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2-[(5-Chloro-2-oxidobenzylidene)-azaniumyl]-2-methylpropane-1,3-diol

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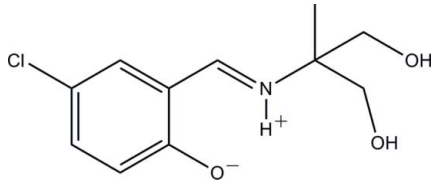
Key indicators: single-crystal X-ray study; T = 298 K; mean σ(C–C) = 0.004 Å; R factor = 0.039; wR factor = 0.076; data-to-parameter ratio = 13.6.

The title compound, $C_{11}H_{14}ClNO_3$, was prepared by the condensation of equimolar quantities of 5-chlorosalicylaldehyde and 2-amino-2-methylpropane-1,3-diol in methanol. In the crystal, it exists in the zwitterionic form, with nominal proton transfer from the phenol group to the imine N atom. This results in the formation of an intramolecular N—H···O hydrogen bond, which generates an S(6) ring. Intermolecular O—H···O hydrogen bonds arise from the hydroxy groups, forming (001) sheets.

Related literature

For a related structure we have reported recently and for background to Schiff bases, see: Wang *et al.* (2011). For standard bond lengths, see: Allen *et al.* (1987).



Experimental

Crystal data

$C_{11}H_{14}ClNO_3$
M_r = 243.68
Orthorhombic, $P2_12_12_1$
a = 6.0019 (16) Å
b = 8.838 (3) Å
c = 21.555 (3) Å
V = 1143.4 (5) Å^3
Z = 4
Mo $K\alpha$ radiation
μ = 0.33 mm^{-1}
T = 298 K
0.13 × 0.12 × 0.10 mm

Data collection

Bruker SMART 1K CCD diffractometer
Absorption correction: multi-scan (*SADABS*; Sheldrick, 1996)
T_{min} = 0.959, T_{max} = 0.968
5744 measured reflections
2105 independent reflections
1745 reflections with $I > 2\sigma(I)$
R_{int} = 0.033

Refinement

$R[F^2 > 2\sigma(F^2)]$ = 0.039
$wR(F^2)$ = 0.076
S = 1.05
2105 reflections
155 parameters
3 restraints
H atoms treated by a mixture of independent and constrained refinement
$\Delta\rho_{max}$ = 0.17 e Å^{-3}
$\Delta\rho_{min}$ = −0.15 e Å^{-3}
Absolute structure: Flack (1983), 842 Friedel pairs
Flack parameter: −0.06 (8)

Table 1
Hydrogen-bond geometry (Å, °).

D—H···A	D—H	H···A	D···A	D—H···A
N1—H1···O1	0.90 (1)	1.84 (2)	2.606 (3)	142 (2)
O4—H4···O3^{i}	0.85 (1)	1.87 (1)	2.680 (2)	160 (3)
O3—H3A···O1^{ii}	0.85 (1)	1.80 (1)	2.648 (2)	176 (3)

Symmetry codes: (i) $x+1, y, z$; (ii) $-x+2, y+\frac{1}{2}, -z+\frac{1}{2}$.

Data collection: *SMART* (Bruker, 2007); cell refinement: *SAINT* (Bruker, 2007); data reduction: *SAINT*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 2008); program(s) used to refine structure: *SHELXL97* (Sheldrick, 2008); molecular graphics: *SHELXTL* (Sheldrick, 2008); software used to prepare material for publication: *SHELXTL*.

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Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: HB6578).

References

Allen, F. H., Kennard, O., Watson, D. G., Brammer, L., Orphen, A. G. & Taylor, R. (1987). *J. Chem. Soc. Perkin Trans. 2*, pp. S1–19.
Bruker (2007). *SMART* and *SAINT*. Bruker AXS Inc., Madison, Wisconsin, USA.
Flack, H. D. (1983). *Acta Cryst.* A**39**, 876–881.
Sheldrick, G. M. (1996). *SADABS*. University of Göttingen, Germany.
Sheldrick, G. M. (2008). *Acta Cryst.* A**64**, 112–122.
Wang, D.-Y., Meng, X.-F. & Ma, J.-J. (2011). *Acta Cryst.* E**67**, o3150.

supporting information

Acta Cryst. (2012). E**68**, o295 [doi:10.1107/S1600536811055784]

2-[(5-Chloro-2-oxidobenzylidene)azaniumyl]-2-methylpropane-1,3-diol

Dong-Yue Wang, Min Liu and Jing-Jun Ma

S1. Comment

Recently, we have reported the structures of a few Schiff base compounds (e.g. Wang *et al.*, 2011). As a continuation of the work, we present here the crystal structure of the title compound, that was obtained as the product of the reaction of 5-chlorosalicylaldehyde with 2-amino-2-methylpropane-1,3-diol in methanol.

In the title compound, Fig. 1, there in an intramolecular N1—H1···O1 hydrogen bond (Table 1). The bond distances and angles are within normal ranges (Allen *et al.*, 1987).

In the crystal of the compound, the Schiff base molecules are linked through intermolecular O—H···O hydrogen bond, to form (001) sheets. (Table 1 and Fig. 2).

S2. Experimental

To a methanol solution (10 ml) of 5-chlorosalicylaldehyde (0.1 mmol, 15.6 mg) and 2-amino-2-methylpropane-1,3-diol (0.1 mmol, 10.5 mg), a few drops of acetic acid were added. The mixture was refluxed for 1 h and then cooled to room temperature. The yellow crystalline solid was collected by filtration, washed with cold methanol and dried in air. Yellow blocks were obtained by slow evaporation of a methanol solution of the product in air.

S3. Refinement

The NH and OH H-atoms were located in a difference Fourier map and were refined with distance restraints, N—H = 0.90 (1) Å, and O—H = 0.85 (1) Å. The C-bound H atoms were positioned geometrically and refined using a riding model, with C—H = 0.93–0.97 Å, with U_{iso}(H) = 1.2U_{eq}(C).

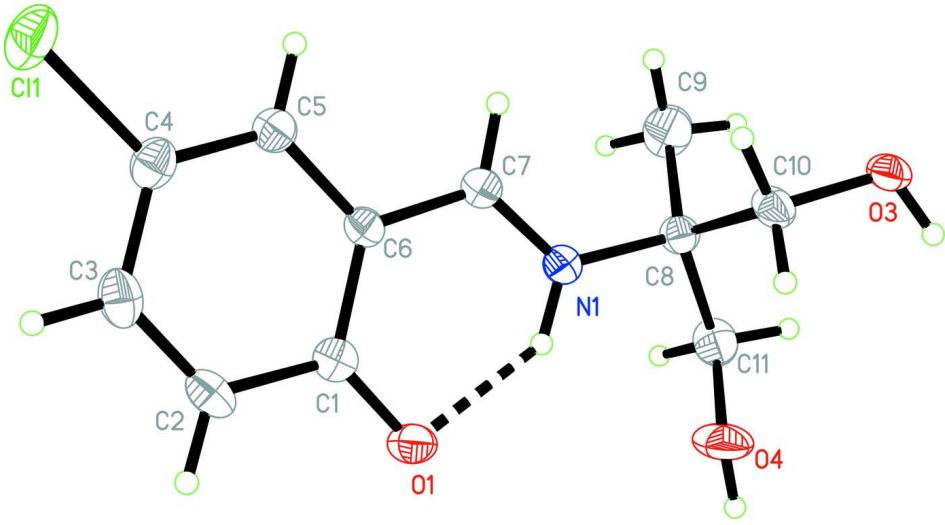

Figure 1

The molecular structure of the title compound, with displacement ellipsoids drawn at the 30% probability level. Intramolecular N—H···O hydrogen bond is drawn as a dashed line.

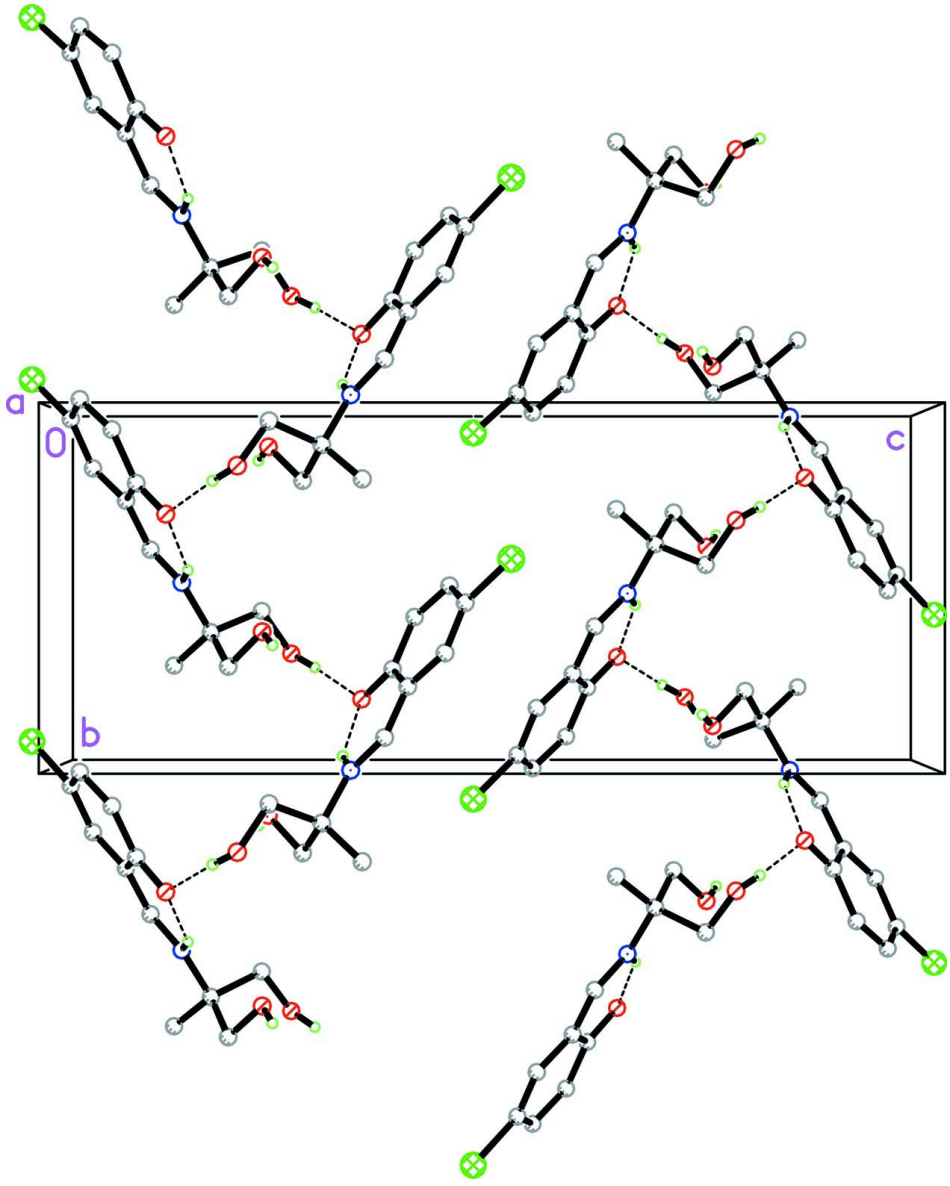

Figure 2

Fragments of (001) sheets moleucles of the title compound, viewed along the *a* axis.

2-[(5-Chloro-2-oxidobenzylidene)azaniumyl]-2-methylpropane-1,3-diol

Crystal data

$C_{11}H_{14}ClNO_3$
$M_r = 243.68$
Orthorhombic, $P2_12_12_1$
$a = 6.0019\ (16)$ Å
$b = 8.838\ (3)$ Å
$c = 21.555\ (3)$ Å
$V = 1143.4\ (5)$ Å^3
$Z = 4$
$F(000) = 512$

$D_x = 1.416$ Mg m^{-3}
Mo $K\alpha$ radiation, $\lambda = 0.71073$ Å
Cell parameters from 1733 reflections
$\theta = 2.5–24.2°$
$\mu = 0.33$ mm^{-1}
$T = 298$ K
Block, yellow
0.13 × 0.12 × 0.10 mm

Data collection

Bruker SMART 1K CCD diffractometer
Radiation source: fine-focus sealed tube
Graphite monochromator
ω scan
Absorption correction: multi-scan (*SADABS*; Sheldrick, 1996)
T_{min} = 0.959, T_{max} = 0.968
5744 measured reflections
2105 independent reflections
1745 reflections with $I > 2\sigma(I)$
R_{int} = 0.033
θ_{max} = 25.5°, θ_{min} = 2.5°
h = −5→7
k = −10→10
l = −26→21

Refinement

Refinement on F^2
Least-squares matrix: full
$R[F^2 > 2\sigma(F^2)]$ = 0.039
$wR(F^2)$ = 0.076
S = 1.05
2105 reflections
155 parameters
3 restraints
Primary atom site location: structure-invariant direct methods
Secondary atom site location: difference Fourier map
Hydrogen site location: inferred from neighbouring sites
H atoms treated by a mixture of independent and constrained refinement
$w = 1/[\sigma^2(F_o^2) + (0.0287P)^2 + 0.113P]$ where $P = (F_o^2 + 2F_c^2)/3$
$(\Delta/\sigma)_{max} < 0.001$
$\Delta\rho_{max}$ = 0.17 e Å^{-3}
$\Delta\rho_{min}$ = −0.15 e Å^{-3}
Absolute structure: Flack (1983), 842 Friedel pairs
Absolute structure parameter: −0.06 (8)

Special details

Geometry. All e.s.d.'s (except the e.s.d. in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell e.s.d.'s are taken into account individually in the estimation of e.s.d.'s in distances, angles and torsion angles; correlations between e.s.d.'s in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell e.s.d.'s is used for estimating e.s.d.'s involving l.s. planes.

Refinement. Refinement of F^2 against ALL reflections. The weighted R-factor wR and goodness of fit S are based on F^2, conventional R-factors R are based on F, with F set to zero for negative F^2. The threshold expression of $F^2 > \sigma(F^2)$ is used only for calculating R-factors(gt) *etc*. and is not relevant to the choice of reflections for refinement. R-factors based on F^2 are statistically about twice as large as those based on F, and R- factors based on ALL data will be even larger.

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (Å^2)

	x	y	z	U_{iso}*/U_{eq}
Cl1	0.67835 (15)	−0.07637 (8)	−0.02077 (3)	0.0613 (2)
N1	0.8677 (3)	0.4851 (2)	0.15414 (9)	0.0356 (5)
O1	1.2095 (3)	0.30421 (19)	0.14623 (7)	0.0422 (4)
O3	0.5214 (3)	0.6832 (2)	0.27003 (8)	0.0462 (5)
O4	1.0926 (3)	0.6167 (2)	0.24842 (11)	0.0609 (6)
C1	1.0897 (4)	0.2197 (3)	0.11037 (11)	0.0350 (6)
C2	1.1701 (5)	0.0789 (3)	0.08645 (11)	0.0442 (6)
H2	1.3107	0.0449	0.0981	0.053*
C3	1.0452 (5)	−0.0064 (3)	0.04693 (12)	0.0470 (7)
H3	1.1032	−0.0965	0.0316	0.056*
C4	0.8320 (5)	0.0388 (3)	0.02898 (11)	0.0412 (6)
C5	0.7447 (4)	0.1709 (3)	0.05062 (10)	0.0366 (6)
H5	0.6031	0.2012	0.0383	0.044*
C6	0.8698 (4)	0.2620 (3)	0.09192 (10)	0.0315 (5)
C7	0.7671 (4)	0.3945 (3)	0.11616 (10)	0.0352 (6)

H7	0.6219	0.4170	0.1042	0.042*
C8	0.7731 (4)	0.6211 (3)	0.18423 (10)	0.0350 (6)
C9	0.6345 (6)	0.7147 (3)	0.13884 (12)	0.0603 (8)
H9A	0.7202	0.7341	0.1021	0.090*
H9B	0.5940	0.8089	0.1579	0.090*
H9C	0.5020	0.6598	0.1280	0.090*
C10	0.6293 (4)	0.5634 (3)	0.23840 (11)	0.0384 (6)
H10A	0.5181	0.4938	0.2226	0.046*
H10B	0.7229	0.5085	0.2674	0.046*
C11	0.9699 (4)	0.7115 (3)	0.20833 (12)	0.0462 (7)
H11A	1.0627	0.7446	0.1741	0.055*
H11B	0.9180	0.8002	0.2306	0.055*
H3A	0.603 (4)	0.720 (3)	0.2983 (10)	0.069*
H4	1.215 (3)	0.651 (3)	0.2621 (12)	0.069*
H1	1.005 (2)	0.452 (3)	0.1646 (11)	0.055*

Atomic displacement parameters (Å²)

	U^{11}	U^{22}	U^{33}	U^{12}	U^{13}	U^{23}
Cl1	0.0825 (6)	0.0435 (4)	0.0579 (4)	−0.0035 (4)	−0.0139 (4)	−0.0124 (3)
N1	0.0328 (13)	0.0364 (11)	0.0375 (11)	0.0047 (10)	−0.0019 (11)	−0.0049 (9)
O1	0.0355 (11)	0.0420 (10)	0.0492 (10)	0.0030 (8)	−0.0089 (9)	0.0012 (8)
O3	0.0260 (10)	0.0535 (11)	0.0593 (13)	0.0027 (9)	−0.0050 (9)	−0.0220 (9)
O4	0.0310 (11)	0.0618 (13)	0.0897 (15)	−0.0013 (10)	−0.0201 (11)	−0.0092 (12)
C1	0.0362 (16)	0.0340 (13)	0.0349 (13)	−0.0015 (11)	0.0023 (11)	0.0063 (11)
C2	0.0376 (15)	0.0406 (14)	0.0543 (15)	0.0119 (14)	0.0050 (14)	0.0048 (13)
C3	0.0590 (19)	0.0330 (13)	0.0491 (16)	0.0070 (14)	0.0076 (15)	−0.0023 (13)
C4	0.0514 (17)	0.0346 (13)	0.0376 (13)	−0.0026 (13)	−0.0027 (13)	−0.0006 (11)
C5	0.0362 (16)	0.0358 (13)	0.0377 (13)	−0.0003 (11)	0.0001 (12)	0.0000 (11)
C6	0.0285 (14)	0.0331 (12)	0.0329 (12)	0.0010 (11)	0.0026 (11)	0.0009 (11)
C7	0.0318 (14)	0.0378 (13)	0.0360 (13)	0.0002 (11)	0.0009 (11)	−0.0011 (11)
C8	0.0351 (15)	0.0286 (12)	0.0412 (13)	0.0039 (11)	−0.0039 (12)	−0.0041 (10)
C9	0.076 (2)	0.0487 (16)	0.0564 (17)	0.0194 (16)	−0.0117 (16)	0.0018 (14)
C10	0.0296 (14)	0.0332 (12)	0.0524 (15)	−0.0012 (12)	0.0023 (12)	−0.0090 (12)
C11	0.0412 (17)	0.0391 (14)	0.0585 (17)	−0.0131 (13)	0.0074 (15)	−0.0071 (13)

Geometric parameters (Å, º)

Cl1—C4	1.743 (2)	C4—C5	1.362 (3)
N1—C7	1.294 (3)	C5—C6	1.416 (3)
N1—C8	1.479 (3)	C5—H5	0.9300
N1—H1	0.903 (10)	C6—C7	1.423 (3)
O1—C1	1.293 (3)	C7—H7	0.9300
O3—C10	1.416 (3)	C8—C11	1.517 (3)
O3—H3A	0.847 (10)	C8—C9	1.528 (3)
O4—C11	1.411 (3)	C8—C10	1.539 (3)
O4—H4	0.847 (10)	C9—H9A	0.9600
C1—C6	1.428 (3)	C9—H9B	0.9600

C1—C2	1.431 (3)	C9—H9C	0.9600
C2—C3	1.362 (3)	C10—H10A	0.9700
C2—H2	0.9300	C10—H10B	0.9700
C3—C4	1.395 (4)	C11—H11A	0.9700
C3—H3	0.9300	C11—H11B	0.9700
C7—N1—C8	126.9 (2)	C6—C7—H7	118.7
C7—N1—H1	112.7 (17)	N1—C8—C11	106.2 (2)
C8—N1—H1	120.1 (17)	N1—C8—C9	111.61 (19)
C10—O3—H3A	112 (2)	C11—C8—C9	111.0 (2)
C11—O4—H4	117 (2)	N1—C8—C10	106.20 (18)
O1—C1—C6	121.9 (2)	C11—C8—C10	110.55 (19)
O1—C1—C2	122.0 (2)	C9—C8—C10	111.1 (2)
C6—C1—C2	116.1 (2)	C8—C9—H9A	109.5
C3—C2—C1	121.4 (2)	C8—C9—H9B	109.5
C3—C2—H2	119.3	H9A—C9—H9B	109.5
C1—C2—H2	119.3	C8—C9—H9C	109.5
C2—C3—C4	121.3 (2)	H9A—C9—H9C	109.5
C2—C3—H3	119.3	H9B—C9—H9C	109.5
C4—C3—H3	119.3	O3—C10—C8	111.98 (19)
C5—C4—C3	120.2 (2)	O3—C10—H10A	109.2
C5—C4—Cl1	120.5 (2)	C8—C10—H10A	109.2
C3—C4—Cl1	119.27 (19)	O3—C10—H10B	109.2
C4—C5—C6	119.9 (2)	C8—C10—H10B	109.2
C4—C5—H5	120.0	H10A—C10—H10B	107.9
C6—C5—H5	120.0	O4—C11—C8	107.65 (19)
C5—C6—C7	118.0 (2)	O4—C11—H11A	110.2
C5—C6—C1	121.1 (2)	C8—C11—H11A	110.2
C7—C6—C1	120.9 (2)	O4—C11—H11B	110.2
N1—C7—C6	122.6 (2)	C8—C11—H11B	110.2
N1—C7—H7	118.7	H11A—C11—H11B	108.5

Hydrogen-bond geometry (Å, º)

D—H···*A*	*D*—H	H···*A*	*D*···*A*	*D*—H···*A*
N1—H1···O1	0.90 (1)	1.84 (2)	2.606 (3)	142 (2)
O4—H4···O3^{i}	0.85 (1)	1.87 (1)	2.680 (2)	160 (3)
O3—H3*A*···O1^{ii}	0.85 (1)	1.80 (1)	2.648 (2)	176 (3)

Symmetry codes: (i) *x*+1, *y*, *z*; (ii) −*x*+2, *y*+1/2, −*z*+1/2.