

Acta Crystallographica Section E

Structure Reports

Online

ISSN 1600-5368

(1-Oxo-2,6,7-trioxa-1-phosfabicyclo-[2.2.2]octan-4-yl)methyl 4-methylbenzenesulfonate

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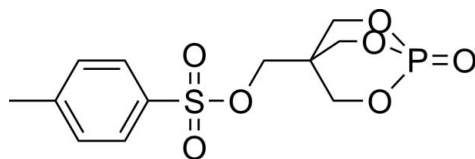
Received 13 April 2012; accepted 17 April 2012

 Key indicators: single-crystal X-ray study; $T = 113$ K; mean $\sigma(\text{C}-\text{C}) = 0.002$ Å; R factor = 0.034; wR factor = 0.095; data-to-parameter ratio = 17.6.

In the title compound, $\text{C}_{12}\text{H}_{15}\text{O}_7\text{PS}$, the P atom has a distorted tetrahedral environment. The P—O—C—C torsion angles deviate significantly from zero [average = 12.0 (3)°], indicating that the bicyclic $\text{OP}(\text{OCH}_2)_3\text{C}$ cage is strained. In the crystal, weak C—H···O interactions consolidate the packing.

Related literature

For related structures, see: Miu *et al.* (1991); Sheng & He (2006); Guo & Zang (2007). For applications of caged bicyclic phosphates and *p*-toluenesulfonates, see: Li *et al.* (2000); Yachi *et al.* (1989); Spungin *et al.* (1992).



Experimental

Crystal data

$\text{C}_{12}\text{H}_{15}\text{O}_7\text{PS}$
 $M_r = 334.27$
 Monoclinic, $P2_1/c$
 $a = 5.8884$ (17) Å
 $b = 19.440$ (5) Å
 $c = 12.469$ (4) Å
 $\beta = 100.614$ (4)°

$V = 1402.8$ (7) Å³
 $Z = 4$
 Mo $K\alpha$ radiation
 $\mu = 0.38$ mm⁻¹
 $T = 113$ K
 $0.20 \times 0.18 \times 0.12$ mm

Data collection

Rigaku Saturn CCD area-detector diffractometer
 Absorption correction: multi-scan (*CrystalClear*; Rigaku/MS, 2005)
 $T_{\min} = 0.929$, $T_{\max} = 0.956$
 17318 measured reflections
 3353 independent reflections
 2808 reflections with $I > 2\sigma(I)$
 $R_{\text{int}} = 0.042$

Refinement

$R[F^2 > 2\sigma(F^2)] = 0.034$
 $wR(F^2) = 0.095$
 $S = 1.03$
 3353 reflections
 191 parameters
 H-atom parameters constrained
 $\Delta\rho_{\max} = 0.41$ e Å⁻³
 $\Delta\rho_{\min} = -0.36$ e Å⁻³

Table 1

Hydrogen-bond geometry (Å, °).

D—H···A	D—H	H···A	D···A	D—H···A
C7—H7A···O5 ⁱ	0.98	2.59	3.288 (2)	128
C8—H8A···O1 ⁱⁱ	0.99	2.38	3.180 (2)	138
C10—H10A···O7 ⁱⁱⁱ	0.99	2.27	3.211 (2)	158
C10—H10B···O2 ^{iv}	0.99	2.44	3.348 (2)	152

Symmetry codes: (i) $x, y, z + 1$; (ii) $x - 1, y, z$; (iii) $x, -y + \frac{1}{2}, z + \frac{1}{2}$; (iv) $-x + 1, -y + 1, -z + 1$.

Data collection: *CrystalClear* (Rigaku/MS, 2005); cell refinement: *CrystalClear*; data reduction: *CrystalClear*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 2008); program(s) used to refine structure: *SHELXL97* (Sheldrick, 2008); molecular graphics: *CrystalStructure* (Rigaku/MS, 2005); software used to prepare material for publication: *CrystalStructure*.

This work was supported by the Natural Science Foundation of Henan Province, China (grant No. 082300420110) and the Natural Science Foundation of Henan Province Education Department, China (grant No. 2007150036).

Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: CV5286).

References

- Guo, M.-L. & Zang, H.-J. (2007). *Acta Cryst.* **E63**, o1967–o1968.
 Li, X., Ou, Y.-X., Zhang, Y.-H. & Lian, D.-J. (2000). *Chin. Chem. Lett.* **11**, 887–890.
 Miu, F. M., Liu, X. L., Li, Y. G., Wang, J. J., Liu, Y. S., Bao, J. C., Cao, J. H. & Zhou, W. (1991). *Acta Chim. Sin.* **49**, 870–875.
 Rigaku/MS (2005). *CrystalClear* and *CrystalStructure*. Rigaku/MS, The Woodlands, Texas, USA.
 Sheldrick, G. M. (2008). *Acta Cryst.* **A64**, 112–122.
 Sheng, X.-J. & He, H.-W. (2006). *Acta Cryst.* **E62**, o4398–o4399.
 Spungin, B., Levinshal, T., Rubenstein, S. & Breitbart, H. (1992). *FEBS Lett.* **311**, 155–160.
 Yachi, K., Sugiyama, Y., Sawada, Y., Iga, T., Ikeda, Y., Toda, G. & Hanano, M. (1989). *Biochim. Biophys. Acta*, **978**, 1–7.

supporting information

Acta Cryst. (2012). E68, o1473 [doi:10.1107/S1600536812016674]

(1-Oxo-2,6,7-trioxa-1-phosphabicyclo[2.2.2]octan-4-yl)methyl 4-methylbenzenesulfonate**Xu-Feng Hou and Jin-Long Yan****S1. Comment**

Caged bicyclic phosphates are widely used as flame retardants or pesticides (Li *et al.*, 2000). They have been studied in the context of hydrogen-bond patterns (Guo & Zang, 2007). *p*-Toluenesulfonates are also used in monitoring the merging of lipids (Yachi *et al.*, 1989), studying membrane fusion during acrosome reaction (Spungin *et al.*, 1992). We report here the crystal structure of the title compound (I).

In (I) (Fig. 1), all bond lengths and angles are normal and correspond to those observed in the related compounds (Miu *et al.*, 1991; Sheng & He, 2006; Guo & Zang, 2007). The P—O—C—C torsion angles deviate significantly from zero [averaged $-12.0(3)^\circ$] indicating that bicyclic OP(OCH₂)₃C cage is strained. In the crystal, weak C—H \cdots O interactions (Table 1) consolidate the packing.

S2. Experimental

The title compound was obtained in the reaction of 4-(hydroxymethyl)-2,6,7-trioxa-1-phosphabicyclo[2.2.2]octane 1-oxide (5 mmol) and 4-toluenesulfonyl chloride(4.7 mmol) in refluxing acetonitrile (50 ml). The solvent was evaporated *in vacuo*. The title compound was recrystallized from ethanol and single crystals of (I) were obtained by slow evaporation.

S3. Refinement

All H atoms were placed in calculated positions, with C—H = 0.95 - 0.99 Å, and included in the final cycles of refinement using a riding model, with $U_{\text{iso}}(\text{H}) = 1.2\text{-}1.5 U_{\text{eq}}(\text{C})$.

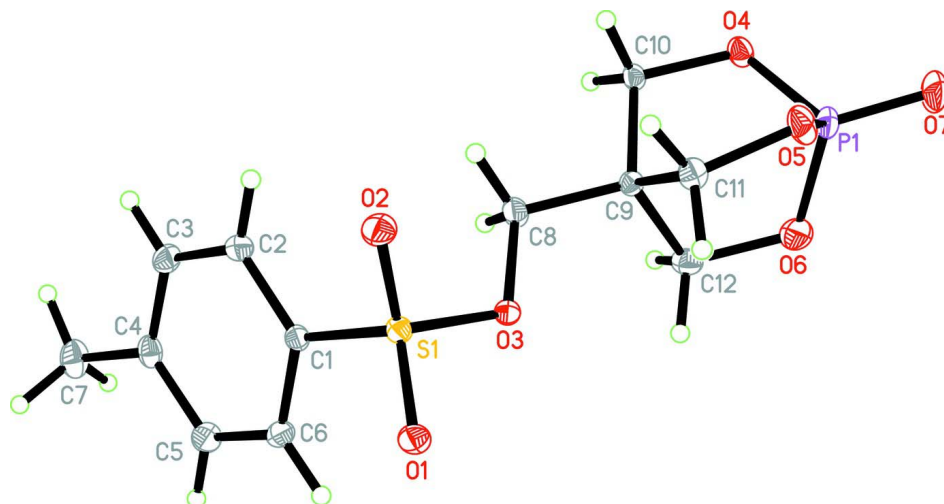


Figure 1

The molecular structure of (I) showing the atomic numbering and 30% probability displacement ellipsoids.

(1-Oxo-2,6,7-trioxa-1-phospha-bicyclo[2.2.2]octan-4-yl)methyl 4-methylbenzenesulfonate

Crystal data

$C_{12}H_{15}O_7PS$

$M_r = 334.27$

Monoclinic, $P2_1/c$

$a = 5.8884$ (17) Å

$b = 19.440$ (5) Å

$c = 12.469$ (4) Å

$\beta = 100.614$ (4)°

$V = 1402.8$ (7) Å³

$Z = 4$

$F(000) = 696$

$D_x = 1.583$ Mg m⁻³

Mo $K\alpha$ radiation, $\lambda = 0.71073$ Å

Cell parameters from 4690 reflections

$\theta = 1.0$ – 27.9 °

$\mu = 0.38$ mm⁻¹

$T = 113$ K

Prism, colorless

$0.20 \times 0.18 \times 0.12$ mm

Data collection

Rigaku Saturn CCD area-detector
diffractometer

Radiation source: rotating anode

Multilayer monochromator

Detector resolution: 14.63 pixels mm⁻¹

ω and φ scans

Absorption correction: multi-scan

(*CrystalClear*; Rigaku/MS, 2005)

$T_{\min} = 0.929$, $T_{\max} = 0.956$

17318 measured reflections

3353 independent reflections

2808 reflections with $I > 2\sigma(I)$

$R_{\text{int}} = 0.042$

$\theta_{\max} = 27.9$ °, $\theta_{\min} = 2.0$ °

$h = -7 \rightarrow 7$

$k = -25 \rightarrow 25$

$l = -16 \rightarrow 16$

Refinement

Refinement on F^2

Least-squares matrix: full

$R[F^2 > 2\sigma(F^2)] = 0.034$

$wR(F^2) = 0.095$

$S = 1.03$

3353 reflections

191 parameters

0 restraints

Primary atom site location: structure-invariant
direct methods

Secondary atom site location: difference Fourier
map

Hydrogen site location: inferred from
neighbouring sites

H-atom parameters constrained

$w = 1/[\sigma^2(F_o^2) + (0.056P)^2 + 0.0836P]$

where $P = (F_o^2 + 2F_c^2)/3$

$(\Delta/\sigma)_{\max} = 0.001$

$\Delta\rho_{\max} = 0.41$ e Å⁻³

$\Delta\rho_{\min} = -0.36$ e Å⁻³

Special details

Geometry. All e.s.d.'s (except the e.s.d. in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell e.s.d.'s are taken into account individually in the estimation of e.s.d.'s in distances, angles and torsion angles; correlations between e.s.d.'s in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell e.s.d.'s is used for estimating e.s.d.'s involving l.s. planes.

Refinement. Refinement of F^2 against ALL reflections. The weighted R -factor wR and goodness of fit S are based on F^2 , conventional R -factors R are based on F , with F set to zero for negative F^2 . The threshold expression of $F^2 > \sigma(F^2)$ is used only for calculating R -factors(gt) *etc.* and is not relevant to the choice of reflections for refinement. R -factors based on F^2 are statistically about twice as large as those based on F , and R -factors based on ALL data will be even larger.

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (\AA^2)

	<i>x</i>	<i>y</i>	<i>z</i>	$U_{\text{iso}}^*/U_{\text{eq}}$
S1	0.99063 (6)	0.451050 (18)	0.67038 (3)	0.01842 (11)
P1	0.39289 (8)	0.28254 (2)	0.29030 (3)	0.02821 (13)
O1	1.22894 (17)	0.45267 (5)	0.66257 (9)	0.0236 (2)
O2	0.84692 (19)	0.50895 (5)	0.63399 (9)	0.0278 (3)
O3	0.89302 (16)	0.38603 (5)	0.60079 (8)	0.0214 (2)
O4	0.20805 (18)	0.29396 (5)	0.36462 (8)	0.0237 (3)
O5	0.50318 (19)	0.35557 (6)	0.28262 (8)	0.0312 (3)
O6	0.5883 (2)	0.23944 (6)	0.36336 (10)	0.0380 (3)
O7	0.3039 (2)	0.25143 (8)	0.18528 (10)	0.0495 (4)
C1	0.9609 (2)	0.43084 (7)	0.80401 (11)	0.0181 (3)
C2	0.7696 (3)	0.45395 (8)	0.84399 (13)	0.0244 (3)
H2	0.6535	0.4805	0.7994	0.029*
C3	0.7513 (3)	0.43764 (8)	0.95002 (13)	0.0263 (3)
H3	0.6201	0.4530	0.9776	0.032*
C4	0.9193 (3)	0.39949 (8)	1.01712 (12)	0.0237 (3)
C5	1.1086 (3)	0.37578 (8)	0.97464 (13)	0.0261 (3)
H5	1.2237	0.3488	1.0191	0.031*
C6	1.1302 (3)	0.39112 (8)	0.86861 (12)	0.0229 (3)
H6	1.2591	0.3748	0.8401	0.028*
C7	0.8998 (3)	0.38445 (10)	1.13373 (13)	0.0340 (4)
H7A	0.7525	0.4022	1.1480	0.051*
H7B	1.0276	0.4067	1.1830	0.051*
H7C	0.9068	0.3346	1.1460	0.051*
C8	0.6458 (2)	0.37224 (9)	0.58979 (12)	0.0258 (4)
H8A	0.5611	0.4157	0.5953	0.031*
H8B	0.6176	0.3411	0.6488	0.031*
C9	0.5622 (2)	0.33904 (7)	0.47941 (11)	0.0170 (3)
C10	0.2981 (2)	0.33360 (8)	0.46400 (11)	0.0186 (3)
H10A	0.2546	0.3106	0.5281	0.022*
H10B	0.2294	0.3802	0.4580	0.022*
C11	0.6235 (3)	0.38303 (8)	0.38722 (12)	0.0224 (3)
H11A	0.5763	0.4313	0.3957	0.027*
H11B	0.7925	0.3821	0.3898	0.027*
C12	0.6632 (3)	0.26691 (8)	0.47338 (13)	0.0255 (3)
H12A	0.8340	0.2690	0.4908	0.031*

H12B 0.6096 0.2364 0.5273 0.031*

Atomic displacement parameters (Å²)

	U^{11}	U^{22}	U^{33}	U^{12}	U^{13}	U^{23}
S1	0.0172 (2)	0.0208 (2)	0.01733 (19)	0.00029 (14)	0.00335 (14)	0.00003 (13)
P1	0.0281 (2)	0.0345 (3)	0.0219 (2)	0.00056 (18)	0.00434 (18)	-0.01078 (18)
O1	0.0179 (6)	0.0302 (6)	0.0238 (6)	-0.0038 (4)	0.0065 (4)	0.0001 (4)
O2	0.0312 (6)	0.0268 (6)	0.0249 (6)	0.0084 (5)	0.0036 (5)	0.0044 (5)
O3	0.0144 (5)	0.0287 (6)	0.0210 (5)	-0.0011 (4)	0.0032 (4)	-0.0062 (4)
O4	0.0219 (6)	0.0286 (6)	0.0199 (5)	-0.0056 (4)	0.0022 (4)	-0.0036 (4)
O5	0.0316 (6)	0.0469 (8)	0.0159 (5)	-0.0085 (5)	0.0063 (5)	0.0038 (5)
O6	0.0359 (7)	0.0306 (7)	0.0440 (7)	0.0096 (5)	-0.0019 (6)	-0.0168 (6)
O7	0.0439 (8)	0.0701 (10)	0.0329 (7)	-0.0012 (7)	0.0026 (6)	-0.0301 (7)
C1	0.0176 (7)	0.0202 (7)	0.0164 (7)	-0.0015 (6)	0.0029 (6)	-0.0019 (6)
C2	0.0186 (8)	0.0310 (9)	0.0234 (8)	0.0028 (6)	0.0032 (6)	-0.0026 (6)
C3	0.0214 (8)	0.0332 (9)	0.0267 (8)	-0.0017 (6)	0.0105 (7)	-0.0058 (7)
C4	0.0277 (8)	0.0230 (8)	0.0212 (8)	-0.0085 (6)	0.0064 (6)	-0.0037 (6)
C5	0.0294 (8)	0.0260 (8)	0.0225 (8)	0.0035 (7)	0.0040 (6)	0.0037 (6)
C6	0.0207 (7)	0.0259 (8)	0.0227 (8)	0.0041 (6)	0.0053 (6)	-0.0003 (6)
C7	0.0434 (11)	0.0379 (10)	0.0231 (8)	-0.0095 (8)	0.0128 (8)	-0.0014 (7)
C8	0.0160 (7)	0.0417 (10)	0.0204 (8)	-0.0065 (6)	0.0051 (6)	-0.0080 (7)
C9	0.0167 (7)	0.0192 (7)	0.0158 (7)	-0.0011 (5)	0.0044 (5)	0.0004 (5)
C10	0.0179 (7)	0.0240 (7)	0.0143 (7)	-0.0017 (6)	0.0036 (5)	-0.0009 (6)
C11	0.0227 (8)	0.0244 (8)	0.0212 (7)	-0.0049 (6)	0.0072 (6)	0.0018 (6)
C12	0.0226 (8)	0.0227 (8)	0.0297 (8)	0.0029 (6)	0.0006 (7)	0.0013 (6)

Geometric parameters (Å, °)

S1—O1	1.4248 (11)	C4—C7	1.508 (2)
S1—O2	1.4309 (11)	C5—C6	1.384 (2)
S1—O3	1.5803 (11)	C5—H5	0.9500
S1—C1	1.7518 (15)	C6—H6	0.9500
P1—O7	1.4499 (13)	C7—H7A	0.9800
P1—O4	1.5698 (11)	C7—H7B	0.9800
P1—O5	1.5714 (13)	C7—H7C	0.9800
P1—O6	1.5717 (13)	C8—C9	1.5177 (19)
O3—C8	1.4617 (17)	C8—H8A	0.9900
O4—C10	1.4724 (16)	C8—H8B	0.9900
O5—C11	1.4654 (18)	C9—C11	1.528 (2)
O6—C12	1.4623 (19)	C9—C12	1.530 (2)
C1—C2	1.387 (2)	C9—C10	1.5352 (19)
C1—C6	1.394 (2)	C10—H10A	0.9900
C2—C3	1.383 (2)	C10—H10B	0.9900
C2—H2	0.9500	C11—H11A	0.9900
C3—C4	1.387 (2)	C11—H11B	0.9900
C3—H3	0.9500	C12—H12A	0.9900
C4—C5	1.397 (2)	C12—H12B	0.9900

O1—S1—O2	119.56 (7)	C4—C7—H7B	109.5
O1—S1—O3	104.08 (6)	H7A—C7—H7B	109.5
O2—S1—O3	108.73 (7)	C4—C7—H7C	109.5
O1—S1—C1	110.02 (7)	H7A—C7—H7C	109.5
O2—S1—C1	108.73 (7)	H7B—C7—H7C	109.5
O3—S1—C1	104.64 (7)	O3—C8—C9	108.30 (11)
O7—P1—O4	114.69 (7)	O3—C8—H8A	110.0
O7—P1—O5	113.99 (8)	C9—C8—H8A	110.0
O4—P1—O5	104.33 (6)	O3—C8—H8B	110.0
O7—P1—O6	113.91 (8)	C9—C8—H8B	110.0
O4—P1—O6	104.27 (7)	H8A—C8—H8B	108.4
O5—P1—O6	104.49 (7)	C8—C9—C11	111.23 (12)
C8—O3—S1	116.97 (9)	C8—C9—C12	111.73 (12)
C10—O4—P1	112.92 (8)	C11—C9—C12	109.15 (12)
C11—O5—P1	114.35 (9)	C8—C9—C10	107.13 (11)
C12—O6—P1	114.27 (9)	C11—C9—C10	108.58 (12)
C2—C1—C6	120.88 (14)	C12—C9—C10	108.94 (12)
C2—C1—S1	119.94 (11)	O4—C10—C9	109.93 (11)
C6—C1—S1	119.18 (11)	O4—C10—H10A	109.7
C3—C2—C1	118.73 (14)	C9—C10—H10A	109.7
C3—C2—H2	120.6	O4—C10—H10B	109.7
C1—C2—H2	120.6	C9—C10—H10B	109.7
C2—C3—C4	121.78 (15)	H10A—C10—H10B	108.2
C2—C3—H3	119.1	O5—C11—C9	108.85 (12)
C4—C3—H3	119.1	O5—C11—H11A	109.9
C3—C4—C5	118.54 (14)	C9—C11—H11A	109.9
C3—C4—C7	120.85 (15)	O5—C11—H11B	109.9
C5—C4—C7	120.60 (15)	C9—C11—H11B	109.9
C6—C5—C4	120.75 (15)	H11A—C11—H11B	108.3
C6—C5—H5	119.6	O6—C12—C9	109.14 (12)
C4—C5—H5	119.6	O6—C12—H12A	109.9
C5—C6—C1	119.30 (14)	C9—C12—H12A	109.9
C5—C6—H6	120.4	O6—C12—H12B	109.9
C1—C6—H6	120.4	C9—C12—H12B	109.9
C4—C7—H7A	109.5	H12A—C12—H12B	108.3
O1—S1—O3—C8	-177.65 (10)	C2—C3—C4—C7	177.82 (15)
O2—S1—O3—C8	-49.18 (12)	C3—C4—C5—C6	1.2 (2)
C1—S1—O3—C8	66.86 (11)	C7—C4—C5—C6	-178.19 (15)
O7—P1—O4—C10	-172.57 (10)	C4—C5—C6—C1	0.1 (2)
O5—P1—O4—C10	-47.17 (10)	C2—C1—C6—C5	-1.2 (2)
O6—P1—O4—C10	62.18 (11)	S1—C1—C6—C5	179.22 (12)
O7—P1—O5—C11	-171.46 (10)	S1—O3—C8—C9	148.31 (10)
O4—P1—O5—C11	62.71 (11)	O3—C8—C9—C11	-54.43 (17)
O6—P1—O5—C11	-46.49 (11)	O3—C8—C9—C12	67.82 (16)
O7—P1—O6—C12	-174.29 (11)	O3—C8—C9—C10	-172.94 (12)
O4—P1—O6—C12	-48.55 (12)	P1—O4—C10—C9	-12.41 (14)

O5—P1—O6—C12	60.69 (12)	C8—C9—C10—O4	-172.94 (12)
O1—S1—C1—C2	152.36 (12)	C11—C9—C10—O4	66.84 (15)
O2—S1—C1—C2	19.67 (15)	C12—C9—C10—O4	-51.92 (15)
O3—S1—C1—C2	-96.37 (13)	P1—O5—C11—C9	-12.39 (15)
O1—S1—C1—C6	-28.07 (14)	C8—C9—C11—O5	-169.66 (12)
O2—S1—C1—C6	-160.76 (12)	C12—C9—C11—O5	66.61 (15)
O3—S1—C1—C6	83.21 (13)	C10—C9—C11—O5	-52.01 (15)
C6—C1—C2—C3	0.9 (2)	P1—O6—C12—C9	-11.31 (16)
S1—C1—C2—C3	-179.56 (12)	C8—C9—C12—O6	-176.46 (12)
C1—C2—C3—C4	0.6 (2)	C11—C9—C12—O6	-53.03 (16)
C2—C3—C4—C5	-1.6 (2)	C10—C9—C12—O6	65.37 (15)

Hydrogen-bond geometry (Å, °)

<i>D</i> —H... <i>A</i>	<i>D</i> —H	H... <i>A</i>	<i>D</i> ... <i>A</i>	<i>D</i> —H... <i>A</i>
C7—H7 <i>A</i> ...O5 ⁱ	0.98	2.59	3.288 (2)	128
C8—H8 <i>A</i> ...O1 ⁱⁱ	0.99	2.38	3.180 (2)	138
C10—H10 <i>A</i> ...O7 ⁱⁱⁱ	0.99	2.27	3.211 (2)	158
C10—H10 <i>B</i> ...O2 ^{iv}	0.99	2.44	3.348 (2)	152

Symmetry codes: (i) $x, y, z+1$; (ii) $x-1, y, z$; (iii) $x, -y+1/2, z+1/2$; (iv) $-x+1, -y+1, -z+1$.