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(E)-(2,4-Dichlorobenzylidene)amino cyclopropanecarboxylate

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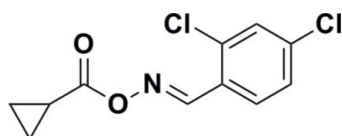
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Key indicators: single-crystal X-ray study; $T = 113$ K; mean $\sigma(\text{C}-\text{C}) = 0.002$ Å; R factor = 0.027; wR factor = 0.076; data-to-parameter ratio = 13.4.

In the title compound $\text{C}_{11}\text{H}_9\text{Cl}_2\text{NO}_2$, the dihedral angle between the benzene and cyclopropane ring planes is 89.95 (13)°. The carbonyl–oxime grouping is almost coplanar with the benzene ring [dihedral angle = 4.08 (6)°]. In the crystal, molecules are linked by $\text{C}-\text{H}\cdots\text{O}$ interactions into [100] chains.

Related literature

For further synthetic details, see: Liu *et al.* (2011*b*, 2012). For related structures, see: Liu & Liu (2011) Liu *et al.* (2011*d*). For the biological activity of related compounds, see: Liu *et al.* (2010, 2011*a,c*).



Experimental

Crystal data

 $\text{C}_{11}\text{H}_9\text{Cl}_2\text{NO}_2$ $M_r = 258.09$ Triclinic, $P\bar{1}$ $a = 6.4381$ (13) Å $b = 7.6030$ (15) Å $c = 11.956$ (2) Å $\alpha = 94.90$ (3)° $\beta = 100.42$ (3)° $\gamma = 102.70$ (3)° $V = 556.71$ (19) Å³ $Z = 2$ Mo $K\alpha$ radiation $\mu = 0.57$ mm⁻¹ $T = 113$ K $0.24 \times 0.20 \times 0.10$ mm

Data collection

Rigaku Saturn CCD diffractometer

Absorption correction: multi-scan

*(CrystalClear; Rigaku/MS**2005)* $T_{\min} = 0.876$, $T_{\max} = 0.946$

3737 measured reflections

1937 independent reflections

1563 reflections with $I > 2\sigma(I)$ $R_{\text{int}} = 0.030$

Refinement

 $R[F^2 > 2\sigma(F^2)] = 0.027$ $wR(F^2) = 0.076$ $S = 1.03$

1937 reflections

145 parameters

H-atom parameters constrained

 $\Delta\rho_{\text{max}} = 0.24$ e Å⁻³ $\Delta\rho_{\text{min}} = -0.21$ e Å⁻³

Table 1

Hydrogen-bond geometry (Å, °).

$D-\text{H}\cdots A$	$D-\text{H}$	$\text{H}\cdots A$	$D\cdots A$	$D-\text{H}\cdots A$
$\text{C}1-\text{H}1\text{B}\cdots\text{O}1^i$	0.97	2.57	3.5008 (19)	161

Symmetry code: (i) $x - 1, y, z$.

Data collection: *CrystalClear* (Rigaku/MS, 2005); cell refinement: *CrystalClear*; data reduction: *CrystalClear*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 2008); program(s) used to refine structure: *SHELXL97* (Sheldrick, 2008); molecular graphics: *SHELXTL* (Sheldrick, 2008); software used to prepare material for publication: *SHELXL97*.

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Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: HB6735).

References

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supporting information

Acta Cryst. (2012). E68, o1594 [doi:10.1107/S1600536812018016]

(*E*)-(2,4-Dichlorobenzylidene)amino cyclopropanecarboxylate

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S1. Experimental

Dropwisely the cyclopropanecarbonyl chloride to 2,4-dichlorobenzaldehyde oxime (7.50 mmol in 25 ml THF) and 7.5 mmol Et₃N, then vigorously stirred at ambient temperature for overnight. The corresponding product precipitated immediately. Compound was dissolved in hot alcohol and the resulting solution was allowed to stand in air at room temperature to give colourless blocks of the title compound.

S2. Refinement

All the H atoms were positioned geometrically (C—H = 0.93–0.97 Å) and refined as riding with $U_{\text{iso}}(\text{H}) = 1.2U_{\text{eq}}(\text{C})$ or $1.5U_{\text{eq}}(\text{methyl C})$.

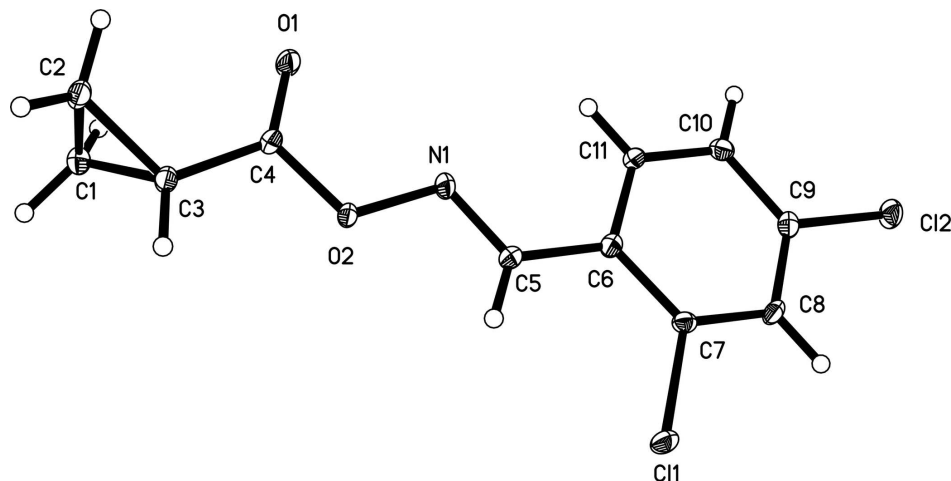


Figure 1

The molecular structure of (I). Displacement ellipsoids are drawn at the 30% probability level.

(*E*)-(2,4-Dichlorobenzylidene)amino cyclopropanecarboxylate

Crystal data

C₁₁H₉Cl₂NO₂

$M_r = 258.09$

Triclinic, $P\bar{1}$

$a = 6.4381(13)$ Å

$b = 7.6030(15)$ Å

$c = 11.956(2)$ Å

$\alpha = 94.90(3)^\circ$

$\beta = 100.42(3)^\circ$

$\gamma = 102.70(3)^\circ$

$V = 556.71(19)$ Å³

$Z = 2$

$F(000) = 264$

$D_x = 1.540$ Mg m⁻³

Mo $K\alpha$ radiation, $\lambda = 0.71073$ Å

Cell parameters from 1887 reflections

$\theta = 1.8$ – 27.9°

$\mu = 0.57 \text{ mm}^{-1}$
 $T = 113 \text{ K}$

Block, colorless
 $0.24 \times 0.20 \times 0.10 \text{ mm}$

Data collection

Rigaku Saturn CCD
 diffractometer
 Radiation source: rotating anode
 Confocal monochromator
 ω scans
 Absorption correction: multi-scan
 (*CrystalClear*; Rigaku/MSO, 2005)
 $T_{\min} = 0.876$, $T_{\max} = 0.946$

3737 measured reflections
 1937 independent reflections
 1563 reflections with $I > 2\sigma(I)$
 $R_{\text{int}} = 0.030$
 $\theta_{\max} = 25.0^\circ$, $\theta_{\min} = 1.8^\circ$
 $h = -7 \rightarrow 7$
 $k = -9 \rightarrow 6$
 $l = -14 \rightarrow 14$

Refinement

Refinement on F^2
 Least-squares matrix: full
 $R[F^2 > 2\sigma(F^2)] = 0.027$
 $wR(F^2) = 0.076$
 $S = 1.03$
 1937 reflections
 145 parameters
 0 restraints
 Primary atom site location: structure-invariant
 direct methods

Secondary atom site location: difference Fourier
 map
 Hydrogen site location: inferred from
 neighbouring sites
 H-atom parameters constrained
 $w = 1/[\sigma^2(F_o^2) + (0.0407P)^2]$
 where $P = (F_o^2 + 2F_c^2)/3$
 $(\Delta/\sigma)_{\max} = 0.001$
 $\Delta\rho_{\max} = 0.24 \text{ e } \text{\AA}^{-3}$
 $\Delta\rho_{\min} = -0.21 \text{ e } \text{\AA}^{-3}$

Special details

Geometry. All e.s.d.'s (except the e.s.d. in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell e.s.d.'s are taken into account individually in the estimation of e.s.d.'s in distances, angles and torsion angles; correlations between e.s.d.'s in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell e.s.d.'s is used for estimating e.s.d.'s involving l.s. planes.

Refinement. Refinement of F^2 against ALL reflections. The weighted R -factor wR and goodness of fit S are based on F^2 , conventional R -factors R are based on F , with F set to zero for negative F^2 . The threshold expression of $F^2 > \sigma(F^2)$ is used only for calculating R -factors(gt) etc. and is not relevant to the choice of reflections for refinement. R -factors based on F^2 are statistically about twice as large as those based on F , and R -factors based on ALL data will be even larger.

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (\AA^2)

	x	y	z	$U_{\text{iso}}^*/U_{\text{eq}}$
Cl1	0.32359 (6)	0.70035 (5)	0.47089 (3)	0.02317 (14)
Cl2	1.15092 (6)	0.98054 (6)	0.68823 (4)	0.02531 (14)
O1	0.08882 (17)	0.30779 (16)	0.94813 (9)	0.0263 (3)
O2	0.00692 (16)	0.37138 (15)	0.76492 (9)	0.0186 (3)
N1	0.2306 (2)	0.46280 (18)	0.78101 (11)	0.0202 (3)
C1	-0.3868 (2)	0.2548 (2)	0.94150 (14)	0.0217 (4)
H1A	-0.2971	0.3293	1.0106	0.026*
H1B	-0.5348	0.2689	0.9234	0.026*
C2	-0.3504 (3)	0.0724 (2)	0.91336 (14)	0.0208 (4)
H2A	-0.4764	-0.0249	0.8783	0.025*
H2B	-0.2388	0.0354	0.9655	0.025*
C3	-0.2777 (2)	0.2198 (2)	0.84222 (14)	0.0198 (4)
H3	-0.3629	0.2113	0.7644	0.024*
C4	-0.0426 (2)	0.3004 (2)	0.86199 (13)	0.0177 (4)

C5	0.2637 (2)	0.5304 (2)	0.69054 (14)	0.0177 (4)
H5	0.1511	0.5129	0.6268	0.021*
C6	0.4823 (2)	0.6365 (2)	0.68810 (14)	0.0164 (4)
C7	0.5265 (3)	0.7218 (2)	0.59348 (13)	0.0168 (4)
C8	0.7305 (2)	0.8271 (2)	0.59131 (13)	0.0188 (4)
H8	0.7566	0.8836	0.5275	0.023*
C9	0.8947 (2)	0.8453 (2)	0.68750 (14)	0.0187 (4)
C10	0.8595 (3)	0.7612 (2)	0.78232 (14)	0.0196 (4)
H10	0.9725	0.7736	0.8453	0.024*
C11	0.6539 (2)	0.6579 (2)	0.78256 (14)	0.0177 (4)
H11	0.6293	0.6016	0.8466	0.021*

Atomic displacement parameters (Å²)

	U^{11}	U^{22}	U^{33}	U^{12}	U^{13}	U^{23}
C11	0.0227 (2)	0.0314 (3)	0.0143 (2)	0.00420 (18)	0.00140 (17)	0.00786 (18)
C12	0.0192 (2)	0.0275 (2)	0.0273 (3)	-0.00120 (17)	0.00716 (17)	0.00581 (18)
O1	0.0182 (6)	0.0425 (8)	0.0166 (7)	0.0021 (5)	0.0021 (5)	0.0121 (6)
O2	0.0123 (5)	0.0267 (6)	0.0165 (6)	0.0012 (5)	0.0038 (4)	0.0082 (5)
N1	0.0117 (6)	0.0267 (8)	0.0213 (8)	0.0005 (6)	0.0039 (5)	0.0072 (6)
C1	0.0179 (8)	0.0266 (9)	0.0221 (10)	0.0051 (7)	0.0073 (7)	0.0061 (7)
C2	0.0191 (8)	0.0226 (9)	0.0198 (9)	0.0004 (7)	0.0059 (7)	0.0054 (7)
C3	0.0151 (8)	0.0264 (9)	0.0160 (9)	0.0002 (7)	0.0027 (7)	0.0059 (7)
C4	0.0187 (8)	0.0201 (8)	0.0162 (9)	0.0045 (7)	0.0070 (7)	0.0059 (7)
C5	0.0178 (8)	0.0218 (9)	0.0145 (9)	0.0048 (7)	0.0041 (7)	0.0056 (7)
C6	0.0184 (8)	0.0158 (8)	0.0162 (9)	0.0042 (6)	0.0061 (7)	0.0031 (6)
C7	0.0185 (8)	0.0193 (8)	0.0134 (9)	0.0070 (6)	0.0019 (6)	0.0024 (6)
C8	0.0244 (9)	0.0189 (9)	0.0165 (10)	0.0057 (7)	0.0100 (7)	0.0071 (7)
C9	0.0181 (8)	0.0170 (8)	0.0223 (9)	0.0038 (7)	0.0082 (7)	0.0024 (7)
C10	0.0189 (8)	0.0226 (9)	0.0183 (9)	0.0077 (7)	0.0022 (7)	0.0040 (7)
C11	0.0194 (8)	0.0206 (9)	0.0166 (9)	0.0079 (7)	0.0065 (7)	0.0073 (7)

Geometric parameters (Å, °)

C11—C7	1.7477 (16)	C3—C4	1.470 (2)
C12—C9	1.7394 (16)	C3—H3	0.9800
O1—C4	1.1976 (18)	C5—C6	1.467 (2)
O2—C4	1.3800 (19)	C5—H5	0.9300
O2—N1	1.4253 (16)	C6—C7	1.394 (2)
N1—C5	1.269 (2)	C6—C11	1.401 (2)
C1—C2	1.478 (2)	C7—C8	1.385 (2)
C1—C3	1.517 (2)	C8—C9	1.389 (2)
C1—H1A	0.9700	C8—H8	0.9300
C1—H1B	0.9700	C9—C10	1.378 (2)
C2—C3	1.508 (2)	C10—C11	1.383 (2)
C2—H2A	0.9700	C10—H10	0.9300
C2—H2B	0.9700	C11—H11	0.9300

C4—O2—N1	112.18 (11)	O2—C4—C3	109.31 (12)
C5—N1—O2	109.05 (12)	N1—C5—C6	118.94 (14)
C2—C1—C3	60.45 (11)	N1—C5—H5	120.5
C2—C1—H1A	117.7	C6—C5—H5	120.5
C3—C1—H1A	117.7	C7—C6—C11	117.60 (14)
C2—C1—H1B	117.7	C7—C6—C5	121.71 (14)
C3—C1—H1B	117.7	C11—C6—C5	120.69 (15)
H1A—C1—H1B	114.8	C8—C7—C6	122.37 (14)
C1—C2—C3	61.08 (11)	C8—C7—C11	116.74 (13)
C1—C2—H2A	117.7	C6—C7—C11	120.89 (12)
C3—C2—H2A	117.7	C7—C8—C9	117.79 (15)
C1—C2—H2B	117.7	C7—C8—H8	121.1
C3—C2—H2B	117.7	C9—C8—H8	121.1
H2A—C2—H2B	114.8	C10—C9—C8	121.89 (15)
C4—C3—C2	116.62 (14)	C10—C9—C12	119.36 (12)
C4—C3—C1	115.92 (13)	C8—C9—C12	118.75 (13)
C2—C3—C1	58.47 (11)	C9—C10—C11	119.14 (14)
C4—C3—H3	117.5	C9—C10—H10	120.4
C2—C3—H3	117.5	C11—C10—H10	120.4
C1—C3—H3	117.5	C10—C11—C6	121.20 (15)
O1—C4—O2	123.97 (14)	C10—C11—H11	119.4
O1—C4—C3	126.72 (15)	C6—C11—H11	119.4
C4—O2—N1—C5	-177.60 (13)	C5—C6—C7—C8	177.96 (16)
C1—C2—C3—C4	105.44 (16)	C11—C6—C7—C11	179.11 (13)
C2—C1—C3—C4	-106.64 (16)	C5—C6—C7—C11	-2.0 (2)
N1—O2—C4—O1	-3.1 (2)	C6—C7—C8—C9	0.4 (3)
N1—O2—C4—C3	176.04 (13)	C11—C7—C8—C9	-179.65 (12)
C2—C3—C4—O1	-25.6 (3)	C7—C8—C9—C10	0.6 (3)
C1—C3—C4—O1	40.4 (2)	C7—C8—C9—C12	-178.69 (13)
C2—C3—C4—O2	155.33 (15)	C8—C9—C10—C11	-1.1 (3)
C1—C3—C4—O2	-138.66 (14)	C12—C9—C10—C11	178.24 (13)
O2—N1—C5—C6	178.12 (13)	C9—C10—C11—C6	0.5 (3)
N1—C5—C6—C7	-176.72 (16)	C7—C6—C11—C10	0.5 (2)
N1—C5—C6—C11	2.2 (3)	C5—C6—C11—C10	-178.44 (16)
C11—C6—C7—C8	-1.0 (3)		

Hydrogen-bond geometry (Å, °)

<i>D</i> —H... <i>A</i>	<i>D</i> —H	H... <i>A</i>	<i>D</i> ... <i>A</i>	<i>D</i> —H... <i>A</i>
C1—H1B...O1 ⁱ	0.97	2.57	3.5008 (19)	161

Symmetry code: (i) $x-1, y, z$.