organic compounds

1-(4-Methoxyphenyl)-2-methyl-1H-indole-3-carbonitrile

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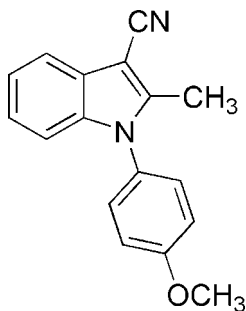
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In the title compound, $C_{17}H_{14}N_2O$, the dihedral angle between the indole ring system and the benzene ring is 58.41 (4)°. The crystal packing features π–π stacking [shortest centroid–centroid separation = 3.8040 (9) Å] and C—H$\cdots\pi$ interactions.

Related literature

For the synthesis of the title compound, see: Du *et al.* (2006). For its precursor, see: Jin *et al.* (2009). For a related structure, see: Yang *et al.* (2011).

Experimental

Crystal data

$C_{17}H_{14}N_2O$	Triclinic, $P\bar{1}$
M_r = 262.30	a = 7.7381 (10) Å
b = 9.4598 (14) Å	Z = 2
c = 9.7976 (16) Å	Mo $K\alpha$ radiation
α = 95.983 (2)°	μ = 0.08 mm^{-1}
β = 95.464 (4)°	T = 113 K
γ = 106.295 (5)°	$0.20 \times 0.18 \times 0.16$ mm
V = 678.79 (17) Å^3	

Data collection

Rigaku Saturn724 CCD diffractometer	8580 measured reflections
Absorption correction: multi-scan (*CrystalClear*; Rigaku, 2009)	3210 independent reflections
T_{min} = 0.984, T_{max} = 0.987	2146 reflections with $I > 2\sigma(I)$
	R_{int} = 0.035

Refinement

$R[F^2 > 2\sigma(F^2)]$ = 0.035	183 parameters
$wR(F^2)$ = 0.090	H-atom parameters constrained
S = 1.01	$\Delta\rho_{max}$ = 0.19 e Å^{-3}
3210 reflections	$\Delta\rho_{min}$ = −0.24 e Å^{-3}

Table 1
Hydrogen-bond geometry (Å, °).

*Cg*2 and *Cg*3 are the centroids of the C2–C7 and C8–C13 rings, respectively.

D—H$\cdots A$	D—H	H$\cdots A$	$D\cdots A$	D—H$\cdots A$
C3—H3$\cdots$*Cg*3^i	0.95	2.83	3.6542 (14)	146
C10—H10$\cdots$*Cg*2^{ii}	0.95	2.95	3.7133 (14)	138

Symmetry codes: (i) $x - 1, y, z$; (ii) $-x + 1, -y + 1, -z + 2$.

Data collection: *CrystalClear* (Rigaku, 2009); cell refinement: *CrystalClear*; data reduction: *CrystalClear*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 2008); program(s) used to refine structure: *SHELXL97* (Sheldrick, 2008); molecular graphics: *CrystalStructure* (Rigaku, 2009); software used to prepare material for publication: *CrystalStructure*.

Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: HB6329).

References

Du, Y., Liu, R., Linn, G. & Zhao, K. (2006). *Org. Lett.* **8**, 5919–5922.
Jin, H., Li, P., Liu, B. & Cheng, X. (2009). *Acta Cryst.* E**65**, o236.
Rigaku (2009). *CrystalClear* and *CrystalStructure*. Rigaku Corporation, Tokyo, Japan.
Sheldrick, G. M. (2008). *Acta Cryst.* A**64**, 112–122.
Yang, K., Li, P.-F., Liu, Y. & Fang, Z.-Z. (2011). *Acta Cryst.* E**67**, o1041.

supporting information

1-(4-Methoxyphenyl)-2-methyl-1*H*-indole-3-carbonitrile

Qiao Yan and Xiuxiang Qi

S1. Comment

In the molecular structure of the title compound, (I), (Fig. 1), the indole ring is almost planar with a dihedral angle of 2.66 (6)° between its pyrrole ring and fused benzene ring, which is greater than that [0.85 (6)°] of the 1-(2-chlorophenyl)-6-fluoro-2-methyl-1*H*-indole-3-carbonitrile reported by Yang *et al.* (2011). The indole ring constructs an angle of 58.41 (4) ° with the methoxylbenzene ring, which is much less than that [80.91 (5)°] reported by Yang *et al.* (2011).

In the crystal packing, π-π stacking interaction and C—H⋯π interaction help establish the molecular packing. The shortest centroid-centroid separation is 3.8040 (9) Å, which occurs between the benzo part and pyrrole part of the molecules.

S2. Experimental

The title compound was prepared according to the method of the literature (Du, *et al.*, 2006). Colourless prisms of (I) were grown from a mixture of ethyl actate and petroleum ether.

S3. Refinement

All H atoms were positioned geometrically (C—H = 0.95 and 0.98 Å)and refined as riding with U_{iso}(H) = 1.2U_{eq}(CH) or 1.5U_{eq}(CH$_3$).
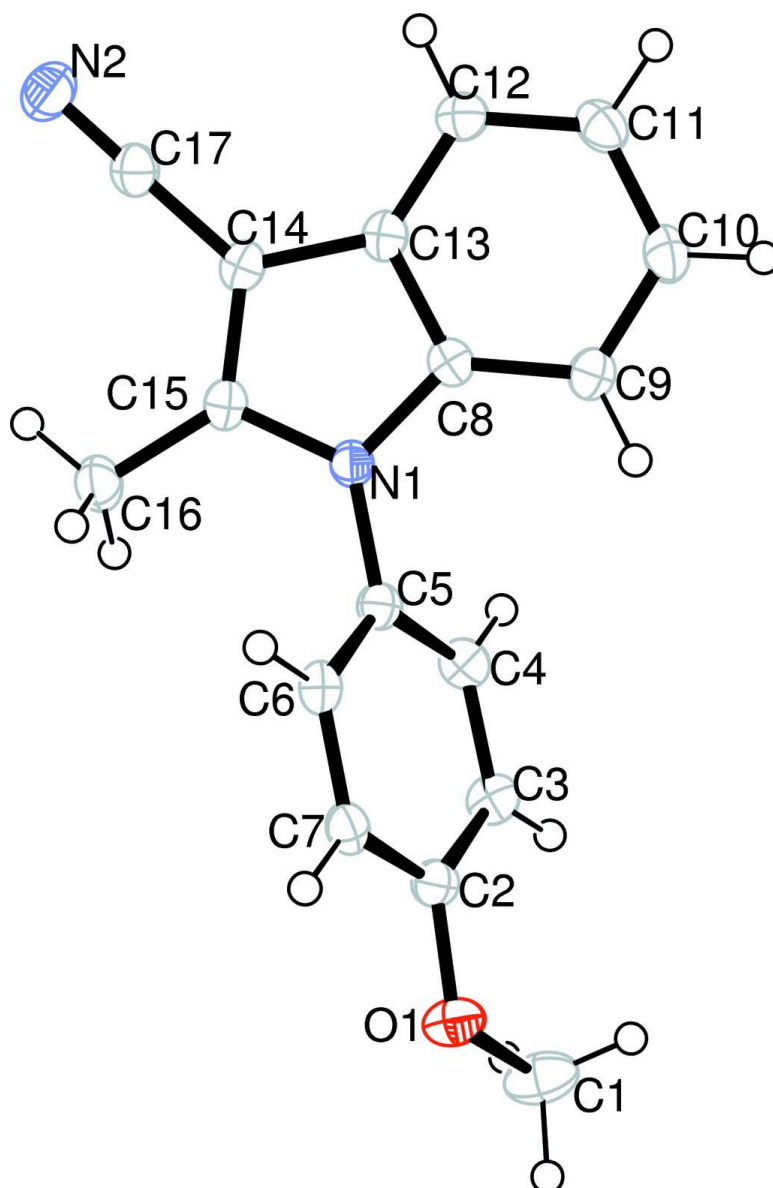
Figure 1

The molecular structure of molecule one of (I) with 50% probability displacement ellipsoids.

1-(4-Methoxyphenyl)-2-methyl-1*H*-indole-3-carbonitrile

Crystal data

$C_{17}H_{14}N_2O$	$\gamma = 106.295\,(5)°$
$M_r = 262.30$	$V = 678.79\,(17)$ Å^3
Triclinic, $P\bar{1}$	$Z = 2$
Hall symbol: -P 1	$F(000) = 276$
$a = 7.7381\,(10)$ Å	$D_x = 1.283$ Mg m^{-3}
$b = 9.4598\,(14)$ Å	Mo $K\alpha$ radiation, $\lambda = 0.71073$ Å
$c = 9.7976\,(16)$ Å	Cell parameters from 2366 reflections
$\alpha = 95.983\,(2)°$	$\theta = 2.1$–27.9°
$\beta = 95.464\,(4)°$	$\mu = 0.08$ mm^{-1}

T = 113 K	$0.20 \times 0.18 \times 0.16$ mm
Prism, colorless	

Data collection

Rigaku Saturn724 CCD diffractometer	8580 measured reflections
Radiation source: rotating anode	3210 independent reflections
Multilayer monochromator	2146 reflections with $I > 2\sigma(I)$
Detector resolution: 14.22 pixels mm^{-1}	R_{int} = 0.035
ω and φ scans	θ_{max} = 27.9°, θ_{min} = 2.1°
Absorption correction: multi-scan (*CrystalClear*; Rigaku, 2009)	$h = -10 \rightarrow 10$
	$k = -12 \rightarrow 12$
	$l = -12 \rightarrow 12$
T_{min} = 0.984, T_{max} = 0.987	

Refinement

Refinement on F^2	Secondary atom site location: difference Fourier map
Least-squares matrix: full	
$R[F^2 > 2\sigma(F^2)]$ = 0.035	Hydrogen site location: inferred from neighbouring sites
$wR(F^2)$ = 0.090	
S = 1.01	H-atom parameters constrained
3210 reflections	$w = 1/[\sigma^2(F_o^2) + (0.038P)^2]$ where $P = (F_o^2 + 2F_c^2)/3$
183 parameters	
0 restraints	$(\Delta/\sigma)_{max}$ = 0.001
Primary atom site location: structure-invariant direct methods	$\Delta\rho_{max}$ = 0.19 e Å^{-3}
	$\Delta\rho_{min}$ = −0.24 e Å^{-3}

Special details

Geometry. All e.s.d.'s (except the e.s.d. in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell e.s.d.'s are taken into account individually in the estimation of e.s.d.'s in distances, angles and torsion angles; correlations between e.s.d.'s in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell e.s.d.'s is used for estimating e.s.d.'s involving l.s. planes.

Refinement. Refinement of F^2 against ALL reflections. The weighted R-factor wR and goodness of fit S are based on F^2, conventional R-factors R are based on F, with F set to zero for negative F^2. The threshold expression of $F^2 > \sigma(F^2)$ is used only for calculating R-factors(gt) *etc.* and is not relevant to the choice of reflections for refinement. R-factors based on F^2 are statistically about twice as large as those based on F, and R- factors based on ALL data will be even larger.

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (Å^2)

	x	y	z	U_{iso}*/U_{eq}
O1	1.23428 (10)	0.89022 (9)	0.06414 (8)	0.0300 (2)
N1	0.63352 (12)	0.69683 (10)	0.34222 (9)	0.0207 (2)
N2	0.27094 (14)	0.77814 (11)	0.68887 (10)	0.0350 (3)
C1	1.34704 (16)	0.79705 (15)	0.03434 (14)	0.0386 (3)
H1A	1.3990	0.7732	0.1211	0.058*
H1B	1.4452	0.8492	−0.0148	0.058*
H1C	1.2741	0.7050	−0.0237	0.058*
C2	1.08888 (14)	0.83382 (12)	0.13188 (11)	0.0223 (3)
C3	1.04548 (15)	0.69409 (12)	0.17549 (11)	0.0246 (3)
H3	1.1173	0.6293	0.1573	0.030*
C4	0.89572 (14)	0.65052 (12)	0.24598 (11)	0.0241 (3)
H4	0.8661	0.5558	0.2771	0.029*
C5	0.78939 (14)	0.74363 (12)	0.27134 (11)	0.0207 (2)
C6	0.83139 (14)	0.88247 (12)	0.22493 (11)	0.0221 (2)

H6	0.7570	0.9457	0.2400	0.026*
C7	0.98146 (14)	0.92716 (12)	0.15712 (11)	0.0226 (2)
H7	1.0119	1.0225	0.1274	0.027*
C8	0.48670 (14)	0.57024 (11)	0.29653 (11)	0.0199 (2)
C9	0.45885 (14)	0.46469 (12)	0.17956 (11)	0.0239 (3)
H9	0.5482	0.4695	0.1186	0.029*
C10	0.29541 (15)	0.35277 (13)	0.15628 (12)	0.0279 (3)
H10	0.2715	0.2794	0.0774	0.033*
C11	0.16419 (15)	0.34554 (12)	0.24707 (12)	0.0273 (3)
H11	0.0531	0.2676	0.2283	0.033*
C12	0.19364 (15)	0.44962 (12)	0.36305 (12)	0.0235 (3)
H12	0.1051	0.4429	0.4248	0.028*
C13	0.35650 (14)	0.56517 (12)	0.38782 (11)	0.0205 (2)
C14	0.42996 (15)	0.69530 (12)	0.48926 (11)	0.0216 (2)
C15	0.59797 (14)	0.77272 (12)	0.45929 (11)	0.0213 (2)
C16	0.72912 (15)	0.91064 (12)	0.53687 (11)	0.0293 (3)
H16A	0.7311	0.9943	0.4852	0.044*
H16B	0.8506	0.8976	0.5486	0.044*
H16C	0.6920	0.9310	0.6280	0.044*
C17	0.34392 (15)	0.74071 (12)	0.60128 (12)	0.0246 (3)

Atomic displacement parameters (Å^2)

	U^{11}	U^{22}	U^{33}	U^{12}	U^{13}	U^{23}
O1	0.0284 (4)	0.0321 (5)	0.0346 (5)	0.0114 (4)	0.0121 (4)	0.0123 (4)
N1	0.0221 (5)	0.0197 (5)	0.0199 (5)	0.0056 (4)	0.0027 (4)	0.0018 (4)
N2	0.0394 (6)	0.0346 (6)	0.0341 (6)	0.0142 (5)	0.0124 (5)	0.0027 (5)
C1	0.0320 (7)	0.0475 (8)	0.0473 (8)	0.0209 (6)	0.0166 (6)	0.0203 (7)
C2	0.0222 (6)	0.0251 (6)	0.0184 (6)	0.0054 (5)	0.0012 (5)	0.0034 (5)
C3	0.0247 (6)	0.0267 (6)	0.0265 (6)	0.0128 (5)	0.0033 (5)	0.0070 (5)
C4	0.0266 (6)	0.0217 (6)	0.0255 (6)	0.0086 (5)	0.0018 (5)	0.0073 (5)
C5	0.0208 (5)	0.0225 (6)	0.0179 (5)	0.0057 (5)	0.0006 (4)	0.0024 (4)
C6	0.0259 (6)	0.0199 (6)	0.0203 (6)	0.0085 (5)	0.0002 (5)	0.0000 (5)
C7	0.0282 (6)	0.0178 (6)	0.0200 (6)	0.0048 (5)	0.0010 (5)	0.0019 (4)
C8	0.0208 (5)	0.0183 (5)	0.0212 (6)	0.0072 (4)	0.0000 (4)	0.0041 (4)
C9	0.0268 (6)	0.0237 (6)	0.0226 (6)	0.0098 (5)	0.0032 (5)	0.0024 (5)
C10	0.0307 (6)	0.0223 (6)	0.0287 (7)	0.0077 (5)	0.0005 (5)	−0.0022 (5)
C11	0.0238 (6)	0.0201 (6)	0.0361 (7)	0.0055 (5)	0.0001 (5)	0.0016 (5)
C12	0.0234 (6)	0.0219 (6)	0.0284 (6)	0.0098 (5)	0.0050 (5)	0.0067 (5)
C13	0.0237 (6)	0.0193 (5)	0.0209 (6)	0.0104 (5)	0.0011 (5)	0.0039 (4)
C14	0.0261 (6)	0.0216 (6)	0.0195 (6)	0.0107 (5)	0.0027 (4)	0.0038 (5)
C15	0.0262 (6)	0.0203 (6)	0.0184 (6)	0.0091 (5)	0.0014 (4)	0.0023 (4)
C16	0.0339 (6)	0.0258 (6)	0.0243 (6)	0.0046 (5)	0.0023 (5)	−0.0006 (5)
C17	0.0279 (6)	0.0213 (6)	0.0258 (6)	0.0090 (5)	0.0028 (5)	0.0040 (5)

Geometric parameters (Å, º)

O1—C2	1.3675 (12)	C7—H7	0.9500
O1—C1	1.4317 (13)	C8—C9	1.3965 (14)
N1—C15	1.3805 (13)	C8—C13	1.4038 (14)
N1—C8	1.3976 (13)	C9—C10	1.3842 (15)
N1—C5	1.4342 (13)	C9—H9	0.9500
N2—C17	1.1497 (13)	C10—C11	1.4045 (15)
C1—H1A	0.9800	C10—H10	0.9500
C1—H1B	0.9800	C11—C12	1.3799 (15)
C1—H1C	0.9800	C11—H11	0.9500
C2—C7	1.3919 (15)	C12—C13	1.3984 (15)
C2—C3	1.3923 (15)	C12—H12	0.9500
C3—C4	1.3905 (15)	C13—C14	1.4391 (15)
C3—H3	0.9500	C14—C15	1.3770 (15)
C4—C5	1.3837 (14)	C14—C17	1.4261 (15)
C4—H4	0.9500	C15—C16	1.4877 (15)
C5—C6	1.3966 (14)	C16—H16A	0.9800
C6—C7	1.3779 (14)	C16—H16B	0.9800
C6—H6	0.9500	C16—H16C	0.9800
C2—O1—C1	117.18 (9)	N1—C8—C13	108.14 (9)
C15—N1—C8	109.22 (9)	C10—C9—C8	117.08 (10)
C15—N1—C5	126.23 (9)	C10—C9—H9	121.5
C8—N1—C5	124.42 (9)	C8—C9—H9	121.5
O1—C1—H1A	109.5	C9—C10—C11	121.28 (10)
O1—C1—H1B	109.5	C9—C10—H10	119.4
H1A—C1—H1B	109.5	C11—C10—H10	119.4
O1—C1—H1C	109.5	C12—C11—C10	121.21 (11)
H1A—C1—H1C	109.5	C12—C11—H11	119.4
H1B—C1—H1C	109.5	C10—C11—H11	119.4
O1—C2—C7	115.32 (10)	C11—C12—C13	118.65 (11)
O1—C2—C3	124.61 (10)	C11—C12—H12	120.7
C7—C2—C3	120.07 (10)	C13—C12—H12	120.7
C4—C3—C2	119.16 (10)	C12—C13—C8	119.40 (10)
C4—C3—H3	120.4	C12—C13—C14	134.75 (10)
C2—C3—H3	120.4	C8—C13—C14	105.80 (10)
C5—C4—C3	120.69 (10)	C15—C14—C17	124.72 (10)
C5—C4—H4	119.7	C15—C14—C13	108.77 (9)
C3—C4—H4	119.7	C17—C14—C13	126.50 (10)
C4—C5—C6	119.91 (10)	C14—C15—N1	108.06 (10)
C4—C5—N1	120.31 (10)	C14—C15—C16	128.79 (10)
C6—C5—N1	119.76 (10)	N1—C15—C16	123.11 (10)
C7—C6—C5	119.59 (10)	C15—C16—H16A	109.5
C7—C6—H6	120.2	C15—C16—H16B	109.5
C5—C6—H6	120.2	H16A—C16—H16B	109.5
C6—C7—C2	120.56 (10)	C15—C16—H16C	109.5
C6—C7—H7	119.7	H16A—C16—H16C	109.5

C2—C7—H7	119.7	H16B—C16—H16C	109.5
C9—C8—N1	129.45 (10)	N2—C17—C14	178.02 (12)
C9—C8—C13	122.36 (10)		
C1—O1—C2—C7	−179.16 (9)	C9—C10—C11—C12	−0.16 (17)
C1—O1—C2—C3	1.16 (16)	C10—C11—C12—C13	1.13 (17)
O1—C2—C3—C4	178.73 (9)	C11—C12—C13—C8	−1.51 (16)
C7—C2—C3—C4	−0.94 (16)	C11—C12—C13—C14	175.65 (11)
C2—C3—C4—C5	0.76 (16)	C9—C8—C13—C12	0.98 (16)
C3—C4—C5—C6	0.52 (16)	N1—C8—C13—C12	178.74 (9)
C3—C4—C5—N1	179.03 (9)	C9—C8—C13—C14	−176.92 (9)
C15—N1—C5—C4	125.90 (12)	N1—C8—C13—C14	0.84 (11)
C8—N1—C5—C4	−58.75 (14)	C12—C13—C14—C15	−178.10 (11)
C15—N1—C5—C6	−55.59 (14)	C8—C13—C14—C15	−0.67 (12)
C8—N1—C5—C6	119.76 (11)	C12—C13—C14—C17	0.6 (2)
C4—C5—C6—C7	−1.62 (15)	C8—C13—C14—C17	178.00 (10)
N1—C5—C6—C7	179.86 (9)	C17—C14—C15—N1	−178.46 (10)
C5—C6—C7—C2	1.45 (16)	C13—C14—C15—N1	0.24 (12)
O1—C2—C7—C6	−179.86 (9)	C17—C14—C15—C16	3.84 (18)
C3—C2—C7—C6	−0.16 (16)	C13—C14—C15—C16	−177.46 (10)
C15—N1—C8—C9	176.83 (10)	C8—N1—C15—C14	0.29 (12)
C5—N1—C8—C9	0.80 (17)	C5—N1—C15—C14	176.23 (9)
C15—N1—C8—C13	−0.72 (12)	C8—N1—C15—C16	178.15 (9)
C5—N1—C8—C13	−176.75 (9)	C5—N1—C15—C16	−5.91 (16)
N1—C8—C9—C10	−177.26 (10)	C15—C14—C17—N2	110 (4)
C13—C8—C9—C10	−0.01 (16)	C13—C14—C17—N2	−68 (4)
C8—C9—C10—C11	−0.40 (16)		

Hydrogen-bond geometry (Å, º)

Cg2 and Cg3 are the centroids of the C2–C7 and C8–C13 rings, respectively.

D—H···A	D—H	H···A	D···A	D—H···A
C3—H3···*Cg*3[i]	0.95	2.83	3.6542 (14)	146
C10—H10···*Cg*2[ii]	0.95	2.95	3.7133 (14)	138

Symmetry codes: (i) *x*−1, *y*, *z*; (ii) −*x*+1, −*y*+1, −*z*+2.