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2-(6-Chloro-1H-indol-3-yl)acetonitrile

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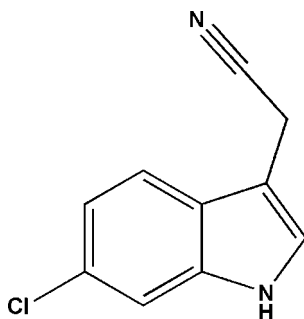
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Key indicators: single-crystal X-ray study; $T = 293$ K; mean $\sigma(\text{C}-\text{C}) = 0.007$ Å; R factor = 0.110; wR factor = 0.312; data-to-parameter ratio = 18.0.

In the title compound, $\text{C}_{10}\text{H}_7\text{ClN}_2$, the carbonitrile group is twisted away from the plane of the indole ring system [$\text{C}_{\text{cy}}-\text{C}_{\text{me}}-\text{C}_{\text{ar}}-\text{C}_{\text{ar}} = -44.7$ (8)°; cy = cyanide, me = methylene and ar = aromatic]. In the crystal, $\text{N}-\text{H}\cdots\text{N}$ hydrogen bonds link the molecules into $C(7)$ chains propagating in [010]. Aromatic $\pi-\pi$ stacking interactions [minimum centroid-centroid separation = 3.663 (3) Å] are also observed.

Related literature

For a related structure, see: Ge *et al.* (2012).

Experimental

Crystal data

$\text{C}_{10}\text{H}_7\text{ClN}_2$	$V = 944.4$ (3) Å ³
$M_r = 190.63$	$Z = 4$
Monoclinic, $P2_1/c$	Mo $K\alpha$ radiation
$a = 9.761$ (2) Å	$\mu = 0.35$ mm ⁻¹
$b = 11.205$ (2) Å	$T = 293$ K
$c = 8.7791$ (18) Å	$0.20 \times 0.12 \times 0.10$ mm
$\beta = 100.39$ (3)°	

Data collection

Rigaku SCXmini CCD diffractometer	8938 measured reflections
Absorption correction: multi-scan (<i>CrystalClear</i> ; Rigaku, 2005)	2122 independent reflections
$T_{\text{min}} = 0.950$, $T_{\text{max}} = 0.965$	1247 reflections with $I > 2\sigma(I)$
	$R_{\text{int}} = 0.134$

Refinement

$R[F^2 > 2\sigma(F^2)] = 0.110$	118 parameters
$wR(F^2) = 0.312$	H-atom parameters constrained
$S = 1.05$	$\Delta\rho_{\text{max}} = 0.51$ e Å ⁻³
2122 reflections	$\Delta\rho_{\text{min}} = -0.46$ e Å ⁻³

Table 1

Hydrogen-bond geometry (Å, °).

$D-\text{H}\cdots A$	$D-\text{H}$	$\text{H}\cdots A$	$D\cdots A$	$D-\text{H}\cdots A$
$\text{N2}-\text{H4A}\cdots\text{N1}^1$	0.86	2.23	3.016 (6)	151

Symmetry code: (i) $-x + 2, y - \frac{1}{2}, -z + \frac{1}{2}$.

Data collection: *CrystalClear* (Rigaku, 2005); cell refinement: *CrystalClear*; data reduction: *CrystalClear*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 2008); program(s) used to refine structure: *SHELXL97* (Sheldrick, 2008); molecular graphics: *DIAMOND* (Brandenburg & Putz, 2005); software used to prepare material for publication: *SHELXL97*.

We thank Southeast University for support.

Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: HB6557).

References

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supporting information

Acta Cryst. (2012). E68, o144 [doi:10.1107/S1600536811053372]

2-(6-Chloro-1*H*-indol-3-yl)acetonitrile

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S1. Experimental

The title compound was obtained commercially from ChemFuture PharmaTech, Ltd (Nanjing, Jiangsu), and were used as received without further purification. Colourless prisms were obtained by slow evaporation of a methanol solution.

S2. Refinement

All H atoms attached to C atoms and N atoms were fixed geometrically and treated as riding with C—H = 0.93 Å (CH), C—H = 0.97 Å (CH₂) and N—H = 0.86 Å with $U_{\text{iso}}(\text{H}) = 1.2U_{\text{eq}}$.

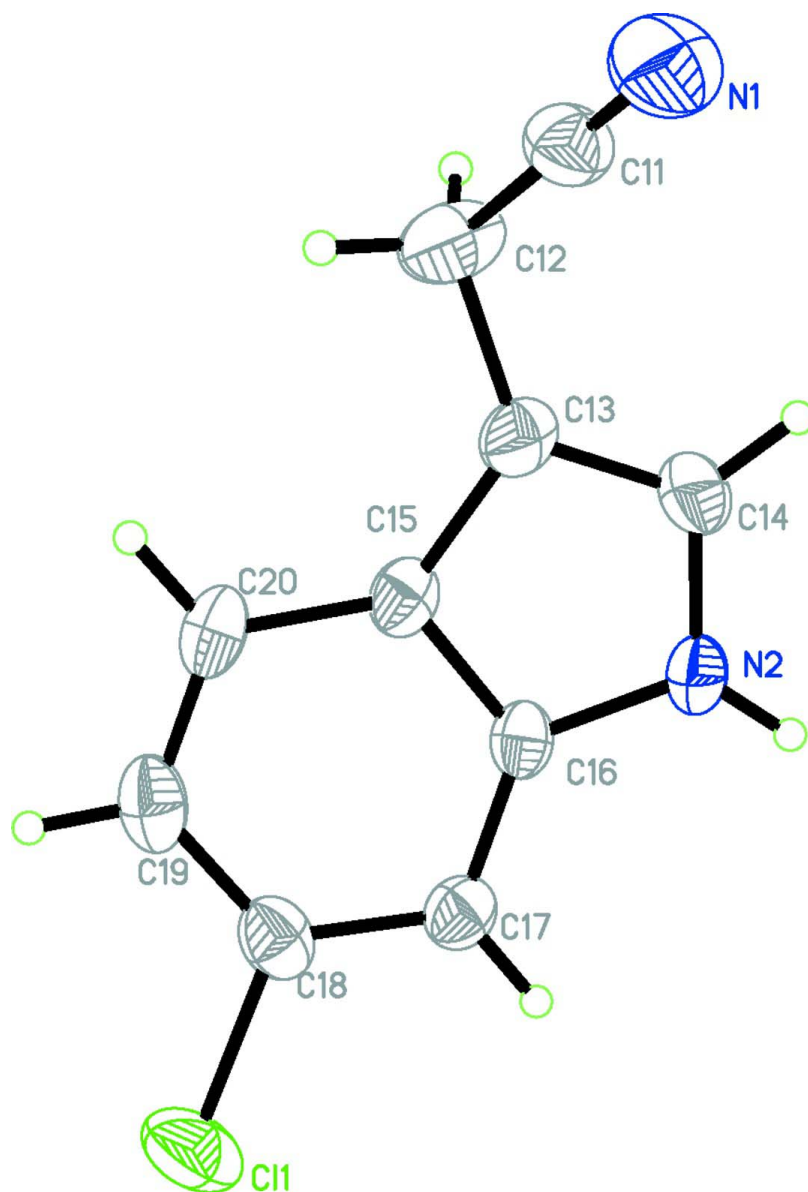
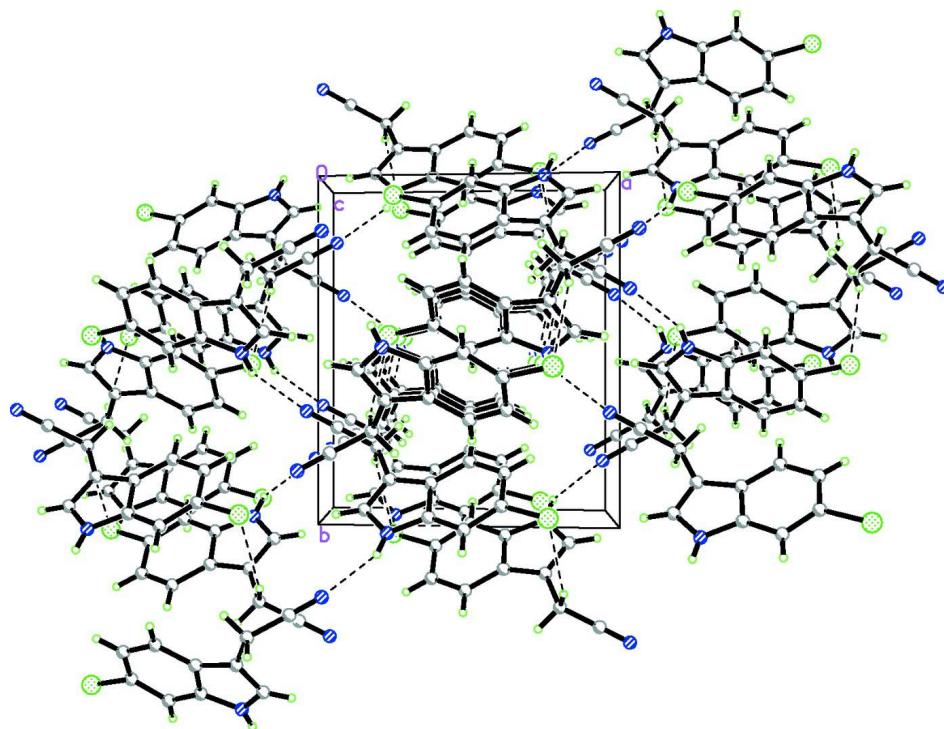


Figure 1

The molecular structure of the title compound with displacement ellipsoids drawn at the 30% probability level.

**Figure 2**

A packing view down the *a* axis showing hydrogen bonds as dashed lines.

2-(6-Chloro-1*H*-indol-3-yl)acetonitrile

Crystal data

$C_{10}H_7ClN_2$

$M_r = 190.63$

Monoclinic, $P2_1/c$

Hall symbol: -P 2ybc

$a = 9.761(2) \text{ \AA}$

$b = 11.205(2) \text{ \AA}$

$c = 8.7791(18) \text{ \AA}$

$\beta = 100.39(3)^\circ$

$V = 944.4(3) \text{ \AA}^3$

$Z = 4$

$F(000) = 392$

$D_x = 1.341 \text{ Mg m}^{-3}$

Mo $K\alpha$ radiation, $\lambda = 0.71073 \text{ \AA}$

Cell parameters from 2122 reflections

$\theta = 3.0\text{--}27.5^\circ$

$\mu = 0.35 \text{ mm}^{-1}$

$T = 293 \text{ K}$

Prism, colourless

$0.20 \times 0.12 \times 0.10 \text{ mm}$

Data collection

Rigaku SCXmini CCD
diffractometer

Radiation source: fine-focus sealed tube

Graphite monochromator

Detector resolution: $13.6612 \text{ pixels mm}^{-1}$

CCD_Profile_fitting scans

Absorption correction: multi-scan

(*CrystalClear*; Rigaku, 2005)

$T_{\min} = 0.950$, $T_{\max} = 0.965$

8938 measured reflections

2122 independent reflections

1247 reflections with $I > 2\sigma(I)$

$R_{\text{int}} = 0.134$

$\theta_{\max} = 27.5^\circ$, $\theta_{\min} = 3.0^\circ$

$h = -12 \rightarrow 12$

$k = -14 \rightarrow 13$

$l = -11 \rightarrow 11$

Refinement

Refinement on F^2
 Least-squares matrix: full
 $R[F^2 > 2\sigma(F^2)] = 0.110$
 $wR(F^2) = 0.312$
 $S = 1.05$
 2122 reflections
 118 parameters
 0 restraints
 Primary atom site location: structure-invariant
 direct methods

Secondary atom site location: difference Fourier
 map
 Hydrogen site location: inferred from
 neighbouring sites
 H-atom parameters constrained
 $w = 1/[\sigma^2(F_o^2) + (0.1465P)^2 + 0.5597P]$
 where $P = (F_o^2 + 2F_c^2)/3$
 $(\Delta/\sigma)_{\max} < 0.001$
 $\Delta\rho_{\max} = 0.51 \text{ e } \text{\AA}^{-3}$
 $\Delta\rho_{\min} = -0.46 \text{ e } \text{\AA}^{-3}$

Special details

Geometry. All e.s.d.'s (except the e.s.d. in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell e.s.d.'s are taken into account individually in the estimation of e.s.d.'s in distances, angles and torsion angles; correlations between e.s.d.'s in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell e.s.d.'s is used for estimating e.s.d.'s involving l.s. planes.

Refinement. Refinement of F^2 against ALL reflections. The weighted R -factor wR and goodness of fit S are based on F^2 , conventional R -factors R are based on F , with F set to zero for negative F^2 . The threshold expression of $F^2 > \sigma(F^2)$ is used only for calculating R -factors(gt) *etc.* and is not relevant to the choice of reflections for refinement. R -factors based on F^2 are statistically about twice as large as those based on F , and R -factors based on ALL data will be even larger.

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (\AA^2)

	<i>x</i>	<i>y</i>	<i>z</i>	$U_{\text{iso}}^*/U_{\text{eq}}$
C11	0.24329 (14)	0.04179 (19)	0.32068 (19)	0.0959 (8)
N1	1.0357 (5)	0.3256 (5)	0.0579 (6)	0.0929 (16)
N2	0.7608 (4)	-0.0082 (4)	0.2478 (4)	0.0578 (10)
H4A	0.7910	-0.0635	0.3133	0.069*
C11	0.9347 (5)	0.2792 (4)	0.0093 (5)	0.0628 (12)
C12	0.8029 (6)	0.2208 (6)	-0.0508 (6)	0.0842 (18)
H12A	0.8115	0.1777	-0.1445	0.101*
H12B	0.7314	0.2810	-0.0781	0.101*
C13	0.7581 (4)	0.1343 (4)	0.0646 (5)	0.0578 (12)
C14	0.8382 (5)	0.0517 (5)	0.1549 (6)	0.0673 (14)
H14A	0.9319	0.0378	0.1539	0.081*
C15	0.6203 (4)	0.1262 (4)	0.1028 (4)	0.0481 (10)
C16	0.6263 (4)	0.0360 (3)	0.2172 (5)	0.0467 (10)
C17	0.5117 (4)	0.0061 (4)	0.2868 (5)	0.0519 (11)
H17A	0.5167	-0.0525	0.3627	0.062*
C18	0.3907 (5)	0.0705 (4)	0.2342 (5)	0.0588 (12)
C19	0.3788 (5)	0.1573 (4)	0.1185 (6)	0.0646 (13)
H19A	0.2943	0.1961	0.0862	0.078*
C20	0.4923 (5)	0.1854 (4)	0.0522 (5)	0.0568 (11)
H20A	0.4848	0.2430	-0.0252	0.068*

Atomic displacement parameters (\AA^2)

	U^{11}	U^{22}	U^{33}	U^{12}	U^{13}	U^{23}
C11	0.0512 (9)	0.1570 (18)	0.0826 (11)	-0.0064 (8)	0.0201 (7)	-0.0179 (9)

N1	0.070 (3)	0.108 (4)	0.096 (4)	-0.037 (3)	0.003 (2)	0.003 (3)
N2	0.042 (2)	0.073 (2)	0.054 (2)	0.0056 (18)	-0.0036 (15)	0.0129 (18)
C11	0.056 (3)	0.067 (3)	0.067 (3)	-0.009 (2)	0.013 (2)	0.001 (2)
C12	0.078 (3)	0.111 (4)	0.056 (3)	-0.046 (3)	-0.006 (2)	0.014 (3)
C13	0.050 (2)	0.074 (3)	0.044 (2)	-0.018 (2)	-0.0083 (17)	0.0047 (19)
C14	0.038 (2)	0.097 (4)	0.064 (3)	-0.003 (2)	0.001 (2)	0.005 (2)
C15	0.046 (2)	0.051 (2)	0.042 (2)	-0.0085 (18)	-0.0082 (16)	-0.0029 (16)
C16	0.041 (2)	0.049 (2)	0.046 (2)	-0.0003 (17)	-0.0047 (16)	-0.0014 (16)
C17	0.054 (3)	0.054 (2)	0.046 (2)	-0.003 (2)	0.0033 (19)	0.0004 (18)
C18	0.047 (2)	0.067 (3)	0.061 (3)	-0.001 (2)	0.0034 (19)	-0.013 (2)
C19	0.053 (3)	0.066 (3)	0.068 (3)	0.013 (2)	-0.008 (2)	-0.012 (2)
C20	0.062 (3)	0.047 (2)	0.053 (2)	0.003 (2)	-0.0123 (19)	0.0017 (17)

Geometric parameters (Å, °)

C11—C18	1.773 (5)	C14—H14A	0.9300
N1—C11	1.128 (6)	C15—C20	1.413 (6)
N2—C14	1.381 (6)	C15—C16	1.419 (6)
N2—C16	1.384 (5)	C16—C17	1.408 (6)
N2—H4A	0.8600	C17—C18	1.390 (6)
C11—C12	1.455 (7)	C17—H17A	0.9300
C12—C13	1.522 (6)	C18—C19	1.396 (7)
C12—H12A	0.9700	C19—C20	1.379 (7)
C12—H12B	0.9700	C19—H19A	0.9300
C13—C14	1.368 (7)	C20—H20A	0.9300
C13—C15	1.447 (6)		
C14—N2—C16	108.2 (4)	C20—C15—C13	134.7 (4)
C14—N2—H4A	125.9	C16—C15—C13	106.7 (4)
C16—N2—H4A	125.9	N2—C16—C17	129.1 (4)
N1—C11—C12	178.7 (6)	N2—C16—C15	108.0 (4)
C11—C12—C13	112.7 (4)	C17—C16—C15	122.8 (4)
C11—C12—H12A	109.0	C18—C17—C16	115.2 (4)
C13—C12—H12A	109.0	C18—C17—H17A	122.4
C11—C12—H12B	109.0	C16—C17—H17A	122.4
C13—C12—H12B	109.0	C17—C18—C19	123.9 (4)
H12A—C12—H12B	107.8	C17—C18—C11	118.0 (4)
C14—C13—C15	106.3 (4)	C19—C18—C11	118.0 (4)
C14—C13—C12	127.8 (5)	C20—C19—C18	119.9 (4)
C15—C13—C12	125.8 (4)	C20—C19—H19A	120.0
C13—C14—N2	110.8 (4)	C18—C19—H19A	120.0
C13—C14—H14A	124.6	C19—C20—C15	119.5 (4)
N2—C14—H14A	124.6	C19—C20—H20A	120.3
C20—C15—C16	118.6 (4)	C15—C20—H20A	120.3
N1—C11—C12—C13	-68 (28)	C13—C15—C16—N2	0.3 (4)
C11—C12—C13—C14	-44.7 (8)	C20—C15—C16—C17	-2.5 (6)
C11—C12—C13—C15	133.6 (5)	C13—C15—C16—C17	177.6 (4)

C15—C13—C14—N2	0.0 (6)	N2—C16—C17—C18	177.5 (4)
C12—C13—C14—N2	178.6 (4)	C15—C16—C17—C18	0.8 (6)
C16—N2—C14—C13	0.2 (6)	C16—C17—C18—C19	1.3 (6)
C14—C13—C15—C20	-180.0 (5)	C16—C17—C18—C11	-177.9 (3)
C12—C13—C15—C20	1.4 (8)	C17—C18—C19—C20	-1.6 (7)
C14—C13—C15—C16	-0.1 (5)	C11—C18—C19—C20	177.6 (4)
C12—C13—C15—C16	-178.8 (4)	C18—C19—C20—C15	-0.3 (7)
C14—N2—C16—C17	-177.4 (4)	C16—C15—C20—C19	2.2 (6)
C14—N2—C16—C15	-0.3 (5)	C13—C15—C20—C19	-178.0 (5)
C20—C15—C16—N2	-179.8 (4)		

Hydrogen-bond geometry (Å, °)

<i>D—H...A</i>	<i>D—H</i>	<i>H...A</i>	<i>D...A</i>	<i>D—H...A</i>
N2—H4A...N1 ⁱ	0.86	2.23	3.016 (6)	151

Symmetry code: (i) $-x+2, y-1/2, -z+1/2$.