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## Structure Reports

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## 2,2',5,5'-Tetrachloro-*N,N'*-diethyl-*N,N'*-[benzene-1,3-diylbis(methylene)]-dibenzenesulfonamide

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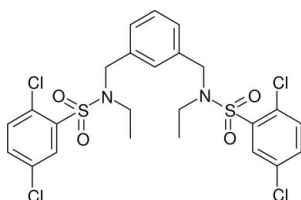
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Key indicators: single-crystal X-ray study;  $T = 296$  K; mean  $\sigma(\text{C}-\text{C}) = 0.003$  Å;  $R$  factor = 0.036;  $wR$  factor = 0.102; data-to-parameter ratio = 21.1.

In the title compound,  $\text{C}_{24}\text{H}_{24}\text{Cl}_4\text{N}_2\text{O}_4\text{S}_2$ , the dihedral angles between the central benzene ring and the pendant rings are  $58.09$  (10) and  $62.59$  (10)°. The dihedral angle between the pendant rings is  $81.64$  (9)°. Both sulfonamide groups lie to the same side of the central ring but the C—S—N—C torsion angles [73.09 (16) and  $-117.35$  (14)] and S—N—C—C torsion angles [ $-143.80$  (14) and  $-111.45$  (16)°] differ significantly for the two pendant chains. The N atoms are close to planar (bond angle sums = 356.4 and 359.5°). In the crystal, weak C—H...O and C—H...Cl interactions link the molecules.

## Related literature

 For related structures, see: Ejaz *et al.* (2011*a,b*).


## Experimental

## Crystal data

 $\text{C}_{24}\text{H}_{24}\text{Cl}_4\text{N}_2\text{O}_4\text{S}_2$ 
 $M_r = 610.37$ 

Triclinic,  $P\bar{1}$   
 $a = 8.0396$  (2) Å  
 $b = 11.1512$  (3) Å  
 $c = 15.5723$  (3) Å  
 $\alpha = 87.454$  (1)°  
 $\beta = 83.378$  (1)°  
 $\gamma = 87.995$  (1)°

$V = 1384.77$  (6) Å<sup>3</sup>  
 $Z = 2$   
 Mo  $K\alpha$  radiation  
 $\mu = 0.61$  mm<sup>-1</sup>  
 $T = 296$  K  
 $0.50 \times 0.35 \times 0.30$  mm

## Data collection

Bruker APEXII CCD diffractometer  
 25271 measured reflections

6905 independent reflections  
 5380 reflections with  $I > 2\sigma(I)$   
 $R_{\text{int}} = 0.020$

## Refinement

$R[F^2 > 2\sigma(F^2)] = 0.036$   
 $wR(F^2) = 0.102$   
 $S = 1.04$   
 6905 reflections

327 parameters  
 H-atom parameters constrained  
 $\Delta\rho_{\text{max}} = 0.47$  e Å<sup>-3</sup>  
 $\Delta\rho_{\text{min}} = -0.38$  e Å<sup>-3</sup>

Table 1

Hydrogen-bond geometry (Å, °).

$D-H\cdots A$	$D-H$	$H\cdots A$	$D\cdots A$	$D-H\cdots A$
$C7-H7B\cdots O3^i$	0.97	2.59	3.511 (2)	158
$C17-H17B\cdots O1^{ii}$	0.97	2.58	3.516 (3)	164
$C24-H24\cdots Cl1^{ii}$	0.93	2.83	3.738 (2)	166

Symmetry codes: (i)  $-x + 1, -y + 1, -z + 1$ ; (ii)  $-x + 1, -y + 2, -z + 1$ .

Data collection: *APEX2* (Bruker, 2007); cell refinement: *SAINT* (Bruker, 2007); data reduction: *SAINT*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 2008); program(s) used to refine structure: *SHELXL97* (Sheldrick, 2008); molecular graphics: *ORTEP-3* (Farrugia, 1997); software used to prepare material for publication: *SHELXL97*.

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Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: SU2337).

## References

- Bruker (2007). *APEX2* and *SAINT*. Bruker AXS Inc., Madison, Wisconsin, USA.  
 Ejaz, Khan, I. U., Ahmad, H. & Harrison, W. T. A. (2011*a*). *Acta Cryst.* **E67**, o3037.  
 Ejaz, Khan, I. U., Ahmad, H., Harrison, W. T. A. & Sheikh, T. A. (2011*b*). *Acta Cryst.* **E67**, o3038.  
 Farrugia, L. J. (1997). *J. Appl. Cryst.* **30**, 565.  
 Sheldrick, G. M. (2008). *Acta Cryst.* **A64**, 112–122.

## supporting information

*Acta Cryst.* (2012). E68, o15 [doi:10.1107/S1600536811045326]

## 2,2',5,5'-Tetrachloro-*N,N'*-diethyl-*N,N'*-[benzene-1,3-diylbis(methylene)]dibenzenesulfonamide

Ejaz, Islam Ullah Khan, Tahir Ali Sheikh and William T. A. Harrison

### S1. Comment

As part of our ongoing studies of symmetrical aryl sulfonamides (Ejaz *et al.*, 2011*a,b*), the synthesis and structure of the title compound are described herein.

In the title compound (Fig. 1), the dihedral angles between the central (C1-C6) benzene ring and the pendant (C10-C15) and (C19-C24) rings are 58.09 (1) and 62.59 (10)°, respectively. The equivalent angle between the pendant rings is 81.64 (9)°. Both sulfonamide groups lie to the same side of the central ring, but the C10—S1—N1—C7 and C19—S2—N2—C16 torsion angles [73.09 (16) and -117.35 (14)°, respectively] and the S1—N1—C7—C2 and S2—N2—C16—C6 torsion angles [-143.80 (14) and -111.45 (16)°, respectively] differ significantly for the two pendant chains. The conformations of the ethyl side chains are also different: the S1—N1—C8—C9 and S2—N2—C17—C18 torsion angles are -89.6 (2) and -126.57 (19)°, respectively. The nitrogen atoms are close to planar (bond angle sums = 356.4 and 359.5° for N1 and N2, respectively), which seems to indicate a valence state close to  $sp^2$  hybridization, as also seen in a related structure (Ejaz *et al.*, 2011*a*).

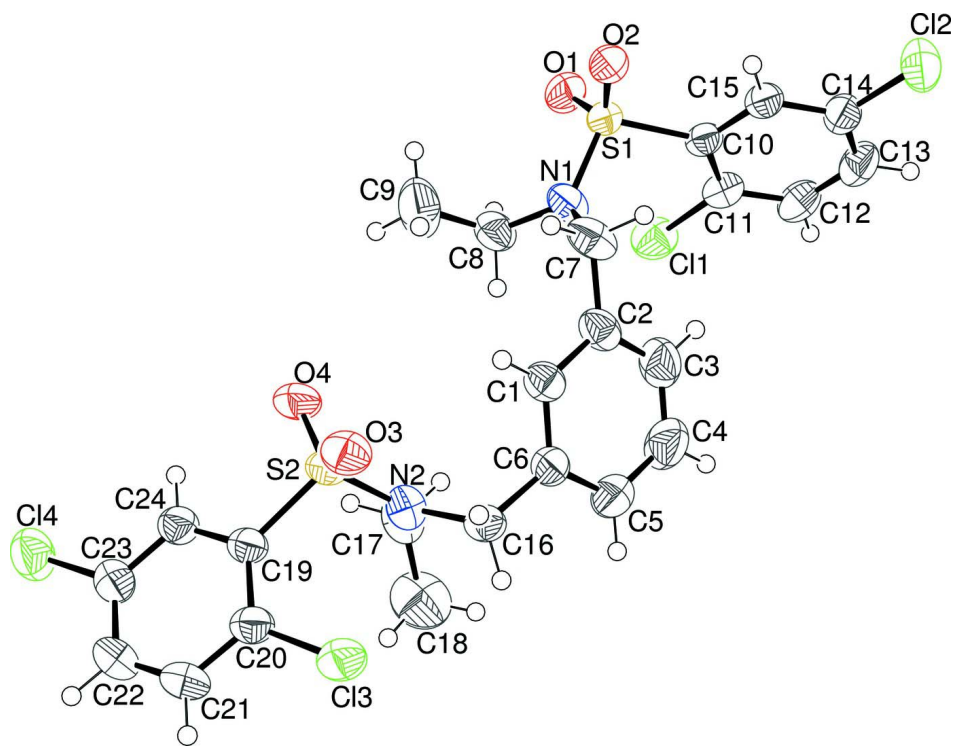
In the crystal, weak C—H···O and C—H···Cl interactions link the molecules (Table 1).

### S2. Experimental

A mixture of *N,N'*-(benzene-1,3-diyl dimethanediyl)bis(2,5-dichlorobenzenesulfonamide) (0.3 g; 0.5 mmol), sodium hydride (0.25 g; 0.9 mmol) and *N,N*-dimethylformamide (10.0 ml) was stirred in a 100-ml round bottom flask at room temperature for half an hour followed by the addition of ethyl iodide (0.15 g; 0.9 mmol). The reaction mixture was further stirred for five hours, and its completion was monitored by TLC. After completion, the contents were poured over crushed ice. The precipitated product was isolated, washed and crystallized from methanol to yield colourless block-like crystals of the title compound.

### S3. Refinement

The hydrogen atoms were placed in calculated positions (C—H = 0.93–0.97 Å) and refined as riding atoms with  $U_{\text{iso}}(\text{H}) = 1.2U_{\text{eq}}(\text{C})$  or  $1.5U_{\text{eq}}(\text{methyl C})$ . The methyl groups were allowed to rotate, but not to tip, to best fit the electron density.

**Figure 1**

The molecular structure of the title molecule, showing 50% displacement ellipsoids and the numbering scheme.

### 2,2',5,5'-Tetrachloro-*N,N'*-diethyl-*N,N'*-[benzene-1,3-diylbis(methylene)]dibenzenesulfonamide

#### Crystal data

$C_{24}H_{24}Cl_4N_2O_4S_2$

$M_r = 610.37$

Triclinic,  $P\bar{1}$

Hall symbol:  $-P\ 1$

$a = 8.0396\ (2)\ \text{\AA}$

$b = 11.1512\ (3)\ \text{\AA}$

$c = 15.5723\ (3)\ \text{\AA}$

$\alpha = 87.454\ (1)^\circ$

$\beta = 83.378\ (1)^\circ$

$\gamma = 87.995\ (1)^\circ$

$V = 1384.77\ (6)\ \text{\AA}^3$

$Z = 2$

$F(000) = 628$

$D_x = 1.464\ \text{Mg m}^{-3}$

Mo  $K\alpha$  radiation,  $\lambda = 0.71073\ \text{\AA}$

Cell parameters from 6905 reflections

$\theta = 2.2\text{--}28.1^\circ$

$\mu = 0.61\ \text{mm}^{-1}$

$T = 296\ \text{K}$

Chnkc, colourless

$0.50 \times 0.35 \times 0.30\ \text{mm}$

#### Data collection

Bruker APEXII CCD  
diffractometer

Radiation source: fine-focus sealed tube

Graphite monochromator

$\omega$  scans

25271 measured reflections

6905 independent reflections

5380 reflections with  $I > 2\sigma(I)$

$R_{\text{int}} = 0.020$

$\theta_{\text{max}} = 28.4^\circ$ ,  $\theta_{\text{min}} = 3.0^\circ$

$h = -8 \rightarrow 10$

$k = -14 \rightarrow 14$

$l = -20 \rightarrow 20$

Refinement

Refinement on  $F^2$   
 Least-squares matrix: full  
 $R[F^2 > 2\sigma(F^2)] = 0.036$   
 $wR(F^2) = 0.102$   
 $S = 1.04$   
 6905 reflections  
 327 parameters  
 0 restraints  
 Primary atom site location: structure-invariant  
 direct methods

Secondary atom site location: difference Fourier  
 map  
 Hydrogen site location: inferred from  
 neighbouring sites  
 H-atom parameters constrained  
 $w = 1/[\sigma^2(F_o^2) + (0.0462P)^2 + 0.4193P]$   
 where  $P = (F_o^2 + 2F_c^2)/3$   
 $(\Delta/\sigma)_{\max} = 0.001$   
 $\Delta\rho_{\max} = 0.47 \text{ e } \text{\AA}^{-3}$   
 $\Delta\rho_{\min} = -0.38 \text{ e } \text{\AA}^{-3}$

Special details

**Geometry.** All e.s.d.'s (except the e.s.d. in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell e.s.d.'s are taken into account individually in the estimation of e.s.d.'s in distances, angles and torsion angles; correlations between e.s.d.'s in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell e.s.d.'s is used for estimating e.s.d.'s involving l.s. planes.

**Refinement.** Refinement of  $F^2$  against ALL reflections. The weighted  $R$ -factor  $wR$  and goodness of fit  $S$  are based on  $F^2$ , conventional  $R$ -factors  $R$  are based on  $F$ , with  $F$  set to zero for negative  $F^2$ . The threshold expression of  $F^2 > \sigma(F^2)$  is used only for calculating  $R$ -factors(gt) *etc.* and is not relevant to the choice of reflections for refinement.  $R$ -factors based on  $F^2$  are statistically about twice as large as those based on  $F$ , and  $R$ -factors based on ALL data will be even larger.

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters ( $\text{\AA}^2$ )

	$x$	$y$	$z$	$U_{\text{iso}}^*/U_{\text{eq}}$
C1	0.6756 (2)	0.63163 (16)	0.43821 (11)	0.0454 (4)
H1	0.5623	0.6242	0.4333	0.055*
C2	0.7280 (3)	0.63672 (17)	0.51940 (12)	0.0502 (4)
C3	0.8970 (3)	0.6480 (2)	0.52549 (15)	0.0695 (6)
H3	0.9340	0.6525	0.5797	0.083*
C4	1.0103 (3)	0.6527 (3)	0.45304 (17)	0.0776 (7)
H4	1.1236	0.6594	0.4583	0.093*
C5	0.9572 (3)	0.6475 (2)	0.37192 (14)	0.0622 (5)
H5	1.0347	0.6509	0.3228	0.075*
C6	0.7890 (2)	0.63737 (16)	0.36384 (11)	0.0452 (4)
C7	0.6016 (3)	0.63174 (18)	0.59876 (12)	0.0579 (5)
H7A	0.4935	0.6113	0.5825	0.069*
H7B	0.6359	0.5696	0.6392	0.069*
C8	0.5407 (3)	0.85534 (19)	0.58997 (13)	0.0571 (5)
H8A	0.6171	0.8607	0.5371	0.069*
H8B	0.5553	0.9256	0.6225	0.069*
C9	0.3646 (4)	0.8577 (3)	0.5668 (2)	0.0967 (9)
H9A	0.3463	0.9274	0.5304	0.145*
H9B	0.2875	0.8603	0.6187	0.145*
H9C	0.3473	0.7867	0.5367	0.145*
C10	0.75316 (19)	0.69614 (16)	0.77750 (10)	0.0395 (3)
C11	0.8994 (2)	0.75963 (17)	0.75576 (12)	0.0462 (4)
C12	1.0466 (2)	0.71855 (19)	0.78590 (14)	0.0567 (5)
H12	1.1438	0.7614	0.7717	0.068*
C13	1.0506 (2)	0.6145 (2)	0.83684 (14)	0.0578 (5)

H13	1.1499	0.5873	0.8574	0.069*
C14	0.9069 (2)	0.55140 (17)	0.85701 (12)	0.0476 (4)
C15	0.7572 (2)	0.59124 (16)	0.82812 (11)	0.0417 (4)
H15	0.6605	0.5479	0.8426	0.050*
C16	0.7269 (2)	0.63890 (16)	0.27608 (11)	0.0462 (4)
H16A	0.8216	0.6338	0.2317	0.055*
H16B	0.6586	0.5697	0.2724	0.055*
C17	0.7074 (3)	0.86554 (18)	0.26202 (15)	0.0604 (5)
H17A	0.7458	0.8739	0.3182	0.073*
H17B	0.6251	0.9295	0.2538	0.073*
C18	0.8521 (4)	0.8787 (3)	0.1935 (2)	0.0934 (9)
H18A	0.8894	0.9598	0.1910	0.140*
H18B	0.8181	0.8602	0.1386	0.140*
H18C	0.9419	0.8246	0.2069	0.140*
C19	0.4012 (2)	0.80340 (16)	0.14908 (10)	0.0414 (4)
C20	0.4657 (2)	0.74445 (17)	0.07480 (11)	0.0470 (4)
C21	0.4306 (3)	0.7893 (2)	-0.00540 (12)	0.0597 (5)
H21	0.4722	0.7489	-0.0547	0.072*
C22	0.3354 (3)	0.8928 (2)	-0.01335 (13)	0.0620 (5)
H22	0.3119	0.9223	-0.0676	0.074*
C23	0.2751 (2)	0.95234 (18)	0.05987 (12)	0.0514 (4)
C24	0.3068 (2)	0.90861 (17)	0.14088 (11)	0.0458 (4)
H24	0.2649	0.9497	0.1898	0.055*
S1	0.55636 (5)	0.74464 (4)	0.74476 (3)	0.04106 (11)
S2	0.42957 (6)	0.74915 (4)	0.25594 (3)	0.04402 (11)
N1	0.5864 (2)	0.74805 (14)	0.64078 (9)	0.0489 (4)
N2	0.62753 (19)	0.74925 (13)	0.26098 (10)	0.0475 (3)
O1	0.51855 (17)	0.86403 (12)	0.77167 (9)	0.0560 (3)
O2	0.44311 (14)	0.65269 (12)	0.77704 (8)	0.0515 (3)
O3	0.37690 (17)	0.62781 (12)	0.26631 (9)	0.0563 (3)
O4	0.34712 (19)	0.83655 (14)	0.31150 (8)	0.0622 (4)
Cl1	0.90284 (7)	0.88984 (5)	0.69119 (4)	0.06287 (14)
Cl2	0.91046 (7)	0.41825 (5)	0.91822 (4)	0.06768 (16)
Cl3	0.59191 (7)	0.61629 (5)	0.07875 (3)	0.06285 (15)
Cl4	0.15880 (8)	1.08554 (5)	0.05174 (4)	0.07195 (17)

Atomic displacement parameters ( $\text{\AA}^2$ )

	$U^{11}$	$U^{22}$	$U^{33}$	$U^{12}$	$U^{13}$	$U^{23}$
C1	0.0522 (10)	0.0428 (10)	0.0418 (9)	-0.0030 (7)	-0.0065 (7)	-0.0035 (7)
C2	0.0699 (12)	0.0418 (10)	0.0399 (9)	0.0025 (8)	-0.0106 (8)	-0.0043 (7)
C3	0.0735 (14)	0.0824 (16)	0.0571 (13)	0.0149 (12)	-0.0270 (11)	-0.0146 (11)
C4	0.0519 (12)	0.107 (2)	0.0773 (16)	0.0142 (12)	-0.0204 (11)	-0.0194 (14)
C5	0.0492 (11)	0.0776 (15)	0.0591 (12)	0.0081 (10)	-0.0047 (9)	-0.0072 (11)
C6	0.0494 (9)	0.0430 (10)	0.0429 (9)	0.0052 (7)	-0.0054 (7)	-0.0032 (7)
C7	0.0879 (15)	0.0477 (11)	0.0384 (9)	-0.0080 (10)	-0.0066 (9)	-0.0019 (8)
C8	0.0692 (13)	0.0528 (11)	0.0463 (10)	0.0054 (9)	-0.0006 (9)	0.0103 (9)
C9	0.0856 (19)	0.106 (2)	0.099 (2)	0.0127 (16)	-0.0304 (16)	0.0276 (18)

C10	0.0340 (7)	0.0486 (10)	0.0360 (8)	-0.0002 (7)	-0.0017 (6)	-0.0088 (7)
C11	0.0416 (9)	0.0473 (10)	0.0486 (10)	-0.0037 (7)	0.0021 (7)	-0.0083 (8)
C12	0.0345 (8)	0.0634 (13)	0.0716 (13)	-0.0056 (8)	-0.0007 (8)	-0.0081 (10)
C13	0.0368 (9)	0.0688 (13)	0.0688 (13)	0.0047 (8)	-0.0101 (8)	-0.0098 (10)
C14	0.0446 (9)	0.0508 (11)	0.0478 (10)	0.0054 (8)	-0.0077 (7)	-0.0067 (8)
C15	0.0363 (8)	0.0506 (10)	0.0383 (8)	-0.0013 (7)	-0.0034 (6)	-0.0068 (7)
C16	0.0526 (10)	0.0469 (10)	0.0382 (9)	0.0034 (8)	-0.0012 (7)	-0.0060 (7)
C17	0.0711 (13)	0.0451 (11)	0.0688 (13)	-0.0084 (9)	-0.0234 (11)	0.0018 (9)
C18	0.098 (2)	0.0845 (19)	0.097 (2)	-0.0371 (16)	-0.0047 (16)	0.0179 (16)
C19	0.0412 (8)	0.0490 (10)	0.0340 (8)	-0.0053 (7)	-0.0023 (6)	-0.0035 (7)
C20	0.0479 (9)	0.0528 (11)	0.0394 (9)	-0.0017 (8)	0.0001 (7)	-0.0069 (8)
C21	0.0703 (13)	0.0731 (14)	0.0347 (9)	0.0004 (11)	-0.0004 (8)	-0.0085 (9)
C22	0.0715 (13)	0.0772 (15)	0.0375 (10)	0.0008 (11)	-0.0103 (9)	0.0033 (9)
C23	0.0531 (10)	0.0530 (11)	0.0486 (10)	-0.0030 (8)	-0.0100 (8)	0.0023 (8)
C24	0.0452 (9)	0.0519 (11)	0.0406 (9)	-0.0028 (8)	-0.0044 (7)	-0.0054 (8)
S1	0.0364 (2)	0.0499 (3)	0.0361 (2)	0.00241 (17)	-0.00185 (15)	-0.00196 (17)
S2	0.0484 (2)	0.0491 (3)	0.0329 (2)	-0.00132 (18)	0.00197 (16)	-0.00125 (17)
N1	0.0683 (10)	0.0431 (8)	0.0353 (7)	0.0015 (7)	-0.0079 (7)	0.0006 (6)
N2	0.0529 (8)	0.0411 (8)	0.0501 (8)	-0.0039 (6)	-0.0123 (7)	-0.0002 (6)
O1	0.0568 (8)	0.0580 (8)	0.0521 (8)	0.0143 (6)	-0.0022 (6)	-0.0118 (6)
O2	0.0339 (6)	0.0686 (9)	0.0508 (7)	-0.0053 (6)	-0.0034 (5)	0.0098 (6)
O3	0.0571 (8)	0.0549 (8)	0.0549 (8)	-0.0138 (6)	0.0022 (6)	0.0074 (6)
O4	0.0727 (9)	0.0730 (10)	0.0378 (7)	0.0149 (7)	0.0046 (6)	-0.0090 (6)
Cl1	0.0584 (3)	0.0556 (3)	0.0720 (3)	-0.0115 (2)	0.0042 (2)	0.0037 (2)
Cl2	0.0640 (3)	0.0647 (3)	0.0758 (4)	0.0053 (2)	-0.0212 (3)	0.0105 (3)
Cl3	0.0713 (3)	0.0645 (3)	0.0507 (3)	0.0141 (3)	0.0007 (2)	-0.0128 (2)
Cl4	0.0829 (4)	0.0666 (4)	0.0675 (3)	0.0136 (3)	-0.0205 (3)	0.0039 (3)

*Geometric parameters (Å, °)*

C1—C2	1.382 (2)	C14—C15	1.385 (2)
C1—C6	1.389 (3)	C14—Cl2	1.730 (2)
C1—H1	0.9300	C15—H15	0.9300
C2—C3	1.383 (3)	C16—N2	1.468 (2)
C2—C7	1.507 (3)	C16—H16A	0.9700
C3—C4	1.366 (3)	C16—H16B	0.9700
C3—H3	0.9300	C17—N2	1.469 (2)
C4—C5	1.383 (3)	C17—C18	1.491 (4)
C4—H4	0.9300	C17—H17A	0.9700
C5—C6	1.381 (3)	C17—H17B	0.9700
C5—H5	0.9300	C18—H18A	0.9600
C6—C16	1.507 (2)	C18—H18B	0.9600
C7—N1	1.474 (2)	C18—H18C	0.9600
C7—H7A	0.9700	C19—C24	1.384 (2)
C7—H7B	0.9700	C19—C20	1.394 (2)
C8—N1	1.465 (2)	C19—S2	1.7820 (17)
C8—C9	1.499 (3)	C20—C21	1.380 (3)
C8—H8A	0.9700	C20—Cl3	1.726 (2)

C8—H8B	0.9700	C21—C22	1.371 (3)
C9—H9A	0.9600	C21—H21	0.9300
C9—H9B	0.9600	C22—C23	1.376 (3)
C9—H9C	0.9600	C22—H22	0.9300
C10—C15	1.383 (2)	C23—C24	1.380 (3)
C10—C11	1.394 (2)	C23—C14	1.735 (2)
C10—S1	1.7769 (16)	C24—H24	0.9300
C11—C12	1.379 (3)	S1—O1	1.4256 (14)
C11—C11	1.7282 (19)	S1—O2	1.4307 (13)
C12—C13	1.378 (3)	S1—N1	1.6082 (15)
C12—H12	0.9300	S2—O4	1.4267 (14)
C13—C14	1.372 (3)	S2—O3	1.4286 (14)
C13—H13	0.9300	S2—N2	1.6025 (16)
C2—C1—C6	121.26 (18)	N2—C16—C6	110.74 (14)
C2—C1—H1	119.4	N2—C16—H16A	109.5
C6—C1—H1	119.4	C6—C16—H16A	109.5
C1—C2—C3	118.56 (19)	N2—C16—H16B	109.5
C1—C2—C7	119.97 (19)	C6—C16—H16B	109.5
C3—C2—C7	121.47 (18)	H16A—C16—H16B	108.1
C4—C3—C2	120.9 (2)	N2—C17—C18	112.5 (2)
C4—C3—H3	119.6	N2—C17—H17A	109.1
C2—C3—H3	119.6	C18—C17—H17A	109.1
C3—C4—C5	120.3 (2)	N2—C17—H17B	109.1
C3—C4—H4	119.8	C18—C17—H17B	109.1
C5—C4—H4	119.8	H17A—C17—H17B	107.8
C6—C5—C4	120.1 (2)	C17—C18—H18A	109.5
C6—C5—H5	120.0	C17—C18—H18B	109.5
C4—C5—H5	120.0	H18A—C18—H18B	109.5
C5—C6—C1	118.91 (17)	C17—C18—H18C	109.5
C5—C6—C16	121.04 (17)	H18A—C18—H18C	109.5
C1—C6—C16	119.97 (16)	H18B—C18—H18C	109.5
N1—C7—C2	110.85 (16)	C24—C19—C20	119.20 (16)
N1—C7—H7A	109.5	C24—C19—S2	117.24 (13)
C2—C7—H7A	109.5	C20—C19—S2	123.55 (14)
N1—C7—H7B	109.5	C21—C20—C19	119.87 (18)
C2—C7—H7B	109.5	C21—C20—C13	117.85 (15)
H7A—C7—H7B	108.1	C19—C20—C13	122.28 (14)
N1—C8—C9	114.22 (19)	C22—C21—C20	120.85 (19)
N1—C8—H8A	108.7	C22—C21—H21	119.6
C9—C8—H8A	108.7	C20—C21—H21	119.6
N1—C8—H8B	108.7	C21—C22—C23	119.20 (18)
C9—C8—H8B	108.7	C21—C22—H22	120.4
H8A—C8—H8B	107.6	C23—C22—H22	120.4
C8—C9—H9A	109.5	C22—C23—C24	121.06 (19)
C8—C9—H9B	109.5	C22—C23—C14	120.31 (15)
H9A—C9—H9B	109.5	C24—C23—C14	118.63 (16)
C8—C9—H9C	109.5	C23—C24—C19	119.79 (17)

H9A—C9—H9C	109.5	C23—C24—H24	120.1
H9B—C9—H9C	109.5	C19—C24—H24	120.1
C15—C10—C11	119.74 (15)	O1—S1—O2	118.08 (8)
C15—C10—S1	117.14 (12)	O1—S1—N1	108.23 (8)
C11—C10—S1	123.11 (14)	O2—S1—N1	110.99 (8)
C12—C11—C10	119.87 (18)	O1—S1—C10	109.11 (8)
C12—C11—C11	118.30 (15)	O2—S1—C10	105.20 (8)
C10—C11—C11	121.83 (14)	N1—S1—C10	104.35 (8)
C13—C12—C11	120.47 (18)	O4—S2—O3	118.77 (9)
C13—C12—H12	119.8	O4—S2—N2	109.82 (9)
C11—C12—H12	119.8	O3—S2—N2	108.25 (8)
C14—C13—C12	119.45 (18)	O4—S2—C19	105.17 (9)
C14—C13—H13	120.3	O3—S2—C19	108.50 (8)
C12—C13—H13	120.3	N2—S2—C19	105.53 (8)
C13—C14—C15	121.21 (18)	C8—N1—C7	118.50 (15)
C13—C14—C12	120.22 (15)	C8—N1—S1	120.83 (13)
C15—C14—C12	118.56 (15)	C7—N1—S1	117.12 (12)
C10—C15—C14	119.25 (16)	C16—N2—C17	118.81 (16)
C10—C15—H15	120.4	C16—N2—S2	122.54 (12)
C14—C15—H15	120.4	C17—N2—S2	118.14 (13)
C6—C1—C2—C3	0.1 (3)	C22—C23—C24—C19	0.2 (3)
C6—C1—C2—C7	179.28 (17)	C14—C23—C24—C19	-179.11 (14)
C1—C2—C3—C4	-0.7 (3)	C20—C19—C24—C23	1.3 (3)
C7—C2—C3—C4	-179.9 (2)	S2—C19—C24—C23	-177.43 (14)
C2—C3—C4—C5	0.7 (4)	C15—C10—S1—O1	124.12 (14)
C3—C4—C5—C6	-0.1 (4)	C11—C10—S1—O1	-54.76 (16)
C4—C5—C6—C1	-0.5 (3)	C15—C10—S1—O2	-3.50 (15)
C4—C5—C6—C16	176.4 (2)	C11—C10—S1—O2	177.63 (14)
C2—C1—C6—C5	0.5 (3)	C15—C10—S1—N1	-120.40 (14)
C2—C1—C6—C16	-176.41 (17)	C11—C10—S1—N1	60.73 (16)
C1—C2—C7—N1	-111.1 (2)	C24—C19—S2—O4	-2.36 (16)
C3—C2—C7—N1	68.0 (2)	C20—C19—S2—O4	178.98 (15)
C15—C10—C11—C12	-1.0 (3)	C24—C19—S2—O3	125.71 (14)
S1—C10—C11—C12	177.85 (15)	C20—C19—S2—O3	-52.96 (17)
C15—C10—C11—C11	178.88 (13)	C24—C19—S2—N2	-118.46 (14)
S1—C10—C11—C11	-2.3 (2)	C20—C19—S2—N2	62.87 (17)
C10—C11—C12—C13	0.5 (3)	C9—C8—N1—C7	68.6 (3)
C11—C11—C12—C13	-179.33 (16)	C9—C8—N1—S1	-89.6 (2)
C11—C12—C13—C14	0.4 (3)	C2—C7—N1—C8	57.3 (2)
C12—C13—C14—C15	-0.9 (3)	C2—C7—N1—S1	-143.80 (14)
C12—C13—C14—C12	178.11 (16)	O1—S1—N1—C8	-12.41 (18)
C11—C10—C15—C14	0.5 (3)	O2—S1—N1—C8	118.67 (16)
S1—C10—C15—C14	-178.43 (13)	C10—S1—N1—C8	-128.51 (16)
C13—C14—C15—C10	0.5 (3)	O1—S1—N1—C7	-170.82 (15)
C12—C14—C15—C10	-178.57 (13)	O2—S1—N1—C7	-39.74 (17)
C5—C6—C16—N2	-111.2 (2)	C10—S1—N1—C7	73.09 (16)
C1—C6—C16—N2	65.6 (2)	C6—C16—N2—C17	60.2 (2)



C24—C19—C20—C21	-2.0 (3)	C6—C16—N2—S2	-111.45 (16)
S2—C19—C20—C21	176.64 (16)	C18—C17—N2—C16	61.4 (2)
C24—C19—C20—C13	177.23 (14)	C18—C17—N2—S2	-126.57 (19)
S2—C19—C20—C13	-4.1 (2)	O4—S2—N2—C16	129.76 (14)
C19—C20—C21—C22	1.2 (3)	O3—S2—N2—C16	-1.35 (16)
C13—C20—C21—C22	-178.04 (17)	C19—S2—N2—C16	-117.35 (14)
C20—C21—C22—C23	0.3 (3)	O4—S2—N2—C17	-41.94 (17)
C21—C22—C23—C24	-1.0 (3)	O3—S2—N2—C17	-173.06 (14)
C21—C22—C23—C14	178.31 (17)	C19—S2—N2—C17	70.94 (16)

*Hydrogen-bond geometry (Å, °)*

<i>D</i> —H $\cdots$ <i>A</i>	<i>D</i> —H	H $\cdots$ <i>A</i>	<i>D</i> $\cdots$ <i>A</i>	<i>D</i> —H $\cdots$ <i>A</i>
C7—H7B $\cdots$ O3 <sup>i</sup>	0.97	2.59	3.511 (2)	158
C17—H17B $\cdots$ O1 <sup>ii</sup>	0.97	2.58	3.516 (3)	164
C24—H24 $\cdots$ C11 <sup>ii</sup>	0.93	2.83	3.738 (2)	166

Symmetry codes: (i)  $-x+1, -y+1, -z+1$ ; (ii)  $-x+1, -y+2, -z+1$ .