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2-Benzyl-6-chloro-1-(4-methylphenyl)-1H-indole-3-carbonitrile

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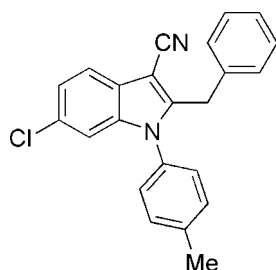
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Key indicators: single-crystal X-ray study; $T = 113$ K; mean $\sigma(\text{C}-\text{C}) = 0.003$ Å; R factor = 0.041; wR factor = 0.103; data-to-parameter ratio = 18.5.

In the title compound, $\text{C}_{23}\text{H}_{17}\text{ClN}_2$, the dihedral angle between the indole ring and the attached tolyl ring is 86.97 (8)°. Weak $\text{C}-\text{H}\cdots\text{N}$ (nitrile) hydrogen bonding, and $\text{C}-\text{H}\cdots\pi$ (aromatic) and short $\text{Cl}\cdots\pi$ (aromatic) [3.628 (1) Å] interactions consolidate the crystal packing.

Related literature

For the synthesis of the title compound, see: Du *et al.* (2006). For its precursor, see: Jin *et al.* (2009). For related structures, see: Yang *et al.* (2011); Yan & Qi (2011a,b,c). For standard bond lengths, see: Allen *et al.* (1987).



Experimental

Crystal data

$\text{C}_{23}\text{H}_{17}\text{ClN}_2$
 $M_r = 356.84$
Monoclinic, $P2_1/c$
 $a = 10.0003$ (10) Å

$b = 9.8565$ (8) Å
 $c = 18.539$ (2) Å
 $\beta = 93.926$ (9)°
 $V = 1823.1$ (3) Å³

$Z = 4$
Mo $K\alpha$ radiation
 $\mu = 0.22$ mm⁻¹

$T = 113$ K
 $0.20 \times 0.16 \times 0.10$ mm

Data collection

Rigaku Saturn724 CCD diffractometer
Absorption correction: multi-scan (*CrystalClear-SM Expert*; Rigaku, 2009)
 $T_{\min} = 0.958$, $T_{\max} = 0.979$

18792 measured reflections
4382 independent reflections
1947 reflections with $I > 2\sigma(I)$
 $R_{\text{int}} = 0.079$

Refinement

$R[F^2 > 2\sigma(F^2)] = 0.041$
 $wR(F^2) = 0.103$
 $S = 0.85$
4382 reflections

237 parameters
H-atom parameters constrained
 $\Delta\rho_{\text{max}} = 0.35$ e Å⁻³
 $\Delta\rho_{\text{min}} = -0.30$ e Å⁻³

Table 1

Hydrogen-bond geometry (Å, °).

C_g is the centroid of the C17–C22 ring.

$D-H\cdots A$	$D-H$	$H\cdots A$	$D\cdots A$	$D-H\cdots A$
$\text{C9}-\text{H9A}\cdots\text{N2}^i$	0.99	2.64	3.557 (3)	154
$\text{C14}-\text{H14}\cdots\text{Cg}^{ii}$	0.95	2.84	3.735 (2)	157

Symmetry codes: (i) $-x + 1, -y + 1, -z$; (ii) $-x, y - \frac{1}{2}, -z + \frac{1}{2}$.

Data collection: *CrystalClear-SM Expert* (Rigaku, 2009); cell refinement: *CrystalClear-SM Expert*; data reduction: *CrystalClear-SM Expert*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 2008); program(s) used to refine structure: *SHELXL97* (Sheldrick, 2008); molecular graphics: *SHELXTL* (Sheldrick, 2008); software used to prepare material for publication: *SHELXL97*.

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Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: ZL2420).

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supporting information

Acta Cryst. (2011). E67, o3388 [https://doi.org/10.1107/S1600536811048422]

2-Benzyl-6-chloro-1-(4-methylphenyl)-1*H*-indole-3-carbonitrile

Qiao Yan and Xiuxiang Qi

S1. Comment

Indoles are widely applied in many fields such as pharmaceuticals. In order to disclose their structure-activity relationship, we have studied the crystal structures of a series of indole derivatives (Yan & Qi, 2011a,b,c). In our continuous investigation of indole derivatives, we herein report the structure of the title compound (Fig. 1). The indole ring forms an angle of 86.97 (8)° with the tolyl ring connected to the N1 atom. This is similar as in ethyl 2-benzyl-1-propyl-1*H*-indole-3-carboxylate [80.91 (5)°] reported by Yang *et al.* (2011), but greater than in 1-(4-methoxyphenyl)-2-methyl-1*H*-indole-3-carbonitrile [58.41 (4)°] (Yan & Qi, 2011a) and 1-(4-bromophenyl)-2-methyl-1*H*-indole-3-carbonitrile [58.85 (11)°] (Yan & Qi, 2011b) reported by our group. All bond distances and angles of the title compound are within the normal ranges (Allen *et al.*, 1987).

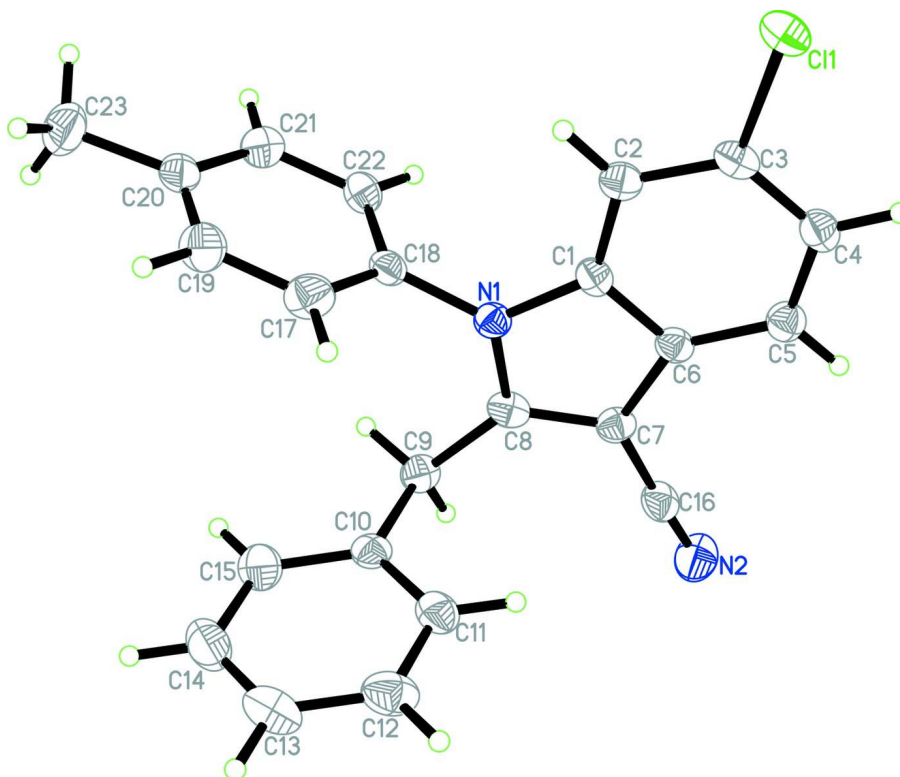
In the crystal packing, a weak intermolecular hydrogen-bond interaction occurs, affording a dimer with the nitrile N atom being a hydrogen-bond acceptor and the methylene being a hydrogen-bond donor (Table 1 & Fig. 2). Significant C—H⋯ π interaction exists between H14 and the aromatic C17—C22 ring (*Cg*) (Table 1). A short Cl⋯ π interaction is detected between Cl1 and the other face of the same tolyl ring. These C—H⋯ π and Cl⋯ π interactions link molecules into columns along the *b* axis (Fig. 3).

S2. Experimental

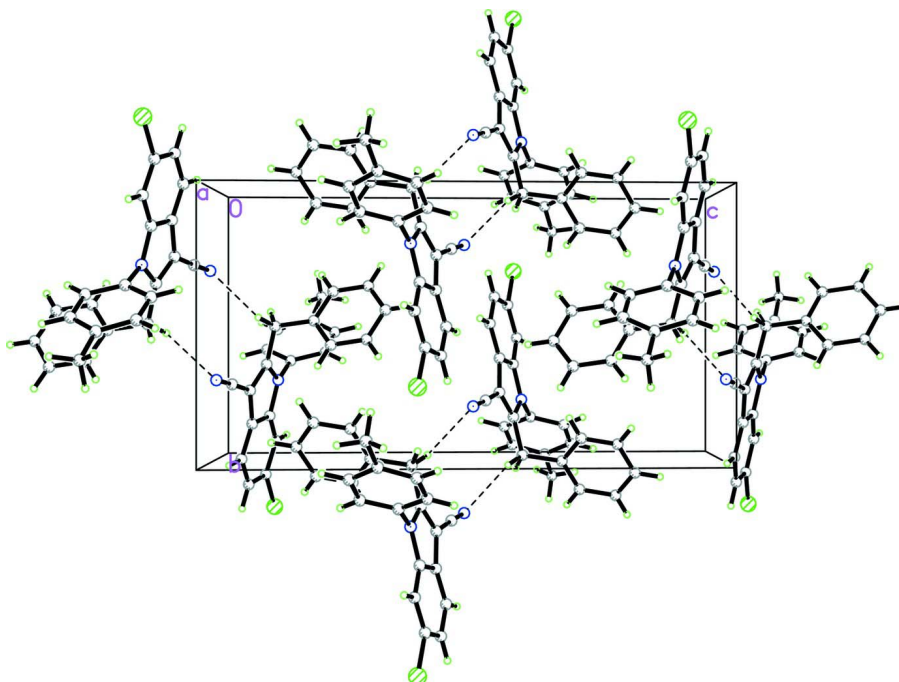
The title compound was prepared according to the method of the literature (Du *et al.*, 2006). Colourless prisms of the title compound were grown by evaporation from a mixture of ethyl acetate and petroleum ether.

S3. Refinement

All H atoms were positioned geometrically (C—H = 0.95, 0.98 and 0.99 Å) and refined as riding with $U_{\text{iso}}(\text{H}) = 1.2U_{\text{eq}}(\text{CH \& CH}_2)$ or $1.5U_{\text{eq}}(\text{CH}_3)$.

**Figure 1**

View of the title compound showing displacement ellipsoids at the 50% probability level.

**Figure 2**

Molecular packing of the title compound, viewed down the *a* axis. The C—H...N H-bonding is represented by dashed lines.

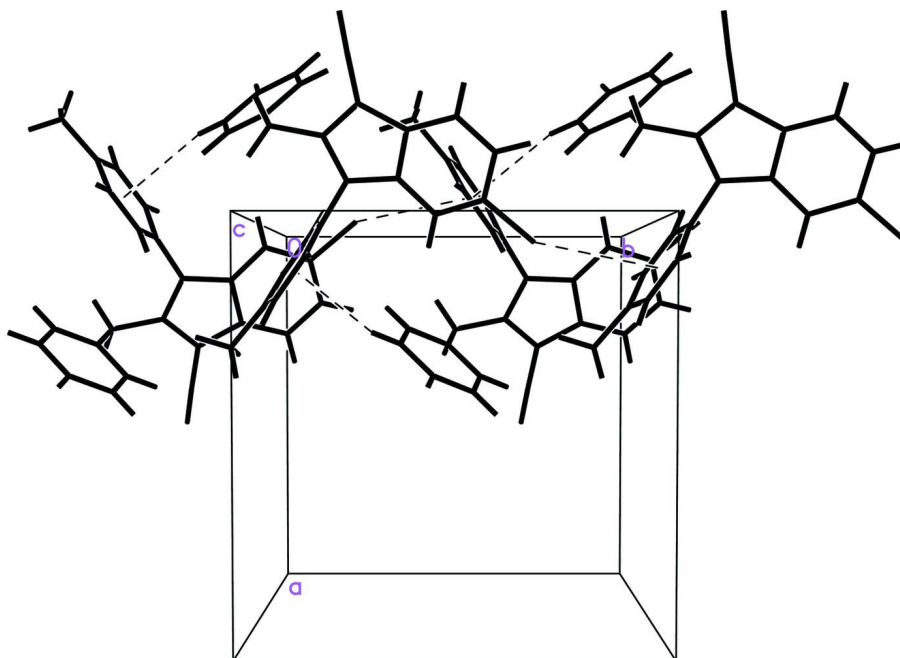


Figure 3

Molecular packing of the title compound, viewed down the *c* axis. The C—H \cdots π and Cl \cdots π interactions are represented by dashed lines.

2-Benzyl-6-chloro-1-(4-methylphenyl)-1*H*-indole-3-carbonitrile

Crystal data

C₂₃H₁₇ClN₂

M_r = 356.84

Monoclinic, *P*2₁/*c*

Hall symbol: -*P* 2ybc

a = 10.0003 (10) Å

b = 9.8565 (8) Å

c = 18.539 (2) Å

β = 93.926 (9)°

V = 1823.1 (3) Å³

Z = 4

F(000) = 744

D_x = 1.300 Mg m⁻³

Mo *K* α radiation, λ = 0.71073 Å

Cell parameters from 6077 reflections

θ = 2.0–28.0°

μ = 0.22 mm⁻¹

T = 113 K

Prism, colorless

0.20 × 0.16 × 0.10 mm

Data collection

Rigaku Saturn724 CCD
diffractometer

Radiation source: rotating anode

Multilayer monochromator

Detector resolution: 14.22 pixels mm⁻¹

ω and ϕ scans

Absorption correction: multi-scan

(*CrystalClear-SM Expert*; Rigaku, 2009)

T_{min} = 0.958, *T_{max}* = 0.979

18792 measured reflections

4382 independent reflections

1947 reflections with *I* > 2 σ (*I*)

R_{int} = 0.079

θ_{\max} = 28.0°, θ_{\min} = 2.0°

h = -12→13

k = -12→12

l = -24→24

Refinement

Refinement on F^2

Least-squares matrix: full

$R[F^2 > 2\sigma(F^2)] = 0.041$

$wR(F^2) = 0.103$

$S = 0.85$

4382 reflections

237 parameters

0 restraints

Primary atom site location: structure-invariant
direct methods

Secondary atom site location: difference Fourier
map

Hydrogen site location: inferred from
neighbouring sites

H-atom parameters constrained

$w = 1/[\sigma^2(F_o^2) + (0.0361P)^2]$

where $P = (F_o^2 + 2F_c^2)/3$

$(\Delta/\sigma)_{\max} < 0.001$

$\Delta\rho_{\max} = 0.35 \text{ e } \text{\AA}^{-3}$

$\Delta\rho_{\min} = -0.30 \text{ e } \text{\AA}^{-3}$

Extinction correction: *SHELXL97* (Sheldrick,
2008), $F_c^* = kF_c[1 + 0.001xF_c^2\lambda^3/\sin(2\theta)]^{-1/4}$

Extinction coefficient: 0.0033 (5)

Special details

Geometry. All e.s.d.'s (except the e.s.d. in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell e.s.d.'s are taken into account individually in the estimation of e.s.d.'s in distances, angles and torsion angles; correlations between e.s.d.'s in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell e.s.d.'s is used for estimating e.s.d.'s involving l.s. planes.

Refinement. Refinement of F^2 against ALL reflections. The weighted R -factor wR and goodness of fit S are based on F^2 , conventional R -factors R are based on F , with F set to zero for negative F^2 . The threshold expression of $F^2 > \sigma(F^2)$ is used only for calculating R -factors(gt) *etc.* and is not relevant to the choice of reflections for refinement. R -factors based on F^2 are statistically about twice as large as those based on F , and R -factors based on ALL data will be even larger.

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (\AA^2)

	x	y	z	$U_{\text{iso}}^*/U_{\text{eq}}$
Cl1	-0.03716 (5)	1.21324 (5)	0.09402 (3)	0.03050 (16)
N1	0.12552 (15)	0.71133 (16)	0.10904 (7)	0.0212 (4)
N2	0.54914 (18)	0.70088 (18)	0.00901 (9)	0.0377 (5)
C1	0.13022 (19)	0.84896 (19)	0.09184 (9)	0.0212 (4)
C2	0.0363 (2)	0.94986 (19)	0.10034 (9)	0.0231 (5)
H2	-0.0485	0.9312	0.1184	0.028*
C3	0.0728 (2)	1.0788 (2)	0.08114 (9)	0.0238 (5)
C4	0.19515 (19)	1.1085 (2)	0.05278 (9)	0.0260 (5)
H4	0.2159	1.1992	0.0403	0.031*
C5	0.2855 (2)	1.0062 (2)	0.04304 (10)	0.0251 (5)
H5	0.3684	1.0253	0.0230	0.030*
C6	0.25444 (19)	0.8741 (2)	0.06279 (9)	0.0216 (4)
C7	0.32207 (19)	0.7456 (2)	0.06210 (9)	0.0230 (5)
C8	0.24230 (19)	0.6490 (2)	0.09126 (9)	0.0236 (5)
C9	0.27113 (19)	0.50407 (19)	0.10918 (10)	0.0252 (5)
H9A	0.3431	0.4712	0.0795	0.030*
H9B	0.1899	0.4493	0.0966	0.030*
C10	0.31403 (19)	0.4831 (2)	0.18884 (10)	0.0235 (5)
C11	0.4009 (2)	0.5740 (2)	0.22559 (11)	0.0303 (5)
H11	0.4327	0.6509	0.2010	0.036*
C12	0.4411 (2)	0.5531 (2)	0.29759 (11)	0.0371 (6)
H12	0.5007	0.6154	0.3221	0.044*
C13	0.3951 (2)	0.4425 (2)	0.33395 (11)	0.0387 (6)

H13	0.4223	0.4290	0.3835	0.046*
C14	0.3092 (2)	0.3513 (2)	0.29807 (11)	0.0397 (6)
H14	0.2778	0.2746	0.3229	0.048*
C15	0.2687 (2)	0.3716 (2)	0.22582 (11)	0.0344 (5)
H15	0.2095	0.3087	0.2015	0.041*
C16	0.4488 (2)	0.7192 (2)	0.03379 (10)	0.0268 (5)
C17	0.00947 (19)	0.6386 (2)	0.12952 (9)	0.0220 (4)
C18	-0.0077 (2)	0.6062 (2)	0.20068 (10)	0.0328 (5)
H18	0.0541	0.6378	0.2381	0.039*
C19	-0.1160 (2)	0.5272 (2)	0.21686 (11)	0.0384 (6)
H19	-0.1279	0.5051	0.2659	0.046*
C20	-0.2076 (2)	0.4795 (2)	0.16341 (11)	0.0290 (5)
C21	-0.1911 (2)	0.5178 (2)	0.09294 (10)	0.0286 (5)
H21	-0.2551	0.4898	0.0557	0.034*
C22	-0.0833 (2)	0.59617 (19)	0.07563 (10)	0.0249 (5)
H22	-0.0730	0.6208	0.0268	0.030*
C23	-0.3186 (2)	0.3845 (2)	0.18130 (12)	0.0472 (7)
H23A	-0.3989	0.4038	0.1496	0.071*
H23B	-0.3389	0.3974	0.2318	0.071*
H23C	-0.2904	0.2906	0.1741	0.071*

Atomic displacement parameters (Å²)

	U^{11}	U^{22}	U^{33}	U^{12}	U^{13}	U^{23}
C11	0.0339 (3)	0.0245 (3)	0.0329 (3)	0.0083 (3)	0.0016 (2)	0.0004 (2)
N1	0.0207 (9)	0.0196 (9)	0.0233 (8)	0.0016 (8)	0.0017 (7)	-0.0009 (7)
N2	0.0329 (11)	0.0306 (11)	0.0507 (12)	0.0012 (10)	0.0121 (9)	-0.0055 (9)
C1	0.0258 (11)	0.0186 (10)	0.0186 (9)	-0.0005 (10)	-0.0025 (8)	-0.0004 (8)
C2	0.0237 (11)	0.0250 (11)	0.0201 (10)	0.0017 (9)	-0.0013 (8)	-0.0017 (9)
C3	0.0273 (12)	0.0218 (11)	0.0218 (10)	0.0062 (10)	-0.0027 (8)	-0.0017 (8)
C4	0.0303 (12)	0.0212 (11)	0.0259 (10)	0.0000 (10)	-0.0011 (9)	0.0008 (9)
C5	0.0251 (12)	0.0255 (12)	0.0248 (10)	-0.0005 (10)	0.0014 (8)	-0.0002 (9)
C6	0.0227 (11)	0.0211 (11)	0.0205 (10)	0.0016 (9)	-0.0016 (8)	-0.0009 (8)
C7	0.0214 (11)	0.0232 (12)	0.0245 (10)	0.0023 (9)	0.0014 (8)	-0.0032 (8)
C8	0.0266 (12)	0.0221 (11)	0.0217 (9)	0.0052 (10)	-0.0012 (8)	-0.0053 (9)
C9	0.0249 (12)	0.0218 (11)	0.0287 (11)	0.0001 (10)	0.0004 (9)	-0.0045 (9)
C10	0.0198 (11)	0.0211 (11)	0.0297 (11)	0.0053 (9)	0.0020 (8)	-0.0014 (9)
C11	0.0317 (13)	0.0227 (12)	0.0360 (12)	0.0037 (10)	-0.0008 (10)	0.0023 (9)
C12	0.0389 (14)	0.0317 (13)	0.0387 (13)	0.0050 (11)	-0.0107 (10)	-0.0059 (11)
C13	0.0474 (15)	0.0377 (14)	0.0303 (12)	0.0129 (12)	-0.0036 (11)	0.0014 (10)
C14	0.0456 (15)	0.0351 (14)	0.0387 (13)	0.0022 (13)	0.0044 (11)	0.0105 (11)
C15	0.0350 (13)	0.0285 (13)	0.0394 (12)	-0.0016 (11)	0.0005 (10)	0.0007 (10)
C16	0.0317 (13)	0.0177 (11)	0.0310 (11)	0.0018 (10)	0.0011 (9)	-0.0025 (9)
C17	0.0230 (11)	0.0197 (11)	0.0234 (10)	0.0043 (9)	0.0023 (8)	0.0028 (8)
C18	0.0313 (13)	0.0446 (14)	0.0222 (11)	0.0006 (11)	-0.0016 (9)	-0.0002 (10)
C19	0.0360 (14)	0.0543 (16)	0.0259 (11)	0.0043 (13)	0.0097 (10)	0.0115 (11)
C20	0.0264 (12)	0.0227 (12)	0.0387 (12)	0.0083 (10)	0.0092 (10)	0.0043 (10)
C21	0.0289 (12)	0.0266 (12)	0.0300 (11)	0.0008 (10)	0.0004 (9)	-0.0045 (9)

C22	0.0304 (12)	0.0239 (11)	0.0207 (10)	0.0009 (10)	0.0032 (8)	0.0006 (9)
C23	0.0372 (15)	0.0389 (15)	0.0681 (17)	0.0018 (13)	0.0230 (12)	0.0092 (13)

Geometric parameters (Å, °)

C11—C3	1.7487 (19)	C11—C12	1.383 (3)
N1—C8	1.379 (2)	C11—H11	0.9500
N1—C1	1.395 (2)	C12—C13	1.377 (3)
N1—C17	1.437 (2)	C12—H12	0.9500
N2—C16	1.146 (2)	C13—C14	1.382 (3)
C1—C2	1.385 (2)	C13—H13	0.9500
C1—C6	1.409 (2)	C14—C15	1.387 (3)
C2—C3	1.375 (2)	C14—H14	0.9500
C2—H2	0.9500	C15—H15	0.9500
C3—C4	1.396 (3)	C17—C18	1.379 (2)
C4—C5	1.374 (2)	C17—C22	1.381 (2)
C4—H4	0.9500	C18—C19	1.384 (3)
C5—C6	1.393 (3)	C18—H18	0.9500
C5—H5	0.9500	C19—C20	1.385 (3)
C6—C7	1.436 (3)	C19—H19	0.9500
C7—C8	1.376 (3)	C20—C21	1.381 (3)
C7—C16	1.428 (3)	C20—C23	1.506 (3)
C8—C9	1.490 (3)	C21—C22	1.382 (3)
C9—C10	1.523 (2)	C21—H21	0.9500
C9—H9A	0.9900	C22—H22	0.9500
C9—H9B	0.9900	C23—H23A	0.9800
C10—C15	1.387 (3)	C23—H23B	0.9800
C10—C11	1.393 (3)	C23—H23C	0.9800
C8—N1—C1	109.63 (16)	C10—C11—H11	119.8
C8—N1—C17	123.60 (16)	C13—C12—C11	120.4 (2)
C1—N1—C17	125.86 (16)	C13—C12—H12	119.8
C2—C1—N1	129.70 (18)	C11—C12—H12	119.8
C2—C1—C6	122.68 (18)	C12—C13—C14	119.8 (2)
N1—C1—C6	107.61 (17)	C12—C13—H13	120.1
C3—C2—C1	116.10 (18)	C14—C13—H13	120.1
C3—C2—H2	121.9	C13—C14—C15	120.1 (2)
C1—C2—H2	121.9	C13—C14—H14	119.9
C2—C3—C4	123.12 (19)	C15—C14—H14	119.9
C2—C3—C11	118.98 (16)	C14—C15—C10	120.5 (2)
C4—C3—C11	117.89 (15)	C14—C15—H15	119.7
C5—C4—C3	119.79 (19)	C10—C15—H15	119.7
C5—C4—H4	120.1	N2—C16—C7	177.6 (2)
C3—C4—H4	120.1	C18—C17—C22	120.01 (19)
C4—C5—C6	119.39 (19)	C18—C17—N1	121.59 (17)
C4—C5—H5	120.3	C22—C17—N1	118.34 (16)
C6—C5—H5	120.3	C17—C18—C19	119.23 (19)
C5—C6—C1	118.87 (18)	C17—C18—H18	120.4

C5—C6—C7	135.07 (18)	C19—C18—H18	120.4
C1—C6—C7	106.05 (17)	C18—C19—C20	121.69 (19)
C8—C7—C16	124.79 (18)	C18—C19—H19	119.2
C8—C7—C6	108.70 (16)	C20—C19—H19	119.2
C16—C7—C6	126.47 (18)	C21—C20—C19	117.9 (2)
C7—C8—N1	107.99 (17)	C21—C20—C23	121.1 (2)
C7—C8—C9	129.96 (18)	C19—C20—C23	120.98 (19)
N1—C8—C9	121.84 (18)	C20—C21—C22	121.22 (19)
C8—C9—C10	112.56 (15)	C20—C21—H21	119.4
C8—C9—H9A	109.1	C22—C21—H21	119.4
C10—C9—H9A	109.1	C17—C22—C21	119.84 (18)
C8—C9—H9B	109.1	C17—C22—H22	120.1
C10—C9—H9B	109.1	C21—C22—H22	120.1
H9A—C9—H9B	107.8	C20—C23—H23A	109.5
C15—C10—C11	118.79 (19)	C20—C23—H23B	109.5
C15—C10—C9	120.45 (18)	H23A—C23—H23B	109.5
C11—C10—C9	120.75 (18)	C20—C23—H23C	109.5
C12—C11—C10	120.42 (19)	H23A—C23—H23C	109.5
C12—C11—H11	119.8	H23B—C23—H23C	109.5
C8—N1—C1—C2	-179.86 (18)	C17—N1—C8—C9	-15.6 (3)
C17—N1—C1—C2	10.8 (3)	C7—C8—C9—C10	97.8 (2)
C8—N1—C1—C6	-0.43 (19)	N1—C8—C9—C10	-76.2 (2)
C17—N1—C1—C6	-169.82 (15)	C8—C9—C10—C15	140.01 (19)
N1—C1—C2—C3	177.05 (17)	C8—C9—C10—C11	-40.9 (2)
C6—C1—C2—C3	-2.3 (3)	C15—C10—C11—C12	0.1 (3)
C1—C2—C3—C4	1.7 (3)	C9—C10—C11—C12	-178.96 (17)
C1—C2—C3—C11	-177.16 (13)	C10—C11—C12—C13	-0.4 (3)
C2—C3—C4—C5	-0.1 (3)	C11—C12—C13—C14	0.6 (3)
C11—C3—C4—C5	178.81 (14)	C12—C13—C14—C15	-0.4 (3)
C3—C4—C5—C6	-1.0 (3)	C13—C14—C15—C10	0.1 (3)
C4—C5—C6—C1	0.5 (3)	C11—C10—C15—C14	0.0 (3)
C4—C5—C6—C7	-178.58 (19)	C9—C10—C15—C14	179.10 (18)
C2—C1—C6—C5	1.3 (3)	C8—N1—C17—C18	89.9 (2)
N1—C1—C6—C5	-178.18 (15)	C1—N1—C17—C18	-102.1 (2)
C2—C1—C6—C7	-179.42 (16)	C8—N1—C17—C22	-87.5 (2)
N1—C1—C6—C7	1.11 (19)	C1—N1—C17—C22	80.5 (2)
C5—C6—C7—C8	177.7 (2)	C22—C17—C18—C19	2.2 (3)
C1—C6—C7—C8	-1.4 (2)	N1—C17—C18—C19	-175.10 (18)
C5—C6—C7—C16	-4.4 (3)	C17—C18—C19—C20	0.1 (3)
C1—C6—C7—C16	176.51 (17)	C18—C19—C20—C21	-2.6 (3)
C16—C7—C8—N1	-176.80 (16)	C18—C19—C20—C23	175.09 (19)
C6—C7—C8—N1	1.2 (2)	C19—C20—C21—C22	2.9 (3)
C16—C7—C8—C9	8.5 (3)	C23—C20—C21—C22	-174.81 (18)
C6—C7—C8—C9	-173.51 (17)	C18—C17—C22—C21	-1.9 (3)
C1—N1—C8—C7	-0.5 (2)	N1—C17—C22—C21	175.46 (17)
C17—N1—C8—C7	169.21 (15)	C20—C21—C22—C17	-0.7 (3)
C1—N1—C8—C9	174.73 (16)		

Hydrogen-bond geometry (Å, °)

Cg is the centroid of the C17–C22 ring.

<i>D</i> —H \cdots <i>A</i>	<i>D</i> —H	H \cdots <i>A</i>	<i>D</i> \cdots <i>A</i>	<i>D</i> —H \cdots <i>A</i>
C9—H9 <i>A</i> \cdots N2 ⁱ	0.99	2.64	3.557 (3)	154
C14—H14 \cdots Cg ⁱⁱ	0.95	2.84	3.735 (2)	157

Symmetry codes: (i) $-x+1, -y+1, -z$; (ii) $-x, y-1/2, -z+1/2$.