

Acta Crystallographica Section E

Structure Reports

Online

ISSN 1600-5368

2-[(2-Acetoxybenzoyl)oxy]benzoic acid

Katarzyna A. Solanko and Andrew D. Bond*

University of Southern Denmark, Department of Physics, Chemistry and Pharmacy,
Campusvej 55, 5230 Odense, Denmark
Correspondence e-mail: adb@chem.sdu.dk

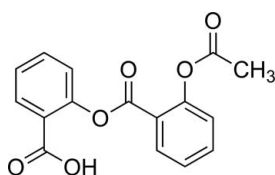
Received 7 May 2012; accepted 11 June 2012

Key indicators: single-crystal X-ray study; $T = 150$ K; mean $\sigma(\text{C}-\text{C}) = 0.002$ Å;
R factor = 0.033; wR factor = 0.080; data-to-parameter ratio = 11.6.

The title compound, $\text{C}_{16}\text{H}_{12}\text{O}_6$, is a common impurity of *ortho*-acetylsalicylic acid (aspirin). The benzene rings form a dihedral angle of $81.9(1)^\circ$ while the acetyl and carboxyl groups form dihedral angles of $74.0(1)$ and $26.4(2)^\circ$, respectively, with the benzene rings to which they are bound. In the crystal, molecules are linked by pairs of $\text{O}-\text{H}\cdots\text{O}$ hydrogen bonds between the carboxyl groups, forming inversion dimers.

Related literature

For background literature concerning the crystallization and crystal structure of aspirin, see: Bond *et al.* (2007, 2011). For a discussion of the pharmacological effects of acetylsalicylic acid, see: Bundgaard (1974). For related structures, see: Greener *et al.* (2000); Cox *et al.* (2000); Iqbal *et al.* (2007).



Experimental

Crystal data

$\text{C}_{16}\text{H}_{12}\text{O}_6$
 $M_r = 300.26$
Monoclinic, $P2_1/c$
 $a = 9.6314(5)$ Å
 $b = 7.7548(3)$ Å
 $c = 18.0763(8)$ Å
 $\beta = 95.572(2)^\circ$

$V = 1343.73(11)$ Å³
 $Z = 4$
Mo $K\alpha$ radiation
 $\mu = 0.12$ mm⁻¹
 $T = 150$ K
 $0.40 \times 0.20 \times 0.02$ mm

Data collection

Bruker Nonius X8 APEXII CCD
diffractometer
Absorption correction: multi-scan
(SADABS; Sheldrick, 2003)
 $T_{\min} = 0.887$, $T_{\max} = 0.998$

15807 measured reflections
2367 independent reflections
1868 reflections with $I > 2\sigma(I)$
 $R_{\text{int}} = 0.031$

Refinement

$R[F^2 > 2\sigma(F^2)] = 0.033$
 $wR(F^2) = 0.080$
 $S = 1.04$
2367 reflections
204 parameters
1 restraint

H atoms treated by a mixture of
independent and constrained
refinement
 $\Delta\rho_{\max} = 0.18$ e Å⁻³
 $\Delta\rho_{\min} = -0.17$ e Å⁻³

Table 1

Hydrogen-bond geometry (Å, °).

$D-H\cdots A$	$D-H$	$H\cdots A$	$D\cdots A$	$D-H\cdots A$
$\text{O5}-\text{H5}\cdots\text{O6}^i$	0.86 (1)	1.81 (1)	2.6660 (16)	176 (2)

Symmetry code: (i) $-x, -y + 1, -z + 1$.

Data collection: APEX2 (Bruker, 2010); cell refinement: SAINT (Bruker, 2010); data reduction: SAINT; program(s) used to solve structure: SHELXTL (Sheldrick, 2008); program(s) used to refine structure: SHELXTL; molecular graphics: SHELXTL; software used to prepare material for publication: SHELXTL.

We are grateful to the Danish Natural Science Research Council and the Carlsberg Foundation for provision of the X-ray equipment.

Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: FY2058).

References

- Bond, A. D., Boese, R. & Desiraju, G. R. (2007). *Angew. Chem. Int. Ed.* **46**, 618–622.
Bond, A. D., Solanko, K. A., Parsons, S., Redder, S. & Boese, R. (2011). *CrystEngComm*, **13**, 399–401.
Bruker (2010). APEX2 and SAINT. Bruker AXS Inc., Madison, Wisconsin, USA.
Bundgaard, H. (1974). *J. Pharm. Pharmacol.* **26**, 18–22.
Cox, P. J., Gilmour, G. I. & MacManus, S. M. (2000). *Int. J. Pharm.* **204**, 133–136.
Greener, B., Archibald, S. J. & Hodgkinson, M. (2000). *Angew. Chem. Int. Ed.* **39**, 3601–3604.
Iqbal, R., Zareef, M., Aziz, S., Qadeer, G. & Arfan, M. (2007). *Acta Cryst.* **E63**, o744–o745.
Sheldrick, G. M. (2003). SADABS. Bruker AXS Inc., Madison, Wisconsin, USA.
Sheldrick, G. M. (2008). *Acta Cryst.* **A64**, 112–122.

supporting information

Acta Cryst. (2012). E68, o2108 [https://doi.org/10.1107/S1600536812026475]

2-[(2-Acetoxybenzoyl)oxy]benzoic acid

Katarzyna A. Solanko and Andrew D. Bond

S1. Comment

Acetylsalicylsalicylic acid is a condensation (dehydration) product of acetylsalicylic acid (aspirin) and salicylic acid, and is a common impurity in commercial aspirin samples. Its pharmacological effects have been examined by Bundgaard (1974), and it has been suggested that the compound is a potentially immunogenic substance involved in the development of allergic reactions to aspirin.

S2. Experimental

The compound was prepared by acetylation of salicylsalicylic acid (purchased from Alfa Aesar) using acetic anhydride. 0.02 mol of salicylsalicylic acid was mixed with 0.01 mol of acetic anhydride with addition of 10% NaOH (5 ml) and *ca* 50 ml ice. The reactants were stirred for *ca* 2 h and the reaction was monitored by thin-layer chromatography. When the reaction was complete, the white solid was filtered and recrystallized from ethanol (yield 90%).

S3. Refinement

H atoms bound to C atoms were placed geometrically and allowed to ride during refinement with C—H = 0.95 (aromatic) or 0.98 Å (methyl) and with $U_{\text{iso}}(\text{H}) = 1.2$ (aromatic) or $1.5U_{\text{eq}}(\text{C})$ (methyl). The H atom bound to O5 was located in a difference Fourier map and refined with an isotropic displacement parameter, with the O—H distance restrained to 0.85 (1) Å.

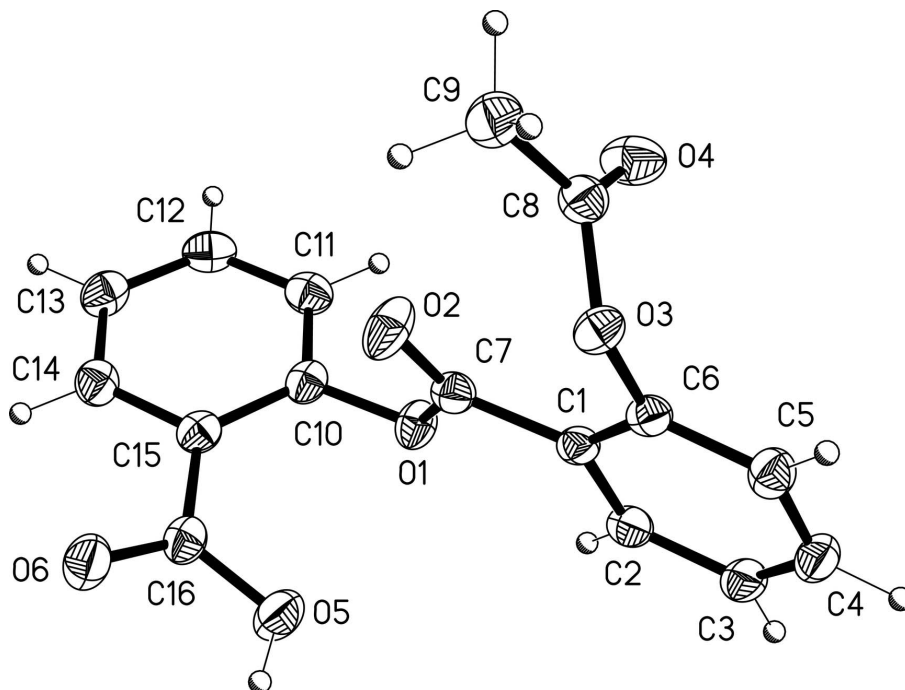


Figure 1

Molecular structure showing displacement ellipsoids at the 50% probability level for non-H atoms.

2-[(2-Acetoxybenzoyl)oxy]benzoic acid

Crystal data

$C_{16}H_{12}O_6$
 $M_r = 300.26$
 Monoclinic, $P2_1/c$
 Hall symbol: $-P\ 2_1/c$
 $a = 9.6314\ (5)\ \text{\AA}$
 $b = 7.7548\ (3)\ \text{\AA}$
 $c = 18.0763\ (8)\ \text{\AA}$
 $\beta = 95.572\ (2)^\circ$
 $V = 1343.73\ (11)\ \text{\AA}^3$
 $Z = 4$

$F(000) = 624$
 $D_x = 1.484\ \text{Mg m}^{-3}$
 Mo $K\alpha$ radiation, $\lambda = 0.71073\ \text{\AA}$
 Cell parameters from 5582 reflections
 $\theta = 2.9\text{--}25.0^\circ$
 $\mu = 0.12\ \text{mm}^{-1}$
 $T = 150\ \text{K}$
 Lath, colourless
 $0.40 \times 0.20 \times 0.02\ \text{mm}$

Data collection

Bruker Nonius X8 APEXII CCD
 diffractometer
 Radiation source: fine-focus sealed tube
 Graphite monochromator
 ω and φ scans
 Absorption correction: multi-scan
 (SADABS; Sheldrick, 2003)
 $T_{\min} = 0.887$, $T_{\max} = 0.998$

15807 measured reflections
 2367 independent reflections
 1868 reflections with $I > 2\sigma(I)$
 $R_{\text{int}} = 0.031$
 $\theta_{\max} = 25.1^\circ$, $\theta_{\min} = 3.6^\circ$
 $h = -11 \rightarrow 11$
 $k = -9 \rightarrow 9$
 $l = -21 \rightarrow 20$

Refinement

Refinement on F^2
 Least-squares matrix: full
 $R[F^2 > 2\sigma(F^2)] = 0.033$
 $wR(F^2) = 0.080$
 $S = 1.04$
 2367 reflections
 204 parameters
 1 restraint
 Primary atom site location: structure-invariant
 direct methods

Secondary atom site location: difference Fourier
 map
 Hydrogen site location: inferred from
 neighbouring sites
 H atoms treated by a mixture of independent
 and constrained refinement
 $w = 1/[\sigma^2(F_o^2) + (0.0341P)^2 + 0.473P]$
 where $P = (F_o^2 + 2F_c^2)/3$
 $(\Delta/\sigma)_{\max} < 0.001$
 $\Delta\rho_{\max} = 0.18 \text{ e } \text{\AA}^{-3}$
 $\Delta\rho_{\min} = -0.17 \text{ e } \text{\AA}^{-3}$

Special details

Geometry. All e.s.d.'s (except the e.s.d. in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell e.s.d.'s are taken into account individually in the estimation of e.s.d.'s in distances, angles and torsion angles; correlations between e.s.d.'s in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell e.s.d.'s is used for estimating e.s.d.'s involving l.s. planes.

Refinement. Refinement of F^2 against ALL reflections. The weighted R -factor wR and goodness of fit S are based on F^2 , conventional R -factors R are based on F , with F set to zero for negative F^2 . The threshold expression of $F^2 > \sigma(F^2)$ is used only for calculating R -factors(gt) *etc.* and is not relevant to the choice of reflections for refinement. R -factors based on F^2 are statistically about twice as large as those based on F , and R -factors based on ALL data will be even larger.

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (\AA^2)

	<i>x</i>	<i>y</i>	<i>z</i>	$U_{\text{iso}}^*/U_{\text{eq}}$
O1	0.25601 (10)	0.66416 (13)	0.32136 (5)	0.0262 (3)
O2	0.38752 (12)	0.75630 (15)	0.42277 (7)	0.0369 (3)
O3	0.62427 (10)	0.57401 (13)	0.46447 (5)	0.0245 (3)
O4	0.69302 (12)	0.76366 (15)	0.38150 (6)	0.0367 (3)
O5	0.07707 (13)	0.52229 (15)	0.41146 (6)	0.0373 (3)
H5	0.052 (3)	0.451 (3)	0.4443 (11)	0.082 (8)*
O6	0.00116 (13)	0.70903 (14)	0.49167 (6)	0.0351 (3)
C1	0.45005 (15)	0.49405 (19)	0.36384 (8)	0.0212 (3)
C2	0.40413 (16)	0.3714 (2)	0.31037 (8)	0.0249 (4)
H2A	0.3206	0.3914	0.2791	0.030*
C3	0.47842 (16)	0.2217 (2)	0.30243 (9)	0.0281 (4)
H3A	0.4461	0.1398	0.2656	0.034*
C4	0.59942 (17)	0.1906 (2)	0.34767 (9)	0.0296 (4)
H4A	0.6494	0.0863	0.3426	0.036*
C5	0.64823 (16)	0.3111 (2)	0.40048 (8)	0.0263 (4)
H5A	0.7320	0.2903	0.4314	0.032*
C6	0.57436 (15)	0.46175 (19)	0.40792 (8)	0.0213 (3)
C7	0.36674 (15)	0.6510 (2)	0.37434 (8)	0.0232 (4)
C8	0.67024 (15)	0.7322 (2)	0.44426 (9)	0.0266 (4)
C9	0.68674 (17)	0.8504 (2)	0.50933 (9)	0.0331 (4)
H9A	0.7316	0.9574	0.4953	0.050*
H9B	0.7447	0.7949	0.5501	0.050*
H9C	0.5948	0.8768	0.5254	0.050*
C10	0.17065 (15)	0.8088 (2)	0.32687 (8)	0.0236 (4)

C11	0.18395 (16)	0.9407 (2)	0.27723 (9)	0.0283 (4)
H11A	0.2497	0.9319	0.2415	0.034*
C12	0.10136 (16)	1.0862 (2)	0.27940 (9)	0.0305 (4)
H12A	0.1091	1.1765	0.2445	0.037*
C13	0.00762 (16)	1.1002 (2)	0.33224 (9)	0.0303 (4)
H13A	-0.0480	1.2009	0.3343	0.036*
C14	-0.00502 (16)	0.9675 (2)	0.38195 (9)	0.0273 (4)
H14A	-0.0691	0.9784	0.4185	0.033*
C15	0.07431 (15)	0.81807 (19)	0.37966 (8)	0.0234 (3)
C16	0.04899 (16)	0.6788 (2)	0.43270 (9)	0.0258 (4)

Atomic displacement parameters (Å²)

	U^{11}	U^{22}	U^{33}	U^{12}	U^{13}	U^{23}
O1	0.0273 (6)	0.0265 (6)	0.0240 (6)	0.0042 (5)	-0.0015 (5)	-0.0055 (5)
O2	0.0365 (7)	0.0319 (7)	0.0394 (7)	0.0092 (5)	-0.0106 (5)	-0.0158 (6)
O3	0.0298 (6)	0.0216 (6)	0.0218 (5)	-0.0006 (5)	0.0007 (4)	-0.0024 (5)
O4	0.0456 (7)	0.0353 (7)	0.0296 (7)	-0.0121 (6)	0.0064 (5)	0.0006 (5)
O5	0.0572 (8)	0.0227 (7)	0.0343 (7)	0.0042 (6)	0.0154 (6)	0.0015 (5)
O6	0.0490 (7)	0.0281 (7)	0.0299 (6)	0.0074 (5)	0.0123 (5)	0.0022 (5)
C1	0.0256 (8)	0.0192 (8)	0.0196 (8)	-0.0030 (6)	0.0066 (6)	0.0000 (6)
C2	0.0259 (8)	0.0258 (9)	0.0235 (8)	-0.0048 (7)	0.0047 (6)	-0.0016 (7)
C3	0.0339 (9)	0.0233 (9)	0.0284 (9)	-0.0048 (7)	0.0097 (7)	-0.0078 (7)
C4	0.0372 (9)	0.0212 (9)	0.0322 (9)	0.0034 (7)	0.0125 (8)	-0.0008 (7)
C5	0.0297 (8)	0.0254 (9)	0.0246 (8)	0.0025 (7)	0.0056 (7)	0.0011 (7)
C6	0.0269 (8)	0.0203 (8)	0.0175 (8)	-0.0032 (6)	0.0062 (6)	-0.0006 (6)
C7	0.0230 (8)	0.0235 (8)	0.0233 (8)	-0.0025 (6)	0.0025 (7)	-0.0007 (7)
C8	0.0227 (8)	0.0262 (9)	0.0303 (9)	-0.0005 (7)	0.0004 (7)	-0.0002 (7)
C9	0.0333 (9)	0.0316 (10)	0.0341 (9)	-0.0021 (7)	0.0019 (7)	-0.0074 (8)
C10	0.0221 (7)	0.0239 (9)	0.0235 (8)	0.0008 (6)	-0.0042 (6)	-0.0055 (7)
C11	0.0274 (8)	0.0316 (10)	0.0253 (8)	-0.0048 (7)	-0.0001 (7)	-0.0007 (7)
C12	0.0326 (9)	0.0275 (9)	0.0299 (9)	-0.0044 (7)	-0.0052 (7)	0.0059 (7)
C13	0.0288 (9)	0.0249 (9)	0.0356 (10)	0.0045 (7)	-0.0044 (7)	0.0019 (8)
C14	0.0232 (8)	0.0287 (9)	0.0293 (8)	0.0022 (7)	-0.0003 (7)	-0.0015 (7)
C15	0.0231 (8)	0.0234 (8)	0.0228 (8)	-0.0009 (6)	-0.0032 (6)	-0.0024 (7)
C16	0.0250 (8)	0.0248 (9)	0.0271 (9)	0.0035 (7)	-0.0002 (7)	-0.0026 (7)

Geometric parameters (Å, °)

O1—C7	1.3662 (18)	C5—C6	1.381 (2)
O1—C10	1.3998 (18)	C5—H5A	0.950
O2—C7	1.1995 (18)	C8—C9	1.488 (2)
O3—C8	1.3659 (19)	C9—H9A	0.980
O3—C6	1.3923 (18)	C9—H9B	0.980
O4—C8	1.2013 (18)	C9—H9C	0.980
O5—C16	1.3090 (19)	C10—C11	1.375 (2)
O5—H5	0.86 (1)	C10—C15	1.396 (2)
O6—C16	1.2242 (18)	C11—C12	1.383 (2)

C1—C6	1.395 (2)	C11—H11A	0.950
C1—C2	1.397 (2)	C12—C13	1.381 (2)
C1—C7	1.480 (2)	C12—H12A	0.950
C2—C3	1.378 (2)	C13—C14	1.379 (2)
C2—H2A	0.950	C13—H13A	0.950
C3—C4	1.378 (2)	C14—C15	1.391 (2)
C3—H3A	0.950	C14—H14A	0.950
C4—C5	1.385 (2)	C15—C16	1.480 (2)
C4—H4A	0.950		
C7—O1—C10	115.71 (11)	C8—C9—H9A	109.5
C8—O3—C6	117.58 (11)	C8—C9—H9B	109.5
C16—O5—H5	108.2 (16)	H9A—C9—H9B	109.5
C6—C1—C2	117.99 (14)	C8—C9—H9C	109.5
C6—C1—C7	121.35 (13)	H9A—C9—H9C	109.5
C2—C1—C7	120.63 (14)	H9B—C9—H9C	109.5
C3—C2—C1	120.84 (15)	C11—C10—C15	121.20 (14)
C3—C2—H2A	119.6	C11—C10—O1	117.20 (13)
C1—C2—H2A	119.6	C15—C10—O1	121.59 (14)
C2—C3—C4	120.23 (15)	C10—C11—C12	119.87 (14)
C2—C3—H3A	119.9	C10—C11—H11A	120.1
C4—C3—H3A	119.9	C12—C11—H11A	120.1
C3—C4—C5	120.08 (14)	C13—C12—C11	120.00 (15)
C3—C4—H4A	120.0	C13—C12—H12A	120.0
C5—C4—H4A	120.0	C11—C12—H12A	120.0
C6—C5—C4	119.63 (15)	C14—C13—C12	119.82 (15)
C6—C5—H5A	120.2	C14—C13—H13A	120.1
C4—C5—H5A	120.2	C12—C13—H13A	120.1
C5—C6—O3	117.13 (13)	C13—C14—C15	121.24 (14)
C5—C6—C1	121.20 (14)	C13—C14—H14A	119.4
O3—C6—C1	121.54 (13)	C15—C14—H14A	119.4
O2—C7—O1	121.63 (14)	C14—C15—C10	117.83 (14)
O2—C7—C1	126.80 (14)	C14—C15—C16	117.59 (13)
O1—C7—C1	111.55 (12)	C10—C15—C16	124.57 (14)
O4—C8—O3	121.91 (14)	O6—C16—O5	122.61 (14)
O4—C8—C9	127.32 (15)	O6—C16—C15	121.59 (14)
O3—C8—C9	110.77 (13)	O5—C16—C15	115.77 (13)
C6—C1—C2—C3	1.1 (2)	C6—O3—C8—O4	-14.2 (2)
C7—C1—C2—C3	-177.36 (13)	C6—O3—C8—C9	166.09 (12)
C1—C2—C3—C4	0.3 (2)	C7—O1—C10—C11	104.36 (15)
C2—C3—C4—C5	-1.2 (2)	C7—O1—C10—C15	-76.66 (17)
C3—C4—C5—C6	0.6 (2)	C15—C10—C11—C12	0.3 (2)
C4—C5—C6—O3	176.69 (13)	O1—C10—C11—C12	179.32 (13)
C4—C5—C6—C1	0.9 (2)	C10—C11—C12—C13	1.2 (2)
C8—O3—C6—C5	115.45 (14)	C11—C12—C13—C14	-1.1 (2)
C8—O3—C6—C1	-68.76 (17)	C12—C13—C14—C15	-0.6 (2)
C2—C1—C6—C5	-1.7 (2)	C13—C14—C15—C10	2.0 (2)

C7—C1—C6—C5	176.74 (13)	C13—C14—C15—C16	-177.04 (14)
C2—C1—C6—O3	-177.31 (12)	C11—C10—C15—C14	-1.9 (2)
C7—C1—C6—O3	1.1 (2)	O1—C10—C15—C14	179.14 (13)
C10—O1—C7—O2	0.9 (2)	C11—C10—C15—C16	177.09 (14)
C10—O1—C7—C1	179.84 (11)	O1—C10—C15—C16	-1.8 (2)
C6—C1—C7—O2	-6.4 (2)	C14—C15—C16—O6	-25.3 (2)
C2—C1—C7—O2	172.01 (15)	C10—C15—C16—O6	155.67 (15)
C6—C1—C7—O1	174.74 (12)	C14—C15—C16—O5	152.75 (14)
C2—C1—C7—O1	-6.87 (18)	C10—C15—C16—O5	-26.3 (2)

Hydrogen-bond geometry (Å, °)

<i>D</i> —H \cdots <i>A</i>	<i>D</i> —H	H \cdots <i>A</i>	<i>D</i> \cdots <i>A</i>	<i>D</i> —H \cdots <i>A</i>
O5—H5 \cdots O6 ⁱ	0.86 (1)	1.81 (1)	2.6660 (16)	176 (2)

Symmetry code: (i) $-x, -y+1, -z+1$.