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2-[(4,6-Dimethoxypyrimidin-2-yl)oxy]-benzaldehyde

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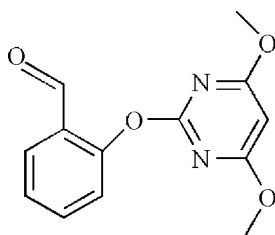
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 Key indicators: single-crystal X-ray study; $T = 293$ K; mean $\sigma(\text{C}-\text{C}) = 0.004$ Å; R factor = 0.046; wR factor = 0.131; data-to-parameter ratio = 12.9.

In the title compound, $\text{C}_{13}\text{H}_{12}\text{N}_2\text{O}_4$, the dihedral angle between the benzene and pyrimidine rings is 55.57 (13)°. The carbonyl group and the two methoxyl groups are approximately coplanar with the benzene ring and pyrimidine ring; the $\text{C}-\text{C}-\text{C}-\text{O}$, $\text{C}-\text{O}-\text{C}-\text{N}$ and $\text{C}-\text{O}-\text{C}-\text{C}$ torsion angles being -6.1 (5), -4.8 (4) and 179.9 (3)°, respectively. In the crystal, molecules are linked *via* $\text{C}-\text{H}\cdots\text{O}$ interactions, forming chains propagating along $[110]$.

Related literature

For the synthesis of the title compound, see: Yang & Lu (2010).



Experimental

Crystal data

 $\text{C}_{13}\text{H}_{12}\text{N}_2\text{O}_4$
 $M_r = 260.25$

 Monoclinic, $P2_1$
 $a = 3.9920$ (8) Å
 $b = 7.3670$ (15) Å
 $c = 20.885$ (4) Å
 $\beta = 94.87$ (3)°
 $V = 612.0$ (2) Å³
 $Z = 2$
 Mo $K\alpha$ radiation
 $\mu = 0.11$ mm⁻¹
 $T = 293$ K
 $0.30 \times 0.20 \times 0.10$ mm

Data collection

 Enraf–Nonius CAD-4
 diffractometer
 Absorption correction: ψ scan
 (SADABS; Sheldrick, 1996)
 $T_{\min} = 0.969$, $T_{\max} = 0.989$
 2563 measured reflections

 2227 independent reflections
 1790 reflections with $I > 2\sigma(I)$
 $R_{\text{int}} = 0.024$
 3 standard reflections every 200
 reflections
 intensity decay: 1%

Refinement

 $R[F^2 > 2\sigma(F^2)] = 0.046$
 $wR(F^2) = 0.131$
 $S = 1.00$
 2227 reflections
 172 parameters

 1 restraint
 H-atom parameters constrained
 $\Delta\rho_{\text{max}} = 0.14$ e Å⁻³
 $\Delta\rho_{\text{min}} = -0.14$ e Å⁻³
Table 1

Hydrogen-bond geometry (Å, °).

$D-H\cdots A$	$D-H$	$H\cdots A$	$D\cdots A$	$D-H\cdots A$
$\text{C}2-\text{H}2B\cdots\text{O}1^i$	0.93	2.54	3.400 (4)	154

 Symmetry code: (i) $x - 1, y + 1, z$.

Data collection: *CAD-4 EXPRESS* (Enraf–Nonius, 1994); cell refinement: *CAD-4 EXPRESS*; data reduction: *XCAD4* (Harms & Wocadlo, 1995); program(s) used to solve structure: *SHELXS97* (Sheldrick, 2008); program(s) used to refine structure: *SHELXL97* (Sheldrick, 2008); molecular graphics: *SHELXTL* (Sheldrick, 2008); software used to prepare material for publication: *SHELXTL*.

We are grateful to Professor Hua-Qin Wang for measuring the data and the Center of Testing and Analysis, Nanjing University, for financial support.

Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: SU2359).

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supporting information

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2-[(4,6-Dimethoxypyrimidin-2-yl)oxy]benzaldehyde

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S1. Comment

The title compound is an important organic intermediate for the synthesis of 2-pyrimidine-oxy-*N*-aryl benzyl amine derivatives, which are important compounds for new pesticides. In the process of synthesising one such derivative we obtained crystals of the title compound, and we report herein on its crystal structure.

As illustrated in Fig. 1, the molecular structure of the title compound is not planar, the dihedral angle between the (C1—C6) benzene and the (N1/N2,C8-C10) pyrimidine rings is 55.57 (13)°. The carbonyl group and the two methoxyl groups are approximately coplanar with the benzene ring and pyrimidine ring, as shown by the torsion angles C4—C5—C7—O1 = -6.1 (5)°, C13—O4—C11—N2 = -4.8 (4)° and C12—O3—C9—C10 = 179.9 (3)°.

In the crystal, there is a C-H...O interaction present (Table 1), that results in the formation of chains that run along direction [110].

S2. Experimental

The title compound was synthesized according to the published procedure (Yang & Lu, 2010). A solution of 12.2 g salicylaldehyde, 21.8 g of 2,6-dimethoxy-4-(Methylsulfonyl) Pyrimidine, and 41.4 g of K₂CO₃, in 150 ml acetonitrile, was heated to 220 K for 4 h. The solution was then filtered, and the filtrates were concentrated under reduced pressure. Colourless block-like crystals of the title compound, suitable for X-ray diffraction, were obtained by slow evaporation of the solvent [Yield 86%].

S3. Refinement

The C-bound H atoms were placed in calculated positions and treated as riding atoms: C—H = 0.93 and 0.96 Å for CH and CH₃ and H atoms, respectively, with $U_{\text{iso}}(\text{H}) = k \times U_{\text{eq}}(\text{C})$, where $k = 1.5$ for CH₃ H-atoms and $k = 1.2$ for all other H-atoms. In the final cycles of refinement, in the absence of significant anomalous scattering effects, 1023 Friedel pairs were merged and $\Delta f''$ set to zero.

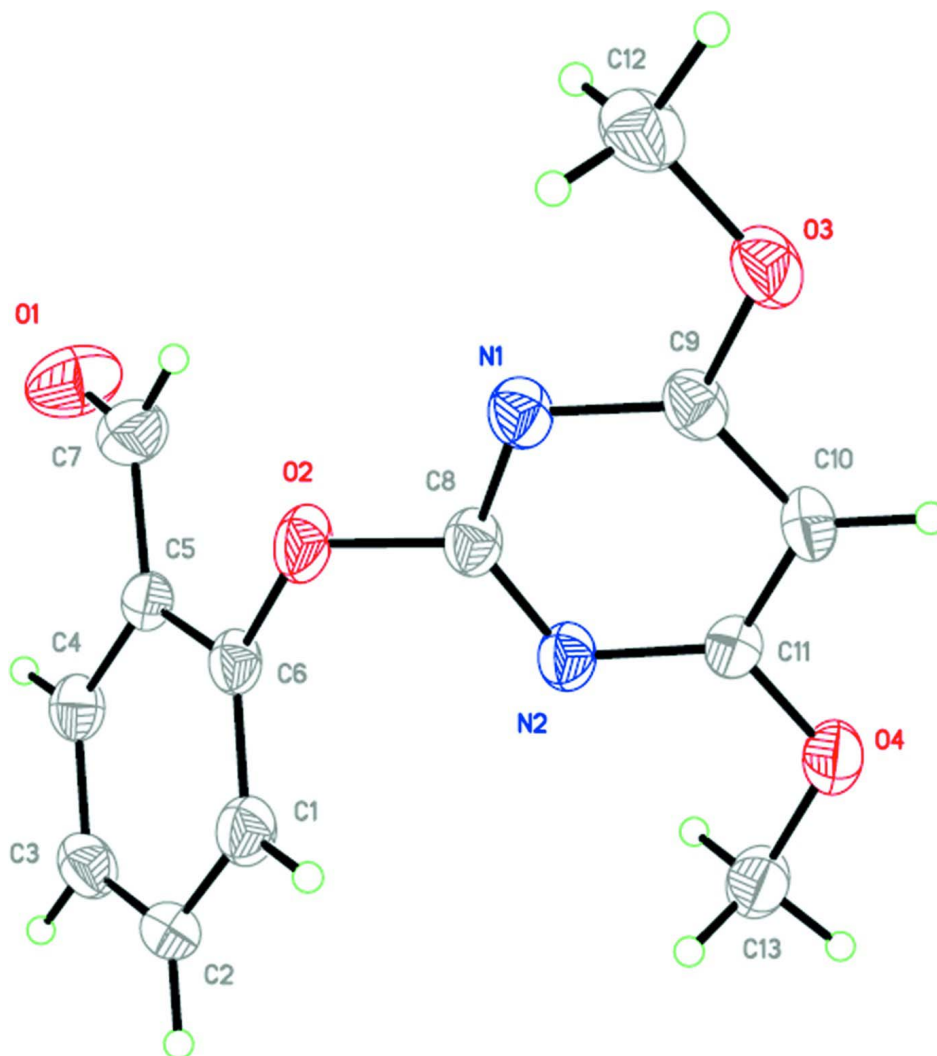


Figure 1

The molecular structure of the title compound, showing the atomic numbering scheme and displacement ellipsoids drawn at the 30% probability level.

2-[(4,6-Dimethoxypyrimidin-2-yl)oxy]benzaldehyde

Crystal data

$C_{13}H_{12}N_2O_4$

$M_r = 260.25$

Monoclinic, $P2_1$

Hall symbol: P 2yb

$a = 3.9920$ (8) Å

$b = 7.3670$ (15) Å

$c = 20.885$ (4) Å

$\beta = 94.87$ (3)°

$V = 612.0$ (2) Å³

$Z = 2$

$F(000) = 272$

$D_x = 1.412$ Mg m⁻³

Mo $K\alpha$ radiation, $\lambda = 0.71073$ Å

Cell parameters from 25 reflections

$\theta = 9\text{--}13^\circ$

$\mu = 0.11$ mm⁻¹

$T = 293$ K

Block, colourless

$0.30 \times 0.20 \times 0.10$ mm

Data collection

Enraf–Nonius CAD-4
diffractometer

Radiation source: fine-focus sealed tube

Graphite monochromator

$\omega/2\theta$ scans

Absorption correction: ψ scan
(SADABS; Sheldrick, 1996)

$T_{\min} = 0.969$, $T_{\max} = 0.989$

2563 measured reflections

2227 independent reflections

1790 reflections with $I > 2\sigma(I)$

$R_{\text{int}} = 0.024$

$\theta_{\max} = 25.4^\circ$, $\theta_{\min} = 2.0^\circ$

$h = 0 \rightarrow 4$

$k = -8 \rightarrow 8$

$l = -25 \rightarrow 25$

3 standard reflections every 200 reflections

intensity decay: 1%

Refinement

Refinement on F^2

Least-squares matrix: full

$R[F^2 > 2\sigma(F^2)] = 0.046$

$wR(F^2) = 0.131$

$S = 1.00$

2227 reflections

172 parameters

1 restraint

Primary atom site location: structure-invariant
direct methods

Secondary atom site location: difference Fourier
map

Hydrogen site location: inferred from
neighbouring sites

H-atom parameters constrained

$w = 1/[\sigma^2(F_o^2) + (0.085P)^2]$

where $P = (F_o^2 + 2F_c^2)/3$

$(\Delta/\sigma)_{\max} < 0.001$

$\Delta\rho_{\max} = 0.14 \text{ e } \text{\AA}^{-3}$

$\Delta\rho_{\min} = -0.14 \text{ e } \text{\AA}^{-3}$

Special details

Geometry. Bond distances, angles etc. have been calculated using the rounded fractional coordinates. All su's are estimated from the variances of the (full) variance-covariance matrix. The cell esds are taken into account in the estimation of distances, angles and torsion angles

Refinement. Refinement of F^2 against ALL reflections. The weighted R -factor wR and goodness of fit S are based on F^2 , conventional R -factors R are based on F , with F set to zero for negative F^2 . The threshold expression of $F^2 > \sigma(F^2)$ is used only for calculating R -factors(gt) etc. and is not relevant to the choice of reflections for refinement. R -factors based on F^2 are statistically about twice as large as those based on F , and R -factors based on ALL data will be even larger.

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (\AA^2)

	x	y	z	$U_{\text{iso}}^*/U_{\text{eq}}$
O1	1.4553 (7)	0.5127 (4)	0.39674 (15)	0.0925 (11)
O2	0.8453 (6)	0.7927 (3)	0.27202 (8)	0.0581 (7)
O3	0.7500 (5)	0.7425 (3)	0.05537 (8)	0.0591 (8)
O4	1.2359 (6)	1.2581 (3)	0.15175 (9)	0.0598 (8)
N1	0.7955 (6)	0.7671 (3)	0.16578 (10)	0.0457 (8)
N2	1.0511 (6)	1.0293 (3)	0.21505 (10)	0.0460 (7)
C1	0.8229 (7)	1.0480 (4)	0.34509 (13)	0.0503 (9)
C2	0.9064 (7)	1.1207 (4)	0.40473 (14)	0.0515 (10)
C3	1.1033 (7)	1.0252 (4)	0.44989 (14)	0.0518 (10)
C4	1.2202 (7)	0.8574 (4)	0.43577 (13)	0.0460 (9)
C5	1.1421 (7)	0.7798 (4)	0.37569 (12)	0.0413 (8)
C6	0.9419 (7)	0.8792 (4)	0.33036 (12)	0.0411 (8)
C7	1.2647 (9)	0.5998 (4)	0.36179 (17)	0.0638 (11)
C8	0.9023 (7)	0.8697 (4)	0.21501 (12)	0.0432 (9)
C9	0.8461 (7)	0.8376 (4)	0.10859 (12)	0.0444 (9)
C10	0.9930 (7)	1.0035 (4)	0.10102 (13)	0.0501 (10)

C11	1.0929 (7)	1.0950 (4)	0.15650 (13)	0.0433 (8)
C12	0.5970 (8)	0.5691 (5)	0.06309 (14)	0.0629 (11)
C13	1.3137 (9)	1.3609 (4)	0.20875 (15)	0.0583 (11)
H1B	0.68640	1.11240	0.31480	0.0600*
H2B	0.82900	1.23560	0.41460	0.0620*
H3A	1.15740	1.07510	0.49040	0.0620*
H4A	1.35410	0.79390	0.46680	0.0550*
H7A	1.18890	0.54810	0.32260	0.0760*
H10A	1.02300	1.05120	0.06070	0.0600*
H12A	0.54180	0.51540	0.02160	0.0940*
H12B	0.39590	0.58390	0.08470	0.0940*
H12C	0.75070	0.49160	0.08810	0.0940*
H13A	1.41400	1.47410	0.19800	0.0880*
H13B	1.46830	1.29400	0.23750	0.0880*
H13C	1.11130	1.38390	0.22920	0.0880*

Atomic displacement parameters (Å²)

	U^{11}	U^{22}	U^{33}	U^{12}	U^{13}	U^{23}
O1	0.1009 (19)	0.0614 (16)	0.112 (2)	0.0237 (15)	-0.0101 (16)	0.0028 (15)
O2	0.0881 (15)	0.0545 (12)	0.0307 (10)	-0.0212 (11)	-0.0003 (9)	0.0054 (8)
O3	0.0734 (15)	0.0684 (14)	0.0350 (11)	-0.0096 (12)	0.0019 (10)	-0.0057 (9)
O4	0.0798 (15)	0.0533 (12)	0.0477 (12)	-0.0107 (12)	0.0134 (11)	0.0053 (9)
N1	0.0502 (13)	0.0510 (14)	0.0349 (12)	-0.0028 (11)	-0.0023 (9)	0.0010 (9)
N2	0.0527 (13)	0.0485 (13)	0.0366 (12)	-0.0027 (12)	0.0026 (10)	0.0050 (10)
C1	0.0556 (17)	0.0529 (17)	0.0426 (15)	0.0039 (14)	0.0052 (13)	0.0098 (13)
C2	0.0619 (18)	0.0447 (16)	0.0505 (17)	0.0076 (14)	0.0196 (14)	0.0002 (12)
C3	0.0563 (16)	0.0596 (19)	0.0405 (15)	-0.0086 (16)	0.0102 (12)	-0.0091 (13)
C4	0.0465 (15)	0.0540 (17)	0.0369 (14)	-0.0036 (13)	-0.0002 (11)	0.0059 (12)
C5	0.0481 (15)	0.0390 (14)	0.0375 (14)	-0.0038 (12)	0.0075 (11)	0.0036 (11)
C6	0.0477 (14)	0.0454 (14)	0.0299 (13)	-0.0074 (13)	0.0022 (11)	0.0029 (11)
C7	0.080 (2)	0.0497 (18)	0.0615 (19)	0.0067 (18)	0.0051 (16)	0.0043 (15)
C8	0.0482 (15)	0.0474 (17)	0.0333 (14)	-0.0010 (14)	-0.0007 (11)	0.0052 (11)
C9	0.0455 (14)	0.0550 (17)	0.0320 (14)	0.0048 (13)	-0.0014 (11)	-0.0005 (12)
C10	0.0630 (18)	0.0565 (18)	0.0317 (14)	0.0012 (15)	0.0092 (13)	0.0071 (12)
C11	0.0485 (15)	0.0397 (14)	0.0426 (15)	0.0008 (13)	0.0085 (11)	0.0046 (11)
C12	0.066 (2)	0.074 (2)	0.0476 (17)	-0.0141 (17)	-0.0017 (15)	-0.0155 (16)
C13	0.072 (2)	0.0521 (18)	0.0505 (17)	-0.0099 (17)	0.0037 (15)	0.0043 (13)

Geometric parameters (Å, °)

O1—C7	1.196 (5)	C5—C6	1.394 (4)
O2—C6	1.400 (3)	C5—C7	1.451 (4)
O2—C8	1.355 (3)	C9—C10	1.371 (4)
O3—C9	1.342 (3)	C10—C11	1.370 (4)
O3—C12	1.431 (4)	C1—H1B	0.9300
O4—C11	1.338 (4)	C2—H2B	0.9300
O4—C13	1.423 (4)	C3—H3A	0.9300

N1—C8	1.318 (3)	C4—H4A	0.9300
N1—C9	1.334 (3)	C7—H7A	0.9300
N2—C8	1.317 (4)	C10—H10A	0.9300
N2—C11	1.339 (3)	C12—H12A	0.9600
C1—C2	1.371 (4)	C12—H12B	0.9600
C1—C6	1.375 (4)	C12—H12C	0.9600
C2—C3	1.370 (4)	C13—H13A	0.9600
C3—C4	1.362 (4)	C13—H13B	0.9600
C4—C5	1.390 (4)	C13—H13C	0.9600
C6—O2—C8	121.3 (2)	N2—C11—C10	123.0 (3)
C9—O3—C12	117.9 (2)	C2—C1—H1B	120.00
C11—O4—C13	118.7 (2)	C6—C1—H1B	120.00
C8—N1—C9	114.3 (2)	C1—C2—H2B	120.00
C8—N2—C11	114.4 (2)	C3—C2—H2B	120.00
C2—C1—C6	119.6 (3)	C2—C3—H3A	120.00
C1—C2—C3	120.4 (3)	C4—C3—H3A	120.00
C2—C3—C4	120.3 (3)	C3—C4—H4A	120.00
C3—C4—C5	120.9 (3)	C5—C4—H4A	120.00
C4—C5—C6	118.0 (3)	O1—C7—H7A	117.00
C4—C5—C7	120.2 (3)	C5—C7—H7A	117.00
C6—C5—C7	121.9 (3)	C9—C10—H10A	122.00
O2—C6—C1	122.1 (2)	C11—C10—H10A	122.00
O2—C6—C5	116.9 (3)	O3—C12—H12A	109.00
C1—C6—C5	120.8 (2)	O3—C12—H12B	109.00
O1—C7—C5	125.2 (3)	O3—C12—H12C	109.00
O2—C8—N1	112.2 (2)	H12A—C12—H12B	109.00
O2—C8—N2	118.8 (2)	H12A—C12—H12C	109.00
N1—C8—N2	129.0 (2)	H12B—C12—H12C	109.00
O3—C9—N1	119.0 (3)	O4—C13—H13A	109.00
O3—C9—C10	117.7 (2)	O4—C13—H13B	109.00
N1—C9—C10	123.3 (2)	O4—C13—H13C	110.00
C9—C10—C11	116.0 (3)	H13A—C13—H13B	109.00
O4—C11—N2	118.7 (2)	H13A—C13—H13C	110.00
O4—C11—C10	118.3 (2)	H13B—C13—H13C	109.00
C8—O2—C6—C5	-126.4 (3)	C2—C1—C6—O2	175.3 (3)
C6—O2—C8—N1	179.6 (2)	C6—C1—C2—C3	-0.9 (4)
C6—O2—C8—N2	-0.1 (4)	C1—C2—C3—C4	0.5 (4)
C8—O2—C6—C1	59.2 (4)	C2—C3—C4—C5	-0.2 (4)
C12—O3—C9—C10	179.9 (3)	C3—C4—C5—C6	0.3 (4)
C12—O3—C9—N1	0.0 (4)	C3—C4—C5—C7	-179.1 (3)
C13—O4—C11—C10	174.6 (3)	C4—C5—C6—O2	-175.2 (3)
C13—O4—C11—N2	-4.8 (4)	C6—C5—C7—O1	174.6 (3)
C9—N1—C8—N2	-0.4 (4)	C7—C5—C6—C1	178.6 (3)
C8—N1—C9—O3	179.2 (2)	C4—C5—C7—O1	-6.1 (5)
C8—N1—C9—C10	-0.7 (4)	C4—C5—C6—C1	-0.7 (4)
C9—N1—C8—O2	179.9 (2)	C7—C5—C6—O2	4.1 (4)

C11—N2—C8—N1	1.0 (4)	O3—C9—C10—C11	-178.9 (3)
C8—N2—C11—C10	-0.7 (4)	N1—C9—C10—C11	1.0 (4)
C11—N2—C8—O2	-179.3 (3)	C9—C10—C11—N2	-0.2 (4)
C8—N2—C11—O4	178.7 (3)	C9—C10—C11—O4	-179.6 (3)
C2—C1—C6—C5	1.0 (4)		

Hydrogen-bond geometry (Å, °)

<i>D</i> —H \cdots <i>A</i>	<i>D</i> —H	H \cdots <i>A</i>	<i>D</i> \cdots <i>A</i>	<i>D</i> —H \cdots <i>A</i>
C2—H2B \cdots O1 ⁱ	0.93	2.54	3.400 (4)	154

Symmetry code: (i) $x-1, y+1, z$.