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## Structure Reports

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## (5*S*,6*R*)-6-Bromo-6-methyl-5-phenyl-3,4,5,6-tetrahydro-2*H*-cyclopenta[*b*]-pyran-7-one

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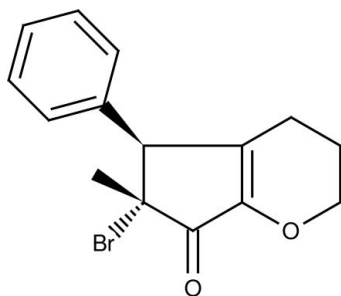
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Key indicators: single-crystal X-ray study;  $T = 173$  K; mean  $\sigma(\text{C}-\text{C}) = 0.012$  Å;  $R$  factor = 0.082;  $wR$  factor = 0.217; data-to-parameter ratio = 14.8.

The title compound,  $\text{C}_{15}\text{H}_{15}\text{BrO}_2$ , was synthesized by a Brønsted acid-catalysed domino electrocyclozation-halogenation reaction. The five-membered ring is essentially planar (r.m.s. deviation 0.006 Å) and forms a dihedral angle of  $72.7(3)^\circ$  with the attached phenyl ring. The six-membered heterocycle adopts a half-chair conformation. The crystal packing is stabilized by a  $\text{C}-\text{H}\cdots\text{O}$  contact.

## Related literature

For background information, see: Rueping & Ieawsuwan (2009); Rueping *et al.* (2007). For the synthesis of the title compound, see: Rueping & Ieawsuwan (2011). For a comparable compound, see: Liang *et al.* (2003).



## Experimental

## Crystal data

$\text{C}_{15}\text{H}_{15}\text{BrO}_2$   
 $M_r = 307.18$   
 Orthorhombic,  $P2_12_12_1$   
 $a = 9.2217(11)$  Å  
 $b = 11.5041(12)$  Å  
 $c = 12.9149(17)$  Å  
 $V = 1370.1(3)$  Å<sup>3</sup>  
 $Z = 4$   
 Mo  $K\alpha$  radiation  
 $\mu = 2.99$  mm<sup>-1</sup>  
 $T = 173$  K  
 $0.21 \times 0.12 \times 0.03$  mm

## Data collection

STOE IPDS II two-circle-diffractometer  
 Absorption correction: multi-scan (MULABS; Spek, 2009; Blessing, 1995)  
 $T_{\min} = 0.572$ ,  $T_{\max} = 0.916$   
 11129 measured reflections  
 2407 independent reflections  
 1849 reflections with  $I > 2\sigma(I)$   
 $R_{\text{int}} = 0.078$

## Refinement

$R[F^2 > 2\sigma(F^2)] = 0.082$   
 $wR(F^2) = 0.217$   
 $S = 1.03$   
 2407 reflections  
 163 parameters  
 H-atom parameters constrained  
 $\Delta\rho_{\text{max}} = 1.07$  e Å<sup>-3</sup>  
 $\Delta\rho_{\text{min}} = -1.13$  e Å<sup>-3</sup>  
 Absolute structure: Flack (1983), 1009 Friedel pairs  
 Flack parameter: 0.02 (3)

Table 1

Hydrogen-bond geometry (Å, °).

| $D-\text{H}\cdots A$                             | $D-\text{H}$ | $\text{H}\cdots A$ | $D\cdots A$ | $D-\text{H}\cdots A$ |
|--|--------------|--------------------|-------------|----------------------|
| $\text{C}1-\text{H}1\cdots\text{O}31^{\text{i}}$ | 1.00         | 2.47               | 3.282 (9)   | 138                  |

Symmetry code: (i)  $-x + 1, y - \frac{1}{2}, -z + \frac{1}{2}$ .

Data collection: *X-Area* (Stoe & Cie, 2001); cell refinement: *X-Area*; data reduction: *X-Area*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 2008); program(s) used to refine structure: *SHELXL97* (Sheldrick, 2008); molecular graphics: *XP* (Sheldrick, 2008); software used to prepare material for publication: *SHELXL97*.

Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: DS2143).

## References

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## supporting information

*Acta Cryst.* (2011). E67, o2748 [https://doi.org/10.1107/S1600536811038232]

**(5*S*,6*R*)-6-Bromo-6-methyl-5-phenyl-3,4,5,6-tetrahydro-2*H*-cyclopenta[*b*]pyran-7-one****Winai Ieawsuwan and Michael Bolte****S1. Comment**

Trans-4,5-substituted 5-bromocyclopentenone derivatives have been prepared by a organocatalyzed cascade protocol (Rueping & Ieawsuwan, 2009; Rueping *et al.*, 2007). The Brønsted acid catalyzed domino electrocyclization-halogenation reaction provides for the first time, a variety of  $\alpha$ -brominated cyclopent-2-enones with a wide substrate scope and with excellent enantioselectivities (Rueping & Ieawsuwan, 2011). Two chiral centers, a tertiary and a quaternary one, can be established during this transformation. The title compound was synthesized for the first time following this reaction and yellow needles suitable for crystal structure determination were obtained.

The five membered ring in the title compound is essentially planar (r.m.s. deviation 0.006 Å) and forms a dihedral angle of 72.7 (3)° with the attached phenyl ring. The six-membered heterocycle adopts a half chair conformation.

A comparable structure, *cis*-6-Methyl-5-phenyl-3,4,5,6-tetrahydro-2*H*-cyclopenta(*b*)pyran-7-one, with an H atom instead of a bromine residue (Liang *et al.*, 2003) has essentially the same conformation (r.m.s. deviation for all C and O atoms 0.183 Å) (Fig. 2).

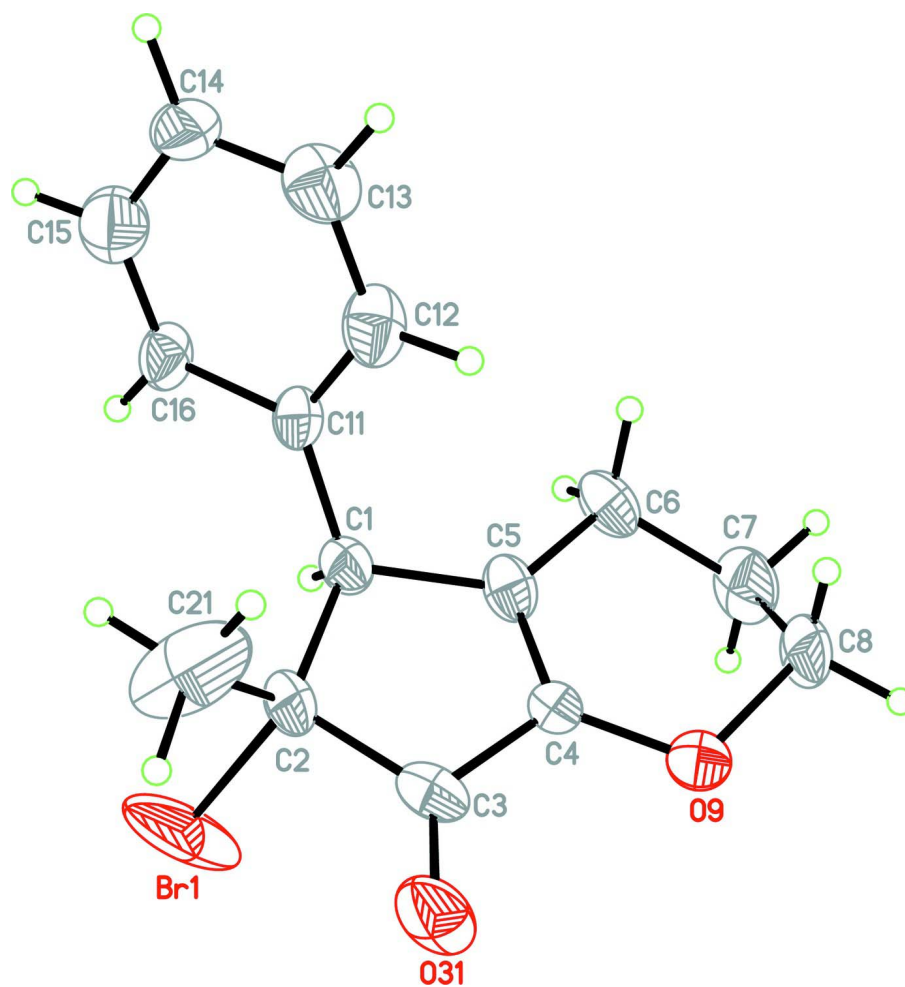
The crystal packing is stabilized by a C—H $\cdots$ O contact (Table 2).

**S2. Experimental**

The title compound has been synthesized as described by Rueping & Ieawsuwan (2011).

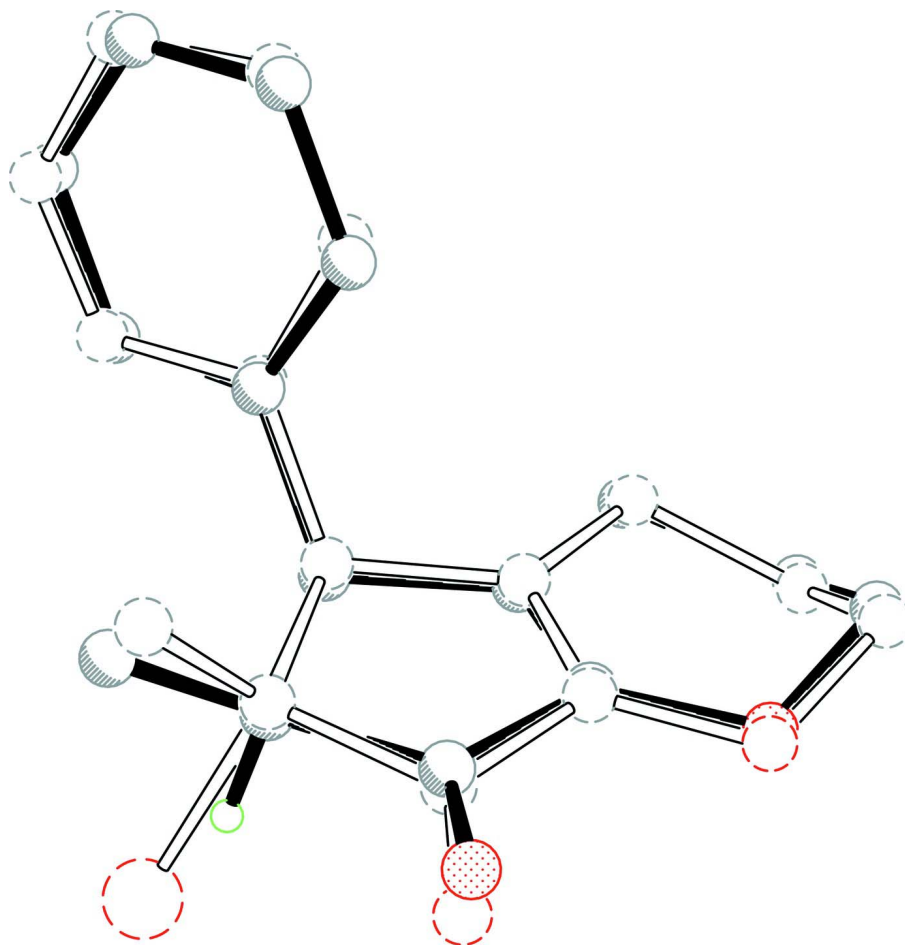
**S3. Refinement**

All H atoms could be located by difference Fourier synthesis. They were refined with fixed individual displacement parameters [ $U(\text{H}) = 1.2 U_{\text{eq}}(\text{C})$  or  $U(\text{H}) = 1.5 U_{\text{eq}}(\text{C}_{\text{methyl}})$ ] using a riding model with C—H ranging from 0.95 Å to 1.00 Å.



**Figure 1**

Perspective view of the title compound with the atom numbering scheme and displacement ellipsoids at the 50% probability level.



**Figure 2**

Least-squares fit of the title compound (open bonds) with *cis*-6-Methyl-5-phenyl-3,4,5,6-tetrahydro-2*H*-cyclopenta(*b*)pyran-7-one (full bonds).

**(5*S*,6*R*)-6-Bromo-6-methyl-5-phenyl-3,4,5,6-tetrahydro-2*H*-cyclopenta[*b*]pyran-7-one**

*Crystal data*

$C_{15}H_{15}BrO_2$

$M_r = 307.18$

Orthorhombic,  $P2_12_12_1$

Hall symbol: P 2ac 2ab

$a = 9.2217 (11) \text{ \AA}$

$b = 11.5041 (12) \text{ \AA}$

$c = 12.9149 (17) \text{ \AA}$

$V = 1370.1 (3) \text{ \AA}^3$

$Z = 4$

$F(000) = 624$

$D_x = 1.489 \text{ Mg m}^{-3}$

Mo  $K\alpha$  radiation,  $\lambda = 0.71073 \text{ \AA}$

Cell parameters from 7315 reflections

$\theta = 3.6\text{--}25.5^\circ$

$\mu = 2.99 \text{ mm}^{-1}$

$T = 173 \text{ K}$

Needle, colourless

$0.21 \times 0.12 \times 0.03 \text{ mm}$

*Data collection*

STOE IPDS II two-circle-diffractometer

Radiation source: fine-focus sealed tube

Graphite monochromator

$\omega$  scans

Absorption correction: multi-scan

(*MULABS*; Spek, 2009; Blessing, 1995)

$T_{\min} = 0.572$ ,  $T_{\max} = 0.916$

11129 measured reflections

2407 independent reflections

1849 reflections with  $I > 2\sigma(I)$   
 $R_{\text{int}} = 0.078$   
 $\theta_{\text{max}} = 25.0^\circ$ ,  $\theta_{\text{min}} = 3.5^\circ$

$h = -10 \rightarrow 10$   
 $k = -13 \rightarrow 13$   
 $l = -15 \rightarrow 15$

### Refinement

Refinement on  $F^2$   
 Least-squares matrix: full  
 $R[F^2 > 2\sigma(F^2)] = 0.082$   
 $wR(F^2) = 0.217$   
 $S = 1.03$   
 2407 reflections  
 163 parameters  
 0 restraints  
 Primary atom site location: structure-invariant  
 direct methods  
 Secondary atom site location: difference Fourier  
 map

Hydrogen site location: inferred from  
 neighbouring sites  
 H-atom parameters constrained  
 $w = 1/[\sigma^2(F_o^2) + (0.1273P)^2 + 1.2132P]$   
 where  $P = (F_o^2 + 2F_c^2)/3$   
 $(\Delta/\sigma)_{\text{max}} < 0.001$   
 $\Delta\rho_{\text{max}} = 1.07 \text{ e } \text{\AA}^{-3}$   
 $\Delta\rho_{\text{min}} = -1.13 \text{ e } \text{\AA}^{-3}$   
 Absolute structure: Flack (1983), 1009 Friedel  
 pairs  
 Absolute structure parameter: 0.02 (3)

### Special details

#### Experimental. ;

**Geometry.** All e.s.d.'s (except the e.s.d. in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell e.s.d.'s are taken into account individually in the estimation of e.s.d.'s in distances, angles and torsion angles; correlations between e.s.d.'s in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell e.s.d.'s is used for estimating e.s.d.'s involving l.s. planes.

**Refinement.** Refinement of  $F^2$  against ALL reflections. The weighted  $R$ -factor  $wR$  and goodness of fit  $S$  are based on  $F^2$ , conventional  $R$ -factors  $R$  are based on  $F$ , with  $F$  set to zero for negative  $F^2$ . The threshold expression of  $F^2 > \sigma(F^2)$  is used only for calculating  $R$ -factors(gt) *etc.* and is not relevant to the choice of reflections for refinement.  $R$ -factors based on  $F^2$  are statistically about twice as large as those based on  $F$ , and  $R$ -factors based on ALL data will be even larger.

### Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters ( $\text{\AA}^2$ )

|     | $x$         | $y$          | $z$          | $U_{\text{iso}}^*/U_{\text{eq}}$ |
|-----|-------------|--------------|--------------|----------------------------------|
| Br1 | 0.4532 (2)  | 0.30553 (14) | 0.06794 (12) | 0.1312 (10)                      |
| C1  | 0.6283 (9)  | 0.3284 (6)   | 0.2534 (5)   | 0.0310 (16)                      |
| H1  | 0.5988      | 0.2457       | 0.2422       | 0.037*                           |
| C2  | 0.5646 (11) | 0.4039 (7)   | 0.1618 (5)   | 0.0404 (19)                      |
| C3  | 0.4546 (10) | 0.4875 (6)   | 0.2110 (6)   | 0.0343 (17)                      |
| C4  | 0.4544 (9)  | 0.4612 (5)   | 0.3206 (5)   | 0.0278 (15)                      |
| C5  | 0.5450 (10) | 0.3752 (6)   | 0.3462 (5)   | 0.0307 (17)                      |
| C6  | 0.5505 (10) | 0.3268 (6)   | 0.4533 (5)   | 0.0369 (18)                      |
| H6A | 0.6408      | 0.3521       | 0.4880       | 0.044*                           |
| H6B | 0.5501      | 0.2408       | 0.4505       | 0.044*                           |
| C7  | 0.4190 (11) | 0.3697 (7)   | 0.5144 (6)   | 0.046 (2)                        |
| H7A | 0.3318      | 0.3257       | 0.4929       | 0.055*                           |
| H7B | 0.4348      | 0.3557       | 0.5892       | 0.055*                           |
| C8  | 0.3945 (10) | 0.4988 (7)   | 0.4959 (6)   | 0.040 (2)                        |
| H8A | 0.3121      | 0.5260       | 0.5386       | 0.048*                           |
| H8B | 0.4819      | 0.5427       | 0.5170       | 0.048*                           |
| O9  | 0.3636 (7)  | 0.5214 (5)   | 0.3851 (4)   | 0.0397 (14)                      |
| C11 | 0.7886 (9)  | 0.3324 (6)   | 0.2644 (5)   | 0.0289 (16)                      |
| C12 | 0.8645 (11) | 0.4156 (7)   | 0.3217 (7)   | 0.040 (2)                        |

|      |             |             |             |             |
|------|-------------|-------------|-------------|-------------|
| H12  | 0.8109      | 0.4724      | 0.3589      | 0.048*      |
| C13  | 1.0115 (11) | 0.4188 (7)  | 0.3266 (7)  | 0.045 (2)   |
| H13  | 1.0581      | 0.4778      | 0.3659      | 0.054*      |
| C14  | 1.0948 (10) | 0.3360 (8)  | 0.2742 (7)  | 0.045 (2)   |
| H14  | 1.1977      | 0.3366      | 0.2781      | 0.054*      |
| C15  | 1.0224 (12) | 0.2534 (7)  | 0.2169 (7)  | 0.047 (2)   |
| H15  | 1.0777      | 0.1984      | 0.1788      | 0.057*      |
| C16  | 0.8782 (10) | 0.2473 (6)  | 0.2126 (7)  | 0.0346 (18) |
| H16  | 0.8336      | 0.1860      | 0.1748      | 0.042*      |
| C21  | 0.6732 (16) | 0.4727 (19) | 0.0990 (12) | 0.137 (9)   |
| H21A | 0.7434      | 0.4194      | 0.0675      | 0.206*      |
| H21B | 0.7241      | 0.5274      | 0.1443      | 0.206*      |
| H21C | 0.6225      | 0.5157      | 0.0444      | 0.206*      |
| O31  | 0.3838 (9)  | 0.5587 (5)  | 0.1641 (5)  | 0.0542 (18) |

*Atomic displacement parameters (Å<sup>2</sup>)*

|     | $U^{11}$    | $U^{22}$    | $U^{33}$    | $U^{12}$    | $U^{13}$     | $U^{23}$    |
|-----|-------------|-------------|-------------|-------------|--------------|-------------|
| Br1 | 0.1929 (19) | 0.1099 (10) | 0.0906 (10) | 0.1023 (12) | -0.1072 (11) | -0.0738 (9) |
| C1  | 0.035 (4)   | 0.035 (4)   | 0.022 (4)   | 0.005 (3)   | -0.001 (3)   | 0.001 (3)   |
| C2  | 0.044 (5)   | 0.060 (5)   | 0.018 (3)   | 0.011 (4)   | 0.003 (4)    | 0.004 (3)   |
| C3  | 0.038 (5)   | 0.037 (4)   | 0.028 (4)   | -0.006 (3)  | -0.009 (4)   | 0.008 (3)   |
| C4  | 0.026 (4)   | 0.033 (3)   | 0.024 (3)   | 0.002 (3)   | -0.003 (3)   | -0.003 (3)  |
| C5  | 0.039 (5)   | 0.030 (3)   | 0.023 (3)   | -0.007 (3)  | 0.004 (3)    | 0.001 (3)   |
| C6  | 0.050 (5)   | 0.040 (4)   | 0.021 (3)   | -0.005 (4)  | -0.003 (4)   | 0.007 (3)   |
| C7  | 0.052 (6)   | 0.058 (5)   | 0.028 (4)   | -0.006 (4)  | 0.011 (4)    | 0.009 (3)   |
| C8  | 0.049 (5)   | 0.053 (4)   | 0.019 (4)   | 0.008 (4)   | 0.010 (4)    | 0.002 (3)   |
| O9  | 0.042 (3)   | 0.046 (3)   | 0.031 (3)   | 0.014 (3)   | -0.005 (3)   | 0.001 (2)   |
| C11 | 0.031 (4)   | 0.033 (4)   | 0.023 (3)   | -0.001 (3)  | 0.006 (3)    | 0.001 (3)   |
| C12 | 0.047 (6)   | 0.031 (4)   | 0.042 (5)   | -0.008 (4)  | 0.009 (4)    | -0.010 (4)  |
| C13 | 0.047 (6)   | 0.042 (4)   | 0.046 (5)   | -0.019 (4)  | -0.001 (4)   | -0.005 (4)  |
| C14 | 0.027 (5)   | 0.071 (6)   | 0.038 (4)   | -0.003 (4)  | 0.001 (3)    | 0.003 (4)   |
| C15 | 0.051 (7)   | 0.048 (4)   | 0.043 (5)   | 0.004 (4)   | 0.003 (4)    | -0.001 (4)  |
| C16 | 0.034 (5)   | 0.034 (4)   | 0.036 (4)   | -0.003 (3)  | 0.005 (4)    | -0.016 (3)  |
| C21 | 0.077 (10)  | 0.24 (2)    | 0.090 (10)  | 0.074 (12)  | 0.048 (8)    | 0.127 (13)  |
| O31 | 0.088 (5)   | 0.038 (3)   | 0.036 (3)   | 0.023 (3)   | -0.015 (3)   | 0.003 (2)   |

*Geometric parameters (Å, °)*

|        |            |         |            |
|--------|------------|---------|------------|
| Br1—C2 | 1.951 (9)  | C8—O9   | 1.481 (9)  |
| C1—C11 | 1.486 (11) | C8—H8A  | 0.9900     |
| C1—C5  | 1.522 (10) | C8—H8B  | 0.9900     |
| C1—C2  | 1.582 (10) | C11—C12 | 1.398 (11) |
| C1—H1  | 1.0000     | C11—C16 | 1.445 (10) |
| C2—C21 | 1.512 (17) | C12—C13 | 1.357 (14) |
| C2—C3  | 1.535 (12) | C12—H12 | 0.9500     |
| C3—O31 | 1.211 (10) | C13—C14 | 1.398 (13) |
| C3—C4  | 1.447 (11) | C13—H13 | 0.9500     |

|               |            |               |            |
|---------------|------------|---------------|------------|
| C4—C5         | 1.337 (11) | C14—C15       | 1.377 (13) |
| C4—O9         | 1.369 (9)  | C14—H14       | 0.9500     |
| C5—C6         | 1.491 (9)  | C15—C16       | 1.333 (13) |
| C6—C7         | 1.528 (13) | C15—H15       | 0.9500     |
| C6—H6A        | 0.9900     | C16—H16       | 0.9500     |
| C6—H6B        | 0.9900     | C21—H21A      | 0.9800     |
| C7—C8         | 1.521 (12) | C21—H21B      | 0.9800     |
| C7—H7A        | 0.9900     | C21—H21C      | 0.9800     |
| C7—H7B        | 0.9900     |               |            |
|               |            |               |            |
| C11—C1—C5     | 114.6 (6)  | H7A—C7—H7B    | 108.1      |
| C11—C1—C2     | 115.1 (7)  | O9—C8—C7      | 110.6 (7)  |
| C5—C1—C2      | 102.0 (6)  | O9—C8—H8A     | 109.5      |
| C11—C1—H1     | 108.3      | C7—C8—H8A     | 109.5      |
| C5—C1—H1      | 108.3      | O9—C8—H8B     | 109.5      |
| C2—C1—H1      | 108.3      | C7—C8—H8B     | 109.5      |
| C21—C2—C3     | 109.4 (9)  | H8A—C8—H8B    | 108.1      |
| C21—C2—C1     | 116.3 (8)  | C4—O9—C8      | 112.4 (6)  |
| C3—C2—C1      | 106.2 (6)  | C12—C11—C16   | 115.0 (8)  |
| C21—C2—Br1    | 108.6 (10) | C12—C11—C1    | 124.7 (7)  |
| C3—C2—Br1     | 105.9 (6)  | C16—C11—C1    | 120.2 (7)  |
| C1—C2—Br1     | 110.0 (5)  | C13—C12—C11   | 122.9 (8)  |
| O31—C3—C4     | 129.1 (8)  | C13—C12—H12   | 118.5      |
| O31—C3—C2     | 124.9 (7)  | C11—C12—H12   | 118.5      |
| C4—C3—C2      | 106.0 (6)  | C12—C13—C14   | 120.5 (8)  |
| C5—C4—O9      | 127.3 (7)  | C12—C13—H13   | 119.7      |
| C5—C4—C3      | 113.3 (6)  | C14—C13—H13   | 119.7      |
| O9—C4—C3      | 119.4 (6)  | C15—C14—C13   | 117.6 (9)  |
| C4—C5—C6      | 121.8 (7)  | C15—C14—H14   | 121.2      |
| C4—C5—C1      | 112.4 (6)  | C13—C14—H14   | 121.2      |
| C6—C5—C1      | 125.5 (7)  | C16—C15—C14   | 122.9 (9)  |
| C5—C6—C7      | 109.3 (7)  | C16—C15—H15   | 118.6      |
| C5—C6—H6A     | 109.8      | C14—C15—H15   | 118.6      |
| C7—C6—H6A     | 109.8      | C15—C16—C11   | 121.0 (8)  |
| C5—C6—H6B     | 109.8      | C15—C16—H16   | 119.5      |
| C7—C6—H6B     | 109.8      | C11—C16—H16   | 119.5      |
| H6A—C6—H6B    | 108.3      | C2—C21—H21A   | 109.5      |
| C8—C7—C6      | 110.6 (7)  | C2—C21—H21B   | 109.5      |
| C8—C7—H7A     | 109.5      | H21A—C21—H21B | 109.5      |
| C6—C7—H7A     | 109.5      | C2—C21—H21C   | 109.5      |
| C8—C7—H7B     | 109.5      | H21A—C21—H21C | 109.5      |
| C6—C7—H7B     | 109.5      | H21B—C21—H21C | 109.5      |
|               |            |               |            |
| C11—C1—C2—C21 | -1.5 (13)  | C11—C1—C5—C6  | -61.6 (10) |
| C5—C1—C2—C21  | 123.2 (12) | C2—C1—C5—C6   | 173.4 (7)  |
| C11—C1—C2—C3  | -123.5 (7) | C4—C5—C6—C7   | 12.9 (10)  |
| C5—C1—C2—C3   | 1.2 (8)    | C1—C5—C6—C7   | -161.6 (8) |
| C11—C1—C2—Br1 | 122.4 (6)  | C5—C6—C7—C8   | -44.3 (9)  |

|               |             |                 |            |
|---------------|-------------|-----------------|------------|
| C5—C1—C2—Br1  | -112.9 (6)  | C6—C7—C8—O9     | 61.9 (10)  |
| C21—C2—C3—O31 | 53.9 (13)   | C5—C4—O9—C8     | 11.8 (11)  |
| C1—C2—C3—O31  | -179.9 (8)  | C3—C4—O9—C8     | -170.1 (7) |
| Br1—C2—C3—O31 | -63.0 (9)   | C7—C8—O9—C4     | -44.0 (10) |
| C21—C2—C3—C4  | -126.8 (10) | C5—C1—C11—C12   | -30.5 (10) |
| C1—C2—C3—C4   | -0.6 (9)    | C2—C1—C11—C12   | 87.3 (9)   |
| Br1—C2—C3—C4  | 116.3 (6)   | C5—C1—C11—C16   | 150.4 (7)  |
| O31—C3—C4—C5  | 178.8 (8)   | C2—C1—C11—C16   | -91.8 (8)  |
| C2—C3—C4—C5   | -0.4 (9)    | C16—C11—C12—C13 | 1.7 (12)   |
| O31—C3—C4—O9  | 0.4 (13)    | C1—C11—C12—C13  | -177.4 (8) |
| C2—C3—C4—O9   | -178.9 (6)  | C11—C12—C13—C14 | -0.9 (14)  |
| O9—C4—C5—C6   | 4.4 (12)    | C12—C13—C14—C15 | 1.1 (13)   |
| C3—C4—C5—C6   | -173.9 (7)  | C13—C14—C15—C16 | -2.4 (14)  |
| O9—C4—C5—C1   | 179.6 (7)   | C14—C15—C16—C11 | 3.4 (14)   |
| C3—C4—C5—C1   | 1.3 (9)     | C12—C11—C16—C15 | -2.9 (12)  |
| C11—C1—C5—C4  | 123.5 (7)   | C1—C11—C16—C15  | 176.3 (8)  |
| C2—C1—C5—C4   | -1.6 (9)    |                 |            |

*Hydrogen-bond geometry (Å, °)*

| <i>D</i> —H... <i>A</i>  | <i>D</i> —H | H... <i>A</i> | <i>D</i> ... <i>A</i> | <i>D</i> —H... <i>A</i> |
|--------------------------|-------------|---------------|-----------------------|-------------------------|
| C1—H1...O31 <sup>i</sup> | 1.00        | 2.47          | 3.282 (9)             | 138                     |

Symmetry code: (i)  $-x+1, y-1/2, -z+1/2$ .