	0.6144 (2)	0.6174 (5)	0.14430 (10)	0.0378 (7)
C17	0.5196 (3)	0.5843 (6)	0.10811 (11)	0.0507 (9)
H17	0.4728	0.4490	0.1084	0.061*
C18	0.4949 (3)	0.7500 (7)	0.07200 (11)	0.0577 (9)
H18	0.4307	0.7263	0.0483	0.069*
C19	0.5635 (3)	0.9510 (6)	0.07023 (11)	0.0559 (9)
H19	0.5454	1.0636	0.0460	0.067*
C20	0.6588 (3)	0.9817 (6)	0.10486 (10)	0.0465 (8)
H20	0.7062	1.1157	0.1035	0.056*
C21	0.6864 (2)	0.8180 (5)	0.14189 (9)	0.0338 (7)

Atomic displacement parameters (Å²)

	U^{11}	U^{22}	U^{33}	U^{12}	U^{13}	U^{23}
S1	0.0366 (4)	0.0552 (6)	0.0419 (5)	0.0055 (4)	0.0013 (3)	-0.0142 (4)
O1	0.0848 (17)	0.0482 (16)	0.0659 (16)	0.0140 (14)	0.0324 (13)	0.0095 (12)
N1	0.0539 (18)	0.055 (2)	0.0551 (19)	-0.0092 (15)	0.0070 (14)	0.0093 (15)
C1	0.0494 (19)	0.046 (2)	0.051 (2)	0.0072 (17)	0.0098 (16)	-0.0029 (16)
C2	0.066 (2)	0.062 (3)	0.049 (2)	-0.007 (2)	0.0129 (18)	0.0073 (17)
C3	0.056 (2)	0.076 (3)	0.043 (2)	-0.011 (2)	0.0160 (17)	-0.011 (2)
C4	0.054 (2)	0.064 (3)	0.054 (2)	0.005 (2)	0.0158 (17)	-0.0135 (19)
C5	0.0485 (18)	0.048 (2)	0.0475 (19)	0.0047 (17)	0.0076 (15)	-0.0042 (15)
C6	0.0360 (16)	0.041 (2)	0.0340 (17)	-0.0026 (16)	0.0022 (13)	-0.0057 (15)
C7	0.0443 (18)	0.041 (2)	0.0407 (19)	0.0012 (17)	0.0034 (14)	-0.0034 (16)
C8	0.0407 (17)	0.045 (2)	0.0359 (17)	0.0067 (16)	0.0029 (14)	-0.0011 (14)
C9	0.0352 (16)	0.0400 (19)	0.0422 (18)	0.0033 (15)	0.0077 (14)	-0.0058 (14)

C10	0.0278 (15)	0.039 (2)	0.0333 (17)	-0.0005 (15)	0.0076 (12)	-0.0016 (14)
C11	0.0399 (17)	0.040 (2)	0.0396 (18)	0.0015 (17)	0.0046 (14)	0.0029 (14)
C12	0.0359 (17)	0.055 (2)	0.054 (2)	-0.0087 (17)	0.0011 (15)	-0.0067 (18)
C13	0.0375 (18)	0.080 (3)	0.0357 (18)	0.0085 (19)	0.0028 (15)	-0.0006 (19)
C14	0.0458 (19)	0.060 (2)	0.046 (2)	0.0026 (18)	0.0103 (16)	0.0154 (16)
C15	0.0386 (17)	0.043 (2)	0.050 (2)	-0.0052 (16)	0.0071 (16)	0.0022 (16)
C16	0.0350 (16)	0.045 (2)	0.0346 (17)	0.0033 (17)	0.0080 (14)	-0.0009 (15)
C17	0.0433 (18)	0.059 (2)	0.050 (2)	-0.0076 (18)	0.0053 (16)	-0.0097 (18)
C18	0.047 (2)	0.081 (3)	0.043 (2)	0.003 (2)	-0.0101 (16)	-0.0156 (19)
C19	0.062 (2)	0.066 (3)	0.039 (2)	0.015 (2)	0.0013 (17)	0.0046 (16)
C20	0.0484 (19)	0.049 (2)	0.0423 (19)	0.0011 (17)	0.0056 (16)	0.0006 (16)
C21	0.0306 (15)	0.0382 (19)	0.0334 (17)	0.0019 (15)	0.0069 (13)	-0.0054 (14)

Geometric parameters (Å, °)

S1—C21	1.767 (3)	C9—H9	0.9800
S1—C9	1.843 (3)	C10—C11	1.382 (3)
O1—C7	1.209 (3)	C10—C15	1.388 (4)
N1—C16	1.374 (4)	C11—C12	1.379 (4)
N1—H1A	0.89 (3)	C11—H11	0.9300
N1—H1B	0.91 (3)	C12—C13	1.374 (4)
C1—C6	1.380 (4)	C12—H12	0.9300
C1—C2	1.384 (4)	C13—C14	1.377 (4)
C1—H1	0.9300	C13—H13	0.9300
C2—C3	1.370 (4)	C14—C15	1.371 (4)
C2—H2	0.9300	C14—H14	0.9300
C3—C4	1.360 (4)	C15—H15	0.9300
C3—H3	0.9300	C16—C17	1.393 (4)
C4—C5	1.377 (4)	C16—C21	1.400 (4)
C4—H4	0.9300	C17—C18	1.373 (4)
C5—C6	1.387 (4)	C17—H17	0.9300
C5—H5	0.9300	C18—C19	1.378 (4)
C6—C7	1.505 (4)	C18—H18	0.9300
C7—C8	1.515 (4)	C19—C20	1.369 (4)
C8—C9	1.522 (3)	C19—H19	0.9300
C8—H8A	0.9700	C20—C21	1.387 (4)
C8—H8B	0.9700	C20—H20	0.9300
C9—C10	1.512 (3)		
C21—S1—C9	102.73 (12)	C11—C10—C15	118.2 (3)
C16—N1—H1A	119 (2)	C11—C10—C9	121.9 (2)
C16—N1—H1B	116.2 (19)	C15—C10—C9	119.8 (3)
H1A—N1—H1B	115 (3)	C12—C11—C10	121.0 (3)
C6—C1—C2	119.8 (3)	C12—C11—H11	119.5
C6—C1—H1	120.1	C10—C11—H11	119.5
C2—C1—H1	120.1	C13—C12—C11	120.1 (3)
C3—C2—C1	120.6 (3)	C13—C12—H12	120.0
C3—C2—H2	119.7	C11—C12—H12	120.0

C1—C2—H2	119.7	C12—C13—C14	119.5 (3)
C4—C3—C2	119.8 (3)	C12—C13—H13	120.2
C4—C3—H3	120.1	C14—C13—H13	120.2
C2—C3—H3	120.1	C15—C14—C13	120.4 (3)
C3—C4—C5	120.5 (3)	C15—C14—H14	119.8
C3—C4—H4	119.7	C13—C14—H14	119.8
C5—C4—H4	119.7	C14—C15—C10	120.8 (3)
C4—C5—C6	120.3 (3)	C14—C15—H15	119.6
C4—C5—H5	119.9	C10—C15—H15	119.6
C6—C5—H5	119.9	N1—C16—C17	120.2 (3)
C1—C6—C5	118.9 (3)	N1—C16—C21	121.1 (3)
C1—C6—C7	123.0 (3)	C17—C16—C21	118.7 (3)
C5—C6—C7	118.0 (3)	C18—C17—C16	120.5 (3)
O1—C7—C6	119.7 (3)	C18—C17—H17	119.8
O1—C7—C8	121.4 (3)	C16—C17—H17	119.8
C6—C7—C8	118.9 (3)	C17—C18—C19	121.1 (3)
C7—C8—C9	112.7 (2)	C17—C18—H18	119.4
C7—C8—H8A	109.1	C19—C18—H18	119.4
C9—C8—H8A	109.1	C20—C19—C18	118.8 (3)
C7—C8—H8B	109.1	C20—C19—H19	120.6
C9—C8—H8B	109.1	C18—C19—H19	120.6
H8A—C8—H8B	107.8	C19—C20—C21	121.7 (3)
C10—C9—C8	115.0 (2)	C19—C20—H20	119.1
C10—C9—S1	104.94 (17)	C21—C20—H20	119.1
C8—C9—S1	111.25 (19)	C20—C21—C16	119.3 (3)
C10—C9—H9	108.5	C20—C21—S1	119.3 (2)
C8—C9—H9	108.5	C16—C21—S1	121.2 (2)
S1—C9—H9	108.5		

Hydrogen-bond geometry (\AA , $^\circ$)

Cg1 and Cg2 are the centroids of the C10—C15 and C16—C21 rings, respectively.

$D-H\cdots A$	$D-H$	$H\cdots A$	$D\cdots A$	$D-H\cdots A$
N1—H1B \cdots Cg1 ⁱ	0.91 (3)	2.55 (3)	3.400 (3)	155 (2)
C12—H12 \cdots Cg1 ⁱⁱ	0.93	2.86	3.581 (4)	135
C14—H14 \cdots Cg2 ⁱ	0.93	2.93	3.587 (3)	128

Symmetry codes: (i) $-x+3/2, y-1/2, -z+1/2$; (ii) $-x+5/2, y+1/2, -z+1/2$.