

Acta Crystallographica Section E

Structure Reports

Online

ISSN 1600-5368

2-[(Indan-1-ylidene)amino]ethanol

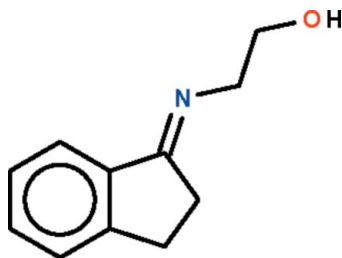
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Received 11 August 2011; accepted 13 August 2011

 Key indicators: single-crystal X-ray study; $T = 100$ K; mean $\sigma(\text{C}-\text{C}) = 0.002$ Å; R factor = 0.037; wR factor = 0.099; data-to-parameter ratio = 14.3.

 The five-membered ring of the title compound, $\text{C}_{11}\text{H}_{13}\text{NO}$, that is fused with the aromatic ring is approximately planar (r.m.s. deviation = 0.037 Å) despite the presence of the sp^3 -hybridized ethylene linkage. The hydroxy group of the N-bound hydroxyethyl chain serves as hydrogen-bond donor to the azomethine N atom of an adjacent molecule, generating a hydrogen-bonded C_2 -symmetric dimer.

Related literature

 The related $\text{C}_{13}\text{H}_{13}\text{NO}$ amine is a reagent in the synthesis of pharmaceuticals, see: Stange *et al.* (1957).


Experimental

Crystal data

 $\text{C}_{11}\text{H}_{13}\text{NO}$
 $M_r = 175.22$
 Monoclinic, $C2/c$
 $a = 16.0207$ (4) Å
 $b = 9.2002$ (2) Å
 $c = 13.0600$ (3) Å
 $\beta = 112.855$ (3)°
 $V = 1773.83$ (7) Å³
 $Z = 8$
 Cu $K\alpha$ radiation
 $\mu = 0.67$ mm⁻¹
 $T = 100$ K
 $0.30 \times 0.30 \times 0.10$ mm

Data collection

 Agilent SuperNova Dual diffractometer with an Atlas detector
 Absorption correction: multi-scan (*CrysAlis PRO*; Agilent, 2010)
 $T_{\min} = 0.825$, $T_{\max} = 0.937$
 3090 measured reflections
 1745 independent reflections
 1590 reflections with $I > 2\sigma(I)$
 $R_{\text{int}} = 0.015$

Refinement

 $R[F^2 > 2\sigma(F^2)] = 0.037$
 $wR(F^2) = 0.099$
 $S = 1.02$
 1745 reflections
 122 parameters
 H atoms treated by a mixture of independent and constrained refinement
 $\Delta\rho_{\max} = 0.29$ e Å⁻³
 $\Delta\rho_{\min} = -0.21$ e Å⁻³

Table 1

Hydrogen-bond geometry (Å, °).

$D-H\cdots A$	$D-H$	$H\cdots A$	$D\cdots A$	$D-H\cdots A$
$\text{O1}-\text{H1}\cdots\text{N1}^i$	0.91 (2)	1.91 (2)	2.820 (1)	173 (2)

 Symmetry code: (i) $-x + 1, y, -z + \frac{1}{2}$.

 Data collection: *CrysAlis PRO* (Agilent, 2010); cell refinement: *CrysAlis PRO*; data reduction: *CrysAlis PRO*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 2008); program(s) used to refine structure: *SHELXL97* (Sheldrick, 2008); molecular graphics: *X-SEED* (Barbour, 2001); software used to prepare material for publication: *publCIF* (Westrip, 2010).

We thank King Abdulaziz University and the University of Malaya for supporting this study.

Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: BT5608).

References

- Agilent (2010). *CrysAlis PRO*. Agilent Technologies, Yarnton, England.
 Barbour, L. J. (2001). *J. Supramol. Chem.* **1**, 189–191.
 Sheldrick, G. M. (2008). *Acta Cryst.* **A64**, 112–122.
 Stange, K., Friederich, H. & Amann, A. (1957). Ger. Patent 955497, 19570103.
 Westrip, S. P. (2010). *J. Appl. Cryst.* **43**, 920–925.

supporting information

Acta Cryst. (2011). E67, o2425 [doi:10.1107/S1600536811032843]

2-[(Indan-1-ylidene)amino]ethanol

Abdulrahman O. Al-Youbi, Abdullah M. Asiri, Hassan M. Faidallah, Khalid A. Alamry and Seik Weng Ng

S1. Comment

A enormously large number of Schiff base derivatives of aldehydes and ketones have been synthesized; however, 1-indanone represents an anomaly as only few have been reported. In the 2-aminoethanol derivative (Scheme I), the five-membered cyclohexene ring is planar despite the presence of sp^3 -hybridized ethylene linkage molecule (Fig. 1). The hydroxy group of the *N*-bound hydroxyethyl chain serves as hydrogen-bond donor to the azomethine N atom of an adjacent molecule to generate a hydrogen-bonded dinuclear molecule (Table 1). However, there is no significant π interaction of the rings as the distances between them exceed 3.5 Å (Fig. 2). The compound has not been reported in the chemical literature; on the other hand, the corresponding reduced amine is a reagent for the synthesis of pharmaceuticals (Stange *et al.*, 1957).

S2. Experimental

A mixture of 2-amino ethanol (0.6 g, 10 mmol) and 1-indanone (1.3 g, 10 mmol) in dry benzene (50 ml) was refluxed in a Dean-Stark apparatus until no more water was collected (in about 2 h). The solvent was then removed under reduced pressure and the residue treated with methanol. The solid which separated out was recrystallized from ethanol to give colorless, 418–419 K.

S3. Refinement

Carbon bound H-atoms were placed in calculated positions [C–H 0.95 to 0.99 Å, $U_{\text{iso}}(\text{H}) 1.2U_{\text{eq}}(\text{C})$] and were included in the refinement in the riding model approximation.

The hydroxy H-atom was located in a difference Fourier map and was freely refined.

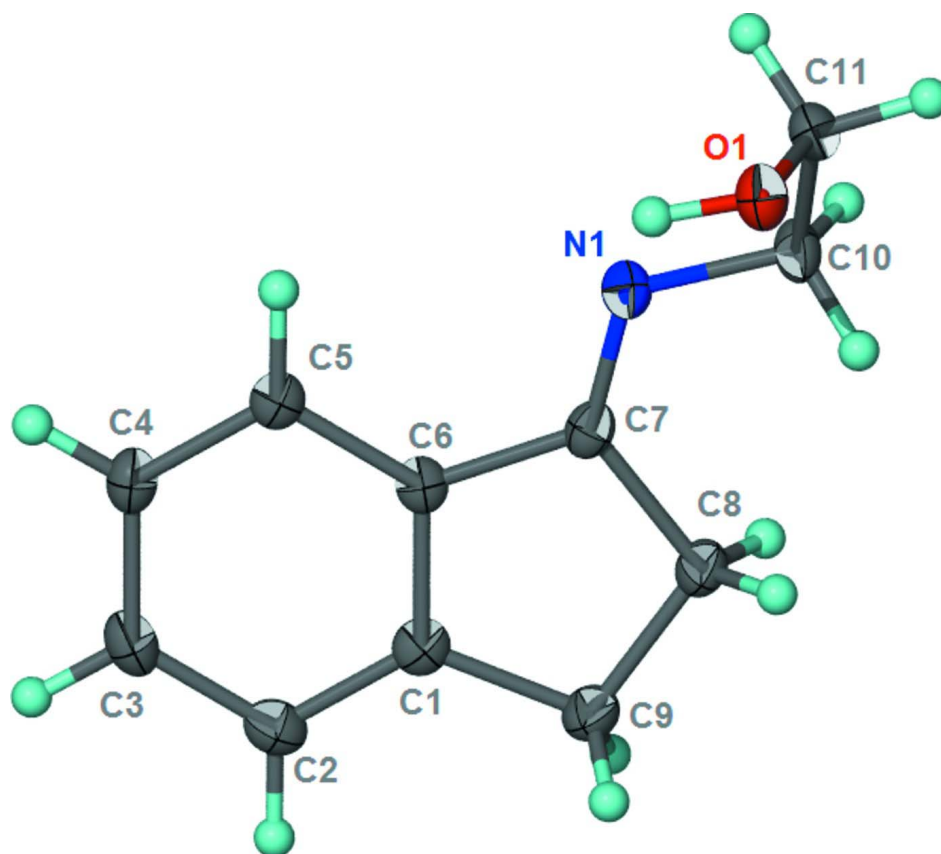
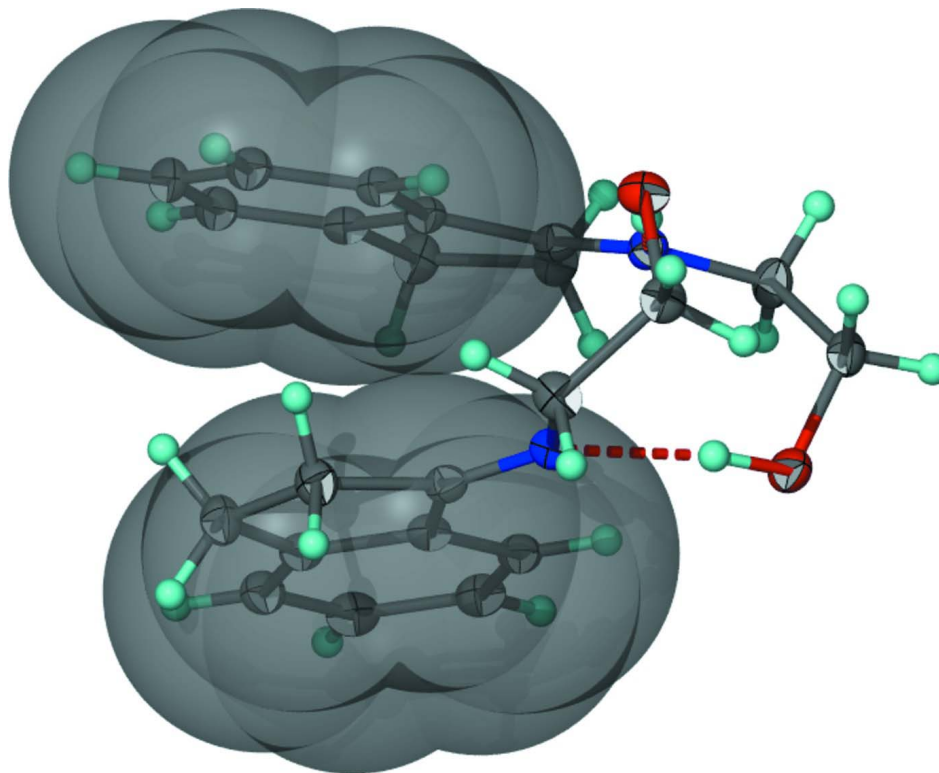


Figure 1

Anisotropic displacement ellipsoid plot (Barbour, 2001) of C₁₃H₁₁NO at the 70% probability level; hydrogen atoms are drawn as spheres of arbitrary radius.

**Figure 2**

Hydrogen-bonded dimer. The atoms of the aromatic rings are shown with their van der Waals surfaces.

2-[(Indan-1-ylidene)amino]ethanol

Crystal data

$C_{11}H_{13}NO$

$M_r = 175.22$

Monoclinic, $C2/c$

Hall symbol: $-C 2yc$

$a = 16.0207 (4) \text{ \AA}$

$b = 9.2002 (2) \text{ \AA}$

$c = 13.0600 (3) \text{ \AA}$

$\beta = 112.855 (3)^\circ$

$V = 1773.83 (7) \text{ \AA}^3$

$Z = 8$

$F(000) = 752$

$D_x = 1.312 \text{ Mg m}^{-3}$

Cu $K\alpha$ radiation, $\lambda = 1.54184 \text{ \AA}$

Cell parameters from 1977 reflections

$\theta = 3.7\text{--}74.2^\circ$

$\mu = 0.67 \text{ mm}^{-1}$

$T = 100 \text{ K}$

Prism, colorless

$0.30 \times 0.30 \times 0.10 \text{ mm}$

Data collection

Agilent SuperNova Dual

diffractometer with an Atlas detector

Radiation source: SuperNova (Cu) X-ray

Source

Mirror monochromator

Detector resolution: $10.4041 \text{ pixels mm}^{-1}$

ω scans

Absorption correction: multi-scan

(*CrysAlis PRO*; Agilent, 2010)

$T_{\min} = 0.825$, $T_{\max} = 0.937$

3090 measured reflections

1745 independent reflections

1590 reflections with $I > 2\sigma(I)$

$R_{\text{int}} = 0.015$

$\theta_{\max} = 74.4^\circ$, $\theta_{\min} = 5.7^\circ$

$h = -19 \rightarrow 19$

$k = -11 \rightarrow 6$

$l = -15 \rightarrow 16$

Refinement

Refinement on F^2

Least-squares matrix: full

$R[F^2 > 2\sigma(F^2)] = 0.037$

$wR(F^2) = 0.099$

$S = 1.02$

1745 reflections

122 parameters

0 restraints

Primary atom site location: structure-invariant
direct methods

Secondary atom site location: difference Fourier
map

Hydrogen site location: inferred from
neighbouring sites

H atoms treated by a mixture of independent
and constrained refinement

$w = 1/[\sigma^2(F_o^2) + (0.0573P)^2 + 1.1159P]$

where $P = (F_o^2 + 2F_c^2)/3$

$(\Delta/\sigma)_{\max} = 0.001$

$\Delta\rho_{\max} = 0.29 \text{ e } \text{\AA}^{-3}$

$\Delta\rho_{\min} = -0.21 \text{ e } \text{\AA}^{-3}$

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (\AA^2)

	x	y	z	$U_{\text{iso}}^*/U_{\text{eq}}$
O1	0.61199 (6)	0.25279 (9)	0.23182 (7)	0.0186 (2)
H1	0.5586 (14)	0.303 (2)	0.2000 (16)	0.051 (6)*
N1	0.54795 (6)	0.41152 (11)	0.38232 (7)	0.0155 (2)
C1	0.50616 (8)	0.78897 (13)	0.39934 (9)	0.0162 (3)
C2	0.44473 (8)	0.90379 (13)	0.36833 (10)	0.0186 (3)
H2	0.4646	1.0007	0.3891	0.022*
C3	0.35384 (8)	0.87441 (13)	0.30649 (10)	0.0194 (3)
H3	0.3114	0.9521	0.2851	0.023*
C4	0.32389 (8)	0.73221 (13)	0.27531 (9)	0.0180 (3)
H4	0.2615	0.7139	0.2334	0.022*
C5	0.38514 (8)	0.61760 (13)	0.30546 (9)	0.0160 (3)
H5	0.3653	0.5209	0.2840	0.019*
C6	0.47636 (7)	0.64727 (12)	0.36786 (9)	0.0148 (3)
C7	0.55414 (7)	0.54679 (13)	0.40624 (9)	0.0147 (3)
C8	0.63802 (7)	0.63476 (13)	0.47441 (9)	0.0174 (3)
H8A	0.6859	0.6219	0.4450	0.021*
H8B	0.6620	0.6034	0.5531	0.021*
C9	0.60731 (8)	0.79464 (13)	0.46423 (10)	0.0192 (3)
H9A	0.6218	0.8381	0.5385	0.023*
H9B	0.6372	0.8525	0.4242	0.023*
C10	0.63052 (8)	0.32330 (13)	0.42013 (9)	0.0181 (3)
H10A	0.6415	0.2820	0.4942	0.022*
H10B	0.6830	0.3850	0.4268	0.022*
C11	0.62087 (8)	0.20122 (12)	0.33805 (9)	0.0173 (3)
H11A	0.6747	0.1373	0.3680	0.021*
H11B	0.5670	0.1423	0.3299	0.021*

Atomic displacement parameters (\AA^2)

	U^{11}	U^{22}	U^{33}	U^{12}	U^{13}	U^{23}
O1	0.0162 (4)	0.0219 (4)	0.0178 (4)	0.0038 (3)	0.0067 (3)	0.0012 (3)
N1	0.0144 (5)	0.0173 (5)	0.0142 (5)	0.0019 (4)	0.0047 (4)	0.0002 (4)
C1	0.0172 (6)	0.0184 (6)	0.0147 (5)	-0.0006 (4)	0.0080 (4)	-0.0002 (4)

C2	0.0223 (6)	0.0158 (5)	0.0194 (6)	0.0006 (5)	0.0102 (5)	0.0001 (4)
C3	0.0198 (6)	0.0199 (6)	0.0197 (6)	0.0062 (5)	0.0090 (5)	0.0043 (5)
C4	0.0144 (5)	0.0230 (6)	0.0163 (5)	0.0023 (5)	0.0055 (4)	0.0018 (4)
C5	0.0158 (6)	0.0182 (6)	0.0147 (5)	-0.0002 (4)	0.0066 (4)	-0.0003 (4)
C6	0.0153 (6)	0.0168 (6)	0.0131 (5)	0.0013 (4)	0.0064 (4)	0.0005 (4)
C7	0.0128 (5)	0.0191 (6)	0.0120 (5)	-0.0007 (4)	0.0044 (4)	0.0000 (4)
C8	0.0139 (5)	0.0187 (6)	0.0171 (5)	-0.0007 (4)	0.0033 (4)	-0.0014 (4)
C9	0.0165 (6)	0.0174 (6)	0.0222 (6)	-0.0017 (4)	0.0058 (5)	-0.0031 (5)
C10	0.0144 (5)	0.0197 (6)	0.0166 (6)	0.0041 (4)	0.0020 (4)	0.0005 (4)
C11	0.0159 (5)	0.0161 (5)	0.0191 (6)	0.0028 (4)	0.0058 (4)	0.0018 (4)

Geometric parameters (Å, °)

O1—C11	1.4201 (14)	C5—H5	0.9500
O1—H1	0.91 (2)	C6—C7	1.4742 (15)
N1—C7	1.2776 (15)	C7—C8	1.5245 (15)
N1—C10	1.4646 (14)	C8—C9	1.5403 (16)
C1—C2	1.3924 (16)	C8—H8A	0.9900
C1—C6	1.3943 (16)	C8—H8B	0.9900
C1—C9	1.5101 (16)	C9—H9A	0.9900
C2—C3	1.3900 (16)	C9—H9B	0.9900
C2—H2	0.9500	C10—C11	1.5185 (16)
C3—C4	1.3985 (17)	C10—H10A	0.9900
C3—H3	0.9500	C10—H10B	0.9900
C4—C5	1.3890 (16)	C11—H11A	0.9900
C4—H4	0.9500	C11—H11B	0.9900
C5—C6	1.3962 (15)		
C11—O1—H1	109.4 (12)	C7—C8—H8A	110.5
C7—N1—C10	118.90 (10)	C9—C8—H8A	110.5
C2—C1—C6	120.08 (11)	C7—C8—H8B	110.5
C2—C1—C9	128.34 (11)	C9—C8—H8B	110.5
C6—C1—C9	111.57 (10)	H8A—C8—H8B	108.7
C3—C2—C1	118.95 (11)	C1—C9—C8	104.63 (9)
C3—C2—H2	120.5	C1—C9—H9A	110.8
C1—C2—H2	120.5	C8—C9—H9A	110.8
C2—C3—C4	120.97 (11)	C1—C9—H9B	110.8
C2—C3—H3	119.5	C8—C9—H9B	110.8
C4—C3—H3	119.5	H9A—C9—H9B	108.9
C5—C4—C3	120.20 (11)	N1—C10—C11	109.93 (9)
C5—C4—H4	119.9	N1—C10—H10A	109.7
C3—C4—H4	119.9	C11—C10—H10A	109.7
C4—C5—C6	118.76 (11)	N1—C10—H10B	109.7
C4—C5—H5	120.6	C11—C10—H10B	109.7
C6—C5—H5	120.6	H10A—C10—H10B	108.2
C1—C6—C5	121.05 (11)	O1—C11—C10	112.74 (9)
C1—C6—C7	109.80 (10)	O1—C11—H11A	109.0
C5—C6—C7	129.11 (11)	C10—C11—H11A	109.0

N1—C7—C6	123.55 (10)	O1—C11—H11B	109.0
N1—C7—C8	128.83 (10)	C10—C11—H11B	109.0
C6—C7—C8	107.61 (10)	H11A—C11—H11B	107.8
C7—C8—C9	106.09 (9)		
C6—C1—C2—C3	0.39 (17)	C10—N1—C7—C8	-2.07 (17)
C9—C1—C2—C3	178.90 (11)	C1—C6—C7—N1	-174.79 (10)
C1—C2—C3—C4	-0.12 (17)	C5—C6—C7—N1	2.73 (18)
C2—C3—C4—C5	-0.35 (18)	C1—C6—C7—C8	4.31 (12)
C3—C4—C5—C6	0.53 (16)	C5—C6—C7—C8	-178.17 (11)
C2—C1—C6—C5	-0.20 (17)	N1—C7—C8—C9	173.44 (11)
C9—C1—C6—C5	-178.95 (10)	C6—C7—C8—C9	-5.60 (12)
C2—C1—C6—C7	177.55 (10)	C2—C1—C9—C8	179.04 (11)
C9—C1—C6—C7	-1.19 (13)	C6—C1—C9—C8	-2.35 (13)
C4—C5—C6—C1	-0.26 (16)	C7—C8—C9—C1	4.78 (12)
C4—C5—C6—C7	-177.54 (10)	C7—N1—C10—C11	-148.98 (10)
C10—N1—C7—C6	176.83 (10)	N1—C10—C11—O1	63.93 (12)

Hydrogen-bond geometry (Å, °)

<i>D</i> —H \cdots <i>A</i>	<i>D</i> —H	H \cdots <i>A</i>	<i>D</i> \cdots <i>A</i>	<i>D</i> —H \cdots <i>A</i>
O1—H1 \cdots N1 ⁱ	0.91 (2)	1.91 (2)	2.820 (1)	173 (2)

Symmetry code: (i) $-x+1, y, -z+1/2$.