

Acta Crystallographica Section E

Structure Reports

Online

ISSN 1600-5368

3-(2-Chloro-3,3,3-trifluoroprop-1-en-1-yl)-2,2-dimethyl-N-[3-(trifluoromethyl)phenyl]cyclopropanecarboxamide

 Guo-Wu Rao,^{a*} Xiao-Min Li^a and Na-Bo Sun^b
^aCollege of Pharmaceutical Science, Zhejiang University of Technology, Hangzhou, 310014, People's Republic of China, and ^bCollege of Biology and Environmental Engineering, Zhejiang Shuren University, Hangzhou, 310015, People's Republic of China

Correspondence e-mail: rgw@zjut.edu.cn

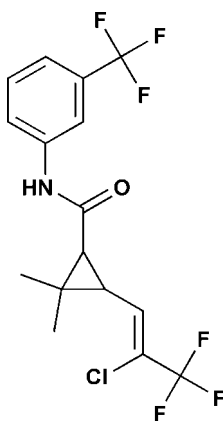
Received 8 May 2012; accepted 8 May 2012

 Key indicators: single-crystal X-ray study; $T = 298$ K; mean $\sigma(\text{C}-\text{C}) = 0.005$ Å; disorder in main residue; R factor = 0.064; wR factor = 0.175; data-to-parameter ratio = 12.2.

In the title molecule, $\text{C}_{16}\text{H}_{14}\text{ClF}_6\text{NO}$, the cyclopropane ring forms a dihedral angle of $70.82(18)^\circ$ with the benzene ring. The torsion angles about the ethylene and amide bonds are $-2.2(5)$ ($\text{Cl}-\text{C}-\text{C}-\text{C}$) and $0.8(5)^\circ$ ($\text{O}-\text{C}-\text{N}-\text{C}$). A supramolecular chain propagated by glide symmetry along $[001]$ and mediated by $\text{N}-\text{H}\cdots\text{O}$ hydrogen bonds is observed in the crystal packing.

Related literature

For the biological activity of pyrethroids, see: Chen *et al.* (1991); Sun *et al.* (2007, 2008). For the synthesis of the title compound, see: Sun *et al.* (2007).



Experimental

Crystal data

$\text{C}_{16}\text{H}_{14}\text{ClF}_6\text{NO}$
 $M_r = 385.73$
 Monoclinic, $P2_1/c$
 $a = 11.006(3)$ Å
 $b = 16.699(4)$ Å
 $c = 9.659(2)$ Å
 $\beta = 93.009(3)^\circ$
 $V = 1772.8(7)$ Å³
 $Z = 4$
 Mo $K\alpha$ radiation
 $\mu = 0.28$ mm⁻¹
 $T = 298$ K
 $0.60 \times 0.13 \times 0.12$ mm

Data collection

Bruker SMART CCD diffractometer
 Absorption correction: multi-scan (SADABS; Bruker, 1997)
 $T_{\min} = 0.846$, $T_{\max} = 0.967$
 7354 measured reflections
 3120 independent reflections
 2247 reflections with $I > 2\sigma(I)$
 $R_{\text{int}} = 0.026$

Refinement

$R[F^2 > 2\sigma(F^2)] = 0.064$
 $wR(F^2) = 0.175$
 $S = 1.04$
 3120 reflections
 256 parameters
 H-atom parameters constrained
 $\Delta\rho_{\max} = 0.52$ e Å⁻³
 $\Delta\rho_{\min} = -0.48$ e Å⁻³

Table 1

Hydrogen-bond geometry (Å, °).

$D-H\cdots A$	$D-H$	$H\cdots A$	$D\cdots A$	$D-H\cdots A$
$\text{N1}-\text{H1}\cdots\text{O1}^i$	0.86	2.15	2.974 (3)	161

 Symmetry code: (i) $x, -y + \frac{1}{2}, z + \frac{1}{2}$.

Data collection: SMART (Bruker, 1997); cell refinement: SAINT (Bruker, 1997); data reduction: SAINT; program(s) used to solve structure: SHELXS97 (Sheldrick, 2008); program(s) used to refine structure: SHELXL97 (Sheldrick, 2008); molecular graphics: ORTEP-3 for Windows (Farrugia, 1997); software used to prepare material for publication: WinGX (Farrugia, 1999).

The authors are very grateful to the Program of the Education Department of Zhejiang Province of China (Y200803060) for financial support.

Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: TK5094).

References

- Bruker (1997). SMART, SAINT and SADABS. Bruker AXS Inc., Madison, Wisconsin, USA.
 Chen, F. H. & Yu, Z. S. (1991). *Chem. J. Chin. Univ.* **12**, 485–487.
 Farrugia, L. J. (1997). *J. Appl. Cryst.* **30**, 565.
 Farrugia, L. J. (1999). *J. Appl. Cryst.* **32**, 837–838.
 Sheldrick, G. M. (2008). *Acta Cryst. A* **64**, 112–122.
 Sun, N. B., Shen, D. L., Tan, C. X., Weng, J. Q., Cong, S. & Fu, H. (2008). *Chin. J. Org. Chem.* **28**, 713–717.
 Sun, N. B., Shen, D. L., Weng, J. Q., Tan, C. X. & Chen, Q. W. (2007). *Chin. Patent* CN 101003493.

supporting information

Acta Cryst. (2012). E68, o1743 [doi:10.1107/S1600536812020922]

3-(2-Chloro-3,3,3-trifluoroprop-1-en-1-yl)-2,2-dimethyl-N-[3-(trifluoromethyl)-phenyl]cyclopropanecarboxamide

Guo-Wu Rao, Xiao-Min Li and Na-Bo Sun

S1. Comment

Pyrethroids have a high potential for biological activity with low toxicity and good environmental compatibility. These have been widely used in pesticides (Chen *et al.*, 1991; Sun *et al.*, 2007, 2008). In continuation of our studies of biological activities in pyrethroids, we have obtained a colourless crystalline compound, (I). The structure was confirmed by single-crystal X-ray diffraction.

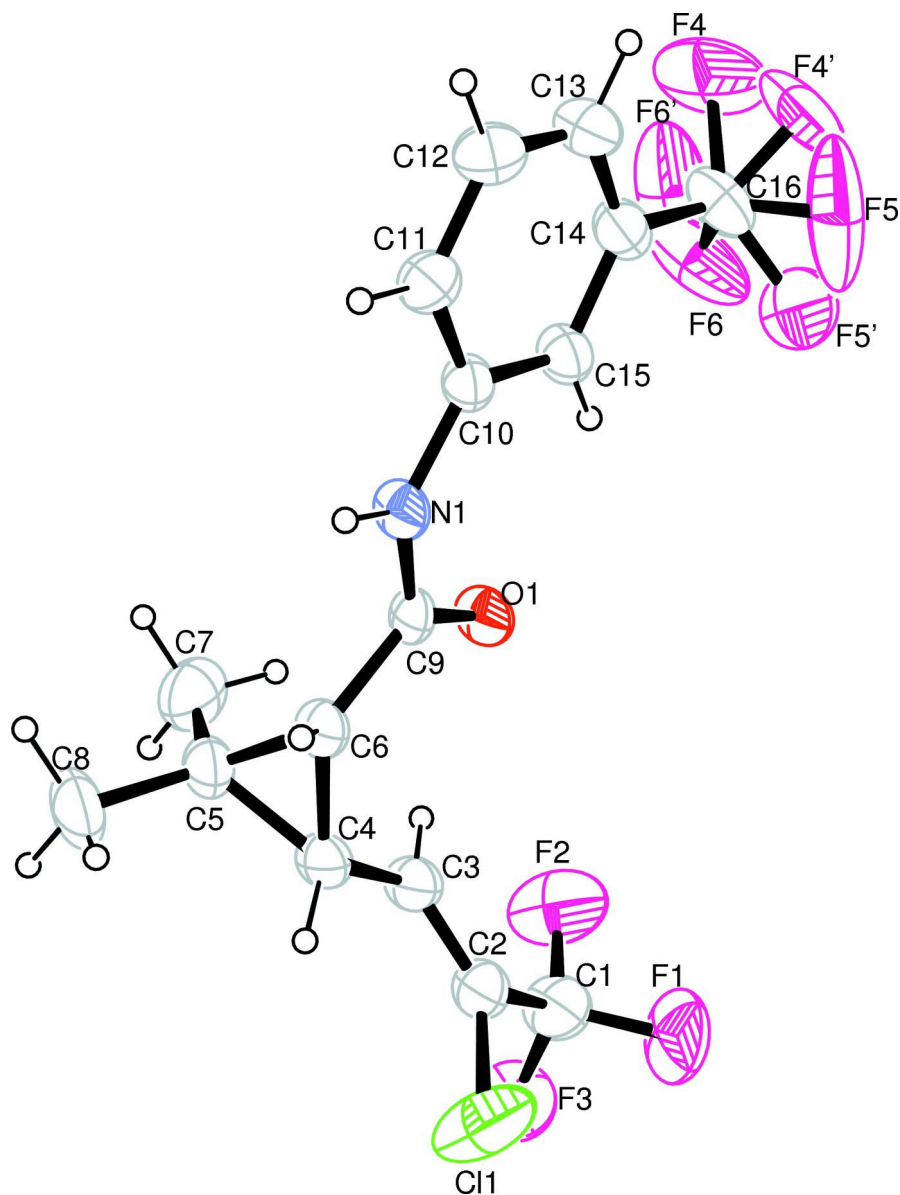
The molecular structure of (I) is illustrated in Fig. 1. In (I), the cyclopropane ring (C4—C6) forms dihedral angles of 89.70 (23) and 70.82 (18)° with the least-squares planes of the (C5,C7,C8) plane and the phenyl ring, respectively. The C11—C2—C3—C4 and O1—C9—N1—C10 torsion angles are -2.2 (5) and 0.8 (5)°, respectively. A supramolecular chain propagated by glide symmetry along [001] and mediated by N—H···O hydrogen bonds, Table 1, is observed in the crystal packing.

S2. Experimental

The title compound was synthesized according to the literature procedure (Sun *et al.*, 2007). A solution of the compound in ethanol was concentrated gradually at room temperature to afford colourless blocks.

S3. Refinement

The H-atoms were placed in calculated positions [N—H = 0.86 Å; C—H = 0.93 to 0.98 Å, $U_{\text{iso}}(\text{H}) = 1.2U_{\text{eq}}(\text{C},\text{N})$] and were included in the refinement in the riding model approximation. The CF₃ group was disordered and modelled over two positions of equal weight.

**Figure 1**

The molecular structure of (I), shown with 30% probability displacement ellipsoids. The F atoms of the CF₃ group have 50% occupancy factors.

**3-(2-Chloro-3,3,3-trifluoroprop-1-en-1-yl)-2,2-dimethyl-
N-[3-(trifluoromethyl)phenyl]cyclopropanecarboxamide**

Crystal data

C₁₆H₁₄ClF₆NO

M_r = 385.73

Monoclinic, *P*2₁/*c*

Hall symbol: -*P* 2ybc

a = 11.006 (3) Å

b = 16.699 (4) Å

c = 9.659 (2) Å

β = 93.009 (3)°

V = 1772.8 (7) Å³

Z = 4

F(000) = 784

D_x = 1.445 Mg m⁻³

Mo *K*α radiation, λ = 0.71073 Å

μ = 0.28 mm⁻¹

$T = 298$ K $0.60 \times 0.13 \times 0.12$ mm
 Block, colourless

Data collection

Bruker SMART CCD diffractometer Radiation source: fine-focus sealed tube Graphite monochromator φ and ω scans Absorption correction: multi-scan (SADABS; Bruker, 1997) $T_{\min} = 0.846$, $T_{\max} = 0.967$	7354 measured reflections 3120 independent reflections 2247 reflections with $I > 2\sigma(I)$ $R_{\text{int}} = 0.026$ $\theta_{\text{max}} = 25.0^\circ$, $\theta_{\text{min}} = 1.9^\circ$ $h = -13 \rightarrow 12$ $k = -19 \rightarrow 19$ $l = -10 \rightarrow 11$
--	---

Refinement

Refinement on F^2 Least-squares matrix: full $R[F^2 > 2\sigma(F^2)] = 0.064$ $wR(F^2) = 0.175$ $S = 1.04$ 3120 reflections 256 parameters 0 restraints Primary atom site location: structure-invariant direct methods Secondary atom site location: difference Fourier map	Hydrogen site location: inferred from neighbouring sites H-atom parameters constrained $w = 1/[\sigma^2(F_o^2) + (0.0682P)^2 + 1.1818P]$ where $P = (F_o^2 + 2F_c^2)/3$ $(\Delta/\sigma)_{\text{max}} < 0.001$ $\Delta\rho_{\text{max}} = 0.52 \text{ e } \text{\AA}^{-3}$ $\Delta\rho_{\text{min}} = -0.48 \text{ e } \text{\AA}^{-3}$ Extinction correction: SHELXL97 (Sheldrick, 2008), $F_c^* = kF_c[1 + 0.001x F_c^2 \lambda^3 / \sin(2\theta)]^{-1/4}$ Extinction coefficient: 0.0120 (19)
---	--

Special details

Geometry. All e.s.d.'s (except the e.s.d. in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell e.s.d.'s are taken into account individually in the estimation of e.s.d.'s in distances, angles and torsion angles; correlations between e.s.d.'s in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell e.s.d.'s is used for estimating e.s.d.'s involving l.s. planes.

Refinement. Refinement of F^2 against ALL reflections. The weighted R -factor wR and goodness of fit S are based on F^2 , conventional R -factors R are based on F , with F set to zero for negative F^2 . The threshold expression of $F^2 > \sigma(F^2)$ is used only for calculating R -factors(gt) etc. and is not relevant to the choice of reflections for refinement. R -factors based on F^2 are statistically about twice as large as those based on F , and R -factors based on ALL data will be even larger.

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (\AA^2)

	x	y	z	$U_{\text{iso}}^*/U_{\text{eq}}$	Occ. (<1)
C1	1.3041 (4)	0.0980 (3)	-0.1232 (4)	0.0913 (12)	
C2	1.3072 (3)	0.1266 (2)	0.0223 (3)	0.0685 (9)	
C3	1.2130 (3)	0.12532 (19)	0.0995 (3)	0.0630 (8)	
H3	1.1393	0.1088	0.0573	0.076*	
C4	1.2130 (3)	0.14749 (19)	0.2459 (3)	0.0619 (8)	
H4	1.2944	0.1552	0.2899	0.074*	
C5	1.1223 (3)	0.1129 (2)	0.3411 (3)	0.0721 (9)	
C8	1.1705 (4)	0.0951 (3)	0.4888 (4)	0.1193 (18)	
H8B	1.2049	0.0423	0.4928	0.179*	
H8A	1.1049	0.0983	0.5503	0.179*	
H8C	1.2319	0.1336	0.5162	0.179*	
C7	1.0243 (4)	0.0577 (2)	0.2842 (5)	0.0998 (13)	

H7A	0.9567	0.0590	0.3432	0.150*	
H7B	1.0556	0.0041	0.2805	0.150*	
H7C	0.9978	0.0747	0.1926	0.150*	
C6	1.1158 (3)	0.20070 (18)	0.3045 (3)	0.0560 (7)	
H6	1.1463	0.2363	0.3792	0.067*	
C9	1.0126 (2)	0.23407 (16)	0.2185 (2)	0.0476 (6)	
C10	0.8386 (3)	0.32746 (16)	0.2473 (3)	0.0527 (7)	
C11	0.7832 (3)	0.37131 (19)	0.3483 (3)	0.0663 (8)	
H11	0.8163	0.3716	0.4389	0.080*	
C12	0.6798 (3)	0.4141 (2)	0.3148 (4)	0.0818 (10)	
H12	0.6429	0.4433	0.3831	0.098*	
C13	0.6299 (3)	0.4144 (2)	0.1813 (4)	0.0798 (10)	
H13	0.5598	0.4437	0.1588	0.096*	
C14	0.6848 (3)	0.37073 (18)	0.0813 (4)	0.0665 (8)	
C15	0.7895 (3)	0.32759 (18)	0.1131 (3)	0.0608 (8)	
H15	0.8267	0.2988	0.0445	0.073*	
C16	0.6311 (4)	0.3696 (3)	-0.0649 (5)	0.0911 (12)	
Cl1	1.44791 (11)	0.15833 (11)	0.08073 (14)	0.1439 (7)	
F1	1.3420 (3)	0.15215 (19)	-0.2109 (3)	0.1418 (12)	
F2	1.1945 (3)	0.0778 (2)	-0.1683 (3)	0.1437 (13)	
F3	1.3764 (3)	0.03595 (16)	-0.1382 (3)	0.1264 (10)	
N1	0.9449 (2)	0.28521 (14)	0.2908 (2)	0.0565 (6)	
H1	0.9703	0.2932	0.3754	0.068*	
O1	0.98962 (18)	0.21704 (13)	0.09703 (18)	0.0643 (6)	
F4	0.5087 (8)	0.3754 (17)	-0.0623 (13)	0.165 (5)	0.50
F5	0.661 (3)	0.4283 (13)	-0.1320 (16)	0.214 (9)	0.50
F6	0.643 (2)	0.3051 (10)	-0.1269 (15)	0.152 (9)	0.50
F4'	0.568 (2)	0.4289 (10)	-0.1007 (11)	0.153 (6)	0.50
F5'	0.7204 (9)	0.3700 (15)	-0.1603 (8)	0.132 (4)	0.50
F6'	0.575 (3)	0.3062 (10)	-0.0982 (18)	0.177 (9)	0.50

Atomic displacement parameters (Å²)

	U^{11}	U^{22}	U^{33}	U^{12}	U^{13}	U^{23}
C1	0.094 (3)	0.121 (3)	0.060 (2)	0.026 (3)	0.017 (2)	-0.008 (2)
C2	0.072 (2)	0.083 (2)	0.0508 (18)	0.0154 (17)	0.0072 (15)	0.0018 (15)
C3	0.0602 (18)	0.081 (2)	0.0475 (16)	0.0132 (15)	0.0012 (14)	-0.0018 (14)
C4	0.0577 (17)	0.086 (2)	0.0414 (15)	0.0140 (15)	-0.0021 (12)	0.0013 (14)
C5	0.084 (2)	0.081 (2)	0.0524 (18)	0.0249 (19)	0.0141 (16)	0.0172 (16)
C8	0.149 (4)	0.148 (4)	0.062 (2)	0.073 (3)	0.023 (2)	0.045 (2)
C7	0.111 (3)	0.072 (2)	0.120 (4)	0.003 (2)	0.037 (3)	0.014 (2)
C6	0.0630 (17)	0.0717 (19)	0.0329 (13)	0.0080 (14)	-0.0013 (12)	-0.0027 (12)
C9	0.0560 (15)	0.0559 (15)	0.0308 (13)	-0.0004 (12)	0.0011 (11)	0.0031 (11)
C10	0.0588 (16)	0.0527 (16)	0.0466 (15)	0.0036 (13)	0.0027 (12)	0.0037 (12)
C11	0.071 (2)	0.073 (2)	0.0561 (18)	0.0065 (16)	0.0095 (15)	-0.0040 (15)
C12	0.079 (2)	0.085 (2)	0.082 (3)	0.0180 (19)	0.020 (2)	-0.0061 (19)
C13	0.067 (2)	0.076 (2)	0.097 (3)	0.0186 (17)	0.0053 (19)	0.010 (2)
C14	0.0655 (19)	0.0607 (18)	0.073 (2)	0.0051 (15)	-0.0035 (16)	0.0097 (16)

C15	0.0709 (19)	0.0603 (17)	0.0510 (17)	0.0113 (15)	0.0013 (14)	0.0012 (14)
C16	0.089 (3)	0.093 (3)	0.088 (3)	0.022 (3)	-0.026 (2)	0.014 (3)
C11	0.0863 (8)	0.2456 (18)	0.1022 (9)	-0.0409 (9)	0.0265 (6)	-0.0419 (10)
F1	0.192 (3)	0.173 (3)	0.0640 (15)	0.027 (2)	0.0439 (17)	0.0278 (16)
F2	0.109 (2)	0.251 (4)	0.0706 (15)	0.001 (2)	-0.0003 (14)	-0.0609 (19)
F3	0.152 (2)	0.133 (2)	0.0976 (19)	0.0426 (18)	0.0355 (16)	-0.0290 (15)
N1	0.0674 (15)	0.0694 (15)	0.0320 (11)	0.0117 (12)	-0.0038 (10)	-0.0053 (10)
O1	0.0720 (13)	0.0879 (15)	0.0323 (10)	0.0193 (11)	-0.0052 (9)	-0.0060 (9)
F4	0.088 (5)	0.259 (15)	0.142 (8)	0.039 (7)	-0.058 (5)	-0.037 (9)
F5	0.34 (2)	0.183 (14)	0.107 (10)	-0.117 (14)	-0.092 (13)	0.088 (10)
F6	0.200 (14)	0.154 (14)	0.093 (6)	0.098 (13)	-0.076 (9)	-0.054 (9)
F4'	0.188 (13)	0.145 (10)	0.122 (6)	0.105 (11)	-0.031 (9)	0.024 (7)
F5'	0.116 (5)	0.212 (11)	0.068 (3)	0.017 (7)	-0.009 (3)	0.024 (6)
F6'	0.239 (17)	0.142 (13)	0.139 (10)	-0.088 (13)	-0.094 (11)	0.044 (9)

Geometric parameters (Å, °)

C1—F2	1.305 (5)	C9—O1	1.221 (3)
C1—F3	1.319 (4)	C9—N1	1.351 (3)
C1—F1	1.321 (5)	C10—C15	1.378 (4)
C1—C2	1.483 (5)	C10—C11	1.387 (4)
C2—C3	1.309 (4)	C10—N1	1.411 (3)
C2—C11	1.704 (4)	C11—C12	1.367 (5)
C3—C4	1.462 (4)	C11—H11	0.9300
C3—H3	0.9300	C12—C13	1.376 (5)
C4—C5	1.507 (4)	C12—H12	0.9300
C4—C6	1.522 (4)	C13—C14	1.375 (5)
C4—H4	0.9800	C13—H13	0.9300
C5—C7	1.500 (5)	C14—C15	1.380 (4)
C5—C6	1.509 (4)	C14—C16	1.502 (5)
C5—C8	1.525 (5)	C15—H15	0.9300
C8—H8B	0.9600	C16—F5	1.230 (10)
C8—H8A	0.9600	C16—F6	1.242 (11)
C8—H8C	0.9600	C16—F4'	1.246 (9)
C7—H7A	0.9600	C16—F6'	1.258 (12)
C7—H7B	0.9600	C16—F4	1.352 (10)
C7—H7C	0.9600	C16—F5'	1.381 (10)
C6—C9	1.481 (4)	N1—H1	0.8600
C6—H6	0.9800		
F2—C1—F3	108.2 (4)	N1—C9—C6	112.0 (2)
F2—C1—F1	106.3 (4)	C15—C10—C11	119.7 (3)
F3—C1—F1	104.8 (3)	C15—C10—N1	123.9 (3)
F2—C1—C2	112.0 (3)	C11—C10—N1	116.3 (3)
F3—C1—C2	111.9 (3)	C12—C11—C10	120.0 (3)
F1—C1—C2	113.3 (4)	C12—C11—H11	120.0
C3—C2—C1	123.7 (3)	C10—C11—H11	120.0
C3—C2—C11	123.3 (3)	C11—C12—C13	120.7 (3)

C1—C2—C11	112.9 (3)	C11—C12—H12	119.7
C2—C3—C4	126.0 (3)	C13—C12—H12	119.7
C2—C3—H3	117.0	C14—C13—C12	119.2 (3)
C4—C3—H3	117.0	C14—C13—H13	120.4
C3—C4—C5	121.8 (3)	C12—C13—H13	120.4
C3—C4—C6	122.9 (2)	C13—C14—C15	120.9 (3)
C5—C4—C6	59.8 (2)	C13—C14—C16	120.1 (3)
C3—C4—H4	114.0	C15—C14—C16	119.0 (3)
C5—C4—H4	114.0	C10—C15—C14	119.4 (3)
C6—C4—H4	114.0	C10—C15—H15	120.3
C7—C5—C4	119.9 (3)	C14—C15—H15	120.3
C7—C5—C6	119.1 (3)	F5—C16—F6	113.5 (12)
C4—C5—C6	60.6 (2)	F5—C16—F4'	51.8 (11)
C7—C5—C8	115.5 (4)	F6—C16—F4'	128.8 (9)
C4—C5—C8	115.9 (3)	F5—C16—F6'	132.7 (8)
C6—C5—C8	114.7 (3)	F6—C16—F6'	37.9 (10)
C5—C8—H8B	109.5	F4'—C16—F6'	110.0 (9)
C5—C8—H8A	109.5	F5—C16—F4	104.5 (11)
H8B—C8—H8A	109.5	F6—C16—F4	101.7 (9)
C5—C8—H8C	109.5	F4'—C16—F4	53.9 (6)
H8B—C8—H8C	109.5	F6'—C16—F4	65.7 (9)
H8A—C8—H8C	109.5	F5—C16—F5'	54.8 (12)
C5—C7—H7A	109.5	F6—C16—F5'	65.5 (8)
C5—C7—H7B	109.5	F4'—C16—F5'	102.4 (8)
H7A—C7—H7B	109.5	F6'—C16—F5'	101.0 (10)
C5—C7—H7C	109.5	F4—C16—F5'	139.2 (7)
H7A—C7—H7C	109.5	F5—C16—C14	112.4 (6)
H7B—C7—H7C	109.5	F6—C16—C14	114.7 (6)
C9—C6—C5	121.5 (3)	F4'—C16—C14	115.8 (7)
C9—C6—C4	123.0 (2)	F6'—C16—C14	114.4 (7)
C5—C6—C4	59.6 (2)	F4—C16—C14	108.9 (6)
C9—C6—H6	114.1	F5'—C16—C14	111.6 (5)
C5—C6—H6	114.1	C9—N1—C10	129.3 (2)
C4—C6—H6	114.1	C9—N1—H1	115.4
O1—C9—N1	123.5 (2)	C10—N1—H1	115.4
O1—C9—C6	124.4 (2)		
F2—C1—C2—C3	5.3 (6)	C15—C10—C11—C12	0.4 (5)
F3—C1—C2—C3	-116.4 (4)	N1—C10—C11—C12	179.9 (3)
F1—C1—C2—C3	125.5 (4)	C10—C11—C12—C13	-0.2 (5)
F2—C1—C2—C11	-176.8 (3)	C11—C12—C13—C14	0.3 (6)
F3—C1—C2—C11	61.6 (4)	C12—C13—C14—C15	-0.7 (5)
F1—C1—C2—C11	-56.6 (4)	C12—C13—C14—C16	179.4 (4)
C1—C2—C3—C4	175.5 (3)	C11—C10—C15—C14	-0.7 (4)
C11—C2—C3—C4	-2.2 (5)	N1—C10—C15—C14	179.8 (3)
C2—C3—C4—C5	-153.3 (3)	C13—C14—C15—C10	0.9 (5)
C2—C3—C4—C6	134.5 (4)	C16—C14—C15—C10	-179.2 (3)
C3—C4—C5—C7	-3.5 (5)	C13—C14—C16—F5	83 (2)

C6—C4—C5—C7	108.7 (3)	C15—C14—C16—F5	-97 (2)
C3—C4—C5—C6	-112.3 (3)	C13—C14—C16—F6	-145.8 (15)
C3—C4—C5—C8	142.7 (3)	C15—C14—C16—F6	34.3 (16)
C6—C4—C5—C8	-105.0 (4)	C13—C14—C16—F4'	25.6 (15)
C7—C5—C6—C9	2.4 (4)	C15—C14—C16—F4'	-154.3 (14)
C4—C5—C6—C9	112.4 (3)	C13—C14—C16—F6'	-103.9 (18)
C8—C5—C6—C9	-140.6 (3)	C15—C14—C16—F6'	76.2 (18)
C7—C5—C6—C4	-110.0 (3)	C13—C14—C16—F4	-32.7 (14)
C8—C5—C6—C4	107.0 (3)	C15—C14—C16—F4	147.4 (14)
C3—C4—C6—C9	0.5 (5)	C13—C14—C16—F5'	142.2 (12)
C5—C4—C6—C9	-109.9 (3)	C15—C14—C16—F5'	-37.7 (13)
C3—C4—C6—C5	110.4 (4)	O1—C9—N1—C10	0.8 (5)
C5—C6—C9—O1	-65.5 (4)	C6—C9—N1—C10	-178.4 (3)
C4—C6—C9—O1	6.5 (5)	C15—C10—N1—C9	-5.4 (5)
C5—C6—C9—N1	113.7 (3)	C11—C10—N1—C9	175.1 (3)
C4—C6—C9—N1	-174.3 (3)		

Hydrogen-bond geometry (\AA , $^\circ$)

$D-H\cdots A$	$D-H$	$H\cdots A$	$D\cdots A$	$D-H\cdots A$
N1—H1 \cdots O1 ⁱ	0.86	2.15	2.974 (3)	161

Symmetry code: (i) $x, -y+1/2, z+1/2$.